## **Supporting Information for**

# An electrochemically controlled supramolecular zip tie based on host-guest chemistry of CB[8]

Iago Neira,<sup>a</sup> Olaya Domarco,<sup>a</sup> Jose L. Barriada,<sup>a</sup> Paola Franchi,<sup>b</sup> Marco Lucarini,<sup>b</sup> Marcos D. García<sup>\*a</sup> and Carlos Peinador<sup>\*a</sup>

<sup>a</sup>Departamento de Química and Centro de Investigaciones, Científicas Avanzadas (CICA), Universidade da Coruña, Facultad de Ciencias, E-15071 A Coruña, Spain.

<sup>b</sup>Department of Chemistry "G. Ciamician" – Alma Mater Studiorum-University of Bologna, Via San Giacomo 11, Bologna, Italy

General	3
Synthetic procedures:	3
Synthesis and characterization of 2-((4-(bromomethyl)benzyl)oxy)naphthalene (4).	3
Synthesis and characterization of 1-(4-((naphthalen-2-yloxy)methyl)benzyl)-[4,4'-bipyridin]- 1-ium bromide <b>(5·Br)</b>	4
Synthesis and characterization of ((oxybis(ethane-2,1-diyl))bis(oxy))bis(ethane-2,1-diyl) bis(4-methylbenzenesulfonate) (6).	4
Synthesis and characterization of 1-bromo-2-(2-(2-(2-bromoethoxy)ethoxy)ethoxy)ethane (7).	5
Synthesis and characterization of 1-methyl-[4,4'-bipyridin]-1-ium hexafluorophosphate (8)	5
Synthesis and characterization of $1-(2-(2-(2-bromoethoxy)ethoxy)ethoxy)ethyl)-1'-(4-((naphthalen-2-yloxy)methyl)benzyl)-[4,4'-bipyridine]-1,1'-diium hexafluorophosphate (9-2PF6).$	6
Synthesis and characterization of 1-methyl-1'-(2-(2-(2-(1'-(4-((naphthalen-2- yloxy)methyl)benzyl)-[4,4'-bipyridin]-1,1'-diium-1-yl)ethoxy)ethoxy)ethoxy)ethyl)-[4,4'- bipyridine]-1,1'-diium hexafluorophosphate <b>(1·4PF<sub>6</sub>)</b>	7
Synthesis and characterization of 3,6,9,12,15-pentaoxaheptadecane-1,17-diyl bis(4- methylbenzenesulfonate) <b>(10)</b> .	8
Synthesis and characterization of 1,17-dibromo-3,6,9,12,15-pentaoxaheptadecane (11)	8
Synthesis and characterization of 1-(17-bromo-3,6,9,12,15-pentaoxaheptadecyl)-1'-(4- ((naphthalen-2-yloxy)methyl)benzyl)-[4,4'-bipyridine]-1,1'-diium hexafluorophosphate (12·2PF <sub>6</sub> )	9
Synthesis and characterization of 1-methyl-1'-(17-(1'-(4-((naphthalen-2- yloxy)methyl)benzyl)-[4,4'-bipyridin]-1,1'-diium-1-yl)-3,6,9,12,15-pentaoxaheptadecyl)-[4,4'- bipyridine]-1,1'-diium <b>(2·4PF<sub>6</sub>)</b>	10
Synthesis and characterization of 3,6,9,12,15,18,21-heptaoxatricosane-1,23-diyl bis(4- methylbenzenesulfonate) <b>(13)</b> .	11
Synthesis and characterization of 1,23-dibromo-3,6,9,12,15,18,21-heptaoxatricosane (14)	11
Synthesis and characterization of 1-(23-bromo-3,6,9,12,15,18,21-heptaoxatricosyl)-1'-(4- ((naphthalen-2-yloxy)methyl)benzyl)-[4,4'-bipyridine]-1,1'-diium hexafluorophosphate <b>(15·2PF<sub>6</sub>)</b>	12
Synthesis and characterization of 1-methyl-1'-(23-(1'-(4-((naphthalen-2- yloxy)methyl)benzyl)-[4,4'-bipyridin]-1,1'-diium-1-yl)-3,6,9,12,15,18,21-heptaoxatricosyl)- [4,4'-bipyridine]-1,1'-diium hexafluorophosphate ( <b>3·4PF<sub>6</sub>)</b> . 13	
Synthesis and characterization of 1⊂CB[8]	14
Synthesis and characterization of <b>2</b> ⊂CB[8]	15
Synthesis and characterization of <b>3</b> ⊂CB[8]	16
NMR spectra	30

#### General

The chemicals used in this work were purchased from commercial suppliers and used without further purification. Compound **8**·PF<sub>6</sub> was prepared using a modified procedure.<sup>1</sup> The purity of the CB[8] was assessed as previously reported by Kaifer *et al.*<sup>2</sup> Milli-Q water was purified with a Millipore Gradient A10 apparatus. Merck 60 F254 foils were used for thin layer chromatography, and Merck 60 (230-400 mesh) silica gel was used for flash chromatography. NMR spectra were recorder on a Bruker Advance 400 or 500 MHz for <sup>1</sup>H, and 100 or 125 MHz for <sup>13</sup>C, equipped each other with a dual cryoprobe. The solvent for NMR experiments was deuterium oxide (D<sub>2</sub>O), methanol (CD<sub>3</sub>OD) and acetonitrile (CD<sub>3</sub>CN). Mass spectrometry experiments were carried out in a LCQ-q-TOF Applied Biosystems QSTAR Elite spectrometer for low and high resolution ESI. UV/Vis spectra were recorded on a Jasco V-650 spectrometer. Microwave-assisted reactions were carried out in an Anton Paar Monowave 300 reactor operating at 2455 MHz in a sealed reaction vial using microwave power between 0-850 W. The samples were irradiated with the microwave power necessary for reaching the temperature of 150 °C heating with a "as fast as possible" protocol. The reaction mixture temperature was monitored via built-in IR sensor.

