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Strain-induced trace element mobility in a quartz-sulphide vein system: An example from the ONKALO™ spent nuclear fuel repository (Olkiluoto, SW Finland)

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*Published Version:*

Strain-induced trace element mobility in a quartz-sulphide vein system: An example from the ONKALO™ spent nuclear fuel repository (Olkiluoto, SW Finland) / Marchesini B.; Menegon L.; Schwarz G.; Neff C.; Keresztes Schmidt P.; Garofalo P.S.; Hattendorf B.; Gunther D.; Mattila J.; Viola G.. - In: JOURNAL OF STRUCTURAL GEOLOGY. - ISSN 0191-8141. - STAMPA. - 154:(2022), pp. 104473.1-104473.15. [10.1016/j.jsg.2021.104473]

*Availability:*

This version is available at: <https://hdl.handle.net/11585/838797> since: 2021-11-17

*Published:*

DOI: <http://doi.org/10.1016/j.jsg.2021.104473>

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Marchesini B.; Menegon L.; Schwarz G.; Neff C.; Keresztes Schmidt P.; Garofalo P.S.; Hattendorf B.; Gunther D.; Mattila J.; Viola G. *Strain-induced trace element mobility in a quartz-sulphide vein system: An example from the ONKALO™ spent nuclear fuel repository (Olkiluoto, SW Finland)*

JOURNAL OF STRUCTURAL GEOLOGY, VOL. 154. ISSN 0191-8141

DOI: 10.1016/j.jsg.2021.104473

The final published version is available online at:

<https://dx.doi.org/10.1016/j.jsg.2021.104473>

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# Type of paper: Original Research Paper

**Title: Strain-induced trace element mobility in a quartz-sulphide vein system: An example from the ONKALO™ spent nuclear fuel repository (Olkiluoto, SW Finland).**

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**Keywords:** sulphides; LA-ICP-TOFMS; diffusion; chemical replacement; trace element mobility

## 37 **Abstract**

38 This work investigates element mobility and deformation mechanisms in sulphide-bearing quartz  
39 veins associated with a strike-slip fault exposed in the ONKALO<sup>TM</sup> Finnish deep repository for spent  
40 nuclear fuel (Olkiluoto Island, southwest Finland). It combines petrography, trace element mapping  
41 by Laser Ablation Inductively Coupled Plasma Time-of-Flight Mass Spectrometry (LA-ICP-TOFMS)  
42 and Electron Backscattered Diffraction (EBSD) analysis of representative microstructures. The fault  
43 core was repeatedly reactivated by multiple brittle deformation episodes assisted by hydrothermal fluid  
44 batches with distinct trace element signatures. LA-ICP-TOFMS element distribution maps and EBSD  
45 on sulphides reveal local, syn-deformational intragrain enrichment of primary and secondary elements  
46 (i.e., As, Co, Cu, Ag, Sn, Sb, Pb, Se, In, Te) by a combination of microscale plastic and brittle  
47 deformation at the reaction fronts. Fluid ingress along microcracks enhancing chemical alteration of  
48 pyrite combined with element diffusion along dislocations and tilt boundaries, controlled trace element  
49 mobility in sulphides at the small scale. At the scale of the vein system, the competence contrast  
50 between inclusions of soft sulphides in the harder host quartz may favor local fracture nucleation and  
51 fluid flow.

52

## 53 **1. Introduction**

54 Tectonic faults and fractures may behave as major fluid conduits (e.g., Caine 1996; Sibson 1996;  
55 Faulkner et al., 2010) favouring fluid flow through the Earth's crust, especially in low-permeability  
56 crystalline basements (e.g., Jamveit et al., 2018; Marchesini et al., 2019; Sibson et al. 2020; Ceccato  
57 et al., 2021). Infiltration of fluids that are in disequilibrium with the mineral assemblage of the host  
58 rock may induce precipitation of authigenic minerals as well as also enhance fluid-rock interaction by  
59 processes ranging from syn-deformational mineralogical alteration to element mobility (e.g., Molli et  
60 al., 2010; Rossetti et al. 2010; Smith et al., 2013; Williams et al., 2017; Velásquez et al., 2018). During  
61 the latter, the chemical and physical properties of the host rock may undergo significant changes (e.g.,  
62 Putnis et al., 2009; Rossetti et al. 2010; Maggi et al. 2012; Fulignati et al. 2019), even affecting the

overall permeability within faulted rock volumes and, as a consequence, their overall mechanical properties (e.g., Sibson, 1975; Wintsch et al., 1995; Olsen and Scholz, 1998; Collettini et al., 2009; Cox, 2010). Faults also govern mineralization processes (e.g., Sibson, 1987; Blundell et al. 2003; Torgersen et al., 2014) in addition to having a crucial role in determining fluid transmissivity in geological reservoirs (e.g., Manzocchi et al. 2008; Agosta et al. 2009). In addition, recent studies emphasized how microscale fluid-mediated deformation of selected minerals, such as sulphides, zircon and monazite, can induce secondary element mobility (e.g., Dubosq et al., 2018; Cugerone et al., 2020; Piazzolo et al. 2016; Fougrouse et al. 2021a). Studies on sulphides showed that fluid-enhanced strain localization may lead to significant secondary enrichments of trace elements and may even be responsible for concentration of economically exploitable precious metal byproducts from sulphide extraction (e.g., Au, Ge; Reddy and Hough, 2013; Dubosq et al., 2018; Cugerone et al., 2020; Fougrouse et al. 2021b). The characterization of the deformation mechanisms active in fluid-mediated brittle-ductile faults associated with quartz-sulphide veins is thus crucial to reveal the processes of deformation-induced mobility of minor and trace elements.

We studied microstructures of quartz-sulphide veins associated with a strike-slip fault system cutting through the crystalline basement exposed in the ONKALO<sup>TM</sup> deep repository. The combination of sulphide imaging by Laser Ablation Inductively Coupled Plasma Time-of Flight Mass Spectrometry (LA-ICP-TOFMS) and Electron Backscattered Diffraction (EBSD) allowed us to gain a refined understanding of how intragrain structures may control trace element mobility.

We anticipate that our analysis reveal that trace element enrichments are potentially controlled by sulphide rheology and enhanced by volume-change reactions. We also observed that, at the studied conditions, some trace elements, such as Ag and Sb, are excellent tracers of intragranular deformation.

## **2. Geological Setting**

### *2.1 Geological framework*

The island of Olkiluoto (Fig. 1a) in southwestern Finland is made of rocks of the Paleoproterozoic Svecofennian orogenic province, which is formed by supracrustal high-grade metamorphic sequences and plutonic rocks (e.g., Kärki and Paulamäki, 2006; Fig. 1a). These rocks underwent a long history of polyphasic ductile deformation between 1.86 and 1.79 Ga that led to repeated structural overprinting and mechanical reactivation of inherited and suitably oriented shear zones under amphibolite-facies metamorphic conditions (Aaltonen et al., 2010 and 2016). During progressive regional exhumation and cooling, the Olkiluoto bedrock was affected by a long brittle deformation history starting at 1.75 Ga (Mattila and Viola, 2014). This involved both the brittle reactivation of pre-existing structures (Viola et al., 2011; Mattila and Viola, 2014; Skyttä and Torvela, 2018; Prando et al., 2020), but also the formation of new faults and joints (Marchesini et al. 2019). Marchesini et al. (2019) and Prando et al. (2020) studied in detail a system of conjugate strike-slip faults, as a representative example of late Svecofennian structures responsible of the initial embrittlement of the basement. These faults nucleated at the brittle-ductile transition and were reactivated multiple times “en route” to the surface concomitantly with the ingress of batches of overpressured hydrothermal fluids. These subvertical strike-slip fault system are oriented N-S and NW-SE, respectively. They are commonly associated with hydrothermal veins, which commonly contain sulphides (Pere et al. 2009; Marchesini et al. 2019). For one of these faults, the NW-SE-striking BFZ300, Marchesini et al. (2019) constrained the composition, temperature and pressure of the synkinematic hydrothermal fluids.

In this study we focus on rocks affected by the latest reactivation of the BFZ300 fault core. Specifically, we analyzed the chlorite-sulphide-bearing quartz veins that formed by hydrofracturing at fluid pressure conditions  $P_f = 140\text{--}180$  MPa and temperatures between 170 and 300 °C (Stage 3 of Marchesini et al. 2019).

## 2.2 Fault zone architecture

The internal architecture of the BFZ300 fault was constrained by combined field structural analysis (Figs. 2a-c) and drill core logging (drill core PH21; Fig. 2d). In the following, we provide a

short description of the fault architecture and its complexities. For further details the reader is referred to Marchesini et al. (2019). The BFZ300 is a NNW-SSE-striking dextral strike-slip fault as determined by Riedel shears and dextral step-over zone (Fig. 2a). The fault cuts through amphibolite-facies migmatite, interlayered with gneiss and pegmatitic granite.

Its damage zone is defined by an increased fracture density compared to the host rock, from which it is separated by two discrete slip surfaces ( $Y_I$  planes; Fig. 2a). The damage zone contains sets of conjugate NW-SE-striking dextral and NNE-SSW-striking sinistral hybrid fractures (Fig. 2b) and laterally continuous NNW-SSE striking joints (Fig. 2b). The joints and the hybrid veins are infilled by a first generation of quartz, referred to as Qtz I. The fault core is bound by two main discrete slip surfaces ( $Y_{II}$ ; Fig. 2a) oriented NNW-SSE and the core is composed by two distinct quartz veins and cataclastic layers (Fig. 2c). The older of the two quartz generations, referred to as Qtz I, contains evidence of cyclic brittle and ductile deformation. The younger generation, Qtz II, has a milky white appearance, a coarser grain size than Qtz I ( $>1$  cm), and lacks of significant evidence of crystal-plastic deformation and recrystallization (Fig. 2c). The vein infilled by Qtz II contains small (1-2 cm) aggregates of sulphides mainly located at its center (Fig. 2d). Sulphides are present as minor phases, making-up ca. 3% of the bulk volume of the vein. Radiate chlorite is observed in textural equilibrium with Qtz II at the edge of the vein (Figs. 2d; 3a).

### 3. Analytical Methods

In this study, we combine petrographic documentation of fault samples with a multi-technique analytical documentation that includes elemental imaging by LA-ICP-TOFMS and determination of crystal structures and orientations by EBSD analysis. Details on accuracy, long-term reproducibility, and determination of limits of detection (LODs) of these techniques are given in the Supplementary Material file. Below, we summarize the main features that are useful for data interpretation.

#### 3.1. Optical petrography

139 Polished thin sections of Qtz II vein were obtained from drill core PH21 (Fig. 2d). Thin sections were  
140 cut orthogonal to the vein wall and parallel to the vein opening direction. The microstructural analysis  
141 was conducted at the Dipartimento di Scienze Biologiche, Geologiche e Ambientali of the University  
142 of Bologna with a Nikon ECLIPSE CI POL petrographic microscope via reflected- and transmitted-  
143 light petrography.

