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Supporting Information

Atomically Precise Rhodium-Gold Carbonyl Nanoclusters: In-Depth Synthesis and Multivalence Investigation of $[\text{Rh}_{16}\text{Au}_6(\text{CO})_{36}]^{6-}$, and its Correlation with $[\text{Rh}_{10}\text{Au}(\text{CO})_{26}]^{3-}$

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SUPPORTING INFORMATION

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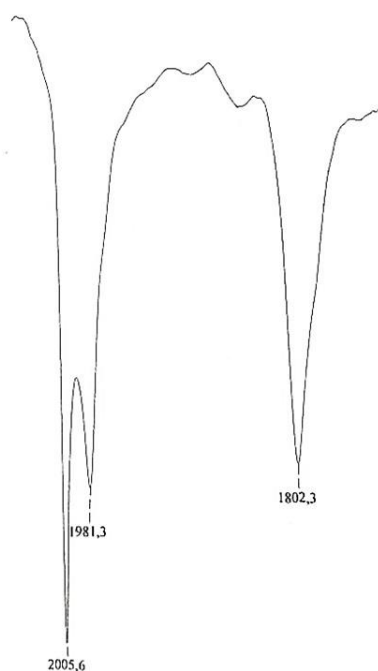


Figure S1a. IR spectrum of $[\text{Rh}_{16}\text{Au}_6(\text{CO})_{36}][\text{TEA}]_6$ registered in CH_3CN solution.

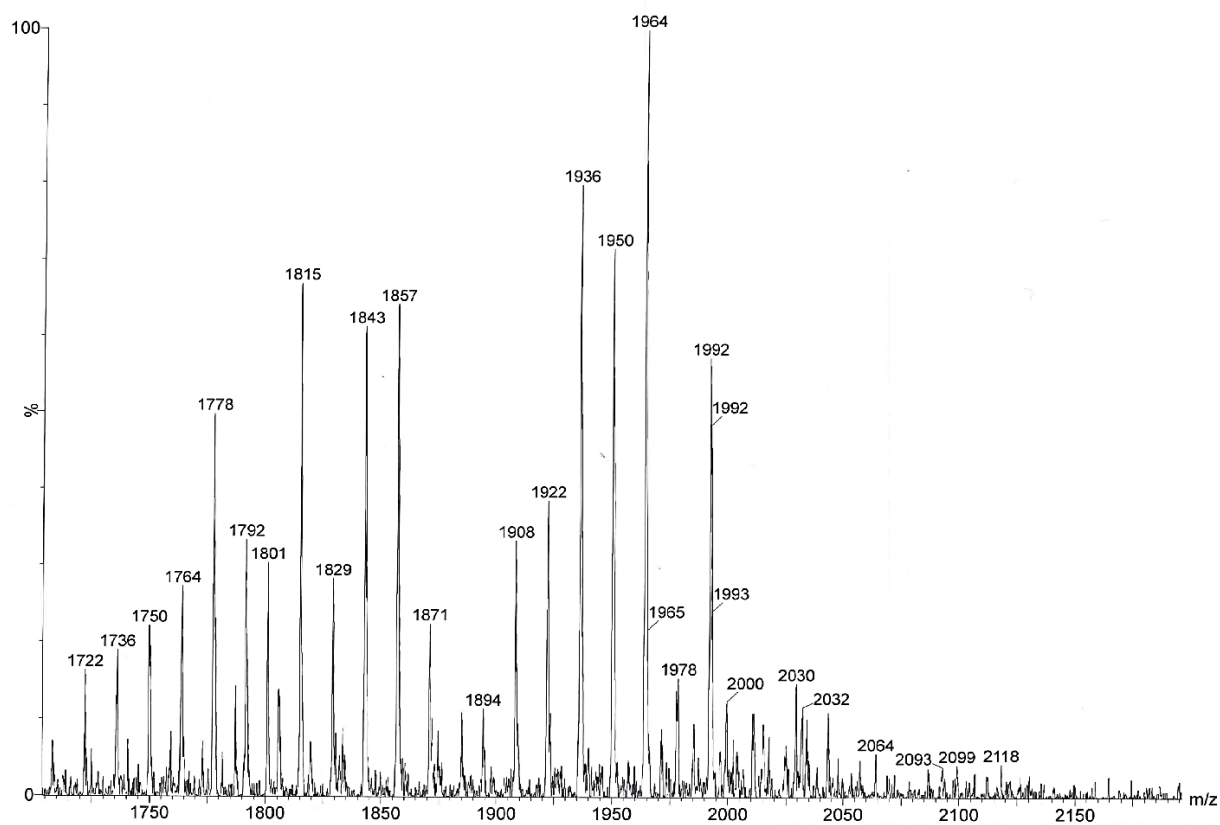


Figure S1b. ESI-MS spectrum of the 2- region of $[\text{Rh}_{16}\text{Au}_6(\text{CO})_{36}][\text{TEA}]_6$.