*EPR measurements.* Compounds  $1^{4+}$ -  $3^{4+}$  were added to a water/CH<sub>3</sub>CN (3/2, v/v) solution saturated with Na<sub>2</sub>S<sub>2</sub>O<sub>4</sub>. The samples under nitrogen were immediately sealed in a capillary EPR tube. The EPR spectra were recorded on a Bruker ESP300 spectrometer equipped with an NMR gaussmeter for field calibration and Bruker ER033M field-frequency lock. The instrument settings were as follows: microwave power 0.79 mW, modulation amplitude 1.0 - 0.2 G, modulation frequency 100 kHz, scan time 180 s, 2 K data points.

### Synthetic procedures:

Synthesis and characterization of 2-((4-(bromomethyl)benzyl)oxy)naphthalene (4).



To a solution of 2-hydroxynaphthalene (1.4 g, 9.8 mmol) and K<sub>2</sub>CO<sub>3</sub> (6.8 g, 50 mmol) in 60 mL of acetone  $\alpha, \alpha'$ -dibromo-*p*-xylene (2.0 g, 7.6 mmol) was added. The mixture was left under stirring at room temperature for 18 hours. After stirring for 20 hours at room temperature, the resulting mixture was filtered off and purified by column chromatography (SiO<sub>2</sub>, EtOAc:Hex 1:7) to give **4** as a

<sup>&</sup>lt;sup>1</sup> J. M. Weber, M. T. Rawls, V. J. MacKenzie, B. R. Limoges, C. M. Elliott, J. Am. Chem. Soc. 2007, **129**, 313.

<sup>&</sup>lt;sup>2</sup> S. Yi and A. E. Kaifer, *J. Org. Chem.* 2011, **76**, 10275.

white solid (0.6 g, 23%). <sup>1</sup>*H NMR* (500 *MHz*, *CDCl*<sub>3</sub>)  $\delta$ : 7.77 (m, 2H), 7.72 (d, *J* = 8.7 Hz, 1H) 7.43 (m, 5H), 7.32 (t, *J* = 7.48 Hz, 1H), 7.21 (m, 2H), 5.17 (s, 2H), 4.51 (s, 2H) ppm. <sup>13</sup>*C NMR* (125 *MHz*, *CDCl*<sub>3</sub>)  $\delta$ : 156.73 (C), 137.69 (C), 137.40 (C), 134.60 (C), 129.67 (CH), 129.48 (C), 129.26 (CH), 128.05 (CH), 127.81 (CH), 126.94 (CH), 126.57 (CH), 123.93 (CH), 119.17 (CH), 107.30 (CH), 69.69 (CH<sub>2</sub>), 33.30 (CH<sub>2</sub>) ppm. *HRMS-ESI* (*m*/*z*): calcd. [M]<sup>+</sup>: 247.1117, found 247.1115.

Synthesis and characterization of 1-(4-((naphthalen-2-yloxy)methyl)benzyl)-[4,4'-bipyridin]-1-ium bromide **(5·Br)**.



To a solution of **4** (470 mg, 1.40 mmol) in 50 mL of CH<sub>3</sub>CN, 4,4'-bipyridine was added (560 mg, 3.60 mmol) and the mixture was left under reflux. After for 4 hours the mixture was cooled down to reach room temperature and 200 mL of Et<sub>2</sub>O was added. The precipitate formed was filtered off and washed with 25 mL of toluene to give **5**·Br (480 mg, 70%) as a yellow solid. <sup>1</sup>*H NMR* (500 *MHz*, *CD*<sub>3</sub>*OD*)  $\delta$ : 9.18 (d, *J* = 7.0 Hz, 2H), 8.83 (d, *J* = 6.3 Hz, 2H), 8.52 (d, *J* = 7.0 Hz, 2H), 7.97 (d, *J* = 6.3 Hz, 2H), 7.77 (d, *J* = 8.9 Hz, 2H), 7.72 (d, *J* = 7.3 Hz, 1H), 7.66 (d, *J* = 8.1 Hz, 2H), 7.60 (d, *J* = 8.3 Hz, 2H), 7.41 (ddd, *J* = 8.1, 6.8, 1.3 Hz, 1H), 7.36 – 7.27 (m, 2H), 7.20 (dd, *J* = 9.0, 2.6 Hz, 1H), 5.92 (s, 2H), 5.27 (s, 2H) ppm. <sup>13</sup>*C NMR* (*125 MHz*, *CD*<sub>3</sub>*OD*)  $\delta$ : 156.4 (C), 154.22 (C), 150.41 (CH), 145.10 (CH), 142.12 (C), 139.63 (C), 134.63 (C), 132.59 (C), 129.10 (CH), 128.99 (CH), 128.35 (CH), 127.19 (CH), 126.42 (CH), 126.01 (CH), 125.98 (CH), 123.42 (CH), 122.14 (CH), 118.36 (CH), 106.97 (CH), 68.74 (CH<sub>2</sub>), 63.73 (CH<sub>2</sub>) ppm. *HRMS-ESI* (*m/z*): calcd: [C<sub>28</sub>H<sub>23</sub>N<sub>2</sub>O]<sup>+</sup>: 403.1804, found 403.1807.