144

### 145 3.2. Laser Ablation Inductively Coupled Plasma Time-of Flight Mass Spectrometry

146 The chemical composition of the authigenic sulphides (pyrite FeS<sub>2</sub>; sphalerite (Zn, Fe)S; galena  
147 PbS) was first determined by Electron Probe Microanalysis (EPMA) analysis (see Supplementary  
148 Material).

149 The variable composition of sulphide phases determined by EPMA motivated the acquisition of  
150 detailed element composition maps by LA-ICP-TOFMS to study the mass fraction and spatial  
151 distribution of trace elements in the sulphide grains. We thus constructed element distribution maps of  
152 a sulphide assemblage using a LA-ICP-TOFMS instrument setup developed at the Department of  
153 Chemistry and Applied Biosciences, Laboratory of Inorganic Chemistry of the ETH of Zürich,  
154 Switzerland. This setup uses an ICP-TOF mass spectrometer instrument (TOFWERK, Thun,  
155 Switzerland) in combination with an argon fluoride (ArF) excimer LA system (193 nm, GeoLas C,  
156 Lambda Physik, Goettingen, Germany) and a low-dispersion LA “tube cell” developed in-house  
157 (Wang et al. 2013), which is designed for high-speed and high-resolution LA-ICPMS imaging  
158 applications. While specific details of this setup are given in the Supplementary Material, we highlight  
159 here that the low-dispersion LA and the ICP-TOFMS detection allows full mass spectrum acquisition  
160 at high speeds.

161 The distribution maps presented here were obtained from regions of ~1 mm<sup>2</sup>, ranging from 500x600  
162 µm to 700x700 µm. These maps were generated with circular laser spot sizes 5 µm in diameter using  
163 a hole-drilling approach (Neff et al.2020) a burst of 25 laser pulses per sample position. This number  
164 of laser pulses leads in total to a high signal intensity per sample position (i.e., pixel), and thereby to



high pixel sensitivity and improved LODs while maintaining the lateral resolution. Elements with signal levels above their LODs at each pixel were quantified based on a 100% mass normalization procedure. The LODs ranged between hundreds of  $\text{ng g}^{-1}$  to tens of  $\mu\text{g g}^{-1}$  per pixel for elements across the periodic table. Across the elemental mass spectrum, 15 major, minor, and trace elements provided concentrations above LODs across a larger area to construct element distribution maps in these samples (Fe, As, Co, Cu, Sn, Sb, Zn, Ni, Pb, Cd, Ag, Mn, In, Se and Te). Only the most significant of these maps are presented and discussed here, while the others are reported in the Supplementary Material.

### 3.3. Electron Backscattered Diffraction

Crystallographic orientation data were obtained from automatically indexed EBSD patterns collected on a JEOL 6610 SEM at the Electron Microscopy Centre of the University of Plymouth, UK. EBSD data are presented in the form of 1) Inverse Pole Figure (IPF) maps; 2) Equal area, lower hemisphere pole figures of crystallographic orientations; 3) Local misorientation (LM) maps, 4) Texture component (TC) maps, and 5) misorientation profiles across low-angle boundaries. Further information about these maps are reported in the Supplementary Material.

## 4. Microstructural analysis

At the microscale, sulphide aggregates appear as complex, composite and interdigitated bodies embedded within coarse Qtz II crystals (Figs. 3a; 4a). The host quartz has a grain size between 200  $\mu\text{m}$  and 5 mm (Figs. 3b-c). Close to the sulphide aggregates, the host quartz is intensely deformed by a dense network of fractures, which are marked by trails of fluid inclusions and only partially healed (Figs. 3c). Close to the sulphide aggregates the host quartz also exhibits a local distortion of the crystal lattice, as documented by undulose extinction (Fig. 3b). A few cm away from the sulphide aggregates, the host quartz exhibits instead a lower degree of fracturing, with mostly intragranular fractures and intact growth structures, and undistorted crystals (Figs. 3d-e).

190 Sulphide aggregates are composed of pyrite, sphalerite and galena in highly variable  
191 proportions. Minor chalcopyrite is locally also observed but it occurs exclusively within sphalerite in  
192 the form of rows of small-sized blebs and rods (i.e., the so called “chalcopyrite disease”; Fig. S1).  
193 Because of its minor occurrence, chalcopyrite is not discussed in the following LA-ICP-TOFMS maps  
194 although its influence on deformation and element distribution is considered in the discussions (see  
195 Section 7.4).

196 Pyrite exhibits a generally euhedral texture in equilibrium with quartz. In thin section, pyrite is seen to  
197 be variably affected by late fractures that are associated with partial replacement of pristine pyrite by  
198 sphalerite and galena (Figs. 4b-e). Sphalerite-galena assemblages are characterized instead by anhedral  
199 textures and represent the infilling of major fractures crosscutting quartz and pyrite (Fig. 4c-e) and/or  
200 the alteration overgrowths on pyrite grains (Fig. 4b). On the basis of [these aforementioned](#)  
201 observations, we selected three representative [portions](#) of the studied Qtz II vein, referred to as  
202 Domains 1 to 3, to document the heterogeneity in fracturing and pyrite alteration within the system  
203 under consideration (Fig. 4a).

204 Pyrite in Domain 1 (marked in green in Fig. 4a) has a euhedral texture in equilibrium with quartz and  
205 [it](#) is characterized by a grain size between 600 and 800  $\mu\text{m}$  (Fig. 4b). The pyrite margins are corroded  
206 when in contact with the sphalerite-galena assemblage (Fig. 4b). Internally, pyrite does not show  
207 evidence of fracturing or replacement into sphalerite and galena. Proximal quartz is affected by  
208 intragrain fractures (Fig. 4b)

209 Pyrite in Domain 2 (yellow in Fig. 4a) exhibits a subhedral texture and has a grain size between 400  
210  $\mu\text{m}$  and 2 mm (Fig. 4c). Dissimilar to pyrite in Domain 1, pyrite in Domain 2 is locally dissected by  
211 intra- and intergranular fractures (red dashed line of Fig. 4c) and it is locally replaced by sphalerite-  
212 galena assemblage. Fractures are both barren and locally sealed by sphalerite and galena (Fig. 4c).  
213 Proximal quartz is affected by intra- and intergranular fractures, that are associated with sphalerite and  
214 galena associations (Fig. 4c).

215 Pyrite in Domain 3 (red in Fig. 4a) exhibits a euhedral to subhedral habit and shows a heterogeneous  
216 grain size distribution (200-800  $\mu\text{m}$ ; Fig. 4d) due to the brittle fragmentation of larger crystals. Pyrite  
217 is deformed by intra- and intergranular fractures (red dashed lines of Figs. 4d-e) along which it is  
218 extensively replaced by sphalerite and galena. Sphalerite-galena assemblages seal intra- and  
219 intergranular fractures (Figs. 3d-e) as documented, albeit to a lesser extent, for pyrite in Domain 2  
220 (Figs. 4d-e). Barren fractures in pyrite are rare (Fig. 4e). Proximal quartz is affected by intra- and  
221 intergranular fractures, which control the precipitation of sphalerite and galena (Fig. 4d).

222

## 223 **5. LA-ICP-TOFMS element distribution maps**

224 We analyzed the chemical composition of major, minor and trace elements within representative pyrite  
225 and sphalerite-galena assemblages from each of the three structural domains. In the following, we  
226 present in detail the distribution maps of selected elements, i.e., As, Co, Cu, Ag, Sn, Sb, Pb, Se, In and  
227 Te. These elements highlight both the primary compositional variability of sulphides and secondary  
228 enrichments. The maps of the other detected elements are available in the Supplementary Material.

229

### 230 *5.1. Sulphides in Domain 1*

231 In the undeformed or weakly deformed sulphides, the compositional maps highlight a direct  
232 association between minor-, trace elements enrichments and specific minerals.. The maps show in  
233 particular that pyrite is As-Co-Pb-Se-Te-Zn-Cd-bearing (Fig. 5; Fig. S2). More in detail, As and Co  
234 show enrichments at grain boundaries and internal bands sub-parallel to grain boundaries, which  
235 indicate that these elements are mostly associated with primary crystal growth (Fig. 5 b,c). Growth  
236 rims are crosscut by thin, branching “chemical structures”, which mostly affect the grain (Figs. 5b-c).  
237 These intragranular chemical structures are thus marked by a local enrichment in As and Co, with mass  
238 fractions ranging between 3000 and 10.000 mg/kg (Figs. 5b-c). The observed textural relationships  
239 between growth structures and intragranular structures are sketched in Fig. 5l.

240 The trace elements Sn, Sb, Ag and Se are mostly associated with galena (with mass fractions  
241 up to 100 mg/kg, Figs. 5 d-j), as inferred from the comparison with optical microstructures (Fig. 5a)  
242 and Pb distribution (Fig. 5h). Se is also associated with pyrite and it occurs in mass fractions between  
243 10 and 300 mg/kg (Fig. 5i). The mass fraction of Te is elevated along the pyrite-sphalerite and pyrite-  
244 galena edges. Cu, In and Cd are primarily associated with sphalerite, in which they occur in mass  
245 fractions as high as 10.000 mg/kg (Figs. 5d, j; S2). Also Mn is primarily associated with sphalerite but  
246 it occurs at lower mass fractions (between 200 and 600 mg/kg; S2). Local enrichments of Ag, Pb and  
247 Sb are visible along the intragrain structures (Figs. 5e, g, h). Irregular distributions of Co, Cu, Ag, Sb,  
248 Sn and Mn in the sphalerite-galena assemblage have an internal polygonal structure (Figs. 5f-g, S2),  
249 indicating sulphide precipitation during vein segregation.

250

## 251 *5.2. Sulphides in Domain 2*

252 Similar to what described for Domain 1, the enrichment (up to 10.000 mg/kg) of As within the  
253 subhedral pyrite from Domain 2 highlights the presence of growth zoning (Figs. 6b-c). However,  
254 growth rims are crosscut by several intragranular structures (Fig. 6b-k) that are marked by high mass  
255 fractions (up to 1.000 mg/kg) of Co, Cu, Ag, Sn, Sb, Pb, Ni and, to a lesser extent, In, Se, and Mo  
256 (Figs. 6 and S3). These structures are best outlined by the distributions of Sb, Ag, Sn and show  
257 irregular branching shapes with variable orientations in the pyrite grain (Figs. 6d-h). The textural  
258 relationships between grow structures and the network of structures in Domain 2 are schematized in  
259 Fig. 6l. A macroscopic open fracture in the right-hand side of the scanned area (Fig. 6a) shows edges  
260 enriched with trace elements associated with sphalerite and galena.

261 Cu, In and Te are mostly associated with sphalerite (Figs. 6d, j, k). High Se mass fractions are found  
262 in galena ( $> 1000$  mg/kg) although it also occurs in pyrite with mass fraction between 10 and 300  
263 mg/kg (Fig. 6i), as in Domain 1.