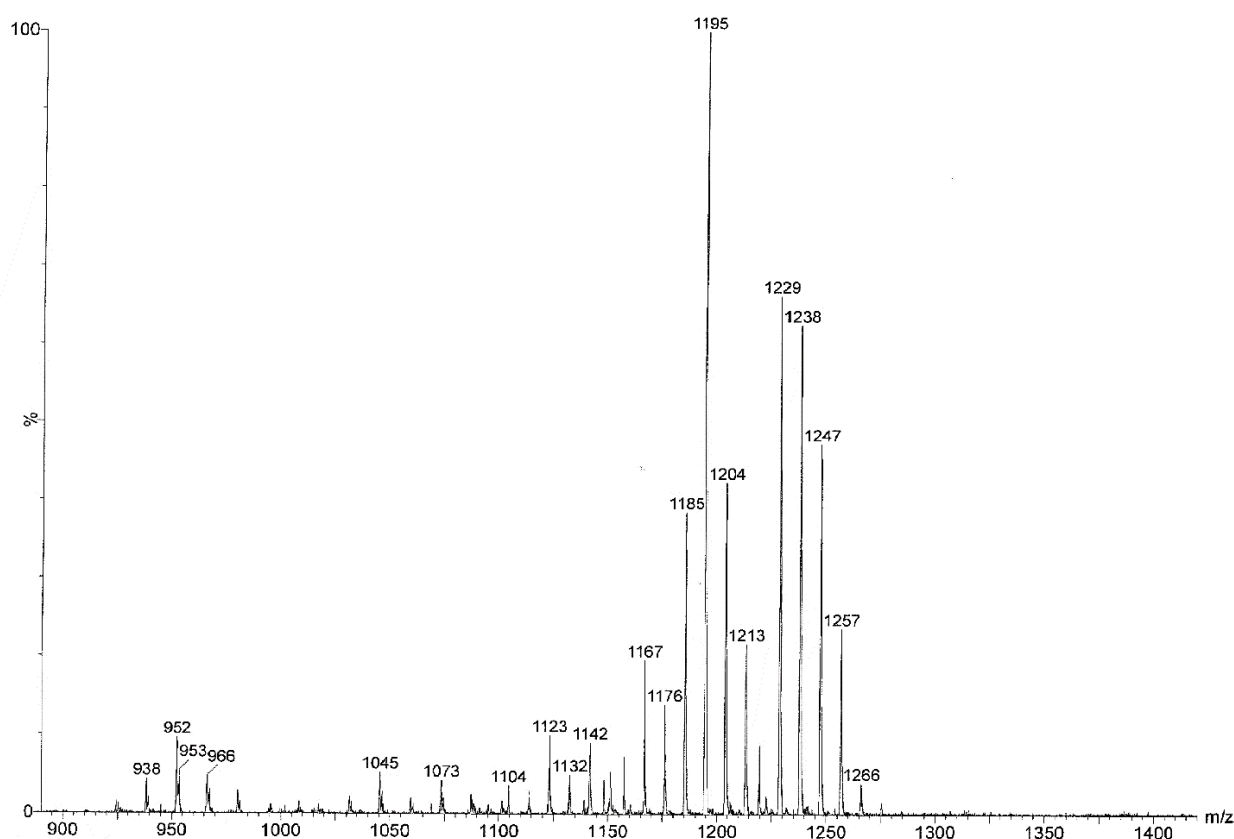


Figure S1c. ESI-MS spectrum of the 3- region of $[\text{Rh}_{16}\text{Au}_6(\text{CO})_{36}][\text{TEA}]_6$.

Signals / group of signals (m/z)	Corresponding ions
1992-1978-1964-1950-1936-1922-1908-1894	$\{[\text{Rh}_{16}\text{Au}_6(\text{CO})_{32-31-30-29-28-27-26-25}][\text{TEA}]_2\}^{2-}$
1871-1857-1843-1829-1815-1801	$\{[\text{Rh}_{16}\text{Au}_6(\text{CO})_{28-27-26-25-24-23}][\text{TEA}]\}^{2-}$
1792-1778-1764-1750-1736-1722	$[\text{Rh}_{16}\text{Au}_6(\text{CO})_{27-26-25-24-23-22}]^{2-}$
1266-1257-1247-1238-1229	$\{[\text{Rh}_{16}\text{Au}_6(\text{CO})_{30-29-28-27-26}][\text{TEA}]\}^{3-}$
1213-1204-1195-1185-1176-1167	$[\text{Rh}_{16}\text{Au}_6(\text{CO})_{29-28-27-26-25-24}]^{3-}$

Table S1a. ESI-MS peak assignments for $[\text{Rh}_{16}\text{Au}_6(\text{CO})_{36}][\text{TEA}]_6$.

Spectroelectrochemical Experiments

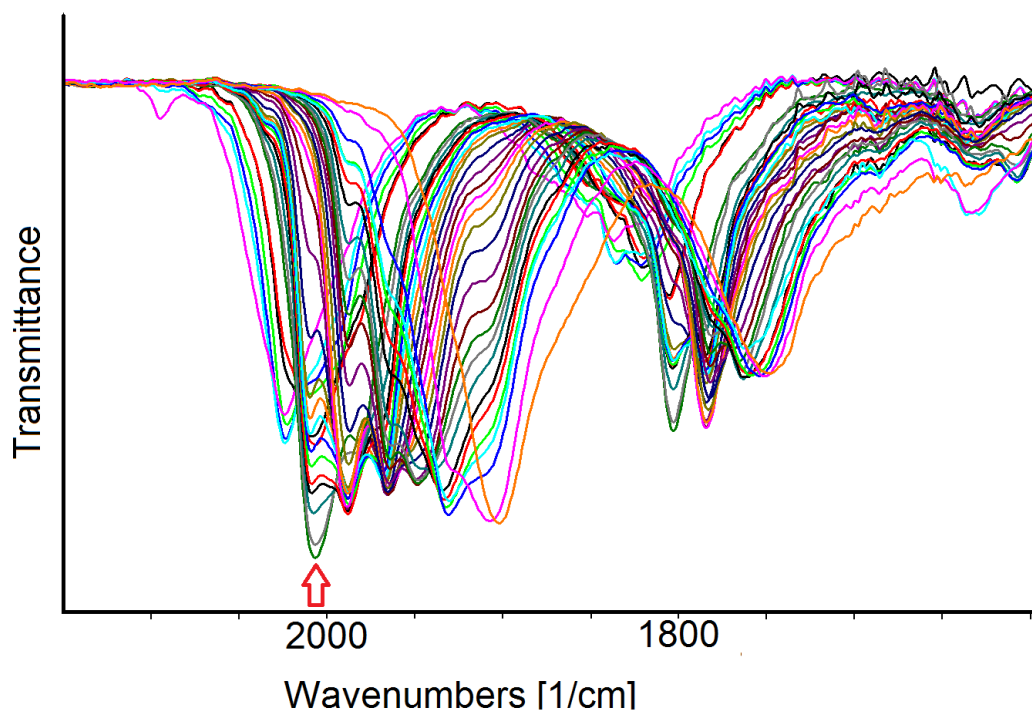


Figure S1d. Infrared spectra of $[\text{Rh}_{16}\text{Au}_6(\text{CO})_{36}]^{6-}$ recorded in an OTTLE cell during a slow WE potential scan between -0.5 to -2.0 V in CH_3CN containing 0.1 mol dm^{-3} $[\text{TBA}][\text{PF}_6]$, under CO atmosphere. The solvent and supporting electrolyte absorptions have been subtracted. The red arrow indicates the initial spectrum.