Finally, **5**·Br was dissolved in 100 mL of MeOH and KPF<sub>6</sub> was added until no more precipitation was observed. The MeOH was removed under reduced pressure to leave a yellow crude. The crude was suspended into 100 mL of H<sub>2</sub>O and the yellow powder was filtered off to afford **5**·PF<sub>6</sub> (540 mg, 98%). <sup>1</sup>*H NMR* (400 MHz, CD<sub>3</sub>CN)  $\delta$ : 8.82 (d, *J* = 6.3 Hz, 4H), 8.29 (d, *J* = 7.0 Hz, 2H), 7.80 (d, *J* = 9.3 Hz, 2H), 7.78 – 7.73 (m, 3H), 7.61 (d, *J* = 8.3 Hz, 2H), 7.49 (d, *J* = 8.3 Hz, 2H), 7.46 – 7.42 (m, 1H), 7.38 – 7.30 (m, 2H), 7.20 (dd, *J* = 9.0, 2.6 Hz, 1H), 5.74 (s, 2H), 5.24 (s, 2H) ppm. <sup>13</sup>*C NMR* (100 MHz, CD<sub>3</sub>CN)  $\delta$ : 157.07 (C), 155.24 (C), 151.76 (CH), 145.53 (CH), 141.72 (C), 139.93 (C), 135.17 (C), 133.13 (C), 130.10 (CH), 129.95 (CH), 129.27 (CH), 128.16 (CH), 127.26 (CH), 127.18 (CH), 126.89 (CH), 124.49 (CH), 122.44 (CH), 119.33 (CH), 107.94 (CH), 69.61 (CH<sub>2</sub>), 64.44 (CH<sub>2</sub>). *HRMS-ESI (m/z*): calcd. [C<sub>28</sub>H<sub>23</sub>N<sub>2</sub>O]<sup>+</sup>: 403.1804, found 403.1807.

Synthesis and characterization of ((oxybis(ethane-2,1-diyl))bis(oxy))bis(ethane-2,1-diyl) bis(4-methylbenzenesulfonate) **(6)**.



A solution of 2,2'-((oxybis(ethane-2,1-diyl))bis(oxy))bis(ethan-1-ol) (5.0 g, 25,7 mmol) and 4methylbenzenesulfonyl chloride (14.7 g, 77.2 mmol) in 100 mL of THF was stirred for 15 minutes at 0°C in an ice bath. Then, a solution of KOH (10.1 g, 180.2 mmol) in 25 mL of H<sub>2</sub>O was added dropwise for 1 hour. After the complete addition, the solution was left 2 h under stirring at room temperature. Consecutively, a mixture of 150 mL H<sub>2</sub>O/Et<sub>2</sub>O (1:3) was added and the organic phase was separated from the aqueous phase. The aqueous phase was extracted again with Et<sub>2</sub>O (3 × 75 mL). The productcontaining organic fractions were combined and washed with 200 mL of a saturate solution of NH<sub>4</sub>Cl and 200 mL of H<sub>2</sub>O. Finally, the organic phase was dried with MgSO<sub>4</sub> and the solvent was removed under reduced pressure to afford **6** as a yellow oil (12.4 g, 96%). <sup>1</sup>*H NMR* (500 *MHz*, *CDCl<sub>3</sub>*)  $\delta$ : 7.79 (d, *J* = 8.4 Hz, 4H), 7.34 (d, *J* = 8.5 Hz, 4H), 4.15 (t, *J* = 5.4, 4.5 Hz, 4H), 3.67 (t, *J* = 5.4, 4.5 Hz, 4H), 3.56 (m, 8H), 2.44 (s, 3H) ppm. <sup>13</sup>*C NMR* (125 *MHz*, *CDCl<sub>3</sub>*)  $\delta$ : 144.96 (C), 132.91 (C), 129.97 (CH), 128.12 (CH), 70.89 (CH<sub>2</sub>), 70.70 (CH<sub>2</sub>), 69.39 (CH<sub>2</sub>), 68.84 (CH<sub>2</sub>), 21.79 (CH<sub>3</sub>) ppm. *HRMS-ESI* (*m/z*): calcd. [C<sub>22</sub>H<sub>30</sub>O<sub>9</sub>S<sub>2</sub> + Na<sup>+</sup>]<sup>+</sup>: 525.1223, found 525.1202.

Synthesis and characterization of 1-bromo-2-(2-(2-(2-bromoethoxy)ethoxy)ethoxy)ethane (7).



A solution of **6** (10.0 g, 19.8 mmol) and LiBr (6.9 g, 79.5 mmol) in 100 mL of acetone was left under reflux for 18 hours. Then, the solvent was removed under reduced pressure and the crude was dissolved in 200 mL of a mixture of  $H_2O/Et_2O$  (1:1). The organic phase was separated from the aqueous phase and this one was extracted again with  $Et_2O$  (3 × 75 mL). The product-containing organic fractions were combined, dried with MgSO<sub>4</sub> and the solvent was removed under reduced pressure to afford **7** as a yellow oil (6.1 g, 95%). <sup>1</sup>*H* NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$ : 3.79 (t, *J* = 6.3 Hz, 4H), 3.65 (d, *J* = 1.1 Hz, 8H), 3.45 (t, *J* = 6.3 Hz, 4H) ppm. <sup>13</sup>*C* NMR (125 MHz, CDCl<sub>3</sub>)  $\delta$ : 71.1 (CH<sub>2</sub>), 70.64 (CH<sub>2</sub>), 70.53 (CH<sub>2</sub>), 30.36 (CH<sub>2</sub>) ppm. HRMS-ESI (m/z): calcd. [C<sub>8</sub>H<sub>17</sub>Br<sub>2</sub>O<sub>3</sub>]<sup>+</sup>: 318.9538, found 318.9536.

Synthesis and characterization of 1-methyl-[4,4'-bipyridin]-1-ium hexafluorophosphate (8).



