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New insights into the glass transition of dried fruits and vegetables and the effect of pulsed electric field treatment

This is the final peer-reviewed author's accepted manuscript (postprint) of the following publication:

*Published Version:*

Iaccheri E., Castagnini J.M., Dalla Rosa M., Rocculi P. (2021). New insights into the glass transition of dried fruits and vegetables and the effect of pulsed electric field treatment. *INNOVATIVE FOOD SCIENCE & EMERGING TECHNOLOGIES*, 67(January 2021), 1-7 [10.1016/j.ifset.2020.102566].

*Availability:*

This version is available at: <https://hdl.handle.net/11585/797856> since: 2021-02-10

*Published:*

DOI: <http://doi.org/10.1016/j.ifset.2020.102566>

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(Article begins on next page)

# Innovative Food Science and Emerging Technologies

## New insights into the glass transition of dried fruits and vegetables and the effect of pulsed electric field treatment

--Manuscript Draft--

<b>Manuscript Number:</b>	IFSET-D-20-00696R2
<b>Article Type:</b>	Research Paper
<b>Keywords:</b>	Pulsed electric fields; Glass transitions; Dried Fruits and vegetables; Air-drying; Water sorption isotherm
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<b>Order of Authors:</b>	Eleonora Iaccheri Juan Manuel Castagnini Marco Dalla Rosa Pietro Rocculi
<b>Abstract:</b>	<p>Pulsed Electric Field (PEF) pre-treatment has been recently studied to be applied to fresh apple, carrot, and potato tissue as novel food technology to improve the drying performances and the quality of final products. Although the modification induced by PEF and the related reactions have been studied as a function of many parameters and related to several quality aspects, no relationship has been investigated with glass transition temperature (<math>T_g</math>) modification of solid products. This study aims to evaluate the effect of PEF treatment on the <math>T_g</math> of different vegetable tissues. Obtained thermograms revealed the existence of two <math>T_g</math> (<math>T_{Ilg}</math>, and <math>T_{IIg}</math>), the first one mainly dependent from the presence of small molecules, such as sugars, organic acids and small amount of amino acids, while high molecular weight starch, fibers and small amount of proteins are associated to higher temperature values of <math>T_{IIg}</math>. PEF treated samples of dried apple, carrot, and potato compared to the untreated ones showed a <math>T_{IIg}</math> shift at 0.22 a w to a lower temperature, from 45.12 to 43.37, from 45.15 to 31.04 and from 90.23 to 85.81°C for apple, carrot and potato respectively. The <math>T_{IIg}</math> shift confirmed the mobility raise of the system promoted by PEF treatment, creating a less stable matrix.</p>
<b>Response to Reviewers:</b>	<p>Editor and Reviewer comments:</p> <p>Reviewer #1: The authors have substantially improved the manuscript and all the suggestions have been considered. However, I would be very careful with the statements about starch modification with such low intensities of PEF. I would suggest making a slight modification in line 274-278. --&gt; In this direction, the effect of PEF on detected in our study was reasonably caused by changes in the interaction of starch with water and other surrounding molecules.</p> <p>According to reviewer suggestion, the sentence has been corrected (line 276-278 of the revised version)</p>

Facultad de Ciencias de la Alimentación  
Universidad Nacional de Entre Ríos  
Concordia, Entre Ríos, Argentina  
Wednesday, July 8th, 2020

D. Knorr  
Innovative Food Science and Emerging Technologies

Dear Editor:

Please find enclosed a manuscript entitled: "New insights into the glass transition of dried fruits and vegetables and effect of pulsed electric field treatment" which I am submitting for exclusive consideration of publication as a research article in the Journal of Food Engineering.

Since no relationship has been studied between the glass transition temperature modification on solid products pretreated with Pulsed Electric Fields, this article aims to investigate the effect of this innovative technology into different fruit and vegetable tissues (apple, carrot and potatoes) by mean of sorption isotherms and differential scanning calorimetry. The most important result is the finding of two glass transitions temperatures, the  $T_g^I$  is probably mainly dependent from the presence of sugars, organic acids and a low concentration of amino acids while high molecular weight starch, fibres and proteins are associated to the  $T_g^{II}$ . Another event visible for apple and carrot was an overshoot area around  $T_g$ , linked to molecular rearrangement, reasonably ascribable to  $\beta$  relaxation phenomenon.

Further experiments are necessary in order to deeply understand the different structure modification inducing glass transition and system mobility change, using optical techniques.  $\beta$  relaxation phenomenon has to be confirmed by dielectric techniques and mechanical measurements, to effort results and gives additional information on product metastability.

Thank you for your consideration of my work. Please address all correspondence concerning this manuscript to me at Facultad de Ciencias de la Alimentación, Universidad Nacional de Entre Ríos, and feel free to correspond with me by e-mail (jmcastagnini@gmail.com).

Sincerely,  
Juan Manuel Castagnini

Editor and Reviewer comments:

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- PEF treated and dried apples, carrots and potatoes show two  $T_g$  ( $T_g^I$  and  $T_g^{II}$ ).
- $T_g^I$  is affected by water plasticization while  $T_g^{II}$  slide as a function of PEF voltage.
- $T_g^{II}$  decrease with PEF enhance metastability of the food matrices analysed.

1 **New insights into the glass transition of dried fruits and vegetables and the effect of pulsed**  
2 **electric field treatment**

3

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5

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14

15 **Abstract**

16 Pulsed Electric Field (PEF) pre-treatment has been recently studied to be applied to fresh apple, carrot, and  
17 potato tissue as novel food technology to improve the drying performances and the quality of final products.  
18 Although the modification induced by PEF and the related reactions have been studied as a function of many  
19 parameters and related to several quality aspects, no relationship has been investigated with glass transition  
20 temperature ( $T_g$ ) modification of solid products. This study aims to evaluate the effect of PEF treatment on the  
21  $T_g$  of different vegetable tissues. Obtained thermograms revealed the existence of two  $T_g$  ( $T_g^I$ , and  $T_g^{II}$ ), the  
22 first one mainly dependent from the presence of small molecules, such as sugars, organic acids and small  
23 amount of amino acids, while high molecular weight starch, fibers and small amount of proteins are associated  
24 to higher temperature values of  $T_g^{II}$ . PEF treated samples of dried apple, carrot, and potato compared to the  
25 untreated ones showed a  $T_g^{II}$  shift at 0.22  $a_w$  to a lower temperature, from 45.12 to 43.37, from 45.15 to 31.04

26 and from 90.23 to 85.81 °C for apple, carrot and potato respectively. The  $T_g^{II}$  shift confirmed the mobility  
27 raise of the system promoted by PEF treatment, creating a less stable matrix.