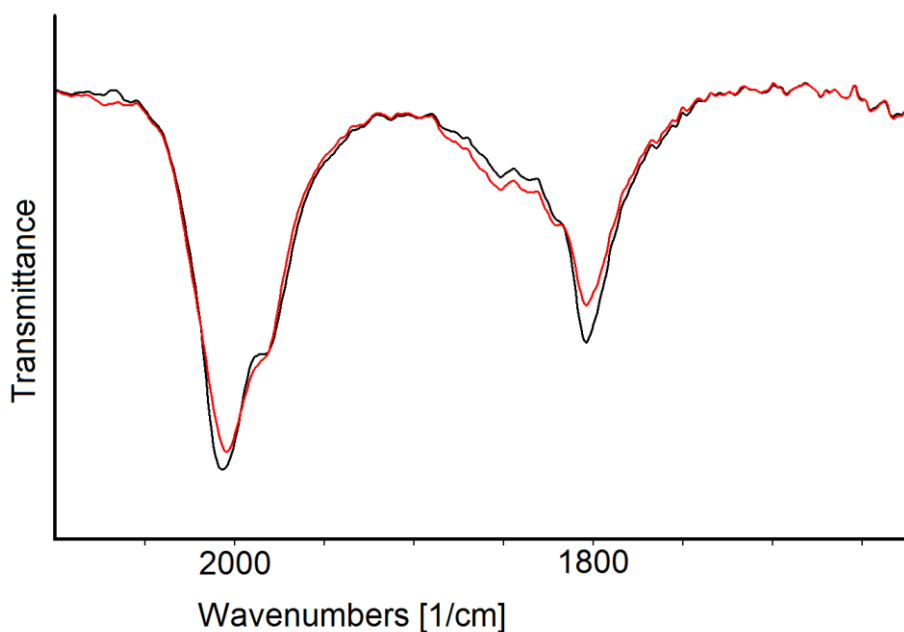


Figure S1e. IR spectra of a CH_3CN solution of $[\text{Rh}_{16}\text{Au}_6(\text{CO})_{36}]^{6-}$ recorded in an OTTLE cell before (black line) and after (red line) a cyclic voltammetry between -0.8 to -0.3 V (scan rate 1 mV s^{-1}), under CO atmosphere. $[\text{TBA}][\text{PF}_6]$ (0.1 mol dm^{-3}) as the supporting electrolyte. The solvent and supporting electrolyte absorptions have been subtracted.

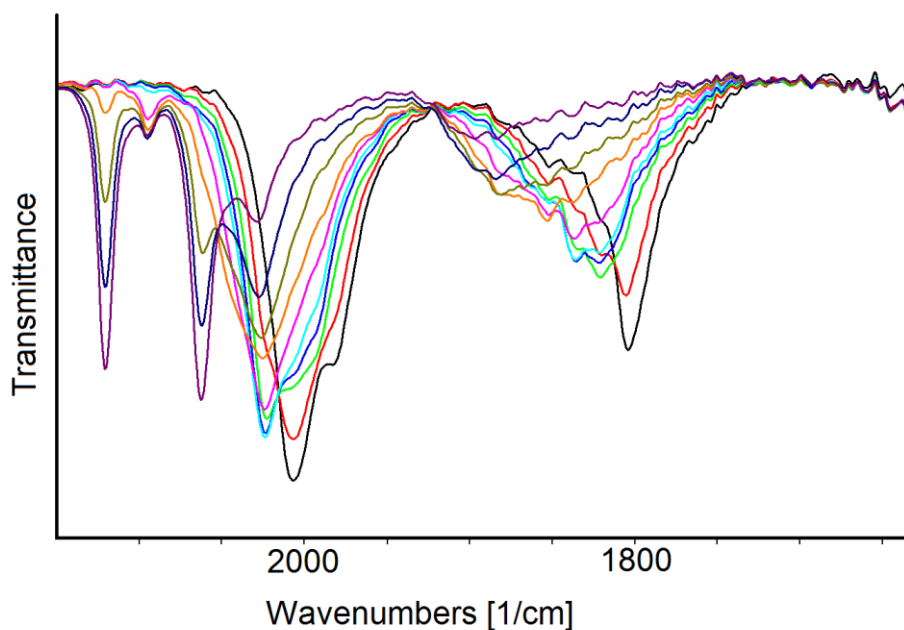


Figure S1f. Infrared spectra of $[\text{Rh}_{16}\text{Au}_6(\text{CO})_{36}]^{6-}$ recorded in an OTTLE cell during the progressive increase of the working electrode potential E from -0.8 to $+0.2$ V in CH_3CN containing 0.1 mol dm^{-3} $[\text{TBA}][\text{PF}_6]$, under CO atmosphere. The solvent and supporting electrolyte absorptions have been subtracted.