A solution of 1-(2,4-dinitrophenyl)-[4,4'-bipyridin]-1-ium<sup>3</sup> (3.2 g, 9 mmol) and methanamine (1.7 mL, 19.8 mmol) in a mixture 100 mL EtOH and 30 mL of  $H_2O$  was stirring at room temperature for 3 hours. Then, the solvent was removed under reduced pressure and the crude was dissolved in 200 mL of a mixture of  $H_2O$ /EtOAc (1:1). The organic phase was separated from the aqueous phase and this one was extracted again with EtOAc (3 × 75 mL). The product-containing organic fractions were

<sup>&</sup>lt;sup>3</sup> D. Bongard, M. Möller, S. Nagaraja Rao, D. Corr, and L. Walder, Helv. Chim. Acta, 2005, 88, 3200-3209.

combined and KPF<sub>6</sub> was added until no more precipitation was observed. The solid was filtered off and washed with 50 mL of H<sub>2</sub>O to leave **8**·PF<sub>6</sub> (1.64 g, 58%) as a brown solid. <sup>1</sup>*H NMR* (500 *MHz*, *CD*<sub>3</sub>*CN*)  $\delta$ : 8.85 (d, *J* = 5.8 Hz, 2H), 8.71 (d, *J* = 7.7 Hz, 2H), 8.29 (d, *J* = 6.4 Hz, 2H), 7.79 (d, *J* = 6.2 Hz, 2H), 4.33 (s, 3H) ppm. <sup>13</sup>*C NMR* (125 *MHz*, *CD*<sub>3</sub>*CN*)  $\delta$ : 155.14 (C), 152.48 (CH), 147.10 (CH), 142.50 (C), 126.68 (CH), 123.10 (CH), 49.23 (CH<sub>3</sub>) ppm. *HRMS-ESI* (*m*/*z*): calcd. [C<sub>11</sub>H<sub>11</sub>N<sub>2</sub>]<sup>+</sup>: 171.0916, found 171.0917.

Synthesis and characterization of 1-(2-(2-(2-(2-bromoethoxy)ethoxy)ethoxy)ethyl)-1'-(4-((naphthalen-2-yloxy)methyl)benzyl)-[4,4'-bipyridine]-1,1'-diium hexafluorophosphate (9·2PF<sub>6</sub>).



A solution of  $5 \cdot PF_6$  (0.48 g, 0.88 mmol) and 7 (1.60 g, 4.44 mmol) in 50 mL of CH<sub>3</sub>CN was left under reflux for 48 hours. Then, the solvent was removed under reduced pressure to leave a solid residue, which was subjected to flash chromatography (SiO<sub>2</sub>, CH<sub>3</sub>CN/NaCl (0.6 M)/MeOH 4/1/1). The productcontaining fractions were combined and the solvents evaporated. The residue was suspended in EtOH and filtered off to remove NaCl. The EtOH was removed under reduced pressure to afford 9.2Cl as a yellow oil. The yellow oil was dissolved in the minimal amount of H<sub>2</sub>O and KPF<sub>6</sub> was added until no more precipitation was observed. The solid was filtered off and washed with 50 mL of  $H_2O$  to leave **9**·2PF<sub>6</sub> (155 mg, 19%) as an orange oil. <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>CN)  $\delta$ : 8.94 (d, J = 6.9 Hz, 2H), 8.90 (d, J = 7.0 Hz, 2H), 8.32 (dd, J = 9.0, 7.0 Hz, 4H), 7.75 (d, J = 9.4 Hz, 2H), 7.71 (dd, J = 8.3, 1.1 Hz, 1H), 7.59 (d, J = 8.5 Hz, 2H), 7.50 (d, J = 8.3 Hz, 2H), 7.40 (ddd, J = 8.2, 6.8, 1.3 Hz, 1H), 7.33 - 7.27 (m, 2H), 7.17 (dd, J = 9.0, 2.6 Hz, 1H), 5.79 (s, 2H), 5.21 (s, 2H), 4.76 - 4.69 (m, 2H), 3.97 - 3.91 (m, 2H), 3.69 (dd, J = 6.1, 5.3 Hz, 2H), 3.60 – 3.56 (m, 2H), 3.56 – 3.51 (m, 2H), 3.51 – 3.46 (m, 4H), 3.43 (dd, J = 6.1, 5.2 Hz, 2H) ppm. <sup>13</sup>C NMR (100 MHz, CD<sub>3</sub>CN) δ: 157.02 (C), 150.99 (C), 150.53 (C), 146.83 (CH), 146.13 (CH), 140.04 (C), 135.12 (C), 132.80 (C), 130.12 (CH), 130.07 (CH), 129.61 (C), 129.28 (CH), 128.13 (CH), 128.05 (CH), 127.32 (CH), 127.16 (CH), 127.16 (CH), 124.46 (CH), 119.32 (CH), 107.89 (CH), 71.25 (CH<sub>2</sub>), 70.88 (CH<sub>2</sub>), 70.54 (CH<sub>2</sub>), 70.53 (CH<sub>2</sub>), 70.49 (CH<sub>2</sub>), 69.56 (CH<sub>2</sub>), 69.11 (CH<sub>2</sub>), 65.02 (CH<sub>2</sub>), 62.23 (CH<sub>2</sub>), 32.23 (CH<sub>2</sub>) ppm. *HRMS-ESI (m/z*): calcd. [C<sub>36</sub>H<sub>39</sub>BrN<sub>2</sub>O<sub>4</sub>F<sub>6</sub>P]<sup>+</sup>: 787.1729, found 787.1749; calcd.  $[C_{36}H_{39}BrN_2O_4]^{+2}$ : 321.1041 , found 321.1050.

Synthesis and characterization of 1-methyl-1'-(2-(2-(2-(2-(1'-(4-((naphthalen-2-yloxy)methyl)benzyl)-[4,4'-bipyridin]-1,1'-diium-1-yl)ethoxy)ethoxy)ethoxy)ethyl)-[4,4'-bipyridine]-1,1'-diium hexafluorophosphate (1·4PF<sub>6</sub>).