28

29 **Keywords:** Pulsed electric fields; Glass transitions; Dried Fruits and vegetables; Air-drying; Water  
30 sorption isotherm

31

## 32 **1. Introduction**

33 Nowadays, pulsed electric fields (PEF) treatment is of growing interest for application in food processing, for  
34 microorganism inactivation in fluid products, as well as for the enhancement of mass transfer phenomena on  
35 solid matrices subjected to further processing steps (e.g. dehydration) (Gongora-Nieto M. et al., 2010). When  
36 PEF is applied to vegetable tissue, the cell membranes can be permeabilized or irreversibly damaged  
37 (Weaver, 1994), with remarkable influence on mass transfer (Ade-Omowaye et al., 2002; Lebovka et al.,  
38 2002). The increasing consumer demand for high-quality fruit and vegetable products has prompted to study  
39 combined methods like PEF as a pre-treatment to dehydration, in order to obtain a stabilized product with high  
40 nutritive and organoleptic quality, reducing energy consumption and minimizing thermal degradation (Ade-  
41 Omowaye et al., 2002). The knowledge of the degree of the induced permeation on cellular food, through the  
42 cell membrane electroporation of the tissue, and its impact on the further processing or storage steps is a key  
43 factor for the final food quality, functionality, stability and safety (Ade-Omowaye et al., 2002). For plant tissue,  
44 a considerable electroporation effect can be observed at moderate electric fields of 0.5-1 kV/cm, as previously  
45 showed on carrots, potatoes and apples (Lebovka et al., 2007, 2002, 2001, 2000). PEF has been reported to  
46 increase the permeability of plant cells, demonstrating its potentiality to reduce the drying time for potatoes  
47 (Liu et al., 2018).

48 From a chemical point of view, the vegetable material subjected to PEF can be considered as a mixture of low  
49 and high molecular weight solids and water. Solids in foods can exist in a crystalline state or is an amorphous  
50 metastable one, the glassy state (Roos, 2010). Crystallization occurs in polymers that have a sufficiently  
51 ordered chain structure. The occurrence of complete crystallization of polymers is unlikely, while the  
52 crystallites are usually embedded in a residual amorphous matrix (Slade et al., 1993). A common feature of  
53 amorphous materials is that they contain excess of free energy and entropy in comparison to their crystalline

54 counterparts at the same temperature and pressure conditions. The study of the relation between water content  
55 and structural modifications in terms of physical events in complex foods is usually approached in terms of  
56 water activity ( $a_w$ ) and glass transition temperature ( $T_g$ ) assessment (Moraga et al., 2011). Macroscopically,  
57 temperature, time and liquid plasticizers have an enormous effect on the physical state and the quality of the  
58 product, particularly above the  $T_g$  value. In this critical temperature range the coexistence of different structures  
59 is present during the phase transition of a glassy solid structure into a rubbery one (Roos, 2010).

60 Gums, proteins, maltodextrins and all high molecular weight substances are characterized by a high  $T_g$ , while  
61 sugar and acids and all low molecular weight substances are recognised for their high hygroscopicity and low  
62  $T_g$  (Djendoubi Mrad et al., 2013; Fan and Roos, 2017; Roos, 2010; Roos and Potes, 2015). Even if sugars are  
63 in a crystalline form at room temperature, they are generally amorphous when in contact with water (Fan and  
64 Roos, 2017; Roos and Karel, 1991). The amorphous sugars are in a high-energy state compared with the  
65 corresponding crystalline form, being ‘metastable’ (Fan and Roos, 2017).

66 The molecular motion does not occur generally below the  $T_g$ , but it changes dramatically within the  $T_g$  range.  
67 Despite this, also in the glassy state, substances retain some degree of molecular mobility that is detectable in  
68 terms of mechanical relaxation. Changes that occur in the solid state are extremely slow, and they are often  
69 referred to rotational movement of glassy structures characterized by Maxwell-Wagner polarization effect  
70 (Perez-De Eulate and Cangialosi, 2018). Intact membrane interfaces can be determined based on impedance  
71 measurements in a frequency range of the so-called  $\beta$  dispersion (Ade-Omowaye et al., 2002).

72 In amorphous material, plasticizers may be considered as compounds that increase the free volume and  
73 therefore depress the  $T_g$ . The free volume of polymers has been discussed by (Sherrington, 2003 and Slade et  
74 al., 1993). These authors hypothesized that the modification of  $T_g$  is caused by the change in availability of  
75 ‘holes or places’ for molecular rearrangement (Sherrington, 2003). The material expansion occurs differently  
76 in the glassy or rubbery states, whereas the free volume is constant at the glass transition (Sherrington, 2003).  
77 Moreover, every phase transition is accompanied by volume modification and it can be observed also as  
78 relaxation process (Roos, 2010).

79 Considering these findings, it is possible to hypothesize that the structural modification induced by PEF can  
80 promote a  $T_g$  modification. Theoretically fruit and vegetable tissue pre-treated by PEF and then dehydrated  
81 becomes glassy and not crystalline, with a consequent limited volume reduction (Figueiredo et al., 1999). Other

82 novel technologies have been combined with drying giving the possibility to relate the effect of the treatment  
83 on the product structure (Aguilera et al., 2003). Chou and Chua (2001) presented methods to generate  
84 ultrasound inducing significant changes in food that reduce external and internal mass transfer during drying.  
85 Controlled ultrasound accelerates diffusion through membranes and changes in concentration and size of sugar  
86 crystals. Osmotic dehydration of foodstuff generally applied to fruits is worth to study in the understanding  
87 of the mass transport mechanisms. Fito (1994) reported the microstructural mechanism of vacuum pulsed  
88 osmotic dehydration in which mass transfer of water and solute are related to pressure gradients in open pores.  
89 A previous work of Castagnini et al. (2020) demonstrated that PEF and drying process enhance system mobility  
90 in terms of water holding capacity and mechanical properties of apple tissue.  
91 Although the effects induced by PEF and related reactions have been studied as a function of many parameters  
92 and even related to several aspects, no studies have been performed in terms of induced  $T_g$  and system meta-  
93 stability modifications.  
94 To our knowledge, this pioneering study is the first investigation on the effect on  $T_g$  induced by PEF on  
95 different vegetable tissues, along with other physical state modifications.

96

## 97 **2. Material and Methods**

### 98 **2.1 Material and samples preparation**

99 Potatoes (*Solanum tuberosum* var. Agata), apples (*Malus pumila* var. Granny Smith) and carrots (*Daucus*  
100 *carota* var. Amsterdam) were purchased from the local market in Cesena, Italy. The water content for fresh  
101 potato, apple and carrot was of  $5.73 \pm 0.64$ ,  $9.66 \pm 0.32$  and  $9.85 \pm 0.44$  kg water/kg dry weight respectively  
102 (measured using vacuum oven at 70°C, according to AOAC official method n. 972.20). Cylindrical samples  
103 ( $9 \pm 0.1$  mm diameter) were obtained using a manual cork borer and then reduced with a manual cutter to a  
104 length of  $10 \pm 0.1$  mm.