264

## 265 *5.3. Sulphides in Domain 3*

266 Pyrite in Domain 3 contains As and Co a mass fractions of <10.000 mg/kg and systematically  
267 lower values of all other trace elements (Figs. 7, S4). The sphalerite-galena assemblage contains the  
268 same minor and trace elements that we documented in Domain 2 (i.e., Cu, Ag, Sn, Sb, Se, In, Te, Ni,  
269 Cd, Mn), with mass fraction values ranging between 1.000 and 100 mg/kg (Figs. 7, S4).  
270 Primary growth structures in pyrite are only faintly outlined by the elemental distribution of As and  
271 Co (Figs. 7b-c). Growth rims are very discontinuous and they are only locally visible (Figs. 7b-c).  
272 Pyrite aggregates are crosscut by a pervasive network of intra- and intergranular structures, which are  
273 marked by high mass fractions of the above-mentioned trace elements (Fig. 7). Some of these  
274 structures are also decorated by galena and sphalerite (Fig 7h). The textural relationships between  
275 growth structures and the observed network of structures are conceptualized in Fig. 7l.  
276 As for Domain 2, Sb, Ag and Sn best define the intragranular microstructure. Sb in particular, not only  
277 highlights the same fracture network as Ag and Sn, but it also shows a spatial correlation with  
278 enrichment structures of Co.  
279 Cu and In are mostly associated with sphalerite, whereas Te and Se are mostly associated with galena  
280 (as in Domain 1). As noted above, Se is also associated with pyrite and it occurs in mass fractions  
281 between 10 and 300 mg/kg (Fig. 7i)

282

## 283 **6. EBSD analysis of pyrite**

284 LA-ICP-TOFMS imaging of pyrite grains documents the presence of several chemical features  
285 that could not be revealed by standard petrographic analysis. We used electron back-scattered  
286 diffraction (EBSD) to verify the nature of intragranular features identified in the trace elements maps.  
287 They can in fact correspond to both lattice distortions or, in general, to misorientation bands or  
288 microcracks that could have represented pathways for intragranular element mobility.

289 Because of the destructive nature of the LA-ICP-TOFMS mass spectrometry, EBSD maps could not  
290 be acquired on exactly the same areas mapped by LA-ICP-TOFMS. EBSD analysis had therefore to

291 be done on areas located as close as possible to those investigated chemically, but still within the same  
292 grain (Figs. 8a,9a). **Pyrite in Domain 1 could not** be analyzed by EBSD.

293 On all EBSD maps and pole figures, the X direction is the pole to the vein wall (Figs. 8b, 9b).

294

#### 295 *6.1. EBSD maps of Domain 2*

296 The inverse pole figure (IPF) coloring scheme of the pyrite grains within Domain 2 shows the  
297 presence of three pyrite grains (Fig. 8b) wherein several low-angle boundaries are identified  
298 (highlighted by white lines in Fig. 8b), corresponding to misorientation values between 1.5 and 10°.  
299 Grain 1 and grain 2 in Fig. 8b display low-angle boundaries oriented at 45°-50° from the base of the  
300 map. Another low-angle boundary **is present** within grain 3; it is oriented at 10°-15° from the base of  
301 the map (Fig. 8b) and it exhibits a sutured geometry.

302 The local misorientation map (LM; see Supplementary Material for method) shows that pyrite grains  
303 contain several substructures characterized by local misorientations of up to 2.5° (Fig. 8c), the most  
304 prominent of which are the identified low-angle boundaries. The texture component map (TC; see  
305 Supplementary material) of Grain 1 highlights up to 4° of intracrystalline distortion, which is  
306 accommodated across misorientation bands and low angle boundaries oriented at 45-70° from the base  
307 of the map (Figs. 8d, e).

308 The TC map of grain 3 shows a more complex and variable misorientation distribution, where the  
309 maximum misorientation is about 4-4.5° and is mainly found along the pyrite grain outer edges (Fig.  
310 8f). The misorientation profile across the low-angle boundary within grain 3 with a sutured geometry  
311 (segment c-d, Fig. 8f) shows that the cumulative lattice misorientation is on average c. 3° with a sharp  
312 misorientation step in the profile across the low-angle boundary located at c. 60 μm from point “c”  
313 (Fig. 8g).

314 To integrate the results from the cumulative misorientation profile collected from grain 1, pole figures  
315 **are** plotted for a subset sampled across the low-angle boundary within grain 1 (Fig. 8h). The location  
316 of the sampled area is highlighted by a dashed white rectangle in Fig. 8d. The boundary trace analysis

of the low-angle boundary provides a solution for the most feasible slip plane and slip direction relative to the rotation axes (Fig. 8h), assuming a tilt boundary geometry. The dispersion paths on the pole figures identify  $\langle 100 \rangle$  as the rotation axis. A tilt boundary plane containing the boundary trace of the low-angle boundary and the rotation axis is consistent with these data (Fig. 8h).

## 6.2. EBSD maps of Domain 3

The inverse pole figure coloring scheme of the pyrite grain mapped within Domain 3 shows the existence of a few low-angle boundaries (highlighted by white lines in Fig. 9b) with orientation spanning from 10-15° from the base of the map (named “I”, Fig. 9b) to c. 140° (named “II”, Fig. 9b). The LM map shows that the grain contains a network of substructures characterized by local misorientation of up to 1.5°, of which the low-angle boundaries shown in Fig. 9b are the most prominent (Fig. 9c). The TC map shows that the lattice distortion is accumulated across the identified low-angle boundaries (Fig. 9d). The misorientation profile across low-angle boundary “I” (a-b, Fig. 9d) shows a progressive lattice distortion along the first half of the profile, followed by a sharp misorientation of ca. 1.5° in correspondence of the low-angle boundary itself. In the second half of the profile, the misorientation accumulated is relatively low ( $<1^\circ$ ), indicating that the two portions of the grain sampled by the profile have a different extent of internal lattice distortion (Fig. 9e).

A second misorientation profile is acquired across low-angle boundary “II” (c-d, Fig. 9d). Here, the cumulative lattice misorientation reaches a maximum value of 7.5° (Fig. 9f) and is accommodated by a sharp misorientation increase across the two low-angle boundaries sampled by the profile (Fig. 9f). Pole figures are plotted for a subset sampled across low-angle boundary “I” (Fig. 9g; the sampled area is delimited by a white dashed rectangle in Fig. 9d). The dispersion paths on the pole figures identify  $\langle 100 \rangle$  as the rotation axis (Fig. 9g). A tilt boundary plane containing the boundary trace of the low-angle boundary and the rotation axis is consistent with these data (Fig. 9g).

## 7. Discussion

343 7.1. *Sulphide composition and trace element distribution*

344 LA-ICP-TOFMS maps show that the distribution of trace elements enrichments in sulphides  
345 (i.e., Co, Cu, Ag, Sn, Pb, Se, In, Te) correspond to a network of intragranular structures, which are  
346 referred to as “chemical structures”. Trace and minor element variability in crystal lattices of sulphides  
347 is extensively documented in literature (e.g., Fleet et al., 1993; Abraitis et al., 2004; Blanchard et al.,  
348 2007; Vukmanovic et al. 2014; Sykora et al., 2018). As and Co are common pyrite components, as they  
349 are incorporated into its crystal structure at mass fractions as high as 10 wt% (e.g., Abraitis et al., 2004;  
350 Blanchard et al, 2007; Sykora et al., 2018). Hence, their occurrence as the most abundant trace elements  
351 in pyrite, at mass fractions that reach about 1 wt% and 0.1 wt%, respectively, is expected.  
352 Considerations similar to those made for pyrite can be made for the occurrence of Cu, In, Cd and Mn  
353 within sphalerite (Figs. 5-7; Figs. S1-S3). These trace elements are reported in the literature among the  
354 most typical for sphalerite, and their mass fractions can be as high as 6-7 wt% (e.g., sphalerite is the  
355 main ore mineral for Cd; Cook et al., 2009). The abundance of these elements in the studied sphalerite  
356 ranges between c. 0.03 wt% (for In) and 1 wt% (Cd and Cu), and is therefore close to the average  
357 values documented in natural samples. Similarly, galena from several geological environments shows  
358 Ag, Sb, Sn, and Se with proportions varying from few tens of ppm to 1.5 wt% (e.g., George et al, 2015;  
359 Palero-Fernandez and Martin-Izard, 2005). The values of these trace elements in the analyzed galena  
360 correspond to the lower values of this range (i.e., Ag: 0.02-0.03 wt%; Sb: c. 0.001 wt%; Sn: c. 0.1  
361 wt%). On the contrary, Se in galena is measured at mass fractions that exceed 1 wt% in our samples  
362 (Figs. 5-7).. The mass fraction of Se in the analyzed galena is close to the highest Se value documented  
363 so far in galena from several ore deposits (i.e., c. 1.8 wt%. George et al, 2015).

364 Our LA-ICP-TOFMS maps reveal how the inclusion of many of these known traces in  
365 sulphides occur during crystallization (e.g., As and Co) or during deformation. In particular, they  
366 reveal which elements (Ag, Sb) are good tracers of intragranular deformation processes in this  
367 geological environment.



368 As and Co are included in pyrite from its crystallization, as suggested by their involvement in primary  
369 growth rims (Figs. 5-6). Growth bands in pyrite highlighted by As and Co have been documented in  
370 several environments (e.g., Fleet et al., 1993). Incorporation of As into the pyrite structure takes place  
371 at the expenses of S, while incorporation of Co takes place at the expenses of mainly Fe (i.e.  
372 stoichiometric substitutions). However, Co is also associated with sphalerite and galena deformation  
373 structures (Figs. 5-7). Although the distributions of As and Co are mainly along growth rims, element  
374 distribution maps show that they are locally enriched also along network of intragranular structures  
375 (Figs. 5l, 6l, 7l). On the contrary, distribution maps of Cu, Ag, Sn, Sb, Pb, Se, In, Te show that these  
376 elements are predominantly associated with sphalerite-galena assemblages (Figs. 5-7). In pyrite these  
377 elements are revealed by intragranular structures.

378

## 379 *7.2. Deformation mechanisms in pyrite*

380 In addition to the fractures that are directly visible by traditional optical microscopy, EBSD mapping  
381 documents the presence of several low-angle boundaries in the pyrite grains, which we discuss further  
382 below. Tilt boundary formation in pyrite is consistent with the misorientation profiles (Fig. 8e and 9e)  
383 and with the boundary trace analysis (Figs. 8h, 9g) of low-angle boundaries identified in Domains 2  
384 and 3. Boundary trace analysis indicates that rotation occurred along a single  $\langle 100 \rangle$  axis. As the (100),  
385 (010) and (001) planes in the cubic crystal system are symmetrically equivalent, which means that the  
386 pole figures are identical, tilt boundaries can be explained by the activity of the  $\{001\}\langle 100 \rangle$  slip  
387 system. This inference is compatible with early work on naturally deformed pyrite from Van Goethem  
388 et al. (1987) that indicates the  $\{001\}\langle 100 \rangle$  slip system as the primary slip system in pyrite. Later  
389 experimental work on pyrite (e.g. Cox, 1981; Barrie et al., 2008) confirmed that the  $\{001\}\langle 100 \rangle$  is  
390 the slip system most easily activated between 600 and 700°C. Here, however, we show that this slip  
391 system may be activated also at a much lower temperature, that is,  $< 300^\circ\text{C}$ , as estimated in Marchesini  
392 et al. (2019).