A mixture of  $9.2PF_6$  (155 mg, 0.16 mmol) and  $8.PF_6$  (525 mg, 1.6 mmol) in 7 mL of CH<sub>3</sub>CN was heated up to 110 °C for 6 hours using microwave-assisted heating. Then, the solvent was removed under reduced pressure to leave a solid residue, which was subjected to flash chromatography (SiO<sub>2</sub>) using two different eluent phases: CH<sub>3</sub>CN/NaCl (0.6 M)/MeOH 4/1/1) to remove impurities and  $CH_3CN/KPF_6$  (0.6 M)/MeOH 4/1/1) to elute the compound. The product-containing fractions were combined and the solvents evaporated. The obtained residue was then suspended in 100 mL H<sub>2</sub>O and filtered off to remove excess KPF<sub>6</sub> and dissolved in CH<sub>3</sub>CN. Finally, the CH<sub>3</sub>CN was removed under reduced pressure to leave  $1.4PF_6$  as a brown oil (51 mg, 23%). <sup>1</sup>H NMR (500 MHz, CD<sub>3</sub>CN)  $\delta$ : 9.01 (d, J = 6.9 Hz, 2H), 8.91 (d, J = 5.7 Hz, 4H), 8.85 (d, J = 6.8 Hz, 2H), 8.41 - 8.34 (m, 8H), 7.81 (d, J = 8.8 Hz, 2H), 7.76 (d, J = 8.9 Hz, 1H), 7.64 (d, J = 8.2 Hz, 2H), 7.57 – 7.52 (m, 2H), 7.46 (ddd, J = 8.2, 6.7, 1.2 Hz, 1H), 7.39 – 7.32 (m, 2H), 7.21 (dd, J = 9.0, 2.6 Hz, 1H), 5.84 (s, 2H), 5.26 (s, 2H), 4.76 (t, J = 4.9, 4.5 Hz, 4H), 4.40 (s, 3H), 3.95 (t, J = 4.6 Hz, 4H), 3.64 – 3.55 (m, 4H), 3.55 – 3.48 (m, 4H) ppm. <sup>13</sup>C NMR (125 *MHz, CD*<sub>3</sub>*CN*) δ: 157.45 (C), 151.42 (C), 151.15 (C), 151.11 (C), 150.58 (C), 147.49 (CH), 147.15 (CH), 147.15 (CH), 146.61 (CH), 140.47 (C), 135.56 (C), 133.26 (C), 130.54 (CH), 130.51 (CH), 130.06 (C), 129.73 (CH), 128.57 (CH), 128.49 (CH), 127.81 (CH), 127.79 (CH), 127.72 (CH), 127.68 (CH), 127.60 (CH), 124.91 (CH), 119.73 (CH), 108.32 (CH), 71.28 (CH<sub>2</sub>), 70.85 (CH<sub>2</sub>), 70.00 (CH<sub>2</sub>), 69.55 (CH<sub>2</sub>), 65.46 (CH<sub>2</sub>), 62.67 (CH<sub>2</sub>), 49.59 (CH<sub>3</sub>) ppm. *HRMS-ESI (m/z*): calcd. [C<sub>47</sub>H<sub>50</sub>N<sub>4</sub>O<sub>4</sub>F<sub>18</sub>P<sub>3</sub>]<sup>+</sup>: 1169.2752, found 1169.2710; calcd.  $[C_{47}H_{50}N_4O_4F_{12}P_2]^{+2}$ : 512.1552, found 512.1531.

Synthesis and characterization of methylbenzenesulfonate) **(10)**.



A solution of 3,6,9,12,15-pentaoxaheptadecane-1,17-diol (2.0 g, 7.1 mmol) and 4methylbenzenesulfonyl chloride (4.0 g, 21.3 mmol) in 50 mL of THF was stirred for 15 minutes at 0 °C in an ice bath. Then, a solution of KOH (2.8 g, 49.7 mmol) in 25 mL of H<sub>2</sub>O was added dropwise for 1 hour. After the complete addition, the solution was left 2 h under stirring at room temperature. Consecutively, a mixture of 150 mL H<sub>2</sub>O/Et<sub>2</sub>O (1:3) was added and the organic phase was separated from the aqueous phase. The aqueous phase was extracted again with Et<sub>2</sub>O (3 × 75 mL). The productcontaining organic fractions were combined and washed with 200 mL of a saturate solution of NH<sub>4</sub>Cl and 200 mL of H<sub>2</sub>O. Finally, the organic phase was dried with MgSO<sub>4</sub> and the solvent was removed under reduced pressure to afford **10** (3.9 g, 94%) as a yellow oil. <sup>1</sup>*H NMR* (500 *MHz*, *CDCl<sub>3</sub>*)  $\delta$ : 7.78 (d, *J* = 8.4 Hz, 4H), 7.33 (d, *J* = 8.5 Hz, 4H), 4.18 – 4.09 (m, 4H), 3.70 – 3.64 (m, 4H), 3.61 (q, *J* = 1.3 Hz, 8H), 3.57 (s, 8H), 2.44 (s, 6H) ppm. <sup>13</sup>*C NMR* (125 *MHz*, *CDCl<sub>3</sub>*)  $\delta$ : 144.86 (C), 132.83 (C), 129.85 (CH), 127.99 (CH), 70.71 (CH<sub>2</sub>), 70.58 (CH<sub>2</sub>), 70.52 (CH<sub>2</sub>), 70.47 (CH<sub>2</sub>), 69.29 (CH<sub>2</sub>), 68.65 (CH<sub>2</sub>), 21.69 (CH<sub>3</sub>). *HRMS-ESI (m/z*): calcd. [C<sub>26</sub>H<sub>39</sub>O<sub>11</sub>S<sub>2</sub>]<sup>+</sup>: 591.1928, found 591.1966.

Synthesis and characterization of 1,17-dibromo-3,6,9,12,15-pentaoxaheptadecane (11).