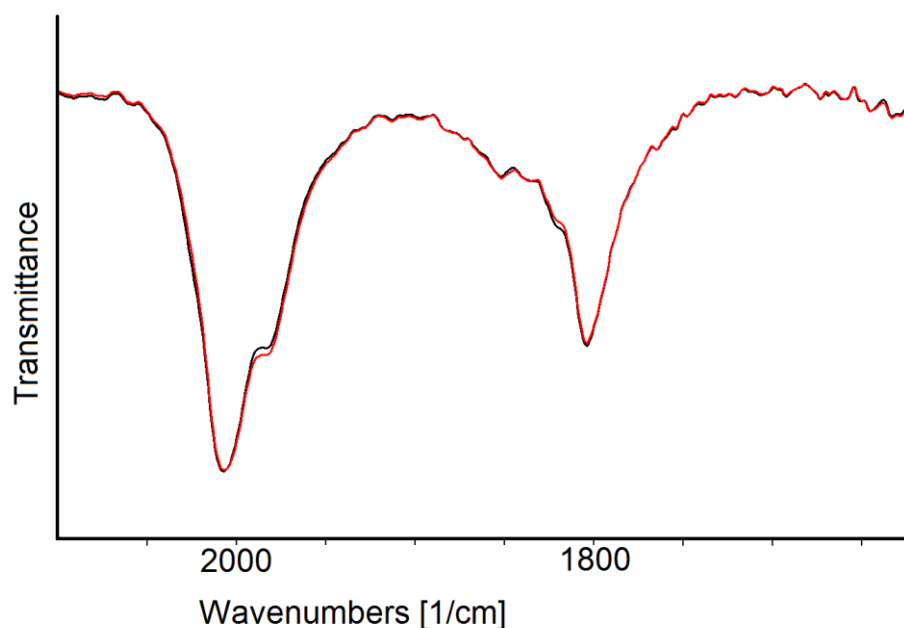
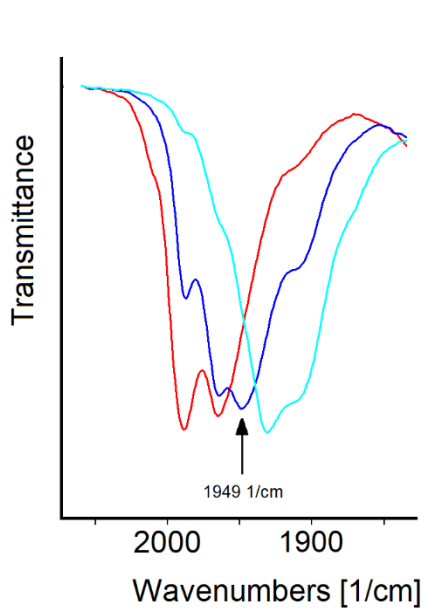
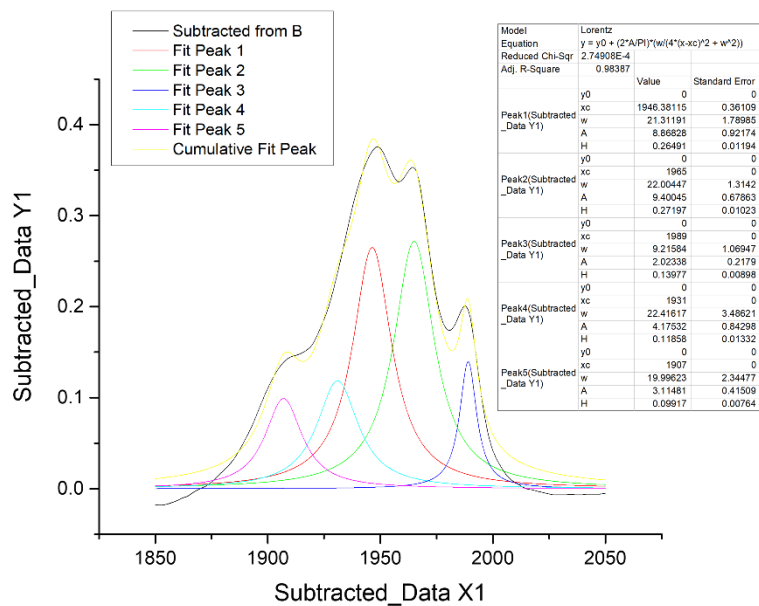


Figure S1g. IR spectra of a CH_3CN solution of $[\text{Rh}_{16}\text{Au}_6(\text{CO})_{36}]^{6-}$ recorded in an OTTLE cell before (black line) and after (red line) a cyclic voltammetry between -0.8 to -1.4 V (scan rate 1 mV s^{-1}), under CO atmosphere. $[\text{TBA}][\text{PF}_6]$ (0.1 mol dm^{-3}) as the supporting electrolyte. The solvent and supporting electrolyte absorptions have been subtracted.



(a)



(b)

Figure S1h. (a) First (red line), intermediate (blue line) and final (turquoise line) spectra in the second reduction sequence of $[\text{Rh}_{16}\text{Au}_6(\text{CO})_{36}]^{6-}$ registered during the progressive potential decrease from -1.4 to -1.7 V (Figure 8c). (b) Spectral deconvolution of the intermediate IR spectrum (blue line in (a)). Its contribution is matched by the combination of main separate bands at 1989, 1965, 1931 and 1907 cm^{-1} , corresponding to red and turquoise curves, respectively and of a new band at 1946 cm^{-1} band due to the now unravelled -8 negative charge.