393 Microcracking of pyrite, possibly leading to enhanced trace element mobility (see section 7.4) is also  
394 revealed by EBSD. The cumulative misorientation profiles collected across some subgrain boundaries  
395 (e.g., Figs. 8g and 9f) show abrupt misorientation jumps of up to 3°. This abrupt misorientation of the  
396 crystalline lattice can be interpreted as due to the effect of incipient microcracking. These effects can  
397 be discriminated from those reflecting lattice distortion and tilt boundary formation as the latter are  
398 more gradual and continuous in space, such as, for example, the case shown in Figures 8e, 9e or close  
399 to the abrupt change in misorientation in Fig. 9f.

400 To conclude, texture component maps of single grains, misorientation profiles across low-angle  
401 boundaries and boundary trace analysis of portions of low-angle boundaries suggest that intragranular  
402 deformation in the studied system was accommodated by a combination of low-T plasticity and  
403 microfracturing. The transition from brittle to crystal-plastic behavior of pyrite was estimated in the  
404 ~200 °C to 450 °C temperature range (Graf et al. 1981; Barrie et al.; 2009; Barrie et al. 2011).  
405 Geothermometric constraints by Marchesini et al. (2019) indicate that the studied Qtz II vein  
406 precipitated and was subsequently deformed between 170 and 300 °C. These conditions are, therefore,  
407 consistent with our observations of competing, yet coexisting, crystal plastic and brittle deformation  
408 mechanisms.

409

### 410 *7.3. Strain-induced trace element mobility*

411 The coexisting brittle and plastic intragranular structures suggest that the observed trace element  
412 enrichments may be related to deformation processes. Trace elements may have been fluid transported  
413 and distributed by fluid percolation with associated mineral-fluid reactions along microcracks.  
414 Alternatively, they may have been concentrated within crystals by the activity of high-diffusivity  
415 pathways, such as tilt boundaries or dislocations. Boudier et al. (2010) suggested that fluids can  
416 preferentially ingress the lattice of mineral phases along dislocation walls. Also, several studies have  
417 already documented that intracrystalline deformation may induce differential element mobility, which,  
418 in turn, results in abrupt changes of element concentrations (e.g., Reddy and Hough, 2013; Piazzolo et

419 al., 2016; Cugerone et al., 2020; Peterman et al., 2016; Fougrouse et al., 2019; Dubosq et al., 2019;  
420 Ribeiro et al., 2020).

421 Segregation of trace elements from the crystalline lattice and local accumulation along single  
422 dislocations or dislocation arrays may occur by means of mechanisms such as pipe diffusion (Love,  
423 1964) and /or dislocation impurity pair diffusion (DIP diffusion; Petukhov and Klyuchnik, 2012)  
424 mechanisms. In the pipe diffusion model dislocation arrays act as “pipes” along which slow-diffusing  
425 components, such as heavy trace elements, are able to preferentially migrate (Vukmanovic et al.,  
426 2014). This process thus requires a chemical potential gradient between the grain and an external phase,  
427 possibly an intergranular fluid phase (Vukmanovic et al., 2014), and is commonly associated with  
428 dislocation arrays (Love, 1964), although it has been shown that it can also be efficient along single  
429 dislocations (Legros et al., 2008). The aforementioned gradient can result from the presence of  
430 chemical or structural sinks, such as precipitates and chemically enriched or depleted volumes of the  
431 crystal (Legros et al., 2008; Piazzolo et al., 2016) or the core-to-core overlap of entangled dislocations  
432 or microfractures (Dubosq et al., 2019). In the DIP diffusion model, solute atoms close to dislocations  
433 are initially captured by the strain field created by the moving dislocations (Cottrell atmosphere;  
434 Cottrell and Bilby 1949) and concentrated along the dislocation itself. This causes a preferential  
435 remobilization of elements by the moving dislocations and their enrichment along the dislocations  
436 themselves (Cottrell and Bilby, 1949; Vukmanovic et al., 2014; Piazzolo et al., 2016; Fougrouse et al.,  
437 2019).

438 Dubosq et al. (2019) suggested that pipe and DIP diffusion processes may be simultaneously active in  
439 pyrite at temperatures typical of greenschist facies conditions (260-450 °C). Conversely, we have  
440 documented that, in pyrite, these mechanisms may activate element mobility even at temperatures as  
441 low as 170-300 °C. In addition, pipe and DIP diffusion could have combined with microcrack  
442 propagation induced by stress concentration at the reaction fronts. The combined effect of these  
443 mechanisms may have contributed to mobilize a far greater range of trace elements in sulphides than

documented in previous studies (Reddy and Hough, 2013; Dubosq et al., 2019; Fougereuse et al., 2019; George et al., 2018).

#### 7.4. *Fracturing and pyrite replacement in the BFZ300 fault system*

The microstructural analysis of the sulphide-bearing quartz vein shows that quartz is texturally in equilibrium with radiate chlorite (Fig. 3a). Additionally, also quartz and pyrite exhibit equilibrium textures (Fig. 4). We interpret this evidence as supporting the primary cogenetic character of pyrite and quartz-chlorite (Fig. 10a), which precipitated from a first batch of fluids rich in As and Co (Figs. 5-7). Conversely, sphalerite-galena-chalcopryrite assemblages are mainly found along fractures and as pyrite replacement (Figs. 4c-e), suggesting that they crystallized at a later stage from a renewed ingress of hydrothermal fluids, possibly enriched in Co, Cu, Ag, Sn, Sb, Pb, Se, In and Te (Figs. 5-7; Fig. 10a). Given the relative mechanical weakness of pyrite with respect to quartz (e.g., Aguilar-Santillan, 2008; Evans, 1984), it is possible that the presence of pyrite inclusions embedded in quartz host may potentially have enhanced local fracturing by stress concentration phenomena. This, in turn, may have favored the later percolation of the fluid batch that caused the sph-gal-ccp precipitation (Fig. 10a).

At the grain scale, replacement of pyrite by the new fluid batch was enhanced by the brittle and plastic deformation mechanisms discussed above (Section 8.3). Within microfractures, the precipitation of new mineral phases may have promoted fracture propagation and further alteration by volume expansion reactions at the crack tips (Fig. 10b). Volume expansion reactions have been documented in a number of geological settings including retrograde metamorphism during exhumation of metamorphic rocks (Jamtveit et al. 2008; van der Straaten et al. 2008; Okamoto et al. 2015), peridotite serpentinization (Iyer et al. 2008; Plümper et al. 2012), hydrothermal mineralization (Zhao et al. 2014; Li et al. 2015; Ulven et al. 2017; Zhang et al. 2020).

In our system, sphalerite, galena and chalcopryrite have a larger molar volumes than the host pyrite, at least at room temperature (e.g., Robie and Bethke 1962). A volume-expansion reaction of these minerals may have exploited microfractures and the (110) and (100) cleavages of pyrite. For

470 chalcopyrite replacement, the process of reaction-induced expansion has been experimentally  
471 documented by Zhang et al. (2020) at  $T \sim 200$  °C. Considering large volume expansion coefficients  
472 (Fei, 1995), the same process of reaction-induced deformation can be inferred also for galena and  
473 sphalerite. These deformations are portrayed by trace element distribution, and in particular by Ag and  
474 Sb in the structural domains with the highest degree of fracturing and replacement of pyrite (Domains  
475 2-3; Figs. 6-7).

476

## 477 **8. Conclusions**

478 This study investigates the trace element content in a sulphide-bearing quartz vein associated  
479 with a strike-slip fault system by combining microstructural analysis and high-spatial resolution  
480 mapping of trace elements in fault rocks. The obtained LA-ICP-TOFMS dataset shows that trace  
481 elements are enriched along several intragranular “chemical features”. EBSD analysis shows that these  
482 enrichments may be related to microcracks and tilt boundaries. At the reaction fronts, trace element  
483 mobility and chemical replacement is promoted by diffusion along tilt boundaries and dislocations and  
484 may be promoted by reactions inducing net volume changes. Precipitation of new mineral phases  
485 within fractures enhances the alteration of pristine minerals and microfracture propagation.

486 We observed that in the case of deformed sulphides, the distributions of As and Co in pyrite can be  
487 used as a marker to study primary growth structures. Conversely, Sb, Ag and Sn are the most effective  
488 tracers of intragranular deformation at the studied conditions. This suggests that, in deformed sulphide  
489 aggregates at the studied conditions, these elements migrate more easily than the other tracers.

490 To conclude, this study highlights the importance of coupling microscale structural analysis  
491 with high-spatial resolution trace element mapping of fault rocks to investigate strain-induced element  
492 mobility within chemically reactive fault zones.

493

## 494 **Author contributions**

BM, GV and LM, conceived the paper. BM collected and processed laboratory data, provided their interpretations, drew the figures and wrote the paper. JM and GV sampled the outcrop and contributed to the interpretation of field data. BM, GS, CN, PKS, PSG, BH and DG acquired, processed and interpreted the LA-ICP-TOFMS data. BM and LM acquired and processed the EBSD maps. All author actively participated in discussing the results and drawing the conclusions.

## Acknowledgement

The Laboratory for Inorganic Chemistry at ETH is acknowledged for hosting Barbara Marchesini as a visiting PhD student between September and December 2018, when the LA-ICP-TOFMS analyses were carried out. The Plymouth Electron Microscopy Centre (PEMC) is acknowledged for support during the EBSD analyses when Barbara Marchesini was a visiting PhD student at the University of Plymouth between January and March 2019. Posiva Oy and the University of Bologna are acknowledged for funding. Sandra Piazzolo, Thomas Blenkinsop, Anna Rogowitz and Francesco Giuntoli are thanked for fruitful discussion concerning data interpretation. [The careful editorial handling by F. Agosta and](#) constructive comments from F. Rossetti and A. D. Lusk helped to significantly improve the manuscript. Francesca Prando is thanked for support during the processing of EBSD data.

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720

## **Highlights**

- Combined geochemical and microstructural study by LA-ICP-TOFMS and EBS analyses on sulphides
- Deformation-induced trace element mobility in natural sulphides under low-temperature and high fluid pressure conditions
- Sb, Ag and Sn as most effective tracers of intragranular deformation

### **Declaration of competing interest**

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

### **Author statement**

BM, GV and LM, conceived the paper. BM collected and processed laboratory data, provided their interpretations, drew the figures and wrote the paper. JM and GV sampled the outcrop and contributed to the interpretation of field data. BM, GS, CN, PKS, PSG, BH and DG acquired, processed and interpreted the LA-ICP-TOFMS data. BM and LM acquired and processed the EBSD maps. All author actively participated in discussing the results, drawing the conclusions, and manuscript revisions.