A solution of **10** (3.9 g, 6.7 mmol) and LiBr (2.3 g, 26.7 mmol) in 60 mL of acetone was left under reflux for 18 hours. Then, the solvent was removed under reduced pressure and the crude was dissolved in 200 mL of a mixture of  $H_2O/Et_2O$  (1:1). The organic phase was separated from the aqueous phase and this one was extracted again with  $Et_2O$  (3 × 75 mL). The product-containing organic fractions were combined, dried with MgSO<sub>4</sub> and the solvent was removed under reduced pressure to afford **11** (1.8 g, 66%) as a yellow oil. <sup>1</sup>*H NMR* (500 MHz, CDCl<sub>3</sub>)  $\delta$ : 3.83 (t, J = 6.3 Hz, 4H), 3.72 – 3.65 (m, 16H), 3.49 (t, J = 6.3 Hz, 4H) ppm. <sup>13</sup>*C NMR* (125 MHz, CDCl<sub>3</sub>)  $\delta$ : 71.19 (CH<sub>2</sub>), 70.66 (CH<sub>2</sub>), 70.57 (CH<sub>2</sub>), 70.52 (CH<sub>2</sub>), 30.38 (CH<sub>2</sub>) ppm. *HRMS-ESI* (*m*/*z*): calcd. [C<sub>12</sub>H<sub>25</sub>O<sub>5</sub>Br<sub>2</sub>]<sup>+</sup>: 407.0063, found 407.0083.

Synthesis and characterization of 1-(17-bromo-3,6,9,12,15-pentaoxaheptadecyl)-1'-(4-((naphthalen-2-yloxy)methyl)benzyl)-[4,4'-bipyridine]-1,1'-diium hexafluorophosphate (12·2PF<sub>6</sub>).



A solution of 5·PF<sub>6</sub> (250 mg, 0.47 mmol) and 11 (950 mg, 2.34 mmol) in 25 mL of CH<sub>3</sub>CN was left under reflux for 48 hours. Then, the solvent was removed under reduced pressure to leave a solid residue, which was subjected to flash chromatography (SiO<sub>2</sub>, CH<sub>3</sub>CN/NaCl (0.6 M)/MeOH 4/1/1). The product-containing fractions were combined and the solvents evaporated. The residue was suspended in EtOH and filtered off to remove NaCl. The EtOH was removed under reduced pressure to afford 12·2Cl as a yellow oil. The yellow oil was dissolved in the minimal amount of  $H_2O$  and  $KPF_6$ was added until no more precipitation was observed. The solid was filtered off and washed with 50 mL of H<sub>2</sub>O to leave **12**·2PF<sub>6</sub> (163 mg, 34%) as an orange oil. <sup>1</sup>H NMR (500 MHz, CD<sub>3</sub>CN)  $\delta$ : 8.99 (d, J = 7.0 Hz, 2H), 8.97 (d, J = 7.0 Hz, 2H), 8.42 (d, J = 7.0 Hz, 2H), 8.40 (d, J = 6.9 Hz, 2H), 7.80 (d, J = 8.7 Hz, 2H), 7.76 (dd, J = 8.3, 1.1 Hz, 1H), 7.64 (d, J = 8.3 Hz, 2H), 7.55 (d, J = 8.3 Hz, 2H), 7.45 (ddd, J = 8.2, 6.8, 1.3 Hz, 1H), 7.40 – 7.31 (m, 2H), 7.21 (dd, J = 9.0, 2.6 Hz, 1H), 5.83 (s, 2H), 5.25 (s, 2H), 4.77 (t, J = 4.7 Hz, 2H), 3.98 (t, J = 4.9 Hz, 2H), 3.70 - 3.64 (m, 2H), 3.64 - 3.59 (m, 2H), 3.55 - 3.46 (m, 14H), 3.46 - 3.41 (m, 2H). <sup>13</sup>C NMR (125 MHz, CD<sub>3</sub>CN) δ: 156.04 (C), 150.05 (C), 149.43 (C), 145.95 (CH), 145.18 (CH), 139.05 (C), 134.16 (C), 131.94 (C), 129.17 (CH), 129.14 (CH), 128.64 (C), 128.35 (CH), 127.20 (CH), 127.15 (CH), 126.41 (CH), 126.31 (CH), 126.23 (CH), 123.53 (CH), 118.39 (CH), 106.85 (CH), 70.22 (CH<sub>2</sub>), 69.80 (CH<sub>2</sub>), 69.69 (CH<sub>2</sub>), 69.68 (CH<sub>2</sub>), 69.62 (CH<sub>2</sub>), 69.61 (CH<sub>2</sub>), 69.51 (CH<sub>2</sub>), 69.50 (CH<sub>2</sub>), 69.47 (CH<sub>2</sub>), 68.55 (CH<sub>2</sub>), 68.16 (CH<sub>2</sub>), 64.03 (CH<sub>2</sub>), 61.21 (CH<sub>2</sub>), 31.20 (CH<sub>2</sub>). HRMS-ESI (m/z): calcd.  $[C_{40}H_{47}N_2O_6F_6PBr]^+$ : 875.2253, found 875.2252.

Synthesis and characterization of 1-methyl-1'-(17-(1'-(4-((naphthalen-2-yloxy)methyl)benzyl)-[4,4'-bipyridin]-1,1'-diium-1-yl)-3,6,9,12,15-pentaoxaheptadecyl)-[4,4'-bipyridine]-1,1'-diium (**2**·**4PF**<sub>6</sub>).