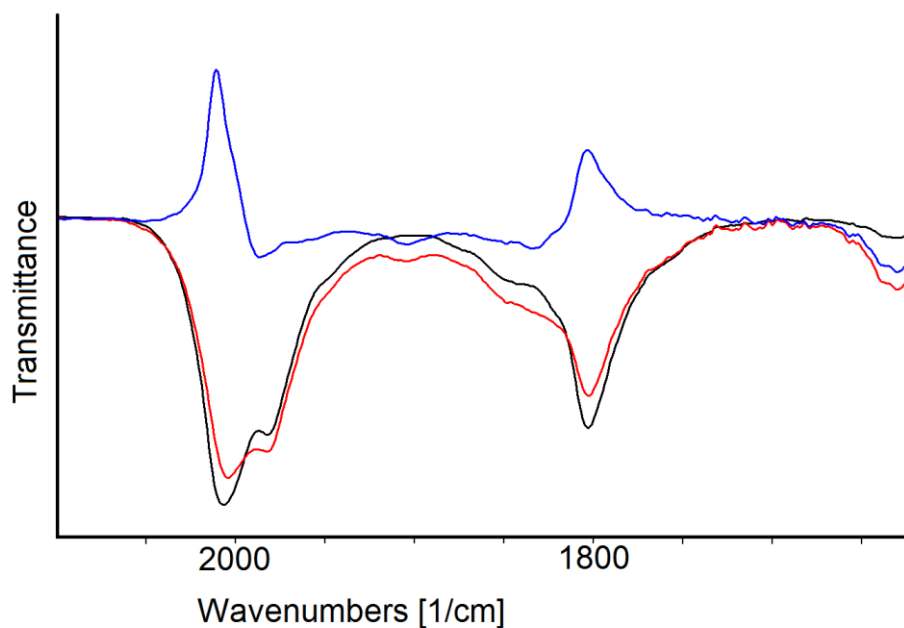


Figure S1i. IR spectra of a CH₃CN solution of [Rh₁₆Au₆(CO)₃₆]⁶⁻ recorded in an OTTLE cell before (black line) and after (red line) a cyclic voltammetry between -1.4 to -1.7 V (scan rate 1 mV s^{-1}). The blue line represents their difference spectrum. [TBA][PF₆] (0.1 mol dm^{-3}) as the supporting electrolyte. The solvent and supporting electrolyte absorptions have been subtracted.

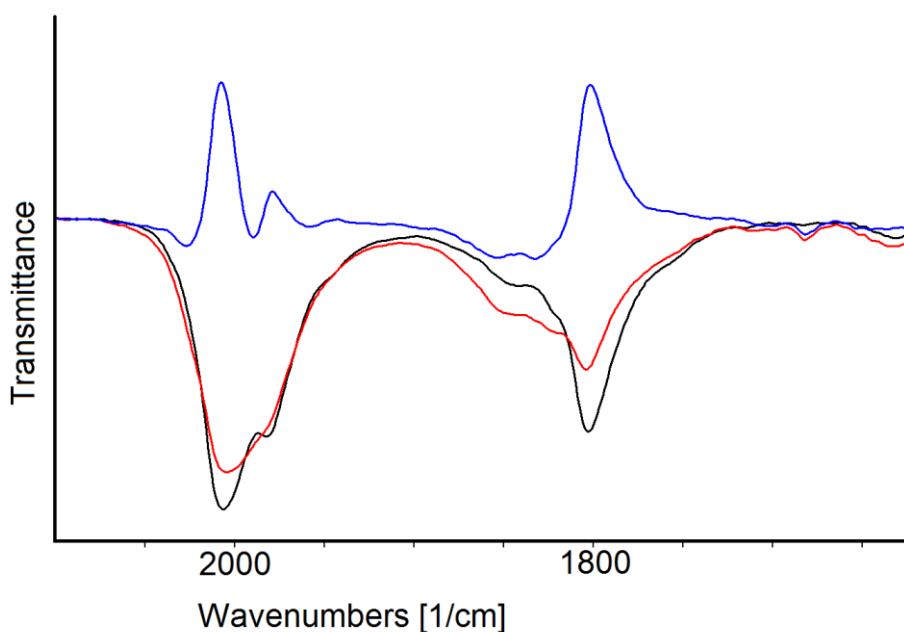


Figure S1i. IR spectra of a CH₃CN solution of [Rh₁₆Au₆(CO)₃₆]⁶⁻ recorded in an OTTLE cell before (black line) and after (red line) a cyclic voltammetry between -1.7 to -2.0 V (scan rate 1 mV s^{-1}). The blue line represents their difference spectrum. [TBA][PF₆] (0.1 mol dm^{-3}) as the supporting electrolyte. The solvent and supporting electrolyte absorptions have been subtracted.