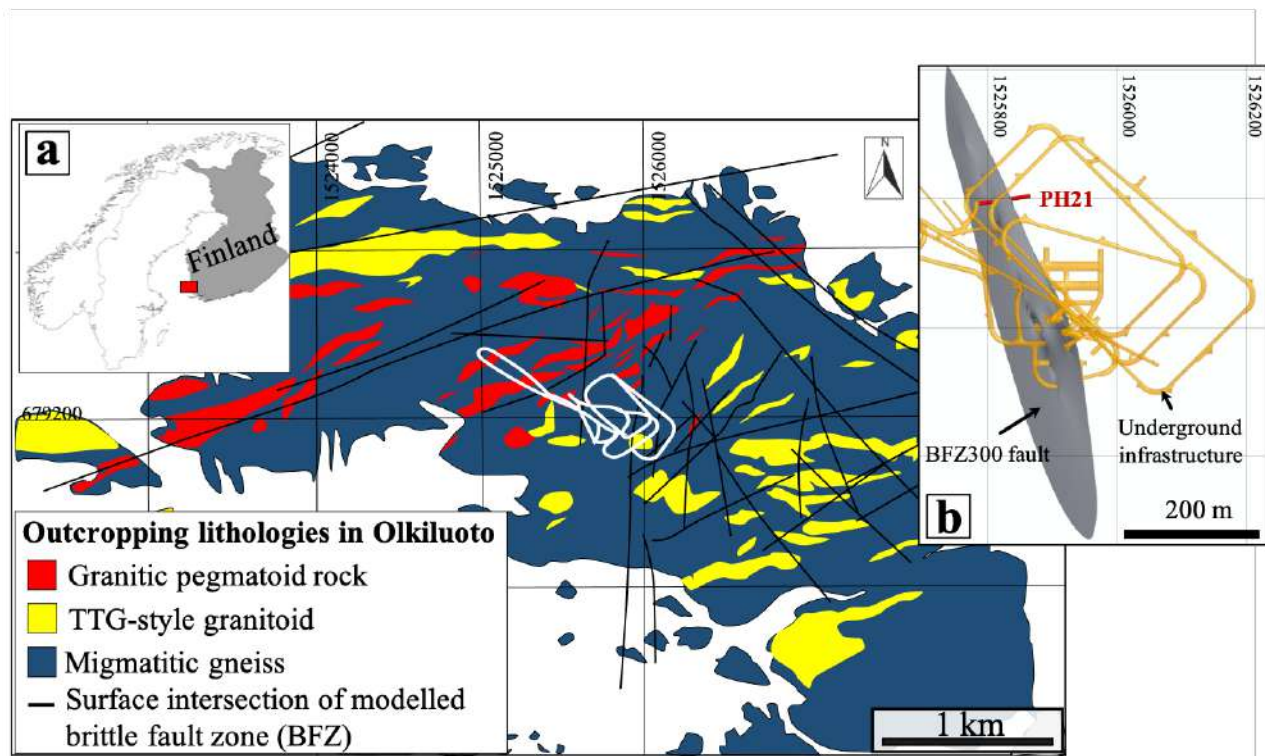


Figure 1. (a) Geological sketch of the Olkiluoto Island (SW Finland) and structural map, showing the surface intersection of modelled brittle fault zones (BFZ, modified from Aaltonen et al., 2016). Surface projection of the underground investigation facility is in white. Coordinates are given in the local KKJ1 coordinate system. (b) Modelled 3D surface of BFZ300 fault plane (grey) intercepted at ca. -421 m b.s.l. by the drill core PH21. Existing underground infrastructure is in yellow.



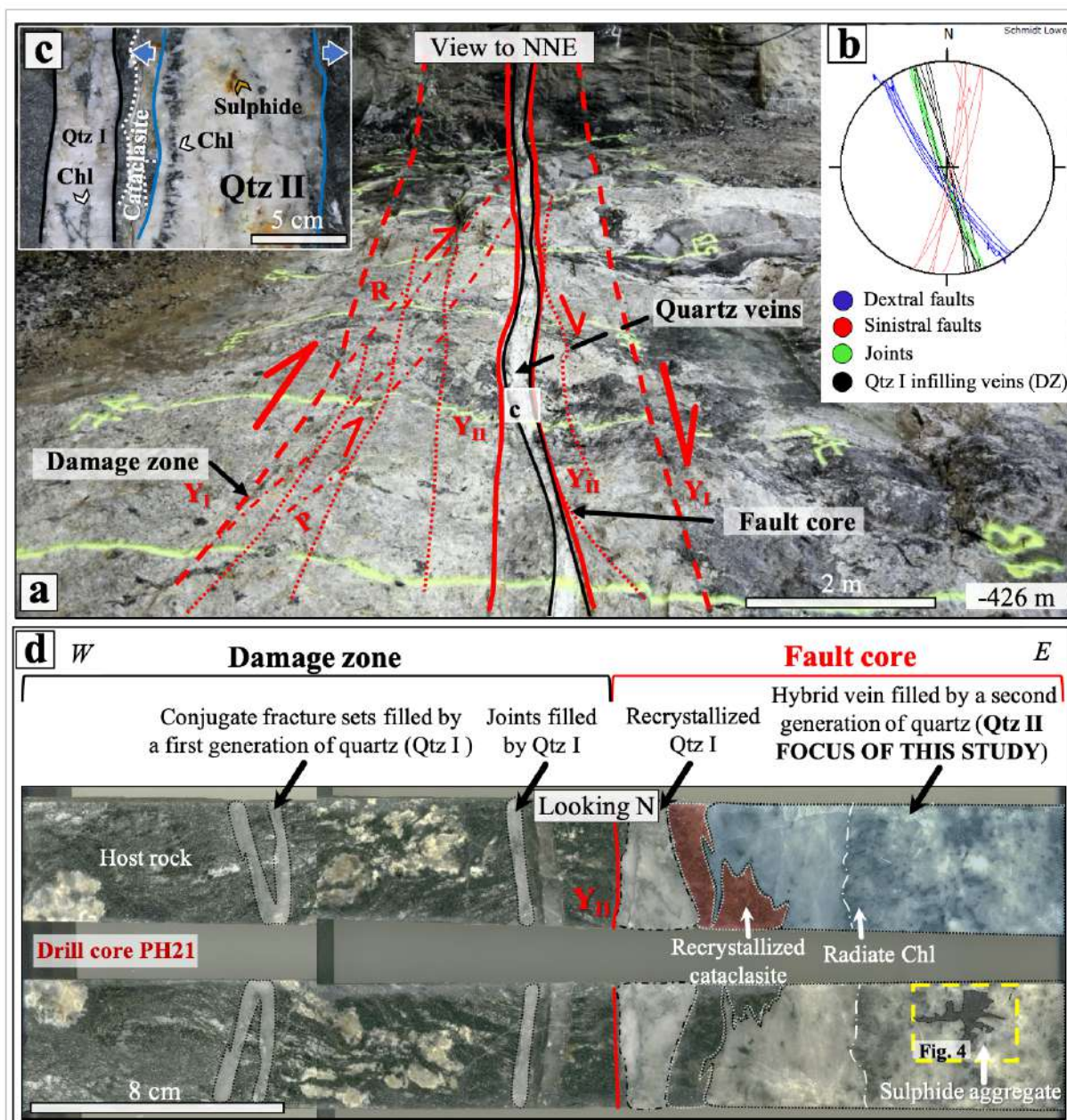


Figure 2. BFZ300 fault architecture. (a) The internal principal slip zone ( $Y_{II}$ ) separates the damage zone from the fault core. (b) Lower-hemisphere, equiangular projection shows the structures in the damage zone. The damage zone is characterized by conjugate sets of fractures and joints infilled by a first generation of quartz (Qtz I), and barren fractures (joints). (c) Fault core is characterized by recrystallized Qtz I and by a younger generation of quartz-sulphides-chlorite bearing vein (Qtz II). The blue arrows indicate the vein opening direction. Modified from Marchesini et al. (2019). (d) PH21 drill core and structural interpretation. Yellow dashed line indicates the location of thin section PH21-3. PH21 was drilled horizontally with a trend of  $080^{\circ}$ . It intercepts BFZ300 at ca. -421 m b.s.l.



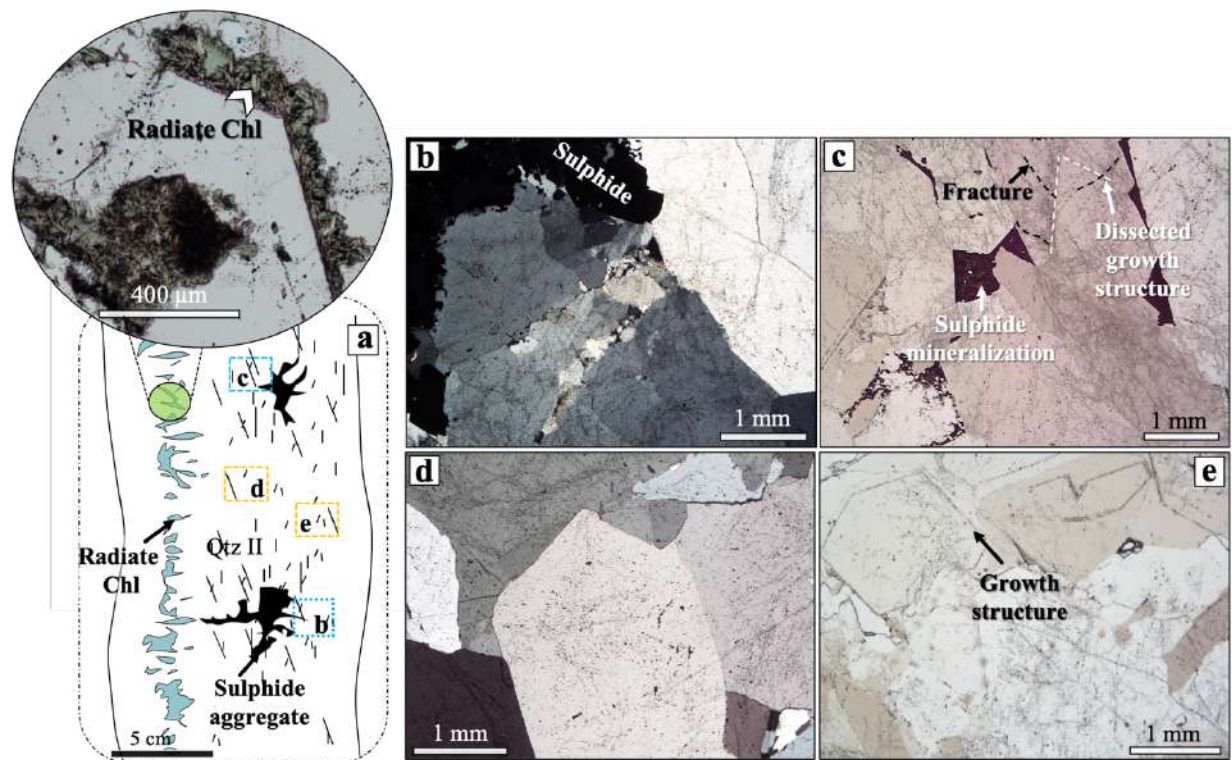


Figure 3. (a) Schematic fabric of the Qtz II vein. Inset shows textural equilibrium between quartz and chlorite. Host quartz is less deformed away from sulphide aggregates. (b) Close to sulphide aggregates, host quartz exhibits plastic distortion (undulose extinction) and it is (c) pervasively fractured by dense and pervasive network of intergranular and transgranular fractures (dashed black lines) dissecting primary growth structures (dashed white lines). (d) In the central part of the vein, at 4-5 cm from the sulphides, Qtz II crystals exhibit a lower degree of fracturing (only intragranular fractures) and no evidence of crystal plasticity. (e) Primary growth structures are well preserved.

