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1	Intracellular phase separation and its role in nickel sensing
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10	biomolecular condensates, transcriptional regulation
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13	Abstract
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15	Nickel homeostasis in many bacteria is controlled by the nickel-sensor NikR. A recent study
16	by Cao et al. found that Escherichia coli NikR undergoes phase separation and that this event
17	enhances its function as a nickel-dependent transcriptional repressor. The results suggest
18	that phase separation is functional for bacterial metal homeostasis.
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Nickel features a dual nature of poisoning and essential element, being required for the biochemical reactions of many prokaryotes, unicellular eukaryotes, and plants. These organisms set up a tight balance between metal accumulation, distribution, and detoxification to control its intracellular quota. This task is usually carried out by specific transcription factors known as Ni(II)-sensors [1]. In *Escherichia coli*, nickel uptake and efflux are controlled by two distinct sensors, NikR and RcnR, respectively [2].

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27 NikR is largely conserved in bacteria and archaea, being crucial for the virulence of 28 pathogens as *Helicobacter pylori*, thus representing a target for antibacterial drug discovery. 29 In E. coli, NikR functions as a repressor, binding to the promoter of the nikABCDE Ni(II)-import 30 system when it is loaded with Ni(II). In the last two decades, several studies have investigated 31 the mechanism through which NikR translates the intracellular Ni(II) concentration into a 32 transcriptional response, based on structural, biophysical and functional data. They found that 33 Ni(II) ions bind in a square-planar coordination geometry at the interface of a homo-34 tetrameric structure, made of the assembly of four C-terminal metal-binding domains (MBDs) 35 (Figure 1A). At the opposite sides of the structure, two dimeric N-terminal DNA-binding 36 domains (DBDs) are attached by flexible likers [3] (Figure 1A). In the apo-structure, different 37 conformers (open, cis and trans) interconvert in solution, depending on the relative 38 orientation of the DBDs to the MBDs (Figure 1B). Such binding heterogeneity is abrogated in 39 the DNA-bound form, in which the DBDs are symmetrically locked to two major grooves in the 40 cis orientation (Figure 1A,C) [2]. Protein-operator interaction requires Ni(II) binding, occurring 41 with affinities in the nanomolar range (Figure 1C) [2]. Metal binding sites lie topologically far 42 from the DBDs, thus Ni(II) binding should be allosterically propagated from the MBDs along 43 the protein structure. This mechanism likely involves a modulation of the protein dynamics: 44 Ni(II) binding does not shift the conformational landscape of the protein toward a single 45 conformer, rather it likely slows down the mobility of the DBDs, increasing the time spent in a 46 conformation competent for DNA binding, as suggested by experiments on H. pylori NikR [4].

47 A recent study published in Cell Reports [5] now adds a further layer to the mechanism 48 of Ni(II)-dependent transcriptional regulation by NikR. Using fluorescence microscopy, Cao et 49 al. found that E. coli NikR, reversibly undergoes concentration and temperature-dependent 50 liquid-liquid phase separation (LLPS) in vitro. The C-terminal domains play a major role in this 51 event, while the N-terminal domains only enhance the LLPS tendency of the full-protein [5]. 52 Initially reported in eukaryotic cells, LLPS has been described as a cellular strategy for 53 subcellular compartmentalization with dynamically forming phase-derived organelles. These 54 biomolecular condensates are built by the co-localization of proteins, nucleic acids, and other regulating molecules [6]. More recently, some bacterial proteins were observed to undergo 55 56 LLPS in vitro and in vivo, supporting a view of the bacterial cytoplasm far from an 57 undifferentiated medium, rather organized in a structure that includes membrane-less 58 organelles [7]. The ability of a protein to undergo phase separation in vitro is an important 59 finding, but its physiological relevance should be proven in living cells, as in vitro systems 60 cannot fully mimic the complexity of intracellular media [8]. To this aim, Cao et al. investigated 61 the occurrence of NikR-driven LLPS in *E. coli*. An over-expressed fusion of GFP-NikR spatially 62 localized near a pole region of the bacterial cells, hinting for the formation of intracellular 63 condensates (Figure 1D). These assemblies were dissolved by 1,6-hexanediol (Hex), suggesting 64 that they were liquid-like condensates and not solid-like aggregates [5].

65 Membrane-less organelles have the potential to be involved in many regulatory 66 pathways, such as transcription, because they increase the local concentrations of all the 67 components involved, like enzymes, transcriptional regulators, and DNA operators. LLPS of 68 bacterial RNA polymerase was indeed reported in vitro and in cell, thus supporting the 69 physiological relevance of this mechanism in bacterial transcription [9]. Thus, how does the 70 observed LLPS impact on the role of NikR as a Ni(II)-dependent regulator? Cao et al. showed 71 that NikR does not need Ni(II) nor DNA to form LLPS and that Ni(II) moderately enhances the 72 formation of droplets in solution and in cell [5]. However, only the presence of Ni(II), and not 73 of other metal ions such as Zn(II) and Mn(II) able to bind the protein in vitro [10], allows the 74 co-localization of promoter DNA in the condensates, indicating that NikR response to Ni(II) 75 remains intact in the phase-derived structures [5]. Disruption of LLPS impaired the regulatory 76 function of NikR relieving the repression from the promoter and leading to an intracellular 77 Ni(II) overload and metal-toxicity [5]. These results hint to a functional role of LLPS in 78 enhancing the activity of NikR, thus required for bacterial resistance to high Ni(II) 79 concentrations, and are coherent with the previously observed ability of E. coli NikR to 80 increase its DNA binding affinity when Ni(II) is in excess [2].

81 The results shown in this study provide the first evidence that intracellular phase 82 separation might be functional to regulate Ni(II) ion homeostasis in bacteria. Further studies 83 are needed to understand how metal-driven changes in protein folding and dynamics 84 influence the formation of biomolecular condensates by NikR. As this protein is central for 85 metal homeostasis of several pathogens, this discovery, if confirmed for other NikR proteins, 86 might open new routes for drug discovery. It is noteworthy that the interaction of NikR with 87 Bi(III), a known drug against H. pylori that impairs bacterial Ni(II) homeostasis, prevents LLPS 88 by NikR both in solution and in cell [5].

The discovery that a bacterial metal-sensor is regulated by LLPS opens new perspectives and will likely stimulate the research, with the evaluation of this event in other metal-dependent systems. Some examples of metal-induced LLPS were reported in the past, but they do not involve physiological processes carried out by metallo-proteins. Considering that one third of all existent proteins require transition metal ions for their function we can envisage that membrane-less organelles that contain metallo-proteins might regulate other physiological processes, possibly influenced by intracellular metal availability.

9697 Figure legend.

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99 Figure 1. NikR forms biomolecular condensates in E. coli cells. A) E. coli NikR binds to the 100 nik operator DNA in a homo-terameric structure (PDB: 2HZV) binding four Ni(II) ions at the 101 tetrameric interface (details in the insert). B) Different conformers of NikR interconverts in 102 solution depending on the position of the DNA binding domains connected to the central 103 metal binding domains by flexible linkers. C) In the presence of Ni(II), NikR binds the nik 104 promoter DNA and represses the transcription of *nikABCDE* Ni(II) uptake system. D) 105 Condensates of *E. coli* NikR accumulate in a pole of the bacterial cells and regulate the 106 activity of NikR as a Ni(II)-sensor. This figure was created using BioRender 107 (https://biorender.com/). Representation of the protein structure was created using 3D 108 Protein Imager (https://3dproteinimaging.com/protein-imager/). 109 110

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