105

### 106 **2.2 Pulsed Electric Field (PEF) treatment**

107 Twenty cylindrical samples per treatment were placed in a 50 x 50 x 50 mm treatment chamber and filled up  
108 with tap water with the ratio of 1:2 (10g sample/20g water) having a conductivity of  $515 \pm 20 \mu\text{s cm}^{-1}$ , as  
109 measured by an electrical conductivity meter (mod. Basic 30, Crison Instrument, Spain). 100 near-rectangular

110 shaped pulses, with a fixed pulse width (10  $\mu$ s) and a pulse frequency of 100 Hz were applied in the present  
111 work. Three electric field strength (0.5, 1 and 1.5 kV.cm<sup>-1</sup>) were tested at 25° (chamber temperature was  
112 monitored before and after treatment). A control sample not subjected to PEF was also prepared. Preliminary  
113 experiment was performed to set the procedure according to previous research (Lebovka et al., 2007, 2002,  
114 2001, 2000). The voltage and current were registered by using a digital oscilloscope (PicoScope 2204a, Pico  
115 Technology, UK) connected to a personal computer.

116 The total treatment time was set to 1s, and the energy specific intake ( $W_T$ ) was calculated 0.121, 0.483 and  
117 1.086 kJ/kg for the samples treated at 0.5, 1 and 1.5 kV/cm respectively (Castagnini et al., 2020).

118 The  $W_T$  was calculated using the equation proposed by Raso et al. (2016) for rectangular pulse:

119

$$120 \quad W_T = \frac{n}{m} \int_0^{\infty} U(t) \cdot I(t) dt \quad \text{eq. 1}$$

121

122 n is the number of pulses applied, m is the mass of treated sample, U(t) is the voltage across the treatment  
123 chamber, and I(t) is the current through the treatment chamber.

124

### 125 **2.3 Air-drying**

126 All the samples were air-dried at 60 °C in a tray drier with transverse airflow, air velocity 2 m/s and an air  
127 renewal fee of 50%, until they reached a water activity typical of a vegetable dried product, of about 0.2 (mod.  
128 CLW 750 TOP + Pol-Eko- Aparatura SP.J., Poland). After drying, the moisture content of samples treated and  
129 untreated with PEF ranged for carrot between 0.26 to 0.13 kg water/kg dry weight, for apple between 0.48 to  
130 0.13 kg water/kg dry weight and potato between 0.31-0.21 kg water/kg dry weight.

131 After dehydration, all the samples were put in hygrometers and conditioned, as reported in the following section.

132

### 133 **2.4 Sample conditioning, moisture, and water activity determination**

134 All the samples were conditioned for about 30 days into a desiccator containing phosphorus pentoxide (P<sub>2</sub>O<sub>5</sub>)  
135 in order to dehydrate and stabilize the samples to the minimum water activity. Next, moisture equilibration  
136 took place at 22 °C inside six sterilised desiccators (Mod. DES 2000), each containing saturated salt solution  
137 in the range of 11-75% relative humidity. Each hydration experiment comprised a reference sample and  
138 samples treated at a specific field strength (0.5, 1 and 1.5 kV.cm<sup>-1</sup>).

139 Apple, carrot and potato samples inserted in open glass container were positioned in the hermetically closed  
140 desiccators containing, on the bottom, the different saturated salt solutions at the required  $a_w$  (0.11 LiCl; 0.22  
141  $\text{CH}_3\text{CO}_2\text{K}$ ; 0.33  $\text{MgCl}_2$ ; 0.44  $\text{K}_2\text{CO}_3$ ; 0.57 NaBr and 0.75 NaCl). Samples were periodically weighed after  
142 closing until they reached a constant weight for three consecutive weightings ( $\Delta w < 0.0005$  g).

143 The water activity of equilibrated sample was measured by a dewpoint hygrometer, mod. Aqualab (Decagon  
144 Devices Inc., Pullman, WA) and the water content percentages are expressed on a dry matter basis.

145

## 146 **2.5 Differential scanning calorimetry (DSC)**

147 The glass transition temperature ( $T_g$ ) was evaluated by a DSC mod. Q20 (TA Instrument, Germany). The DSC  
148 was equipped with a low-temperature cooling unit mod. System90 (TA Instrument, Germany). Heat flow was  
149 calibrated using the heat of fusion of indium ( $\Delta H$  28.71 J/g). For the calibration, the experimental heating rate  
150 was applied under a dry nitrogen gas flux of 50 mL/min. Each sample (about 15 mg) was weighed in 50  $\mu\text{l}$   
151 hermetic aluminium pans and then loaded onto the DSC instrument at room temperature, using an empty pan  
152 of the same type for reference. Samples were equilibrated at  $-40^\circ\text{C}$  for 10 min, and then two heat-cool cycles  
153 were applied from  $-60^\circ\text{C}$  to  $80^\circ\text{C}$  at  $5^\circ\text{C}/\text{min}$ . The  $T_g$  was evaluated by using the automatic tool of the Software  
154 TA-Universal analyser (TA Instrument, Germany).

155

## 156 **2.6 Glass transition model**

157 The Gordon and Taylor model (eq. 1) was applied to  $T_g$  vs  $X_s$  and  $X_w$  experimental data according to the  
158 following equation:

159

$$160 \quad T_g = \frac{T_{gs} X_s + k T_{gw} X_w}{X_s + k X_w} \quad (\text{eq.1})$$

161

162 Where  $T_g$ ,  $T_{gs}$  and  $T_{gw}$  are the glass transition temperatures respectively of the sample, solid and water,  $X_s$  and  
163  $X_w$  are the percentage of solid and water content in the sample on wet basis, and k is an empirical parameter.

164 The  $T_{gw}$  was taken as  $-135^\circ\text{C}$ , as average value of the literature data (from  $-125$  to  $-150^\circ\text{C}$ ) (Roos, 1995).

165

166 **2.7 Statistical analysis**

167 The experimental data were fitted with eq.1 by nonlinear regression using the Marquardt algorithm (Marquardt,  
168 1963) with MATLAB (MathWorks, Natick, USA). The algorithm calculates the set of parameters and their  
169 95% confidence interval. The goodness of the fit was checked by the estimation of the determination  
170 coefficient  $R^2$  and the associated average RMSE values.