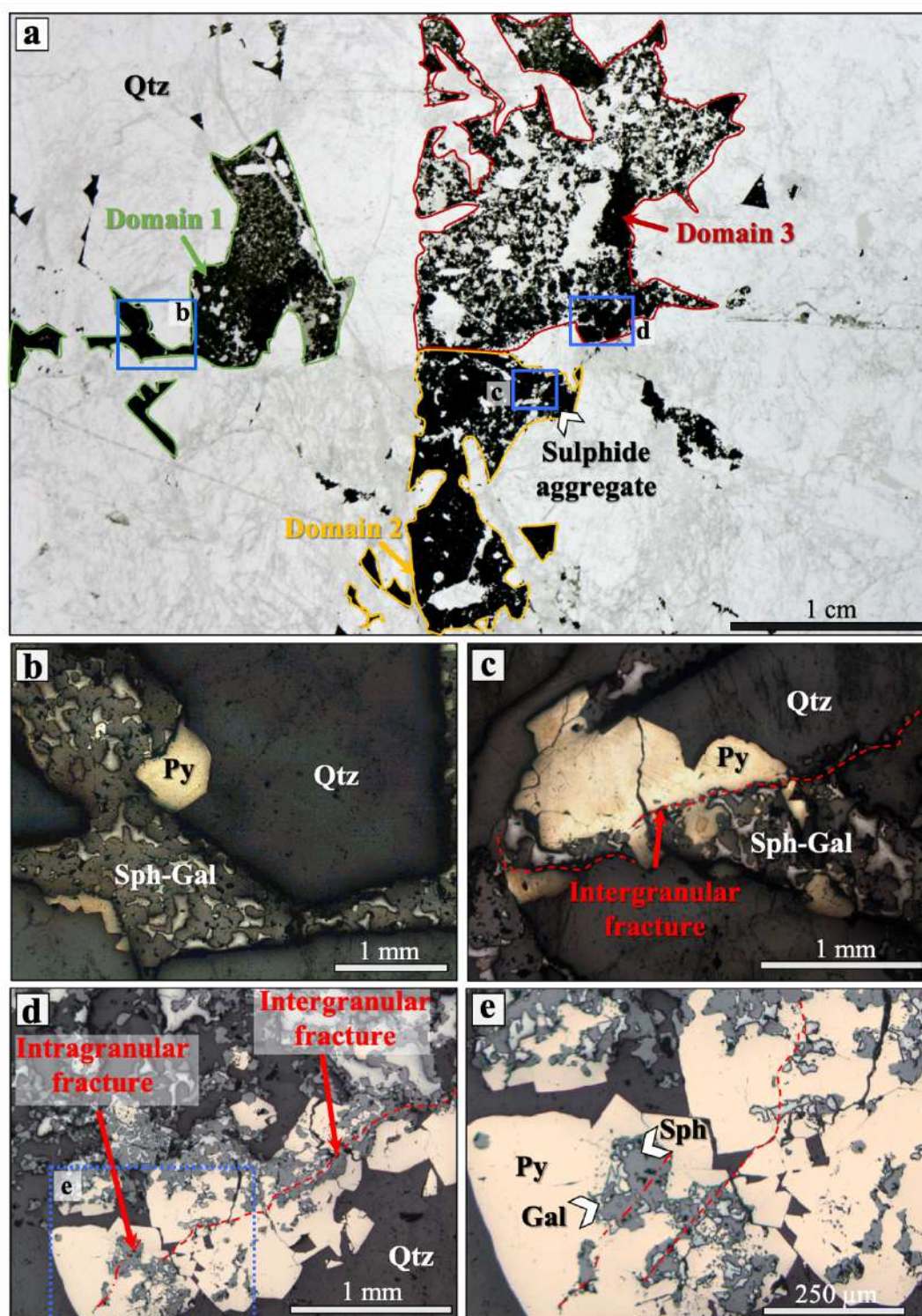


Figure 4. (a) Photomicrograph (plane polarized light) of a representative sulphide aggregate. The host quartz close to sulphide aggregate is highly fractured. Fractures are marked by trails of fluid inclusions. Three different structural domains were identified based on brittle deformation of pyrite and replacement of sphalerite and galena at the expenses of pyrite. (b) In Domain 1, pyrite exhibits a general euhedral texture, with minor anhedral sphalerite and galena deposited at the edge of the pyrite crystal. (c) In Domain 2 pyrite shows a subhedral texture. Replacement of pyrite by sphalerite and galena is locally observed along the pyrite crystal boundaries following the position of thin brittle intergranular fractures. (d) In Domain 3 pyrite displays a fractured texture and extensive replacement by sphalerite and galena along

discrete fractures. (e) Blown-up view of the replacement of pyrite into sphalerite and galena close to intragrain fractures and along crystal edges. Py: pyrite; Gal: galena; Sph: sphalerite; Chl: chlorite.



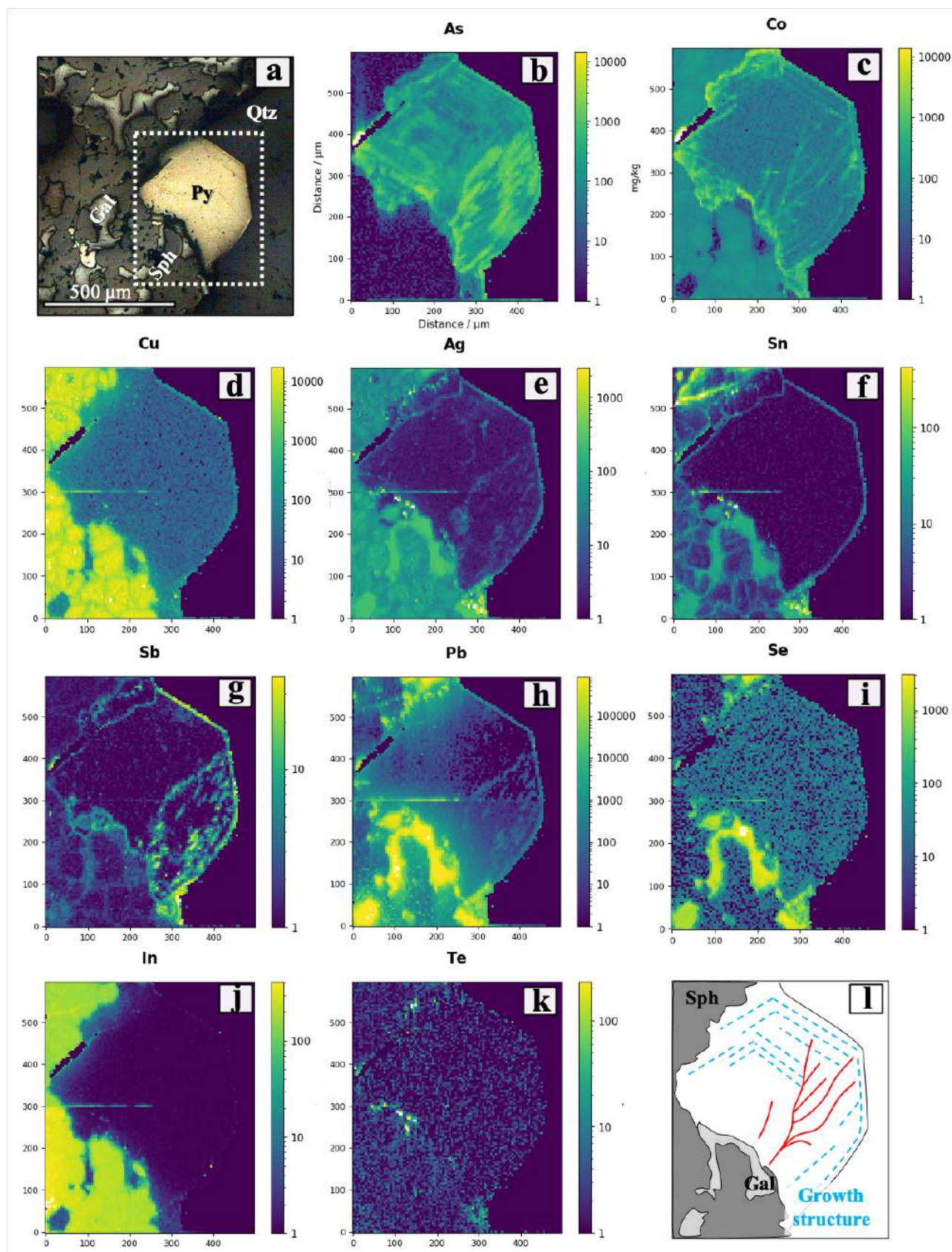


Figure 5. LA-ICP-TOFMS element distribution maps of sulphides in Domain 1. All panels show the same 500 x 600  $\mu\text{m}$  sample area. All the color-coded vertical bars show the range of mass fractions of the indicated elements. (a) Reflected light photomicrograph of the analysed area, where the white dashed line defines the area analysed by LA-ICP-TOFMS. (b-k) Details on element distribution maps As, Co, Cu, Ag, Sn, Sb, Pb, Se, In, Te are given in the main text. (l) Synthesis

of the textural relationships between primary grow structures and the network of chemical structures identified in pyrite. The line with high concentration at 300  $\mu\text{m}$  in the Y-axis is an artifact caused by the acquisition of the maps in sections. This artifact is located at the edge of the maps. Red lines indicate intragranular “chemical structures”. Qtz: quartz; Py: pyrite; Sph-Gal: sphalerite-galena.