A mixture of 12·2PF<sub>6</sub> (163 mg, 0.16 mmol) and 8·PF<sub>6</sub> (505 mg, 1.6 mmol) in 7 mL of CH<sub>3</sub>CN was heated up to 110 °C for 6 hours using microwave-assisted heating. Then, the solvent was removed under reduced pressure to leave a solid residue, which was subjected to flash chromatography (SiO<sub>2</sub>) using two different eluent phases: CH<sub>3</sub>CN/NaCl (0.6 M)/MeOH 4/1/1) to remove impurities and  $CH_3CN/KPF_6$  (0.6 M)/MeOH 4/1/1) to elute the compound. The product-containing fractions were combined and the solvents evaporated. The obtained residue was then suspended in 100 mL H<sub>2</sub>O and filtered off to remove excess KPF<sub>6</sub> and dissolved in CH<sub>3</sub>CN. Finally, the CH<sub>3</sub>CN was removed under reduced pressure to leave 2·4PF<sub>6</sub> as a brown oil (74 mg, 33%). <sup>1</sup>H NMR (500 MHz, CD<sub>3</sub>CN)  $\delta$ : 8.99 (d, J = 6.8 Hz, 2H), 8.95 - 8.89 (m, 4H), 8.84 (d, J = 6.9 Hz, 2H), 8.42 - 8.34 (m, 8H), 7.80 (d, J = 8.4 Hz, 2H), 7.76 (d, J = 8.2 Hz, 1H), 7.64 (d, J = 8.1 Hz, 2H), 7.55 (d, J = 8.3 Hz, 2H), 7.48 – 7.43 (m, 1H), 7.38 – 7.32 (m, 2H), 7.21 (dd, J = 8.9, 2.6 Hz, 1H), 5.84 (s, 2H), 5.25 (s, 2H), 4.74 (q, J = 5.0 Hz, 4H), 4.39 (s, 3H), 3.94 (q, J = 4.6 Hz, 4H), 3.61 - 3.45 (m, 16H). <sup>13</sup>C NMR (125 MHz, CD<sub>3</sub>CN)  $\delta$ : 156.29 (C), 150.29 (C), 149.85 (C), 149.84 (C), 149.41 (C), 146.29 (CH), 146.03 (CH), 146.03 (CH), 145.42 (CH), 139.30 (C), 134.38 (C), 132.10 (C), 129.38 (CH), 129.35 (CH), 128.88 (C), 128.55 (CH), 127.40 (CH), 127.33 (CH), 126.63 (CH), 126.61 (CH), 126.54 (CH), 126.52 (CH), 126.44 (CH), 123.75 (CH), 118.57 (CH), 107.14 (CH), 70.10 (CH<sub>2</sub>), 69.91 (CH<sub>2</sub>), 69.87 (CH<sub>2</sub>), 69.62 (CH<sub>2</sub>), 68.83 (CH<sub>2</sub>), 68.37 (CH<sub>2</sub>), 64.29 (CH<sub>2</sub>), 61.45 (CH<sub>2</sub>), 48.43 (CH<sub>3</sub>). *HRMS-ESI (m/z*): calcd. [C<sub>51</sub>H<sub>58</sub>N<sub>4</sub>O<sub>6</sub>F<sub>18</sub>P<sub>3</sub>]<sup>+</sup>: 1257.3276, found 1257.3273; calcd.  $[C_{51}H_{58}N_4O_6F_{12}P_2]^{+2}$ : 556.1814, found 556.1813; calcd.  $[C_{51}H_{58}N_4O_6F_6P]^{+3}$ : 322.4660, found 322.4662.

Synthesis and characterization of methylbenzenesulfonate) **(13)**.

3,6,9,12,15,18,21-heptaoxatricosane-1,23-diyl bis(4-

A solution of 3,6,9,12,15,18,21-heptaoxatricosane-1,23-diol (1.0 g, 2.7 mmol) and 4methylbenzenesulfonyl chloride (1.5 g, 8.1 mmol) in 50 mL of THF was stirred for 15 minutes at 0°C in an ice bath. Then, a solution of KOH (1.1 g, 18.9 mmol) in 25 mL of H<sub>2</sub>O was added dropwise for 1 hour. After the complete addition, the solution was left 2 h under stirring at room temperature. Consecutively, a mixture of 150 mL H<sub>2</sub>O/CH<sub>2</sub>Cl<sub>2</sub> (1:3) was added and the organic phase was separated from the aqueous phase. The aqueous phase was extracted again with CH<sub>2</sub>Cl<sub>2</sub> (3 × 50 mL). The product-containing organic fractions were combined and washed with 200 mL of a saturate solution of NH<sub>4</sub>Cl and 200 mL of H<sub>2</sub>O. Finally, the organic phase was dried with MgSO<sub>4</sub> and the solvent was removed under reduced pressure to afford **13** as a yellow oil (1.3 g, 72%). <sup>1</sup>*H NMR* (500 *MHz*, *CDCl<sub>3</sub>*)  $\delta$ : 7.79 (d, *J* = 8.3 Hz, 4H), 7.34 (d, *J* = 8.0 Hz, 4H), 4.21 – 4.12 (m, 4H), 3.71 – 3.65 (m, 4H), 3.66 – 3.60 (m, 16H), 3.58 (s, 8H), 2.44 (s, 6H) ppm. <sup>13</sup>*C NMR* (125 *MHz*, *CDCl<sub>3</sub>*)  $\delta$ : 144.79 (C), 132.98 (C), 129.82 (CH), 127.98 (CH), 70.73 (CH<sub>2</sub>), 70.58 (CH<sub>2</sub>), 70.54 (CH<sub>2</sub>), 70.49 (CH<sub>2</sub>), 69.24 (CH<sub>2</sub>), 68.67 (CH<sub>2</sub>), 21.64 (CH<sub>3</sub>) ppm. *HRMS-ESI (m/z)*: calcd. [C<sub>30</sub>H<sub>47</sub>O<sub>13</sub>S<sub>2</sub>]<sup>+</sup>: 679.2452, found 679.2462.

Synthesis and characterization of 1,23-dibromo-3,6,9,12,15,18,21-heptaoxatricosane (14).