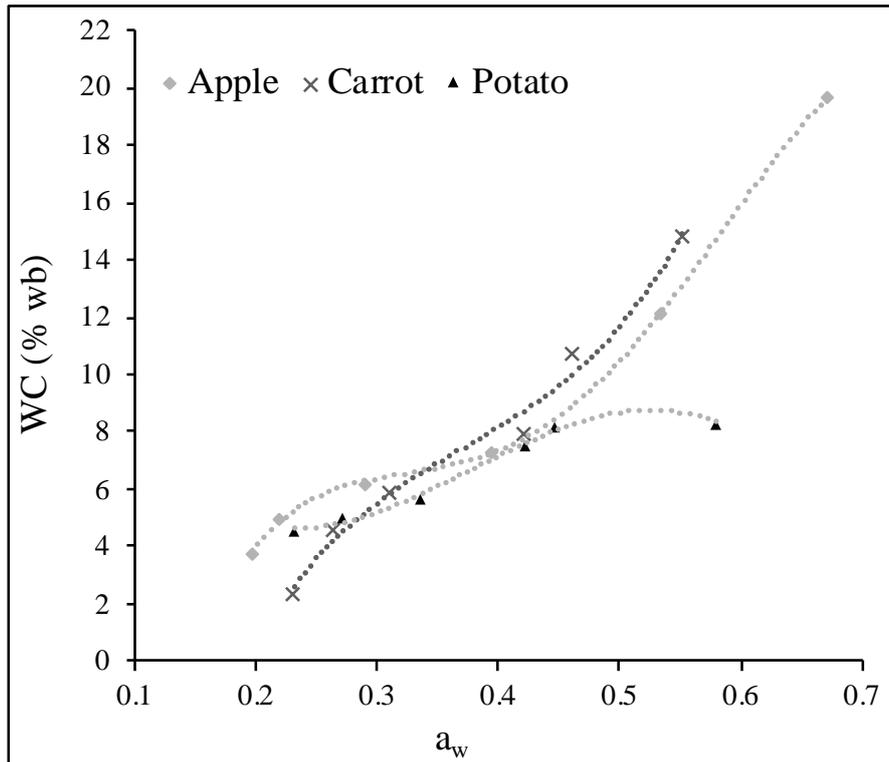
171 Within the same water activity level, significant differences between means of  $T_g^I$  and  $T_g^{II}$  affected by PEF  
172 treatments were analysed by means of ANOVA (analysis of variance, p-level < 0.05, post-hoc Tukey, Stat  
173 graphics Centurion XVI software).

174

175 **3. Results and discussion**

176 Average data of water content (WC) and water activity ( $a_w$ ) for control samples are shown in figure 1. As  
177 expected, the sorption behaviour for apple, carrot and potato was different. The shape of the moisture sorption  
178 isotherms is characterised by the state of the system constituents. Comparing the three curves, a higher water  
179 holding capacity for apple, than that of carrot and potato, is shown. The differences in crystal (e.g. starch,  
180 sugars and proteins) orientation and hydration affect the shape of sorption isotherms, producing different  
181 properties of the solid-state (Barbosa-Canovas et al., 2007).

182



183

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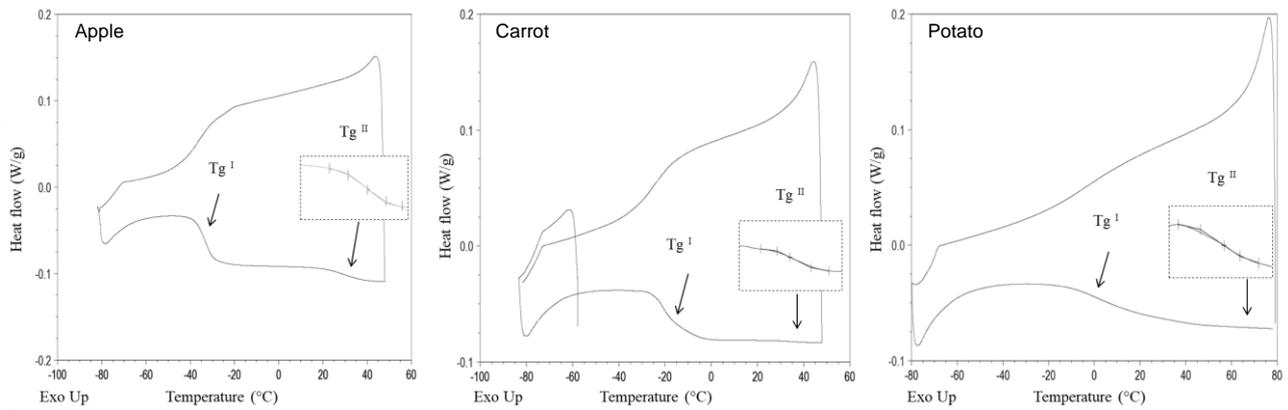
185 Figure 1. Water content (WC) expressed on wet basis and water activity ( $a_w$ ) relations of control samples  
 186 (apple; carrot; potato) at the different levels of hydration showing the sorption behaviour of the different  
 187 vegetables.

188

189 In figure 2, as an example, the obtained thermograms for apple, carrot and potato samples at 0.57  $a_w$  are  
 190 reported, evidencing the presence of two glass transition temperature,  $T_g^I$  and  $T_g^{II}$ . In these food matrixes, the  
 191  $T_g^I$  is probably mainly dependent from the presence of sugars, organic acids and a small amount of amino acids  
 192 while high molecular weight starch, fibres and small amount of proteins are associated to the  $T_g^{II}$  (Fan & Roos,  
 193 2017). The presence of two  $T_g$  has been previously reported in literature also for different fruit and vegetable,  
 194 such as dried papaya (Kurozawa et al., 2012), banana (Katekawa and Silva, 2008), plum skin (Telis et al.,  
 195 2006) and tomato (Telis and Sobral, 2002). The resolution of  $T_g^{II}$  was lower than  $T_g^I$  in the thermograms since  
 196 the accuracy of DSC to measure the  $T_g$  decrease as a less defined enthalpic change takes place in the  
 197 macromolecules (Shrestha et al., 2007). It is worth noting that the  $T_g^{II}$  is a phenomenon of less intensity and  
 198 slighter influenced by the hydration level than the  $T_g^I$ .