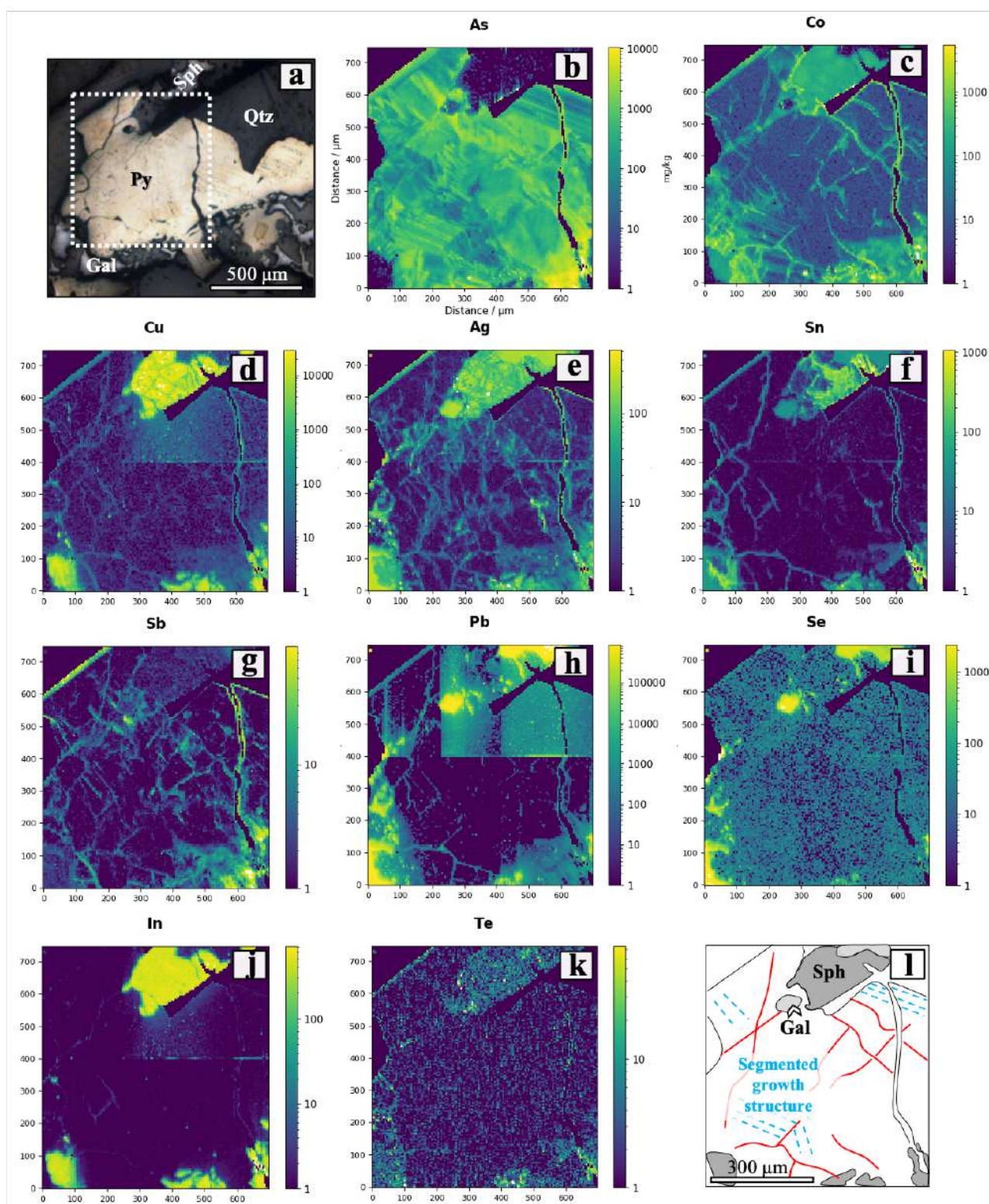


Figure 6. LA-ICP-TOFMS maps of sulphides in Domain 2. All panels show the same 700 x 700  $\mu\text{m}$  sample area. All the color-coded vertical bars show the range of mass fractions of the indicated elements. (a) Reflected light photomicrograph of the analysed area. The dashed white line defines the area analysed with LA-ICP-TOFMS. (b-k) Distribution maps of As, Co, Cu, Ag, Sn, Sb, Pb, Se, In and Te show intragranular chemical patterns. Details are given in the main text. (l) Synthesis of the textural relationships between primary grow structures and the network of chemical structures identified in pyrite. The line with high concentration at 400  $\mu\text{m}$  in the Y-axis is an artifact caused by the acquisition of the maps in sections. Red lines indicate intragranular "chemical structures". Qtz: quartz; Py: pyrite; Sph-Gal: sphalerite-galena.



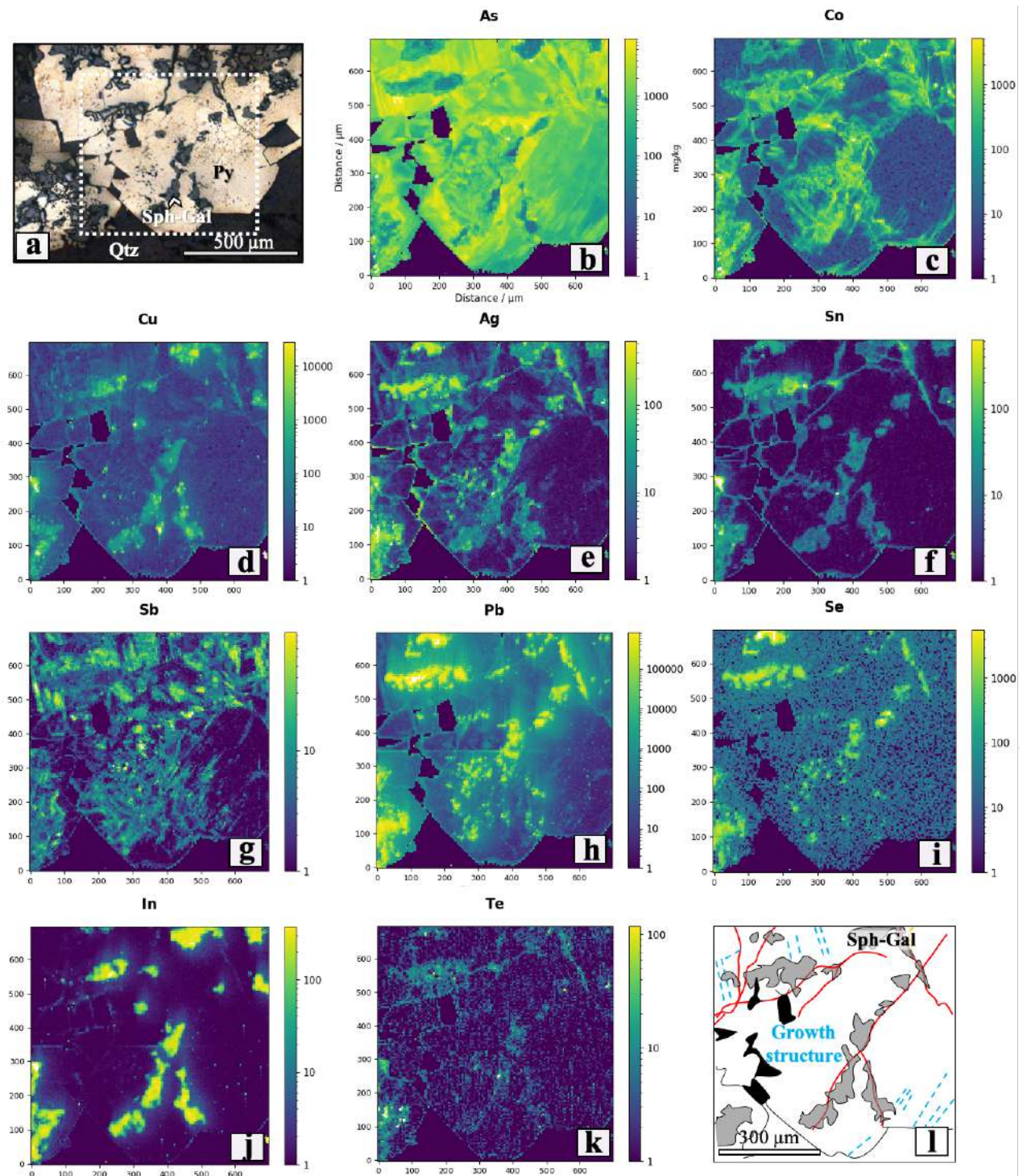


Figure 7. LA-ICP-TOFMS maps of sulphides in Domain 3. (a) Reflected light photomicrograph of the analysed area. The dashed white line highlights the analysed area. All panels show the same 700 x 700 µm sample area. All the color-coded vertical bars show the range of mass fractions of the indicated elements. (b-k) Details on element distribution maps of As, Co, Cu, Ag, Sn, Sb, Pb, Se, In and Te are given in the main text (l) Synthesis of the textural relationships between primary growth structures and the network of chemical structures identified in pyrite. Red lines indicate intragranular "chemical structures". Qtz: quartz; Py: pyrite; Sph-Gal: sphalerite-galena.