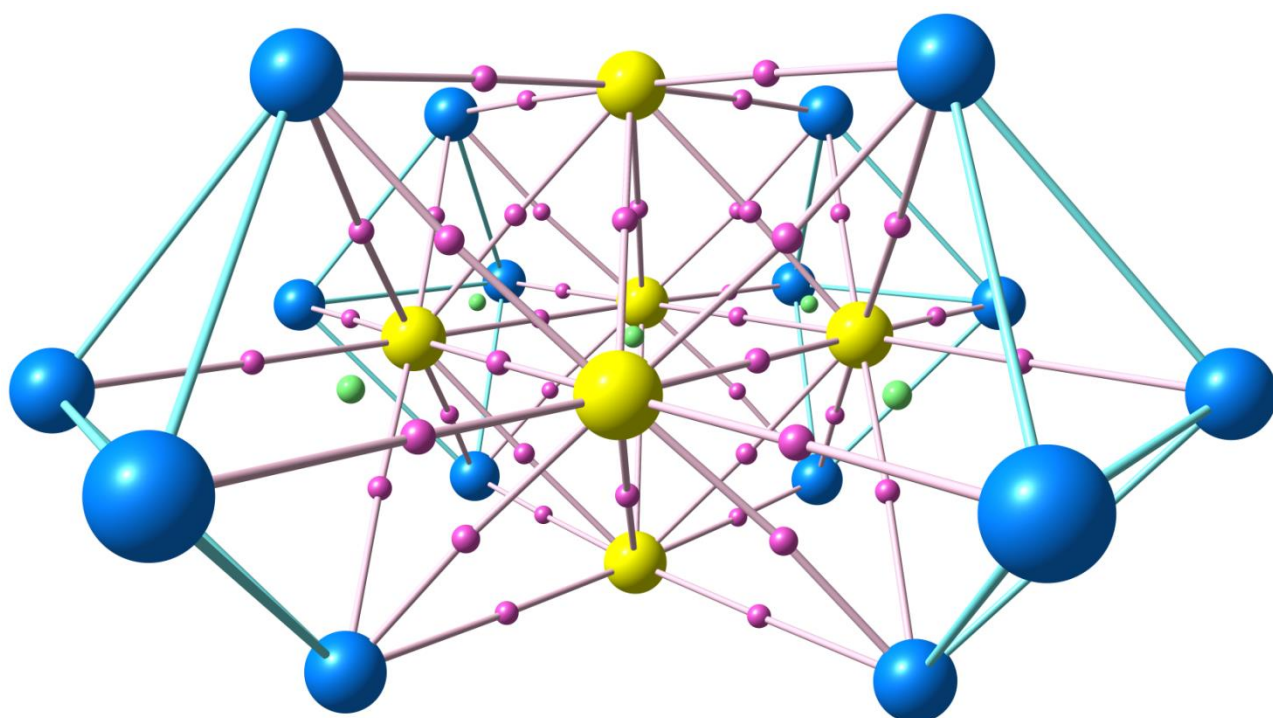


Figure S1m. Au_6Rh_{16} core of $[Rh_{16}Au_6(CO)_{36}]^{6-}$ with (3,-1) BCPs (pink small spheres) and selected (3,+3) CCPs (green small spheres). Rh atoms are in blue, Au atoms in yellow.

Table S1b. Average properties at selected (3,-1) BCPs and (3,+3) CCPs computed for $[Rh_{16}Au_6(CO)_{36}]^{n-}$ ($n = 5, 6, 7$) clusters.

n	ρ ($e \text{ \AA}^{-3}$)	V (hartree \AA^{-3})	E (hartree \AA^{-3})	$\nabla^2\rho$ ($e \text{ \AA}^{-5}$)
<i>(3,-1) Au---Au</i>				
7	0.232	-0.206	-0.029	2.129
6	0.236	-0.212	-0.029	2.197
5	0.227	-0.202	-0.026	2.143
<i>(3,-1) Au---Rh</i>				
7	0.257	-0.240	-0.047	2.090
6	0.262	-0.246	-0.048	2.130
5	0.263	-0.252	-0.049	2.204
<i>(3,+3) Au₆</i>				
7	0.085	-0.043	-0.001	0.583
6	0.092	-0.047	-0.002	0.623
5	0.089	-0.044	-0.002	0.577
<i>(3,+3) Rh₄Au₂</i>				
7	0.131	-0.060	-0.006	0.701
6	0.134	-0.064	-0.006	0.749
5	0.138	-0.065	-0.006	0.736

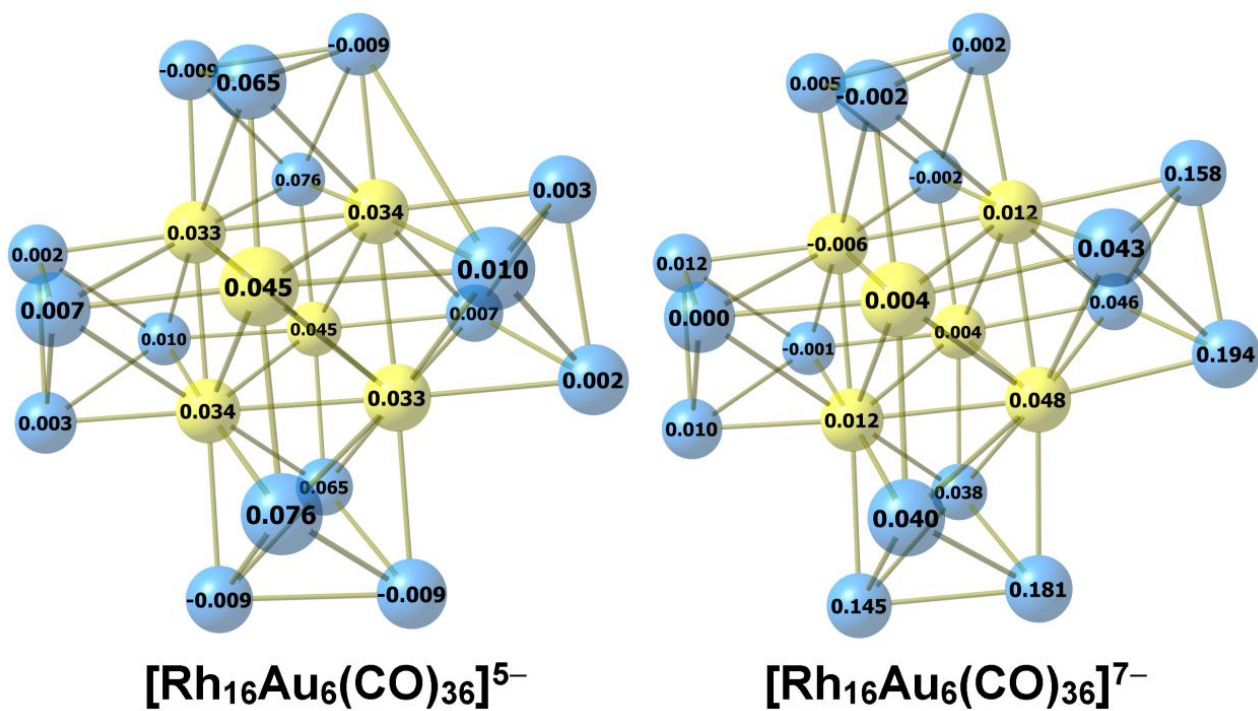


Figure S1n. $\text{Au}_6\text{Rh}_{16}$ cores of $[\text{Rh}_{16}\text{Au}_6(\text{CO})_{36}]^{5-}$ and $[\text{Rh}_{16}\text{Au}_6(\text{CO})_{36}]^{7-}$ with values of Löwdin spin population (a.u.). Rh atoms are in blue, Au atoms in yellow.