199 Phase transitions in low-moisture fruits and vegetables are dominated by sugars present in the material, within  
 200 the same moisture content (Figueiredo et al., 1999; Roos, 2010), and it is recognized that the presence of long

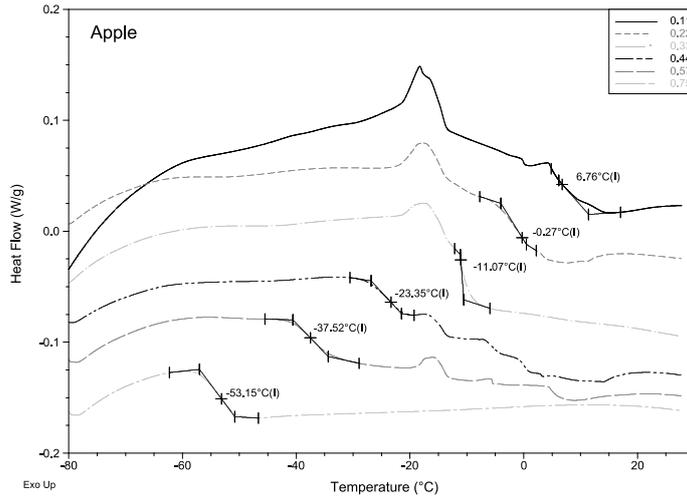
201 chain polymers, like starch, produces a  $T_g$  increase (Caballero-Cerón et al., 2018). Thus, the higher  $T_g$  values  
202 are attributable to high weight molecules. Confirming that, potato had the highest starch content than that of  
203 carrot and apple and it presented the highest  $T_g$  values, maintaining constant  $a_w$  level. Accordingly, the effect  
204 of PEF on  $T_g^{II}$  is mainly attributable to macromolecules interaction in the food matrix.  
205



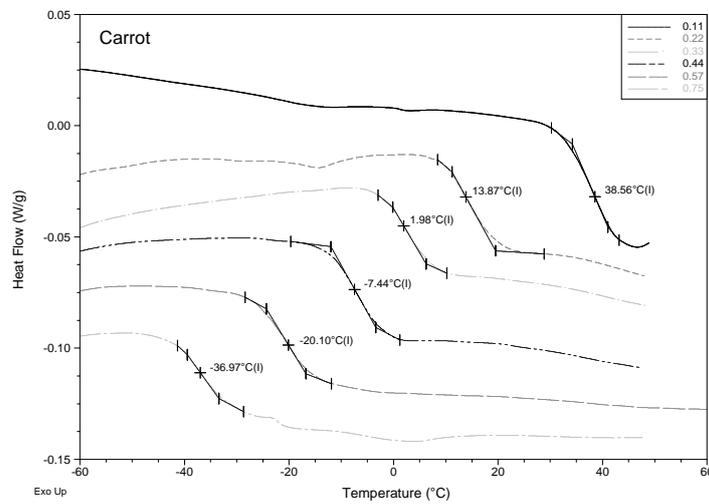
206  
207  
208 Figure 2. Thermograms of control samples at 0.57  $a_w$  of apple, carrot and potato showing the two-glass  
209 transition  $T_g^I$  and  $T_g^{II}$ .  
210

211 In the thermograms of the differently hydrated samples of control apple (A), carrot (C) and potato (P) reported  
212 in figure 3, the water content effect on  $T_g^I$  reduction is evident.

213



214



215

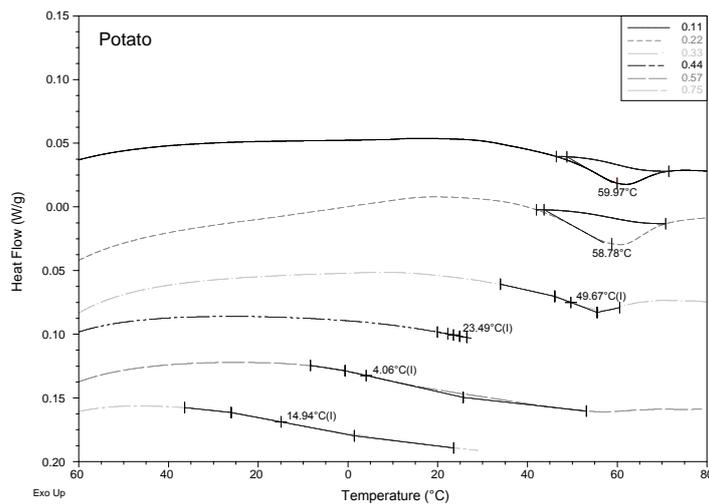
216

217 Figure 3. Thermograms of control apple, carrot, and potato samples at different hydration levels ( $a_w$  values)

218 where  $T_g^I$  values for each thermogram are reported.

219

220



221 As expected, conditioned samples show a hydration level decrease correspondent to a  $T_g^I$  raise, direct  
 222 consequence of water unavailability. Water results mainly involved and compartmentalised among different  
 223 structure (Roos, 2010). In table 1, the results of the obtained  $T_g^I$  and  $T_g^{II}$  values for control and PEF treated  
 224 samples are reported.

225

226 Table 1. Glass transition temperature ( $T_g^I$  and  $T_g^{II}$ ) values for control and PEF treated apple, carrot and potato  
 227 samples at different water activity ( $a_w$ ).