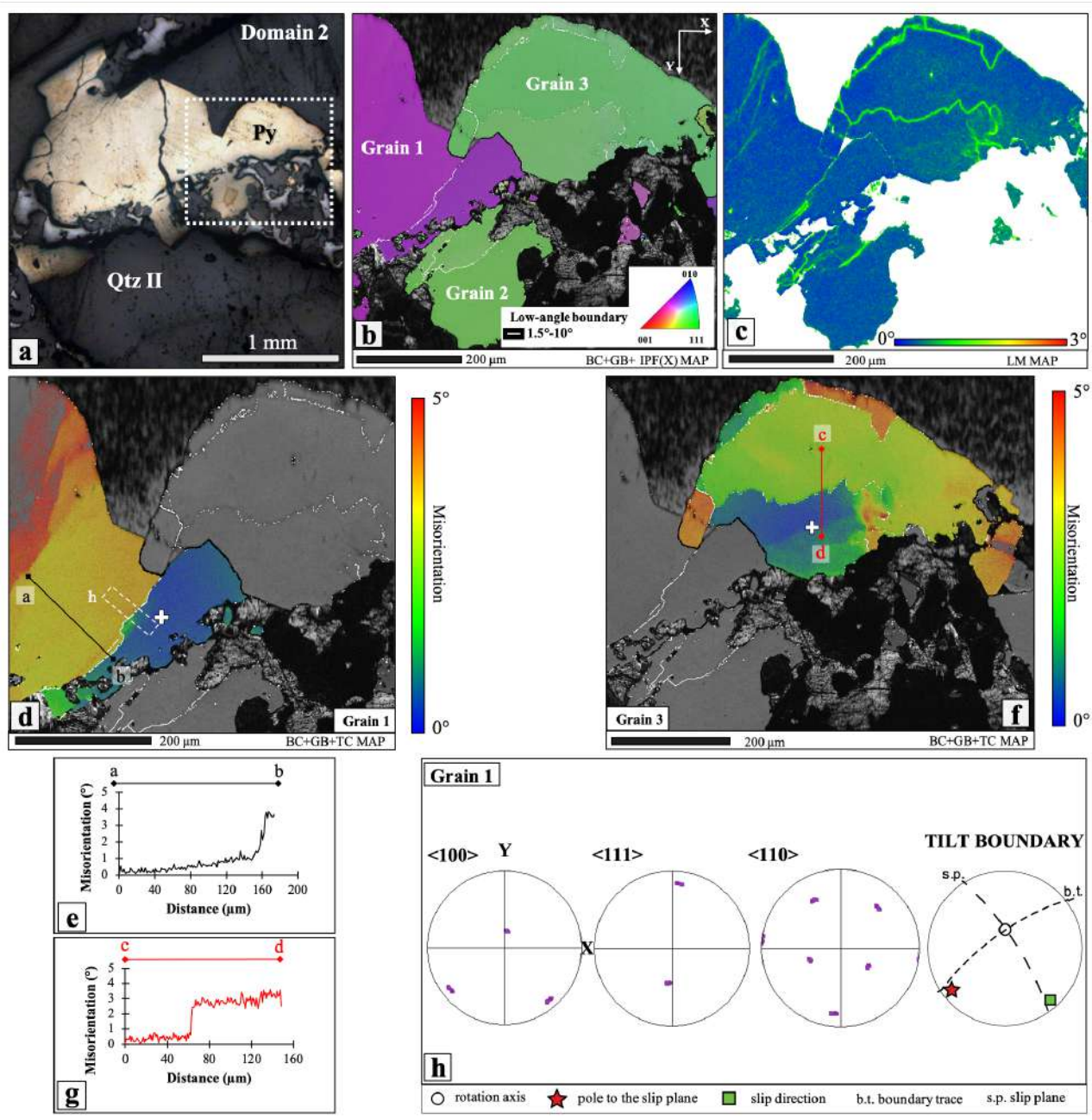


Figure 8. (a) Dashed white box indicates the analysed area with EBSD of a representative pyrite in Domain 2. (b) EBSD orientation map of pyrite, colour-coded according to the inverse pole figure (IPF) with respect to the X-direction (see text for explanation). IPF reference map is shown at the bottom right side. Low-angle ( $1.5\text{--}10^\circ$ ) boundaries are marked with a white continuous line, while grain boundaries ( $>10^\circ$ ) are shown in black. Grey pixels are non-indexed points. (c) Local misorientation (LM) map, colour coded according to the degrees of misorientation between each pixel and its neighbouring clusters of pixels within an area of  $7\times 7$  pixels. Colour scale is shown at the bottom right side. (d) Texture component (TC) map of Grain 1. The map is colour-shaded according to the angular misorientation from the reference point that is marked with a white cross. Trace of misorientation profile a-b oriented perpendicular to the low-angle boundary is shown in black. White dashed rectangle encompasses the sampled area for boundary trace analysis shown in (h). (e) Misorientation profile across the low-angle boundary in Grain 1 (a-b) shows a continuous increase in misorientation up to  $4^\circ$ . (f) TC map of Grain 3 and trace of misorientation profile c-d perpendicular to a representative section of the low-angle boundary. (g) Misorientation profile across the low-grain boundary shows a sharp step at  $60\text{ }\mu\text{m}$ .



(h) Upper hemisphere, equal angle stereographic projection of the sampled dataset and boundary trace analysis. Our interpretation of the most feasible solution of the boundary trace analysis is sketched.

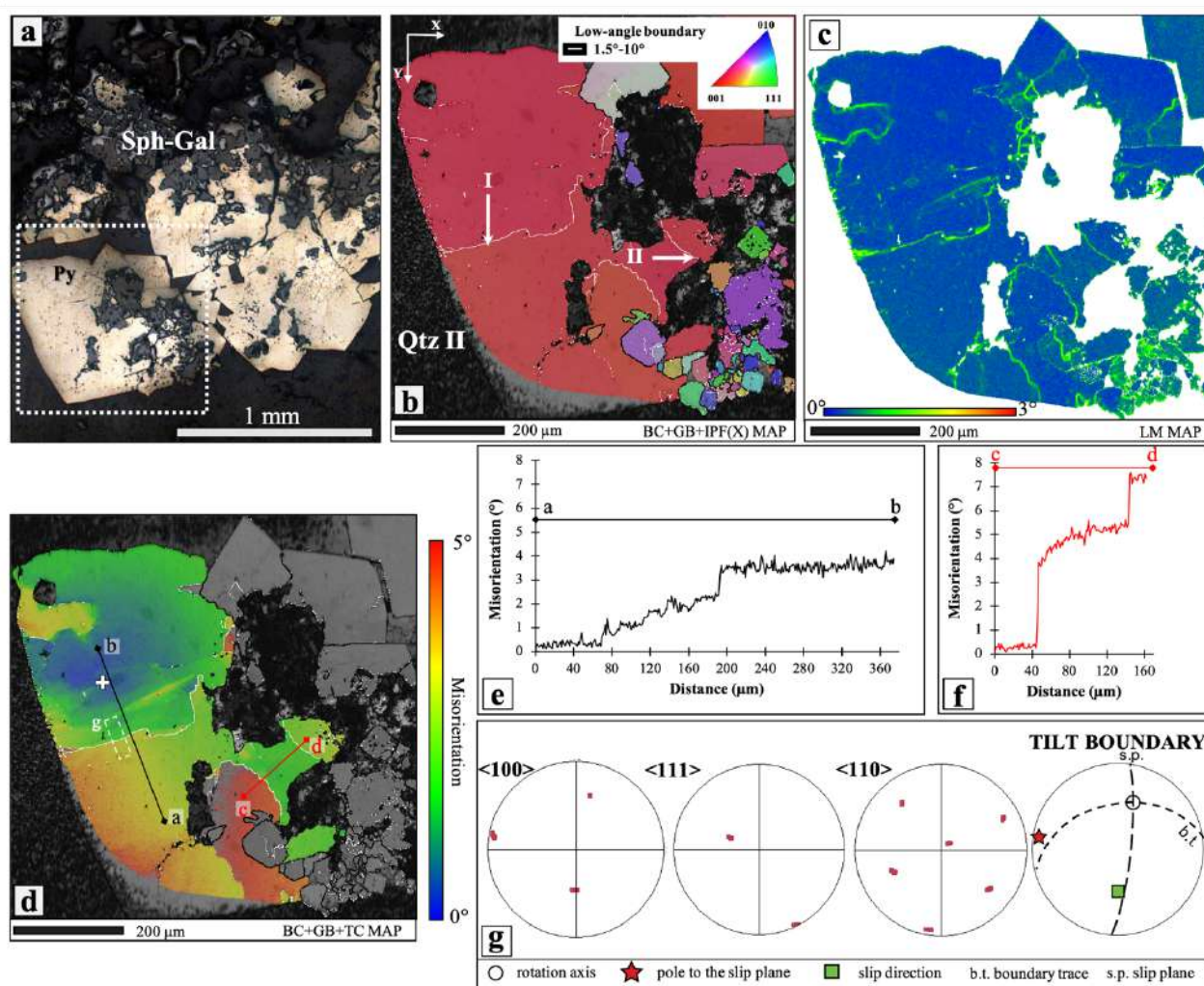


Figure 9. (a) Dashed white box indicates the analysed area with EBSD of a representative pyrite in Domain 3. (b) EBSD orientation map of pyrite, colour-coded according to the inverse pole figure (IPF) with respect to the X-direction (see text for explanation). Low-angle ( $1.5\text{--}10^\circ$ ) boundaries are marked with a white continuous line, whereas grain boundaries  $>10^\circ$  are in black. Grey pixels are non-indexed points. (c) LM map, colour coded according to the degrees of misorientation between each pixel and its neighbouring clusters of pixels within an area of  $7\times 7$  pixels. Colour scale is shown at the bottom right side. (d) TC map highlights the spatial distribution of misorientation across the analysed area. The map is colour-shaded according to the angular misorientation from the reference point marked with a white cross. White dashed rectangle encompasses the sampled area for boundary trace analysis. (e-f) Misorientation profiles across the two distinct sets of low-angle boundaries. (g) Upper hemisphere, equal angle stereographic projection of the sampled dataset and boundary trace analysis. Our interpretation of the most feasible solution of the boundary trace analysis is sketched.

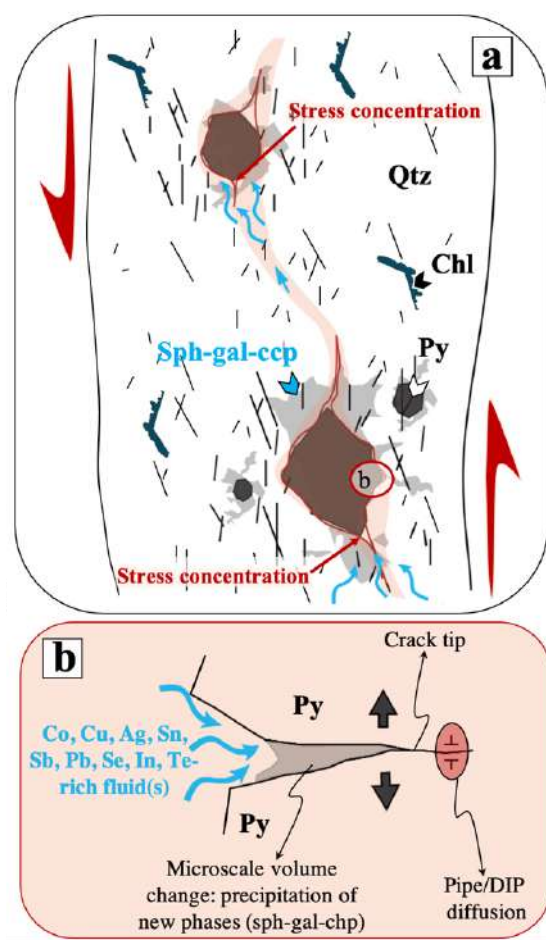


Figure 10. Conceptual model of fracturing and alteration of pyrite. (a) Pyrite (Py) precipitates in equilibrium with quartz (Qtz) and radiate chlorite (Chl) from a first batch of fluid. Sphalerite-galena-chalcopyrite assemblages precipitated as new phases from a later batch of fluids, enriched in Co, Cu, Ag, Sn, Sb, Pb, Se, In, Te. The new assemblages are arranged along intra- and intergranular fractures and they locally alter pyrite. The competence contrast between soft pyrite inclusions and harder host quartz may have enhanced local fracturing by stress concentration at the reaction fronts. This, in turn, may have favored the following percolation of Co, Cu, Ag, Sn, Sb, Pb, Se, In, Te-bearing fluid. Outside the reaction fronts the host quartz is less fractured. (b) Precipitation of sphalerite-galena-chalcopyrite assemblages along major fractures induce local microfracturing of pyrite, promoting fracture propagation and further alteration at the crack tips. Microcracks may thus locally exploit tilt boundaries and dislocations. Chemical alteration and trace element mobility in the system is also favored by the simultaneous activity of pipe and DIP diffusion along dislocations and tilt boundaries and volume change chemical interactions.