IR and ESI mass spectra for $[\text{Rh}_{10}\text{Au}(\text{CO})_{26}]^{3-}$ (2).

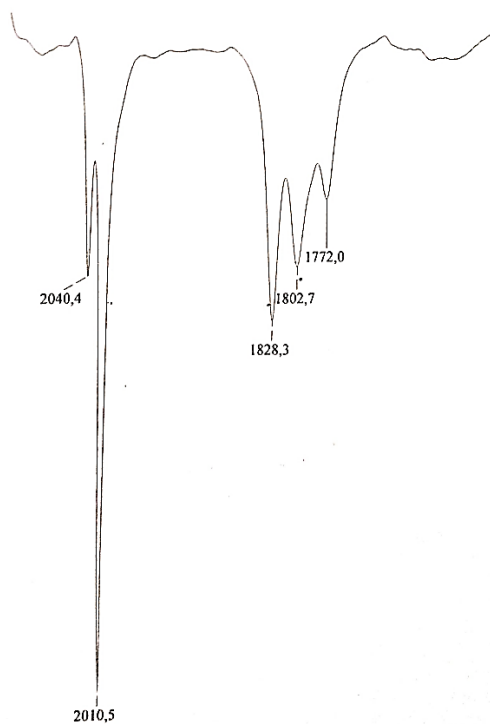


Figure S2a. IR spectrum of $[\text{Rh}_{10}\text{Au}(\text{CO})_{26}][\text{TBMA}]_3$ registered in CH_3CN solution.

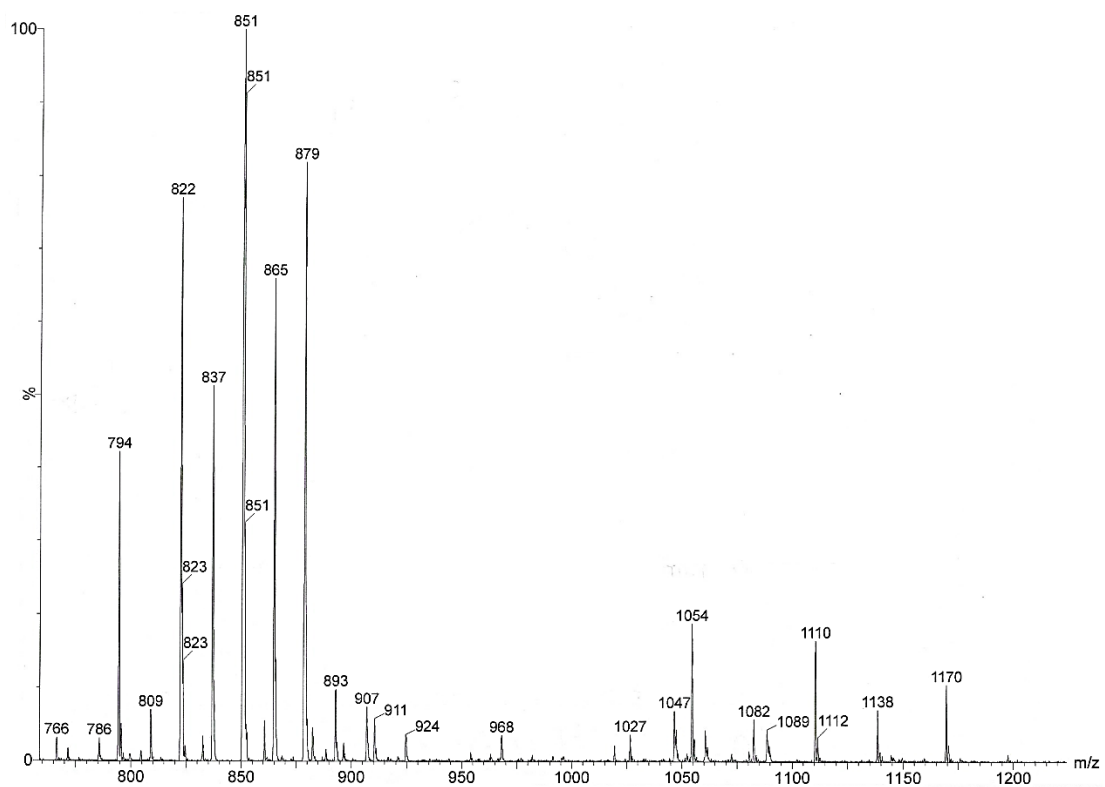


Figure S2b. ESI-MS spectrum of $[\text{Rh}_{10}\text{Au}(\text{CO})_{26}][\text{TBMA}]_3$ registered in CH_3CN solution (higher m/z region).