228

$a_w$	Voltage (kV)	Apple		Carrot		Potato	
		$T_g^I$	$T_g^{II}$	$T_g^I$	$T_g^{II}$	$T_g^I$	$T_g^{II}$
0.75	0	-53.79 ± 0.88 <sup>a</sup>	24.35 ± 0.12 <sup>c</sup>	-41.84 ± 0.22 <sup>a</sup>	28.65 ± 0.11 <sup>c</sup>	-8.52 ± 0.22 <sup>c</sup>	41.44 ± 0.19 <sup>c</sup>
	0.5	-52.98 ± 0.74 <sup>a</sup>	24.11 ± 0.37 <sup>bc</sup>	-41.85 ± 0.76 <sup>a</sup>	18.2 ± 0.82 <sup>b</sup>	-10.72 ± 0.34 <sup>b</sup>	42.69 ± 0.84 <sup>c</sup>
	1	-54.77 ± 0.45 <sup>a</sup>	22.73 ± 0.34 <sup>b</sup>	-37.37 ± 0.52 <sup>b</sup>	15.03 ± 0.43 <sup>a</sup>	-11.91 ± 0.56 <sup>b</sup>	38.37 ± 0.12 <sup>b</sup>
	1.5	-54.75 ± 0.43 <sup>a</sup>	21.11 ± 0.54 <sup>a</sup>	-39.14 ± 0.36 <sup>b</sup>	13.96 ± 0.51 <sup>a</sup>	-21.3 ± 0.11 <sup>a</sup>	34 ± 0.38 <sup>a</sup>
0.57	0	-35.37 ± 0.67 <sup>b</sup>	32.06 ± 0.64 <sup>c</sup>	-20.02 ± 0.67 <sup>b</sup>	38.49 ± 0.56 <sup>c</sup>	9.71 ± 0.23 <sup>a</sup>	60.95 ± 0.91 <sup>b</sup>
	0.5	-38.88 ± 0.5 <sup>a</sup>	30.03 ± 0.43 <sup>b</sup>	-14.17 ± 0.81 <sup>c</sup>	21.71 ± 0.09 <sup>b</sup>	12.03 ± 0.67 <sup>b</sup>	56.36 ± 0.95 <sup>ab</sup>
	1	-35.82 ± 0.67 <sup>b</sup>	28.1 ± 0.33 <sup>b</sup>	-22.43 ± 0.39 <sup>b</sup>	21.29 ± 0.32 <sup>b</sup>	11.75 ± 0.73 <sup>b</sup>	53.98 ± 0.66 <sup>ab</sup>
	1.5	-32.18 ± 0.68 <sup>c</sup>	11.32 ± 0.54 <sup>a</sup>	-30.2 ± 0.69 <sup>a</sup>	18.31 ± 0.21 <sup>a</sup>	9.12 ± 1.02 <sup>b</sup>	53.62 ± 0.99 <sup>a</sup>
0.44	0	-24.29 ± 0.14 <sup>a</sup>	38.7 ± 0.53 <sup>a</sup>	-11.33 ± 0.11 <sup>b</sup>	44.68 ± 0.35 <sup>d</sup>	16.56 ± 0.44 <sup>a</sup>	54.21 ± 0.17 <sup>c</sup>
	0.5	-24.34 ± 0.43 <sup>a</sup>	38.52 ± 0.62 <sup>a</sup>	-13.4 ± 0.65 <sup>a</sup>	41.29 ± 0.24 <sup>c</sup>	21.24 ± 0.23 <sup>b</sup>	47.67 ± 0.33 <sup>b</sup>
	1	-24.11 ± 0.25 <sup>a</sup>	37.78 ± 0.56 <sup>a</sup>	-12.38 ± 0.35 <sup>ab</sup>	38.36 ± 0.34 <sup>b</sup>	16.89 ± 0.34 <sup>a</sup>	42.07 ± 0.26 <sup>a</sup>
	1.5	-23.61 ± 0.91 <sup>a</sup>	37.09 ± 0.32 <sup>a</sup>	-12.4 ± 0.57 <sup>ab</sup>	37.23 ± 0.11 <sup>a</sup>	16.44 ± 0.47 <sup>a</sup>	42.43 ± 0.75 <sup>a</sup>
0.33	0	-12.07 ± 0.56 <sup>a</sup>	43.88 ± 0.26 <sup>c</sup>	-11.36 ± 0.41 <sup>a</sup>	44.39 ± 0.26 <sup>d</sup>	20.48 ± 0.55 <sup>a</sup>	55.3 ± 0.47 <sup>c</sup>
	0.5	-11.07 ± 0.68 <sup>ab</sup>	41.07 ± 0.25 <sup>b</sup>	1.24 ± 0.04 <sup>b</sup>	38.01 ± 0.51 <sup>c</sup>	21.13 ± 0.63 <sup>a</sup>	53.35 ± 0.5 <sup>c</sup>
	1	-10.22 ± 0.54 <sup>ab</sup>	40.91 ± 0.65 <sup>b</sup>	8.1 ± 0.91 <sup>c</sup>	35.73 ± 0.11 <sup>b</sup>	21.29 ± 0.45 <sup>a</sup>	47.25 ± 0.56 <sup>b</sup>
	1.5	-9.39 ± 0.45 <sup>b</sup>	36.93 ± 0.27 <sup>a</sup>	6.53 ± 0.34 <sup>c</sup>	33.53 ± 0.52 <sup>a</sup>	20.96 ± 0.41 <sup>a</sup>	43.29 ± 0.66 <sup>a</sup>
0.22	0	-0.8 ± 0.44 <sup>a</sup>	45.12 ± 0.52 <sup>a</sup>	13.36 ± 0.09 <sup>b</sup>	45.15 ± 0.56 <sup>c</sup>	43.36 ± 0.26 <sup>a</sup>	90.23 ± 0.71 <sup>b</sup>
	0.5	-0.87 ± 0.52 <sup>a</sup>	44.18 ± 0.76 <sup>a</sup>	6.75 ± 0.56 <sup>a</sup>	40.13 ± 0.77 <sup>b</sup>	50.01 ± 0.81 <sup>c</sup>	85.83 ± 0.28 <sup>a</sup>
	1	0.72 ± 0.16 <sup>a</sup>	43.45 ± 0.13 <sup>a</sup>	7.75 ± 0.44 <sup>a</sup>	37.64 ± 0.69 <sup>b</sup>	47.62 ± 0.33 <sup>b</sup>	86.12 ± 0.34 <sup>a</sup>
	1.5	-0.9 ± 0.46 <sup>a</sup>	43.37 ± 0.45 <sup>a</sup>	14.53 ± 0.51 <sup>b</sup>	31.04 ± 0.57 <sup>a</sup>	41.59 ± 0.56 <sup>a</sup>	85.81 ± 0.45 <sup>a</sup>
0.11	0	5.5 ± 0.56 <sup>a</sup>	51.1 ± 0.32 <sup>c</sup>	40.27 ± 0.63 <sup>d</sup>	64.89 ± 0.43 <sup>d</sup>	53.34 ± 0.58 <sup>b</sup>	86.69 ± 0.78 <sup>b</sup>
	0.5	11.2 ± 0.67 <sup>b</sup>	48.58 ± 0.37 <sup>b</sup>	32.87 ± 0.54 <sup>b</sup>	56.37 ± 0.22 <sup>c</sup>	49.17 ± 0.76 <sup>a</sup>	83.12 ± 0.63 <sup>a</sup>
	1	11.44 ± 0.77 <sup>b</sup>	47.18 ± 0.47 <sup>b</sup>	35.96 ± 0.22 <sup>c</sup>	51.59 ± 0.29 <sup>b</sup>	58.14 ± 0.21 <sup>c</sup>	83.06 ± 0.25 <sup>a</sup>
	1.5	9.35 ± 0.34 <sup>b</sup>	45.61 ± 0.41 <sup>a</sup>	15.03 ± 0.18 <sup>a</sup>	41.08 ± 0.37 <sup>a</sup>	51.14 ± 0.53 <sup>a</sup>	83.64 ± 0.01 <sup>a</sup>

229 Averaged values and ± standard deviation of three replicates. Statistics were carried out for each PEF treatment  
 230 and matrix at the same water activity. Different letters in the same column indicates statistical differences.

231

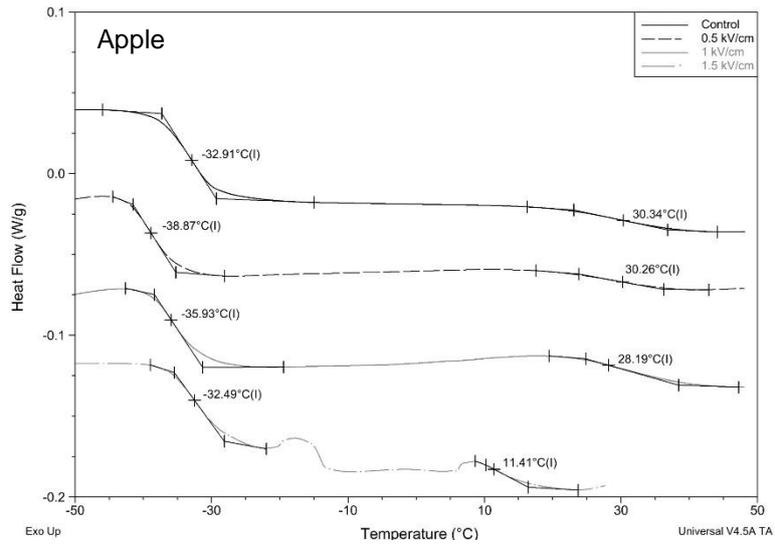
232 The plasticising effect of water on  $T_g^I$  was clearly observed for apple, carrot, and potato while less clear  
233 dependence was related to PEF treatment.  $T_g^{II}$  shifted to lower temperatures showing system mobility  
234 enhancement as a consequence of PEF treatment voltage increase, as also reported by Castagnini (et al. 2020).  
235  $T_g^{II}$  is mainly influenced by PEF treatment because related to big molecules, the main affected by PEF field  
236 strength.