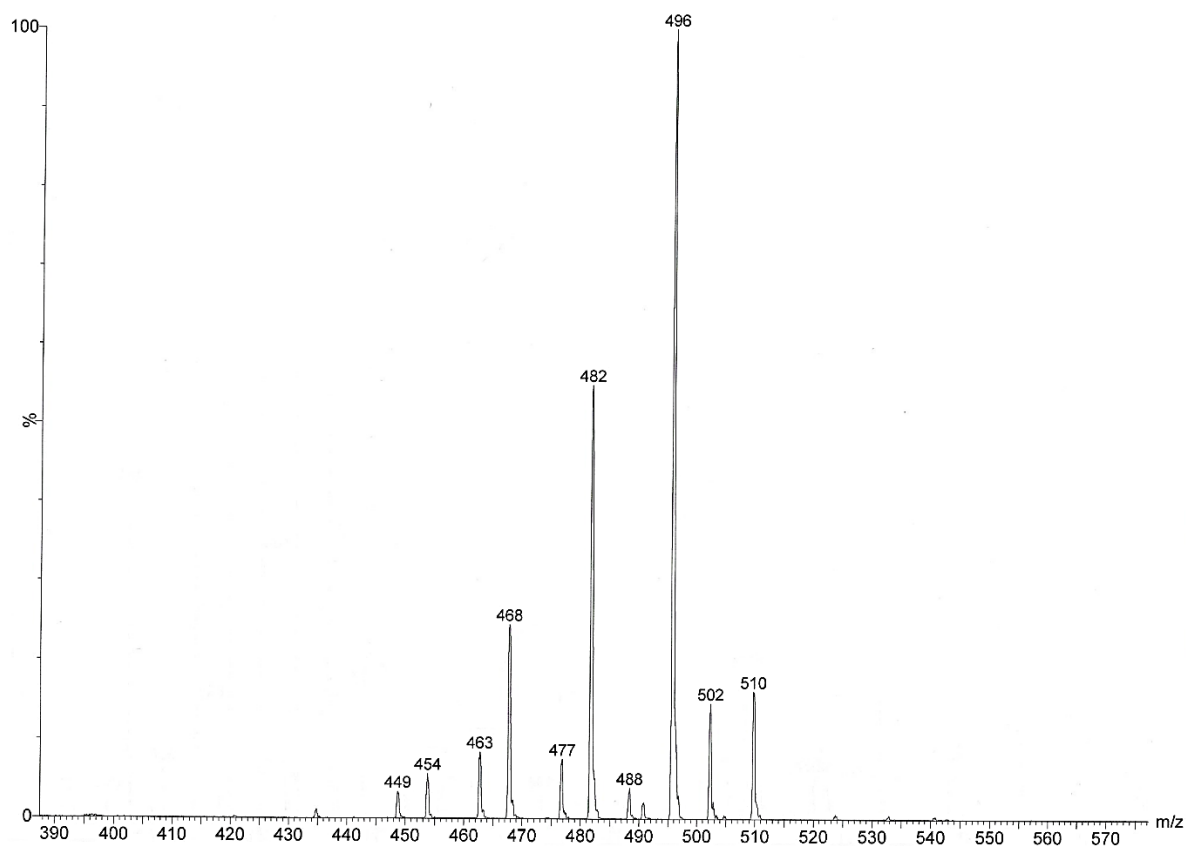


Figure S2c. ESI-MS spectra of $[\text{Rh}_{10}\text{Au}(\text{CO})_{26}][\text{TBMA}]_3$ registered in CH_3CN solution (lower m/z region).

Signals or group of signals (m/z)	Corresponding ions
1170	$\{[\text{Rh}_5\text{Au}(\text{CO})_{11}][\text{TBMA}]\}^-$
1138-1110-1082-1054	$[\text{Rh}_5\text{Au}(\text{CO})_{14-13-12-11}\text{Cl}]^-$
968	$\{[\text{Rh}_{10}\text{Au}(\text{CO})_{20}][\text{TBMA}]\}^{2-}$
907-893-879-865-851-837-822-808	$[\text{Rh}_{10}\text{Au}(\text{CO})_{21-20-19-18-17-16-15-14}]^{2-}$
510-496-482-468-454	$[\text{Rh}_5\text{Au}(\text{CO})_{11-10-9-8-7}]^{2-}$

Table S2a. ESI-MS peak assignments for $[\text{Rh}_{10}\text{Au}(\text{CO})_{26}][\text{TBMA}]_3$.

Table S2b. Crystallographic data for cluster **2**

Formula	C ₅₆ H ₄₈ AuN ₃ O ₂₈ Rh ₁₀
Fw	2405.04
Crystal system	Monoclinic
Space group	<i>P</i> 2 ₁ / <i>n</i>
a (Å)	18.6419(9)
b (Å)	14.5075(7)
c (Å)	25.1273(12)
α (°)	90
β (°)	105.386(2)
γ (°)	90
Cell volume (Å ³)	6552.1(5)
Z	4
D (g/cm ³)	2.438
μ (mm ⁻¹)	4.761
F(000)	4568
θ limits (deg)	1.580 to 25.000
Index ranges	-22<=h<=22, -17<=k<=17, -29<=l<=29
Reflections collected	84774
Independent reflections	11530 [R(int) = 0.1065]
Completeness to θ max	99.9%
Data/restraints/parameters	11530 / 576 / 874
Goodness of fit	1.231
R ₁ (I > 2σ(I))	0.0669
wR ₂ (all data)	0.1227
Largest diff. peak and hole, e Å ⁻³	2.122 and -2.182

Table S2c: Bond lengths (Å) from crystallographic analyses

Au(1)-Rh(5)	2.7135(10)
Au(1)-Rh(11)	2.7168(11)
Au(1)-Rh(4)	2.7244(10)
Au(1)-Rh(3)	2.7389(10)
Au(1)-Rh(8)	2.7430(10)
Au(1)-Rh(7)	2.7497(11)
Au(1)-Rh(6)	2.7592(11)
Au(1)-Rh(9)	2.7899(10)
Rh(2)-Rh(3)	2.8837(13)
Rh(2)-Rh(6)	2.8855(13)
Rh(2)-Rh(5)	2.8910(13)
Rh(2)-Rh(7)	2.9177(13)

Rh(3)-Rh(5)	2.9481(13)
Rh(3)-Rh(6)	2.9494(13)
Rh(4)-Rh(10)	2.8626(13)
Rh(4)-Rh(9)	2.9367(14)
Rh(4)-Rh(11)	2.9679(14)
Rh(5)-Rh(7)	2.9597(13)
Rh(6)-Rh(7)	2.9409(13)
Rh(8)-Rh(10)	2.9025(13)
Rh(8)-Rh(9)	2.9239(13)
Rh(8)-Rh(11)	2.9698(13)
Rh(9)-Rh(10)	2.9016(13)
Rh(10)-Rh(11)	2.8708(13)