237 Free-volume is one of the most recognized theory to fundamental explain glass transition (Roos, 2010).  
238 Accounting for free volume increase, also PEF treatment induces different long-lasting transition processes  
239 inside the tissue, such as mass transfer, moisture and air redistribution (Labuza & Hyman, 1998), or also  
240 resealing of cells (Knorr, 1999). The volume growth of the material occurs differently in the glassy and rubbery  
241 states, whereas the free volume is constant at the glass transition (Sherrington, 2003). In this way, it can be  
242 possible to hypothesise that PEF electroporation combined with thermal treatment improve glass transition  
243 mechanisms allowed to attain additional volume for the investigated vegetables. The resulting change of  
244 volume corresponds to structure modifications promotion, generating a high mobility degree and a consequent  
245  $T_g$  decrease.

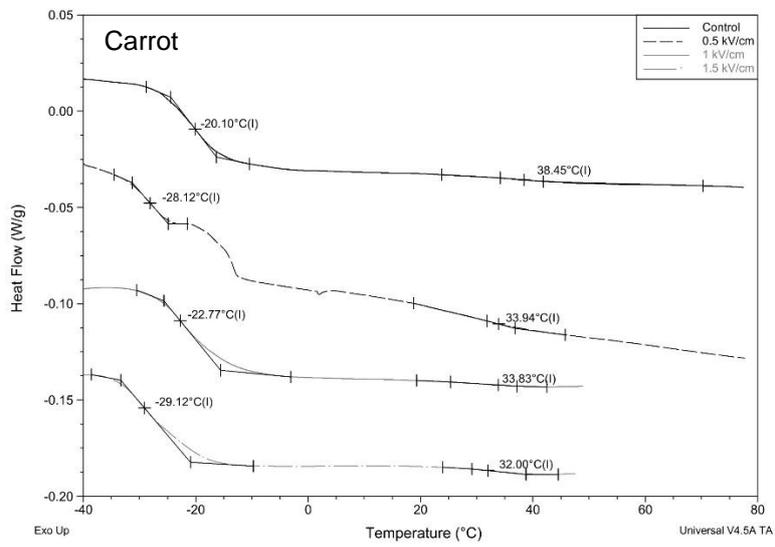
246 In Figure 4, the thermograms of control and PEF treated at the different strength field level (0.5, 1 and 1.5  
247 kV/cm) apple, carrot and potato samples are reported.

248 The most appreciable change for apple is related to the second event, with about 21°C of glass transition  
249 temperature shift. This shift can be attributable to high weight molecules, while  $T_g^I$  has a slighter temperature  
250 shift compared to  $T_g^{II}$ , considering PEF treatment. Therefore, it can be assumed that on apple samples, the  
251 increase of PEF level has a huge effect on macromolecules fraction, allowing a mobility raise due to their  
252 structural rearrangement and modification of water interactions. Differently, carrot and potato samples  
253 evidenced a glass transition shift of about 5-9°C and about 6-7°C for both  $T_g^I$  and  $T_g^{II}$ , respectively. As observed  
254 from sorption isotherm, also  $T_g$  data evidence structural differences among the product analysed, and  
255 consequently a different physical behaviour.

256



257



258

259

260 Figure 4. T<sub>g</sub><sup>I</sup> and T<sub>g</sub><sup>II</sup> of apple, carrot and potato at equilibrated 0.57 a<sub>w</sub>. Control and PEF treated samples  
261 showing a T<sub>g</sub> shift.

262

263

264 The  $T_g^I$  experimental data have been fitted with the Gordon Taylor model, and the estimated parameters are  
265 reported in Table 2. The Gordon and Taylor model show a good fitting for apple and carrot samples,  
266 considering the high coefficient of determination values (ranging from 0.866 to 0.997). The  $T_{gs}$  parameter  
267 represents the glass transition temperature of the solids, confirming a higher value compared to the moisturised  
268 samples. The k value is related to the thermal expansion coefficient and heat capacity of the components while  
269 changing from the glassy to the liquid state during the glass transition (Couchman and Karasz, 1978). At the  
270 applied energy levels, PEF treatment is non-destructive for the cell wall of considered plant materials  
271 (Dellarosa et al., 2018). In these conditions, it enables effective and selective diffusion of low molecular weight  
272 intracellular components even at low temperatures (Shrestha et al., 2007). Different authors (Han et al., 2009;  
273 Wang et al., 2015) demonstrated that PEF treatment could destroy the crystalline region of corn and potatoes  
274 starch resulting in the decrease of relative crystallinity. In these previous researches, higher field strengths  
275 have been applied compared with our PEF conditions. However, according to Caballero-Cerón et al. (2018),  
276 the  $T_g^{II}$  value in potato is mainly influenced by starch amount and conformation. **In this direction, the effect of**  
277 **PEF on  $T_g^{II}$  detected in our study was reasonably caused by changes in the interaction of starch with water and**  
278 **other surrounding molecules.**

279 After PEF treatment, during the heating process in DSC, water molecules could react more easily with the  
280 molecules in the crystalline region (Liu et al., 2018). As a consequence, PEF treatment can lead to a molecular  
281 rearrangement, decreasing the order of the structure (Liu et al., 2018). Pores induced by PEF produce a free  
282 volume gain, this additional space given by treatment affects structure and allows reorganisation, as previously  
283 reported also by Castagnini (et al. 2020).

284

285

286 Table 2. Results of fitting of  $T_g^1$  experimental data with Gordon Taylor models: model parameters ( $p$  level <  
 287 0.05), coefficient of determination ( $R^2$ ) and root mean square error (RMSE).  
 288

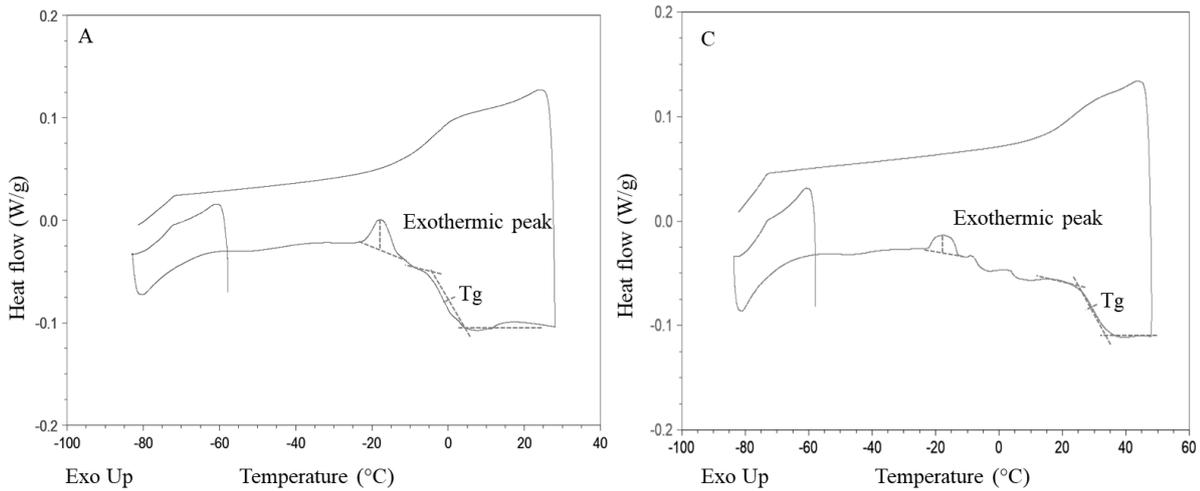
<b>Samples</b>	<b>Gordon Taylor parameters</b>	<b>Control</b>	<b>0.5 kV</b>	<b>1 kV</b>	<b>1.5 kV</b>
<b>Apple</b>	<b>T<sub>g</sub></b>	25.81 ± 7.88	37.80 ± 10.08	43.67 ± 6.45	37.58 ± 10.01
	<b>K</b>	4.46 ± 0.81	5.68 ± 1.12	7.11 ± 0.78	5.66 ± 1.07
	<b>R<sup>2</sup></b>	0.953	0.951	0.985	0.946
	<b>RMSE</b>	4.40	4.75	2.46	4.71
<b>Carrot</b>	<b>T<sub>g</sub></b>	59.00 ± 2.09	65.78 ± 11.73	33.47 ± 10.47	58.08 ± 5.53
	<b>K</b>	5.87 ± 0.22	6.87 ± 1.17	3.66 ± 1.02	6.56 ± 0.58
	<b>R<sup>2</sup></b>	0.997	0.963	0.866	0.988
	<b>RMSE</b>	1.18	4.34	5.89	2.13
<b>Potato</b>	<b>T<sub>g</sub></b>	169.56 ± 52.59	161.45 ± 38.69	179.50 ± 51.62	207.35 ± 85.44
	<b>K</b>	13.72 ± 5.02	12.83 ± 3.70	14.17 ± 4.84	17.11 ± 7.98
	<b>R<sup>2</sup></b>	0.932	0.955	0.943	0.917
	<b>RMSE</b>	8.10	6.84	8.27	9.78

289

290

291 Finally, it is important to underline that DSC analyses of apple and carrot samples evidenced the presence of  
 292 another thermal event, characterised by an exothermic peak at low hydration level (0.11<sub>a<sub>w</sub></sub>), as shown in figure  
 293 5.

294



295  
 296 Figure 5. Apple and carrot control samples thermograms at the lowest  $a_w$  0.11 shown an evident exothermic  
 297 peak just before  $T_g$  region imputable to structural rearrangement before phase transition.  
 298  
 299

300 Often, an overshoot area around  $T_g$  is measured and linked to molecular rearrangement, referred to as  $\beta$   
 301 relaxation (Ade-Omowaye et al., 2002). According to Johari (1976),  $\beta$  relaxation is driven by diffusion of free  
 302 volume at particles interfaces. Only low moisture hydration levels are characterized by relaxation enthalpy  
 303 peak confirming to be a peculiar process of glassy structures. In this research, this phenomenon was nor  
 304 quantified neither considered for PEF treated samples, but for the first time its presence is evidenced on apple  
 305 and carrot tissues at low hydration levels.  $\beta$  relaxation phenomenon has to be confirmed by dielectric  
 306 techniques, modulated DSC and mechanical measurements, to effort results in order to achieve additional  
 307 information on product metastability.  
 308  
 309

### 310 Conclusions

311 Apple, carrot and potato samples were treated with Pulsed Electric Field at increasing filed strength and then  
 312 dried. The Differential Scanning Calorimetric revealed the existence of two stepwise changes in the heat  
 313 capacity at the phase transition, attributable to two  $T_g$  ( $T_g^I$  and  $T_g^{II}$ ). Apple, carrot and potato PEF treated  
 314 samples compared to the untreated ones showed a  $T_g$  shift to lower temperatures, confirming the mobility raise  
 315 of the system after PEF treatment. The most appreciable change for apple was related to the second event, with

316 about 21°C of  $T_g^{II}$  shift, while carrot and potato were characterised by a on  $T_g^I$  shift of about 7-9°C and 2-7°C,  
317 respectively.

318 Another event visible for apple and carrot was an overshoot area around  $T_g$ , linked to molecular rearrangement,  
319 reasonably ascribable to  $\beta$  relaxation phenomenon.

320 Data shows some interesting aspects of structural modifications promoted by PEF pre-treatment to drying, also  
321 opening a lot of question to be deeply understand in future researches. Some structural modification  
322 phenomena promoted by the synergistic interactions of PEF and drying are here showed for the first time.

323 PEF treatment is often used to reduce drying time, but no information's are available on the structural changes  
324 of the product as a consequence of the combination of these treatments. It is worth noting that the quality  
325 characteristics of the dried product, particularly crispness, are strictly related to the relation between  $T_g$  and  
326 storage temperature. Particularly, at critical water activity, when the product  $T_g$  corresponds to the product  
327 temperature, a crispy dried vegetable product can become soft and plastic, losing its peculiar quality  
328 parameters. The  $T_g^{II}$  can be used to define the final drying water activity, because PEF treatment, influencing  
329 this value, can enhance mobility and compromise the stability of the system. In this way,  $T_g^{II}$  shift induced by  
330 PEF has to be carefully considered in storage and packaging conditions management.

331

### 332 **Acknowledgements**

333 The authors are thankful to Ministero degli Affari Esteri e della Cooperazione Internazionale (Italy) for  
334 financial support. J. M. Castagnini had a scholarship from Study in Italy program.

335

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**New insights into the glass transition of dried fruits and vegetables and effect of pulsed electric field treatment**

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**Declaration of interests**

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

The authors declare the following financial interests/personal relationships which may be considered as potential competing interests:

### Credit Author Statement

All the authors contributed in the same way to the development of the research presented in this article.