A solution of **13** (1.30 g, 1.9 mmol) and LiBr (0.67 g, 7.6 mmol) in 50 mL of acetone was left under reflux for 18 hours. Then, the solvent was removed under reduced pressure and the crude was dissolved in 100 mL of a mixture of  $H_2O/CH_2Cl_2$  (1:1). The organic phase was separated from the aqueous phase and this one was extracted again with  $CH_2Cl_2$  (3 × 50 mL). The product-containing organic fractions were combined, dried with MgSO<sub>4</sub> and the solvent was removed under reduced pressure to afford **14** (0.92 g, 95%) as a yellow oil. <sup>1</sup>*H NMR* (500 *MHz*, *CDCl<sub>3</sub>*)  $\delta$ : 3.83 (t, *J* = 6.3 Hz, 4H), 3.69 – 3.65 (m, 24H), 3.49 (t, *J* = 6.3 Hz, 4H) ppm. <sup>13</sup>*C NMR* (125 *MHz*, *CDCl<sub>3</sub>*)  $\delta$ : 71.21 (CH<sub>2</sub>), 70.65 (CH<sub>2</sub>), 70.58 (CH<sub>2</sub>), 70.53 (CH<sub>2</sub>), 30.32 (CH<sub>2</sub>) ppm. *HRMS-ESI* (*m*/*z*): calcd. [C<sub>16</sub>H<sub>33</sub>O<sub>7</sub>Br<sub>2</sub>]<sup>+</sup>: 495.0587, found 495.0600.

Synthesis and characterization of 1-(23-bromo-3,6,9,12,15,18,21-heptaoxatricosyl)-1'-(4-((naphthalen-2-yloxy)methyl)benzyl)-[4,4'-bipyridine]-1,1'-diium hexafluorophosphate (**15**·2PF<sub>6</sub>).



A solution of 5 PF<sub>6</sub> (331 mg, 0.60 mmol) and 14 (1500 mg, 1.95 mmol) in 20 mL of CH<sub>3</sub>CN was left under reflux for 48 hours. Then, the solvent was removed under reduced pressure to leave a solid residue, which was subjected to flash chromatography (SiO<sub>2</sub>, CH<sub>3</sub>CN/NaCl (0.6 M)/MeOH 4/1/1). The product-containing fractions were combined and the solvents evaporated. The residue was suspended in EtOH and filtered off to remove NaCl. The EtOH was removed under reduced pressure to afford 15·2Cl as a yellow oil. The yellow oil was dissolved in the minimal amount of  $H_2O$  and  $KPF_6$ was added until no more precipitation was observed. The solid was filtered off and washed with 50 mL of H<sub>2</sub>O to leave **15**·2PF<sub>6</sub> (80 mg, 7%) as an orange oil. <sup>1</sup>H NMR (500 MHz, CD<sub>3</sub>CN)  $\delta$ : 9.00 (d, J = 7.0 Hz, 2H), 8.95 (d, J = 7.0 Hz, 2H), 8.41 (t, J = 6.9 Hz, 2H), 7.81 (d, J = 9.0 Hz, 2H), 7.76 (dd, J = 8.3, 1.1 Hz, 1H), 7.64 (d, J = 8.3 Hz, 2H), 7.55 (d, J = 8.3 Hz, 2H), 7.46 (ddd, J = 8.2, 6.8, 1.3 Hz, 1H), 7.39 - 7.33 (m, 2H), 7.22 (dd, J = 9.0, 2.6 Hz, 1H), 5.84 (s, 2H), 5.26 (s, 2H), 4.78 (t, J = 4.8 Hz, 2H), 3.97 (t, J = 4.5 Hz, 2H), 3.74 (t, J = 6.2, 5.2 Hz, 2H), 3.64 – 3.62 (m, 2H), 3.60 – 3.48 (m, 24H) ppm. <sup>13</sup>C NMR (125 MHz, *CD*<sub>3</sub>*CN*) δ: 159.21 (C), 153.17 (C), 152.75 (C), 149.03 (CH), 148.38 (CH), 142.25 (C), 137.33 (C), 135.06 (C), 132.30 (CH), 132.28 (CH), 131.83 (C), 131.49 (CH), 130.34 (CH), 130.30 (CH), 129.61 (CH), 129.45 (CH), 129.36 (CH), 126.67 (CH), 121.51 (CH), 110.09 (CH), 73.39 (CH<sub>2</sub>), 72.97 (CH<sub>2</sub>), 72.60 (CH<sub>2</sub>), 72.53 (CH<sub>2</sub>), 72.51 (CH<sub>2</sub>), 72.50 (CH<sub>2</sub>), 72.48 (CH<sub>2</sub>), 72.41 (CH<sub>2</sub>), 72.37 (CH<sub>2</sub>), 72.35 (CH<sub>2</sub>), 72.33 (CH<sub>2</sub>), 71.75 (CH<sub>2</sub>), 71.34 (CH<sub>2</sub>), 67.23 (CH<sub>2</sub>), 64.30 (CH<sub>2</sub>), 34.22 (CH<sub>2</sub>) ppm. HRMS-ESI (m/z): calcd. [C<sub>44</sub>H<sub>55</sub>N<sub>2</sub>O<sub>8</sub>F<sub>6</sub>PBr]<sup>+</sup>: 963.2778, found 963.2787; calcd. [C<sub>44</sub>H<sub>55</sub>N<sub>2</sub>O<sub>8</sub>Br]<sup>+2</sup>: 409.1565, found 409.1571.

Synthesis and characterization of 1-methyl-1'-(23-(1'-(4-((naphthalen-2-yloxy)methyl)benzyl)-[4,4'-bipyridin]-1,1'-diium-1-yl)-3,6,9,12,15,18,21-heptaoxatricosyl)-[4,4'-bipyridine]-1,1'-diium hexafluorophosphate ( $3\cdot4PF_6$ ).



A mixture of 15·2PF<sub>6</sub> (115 mg, 0.10 mmol) and 8·PF<sub>6</sub> (329 mg, 1.0 mmol) in 5 mL of CH<sub>3</sub>CN was heated up to 110 °C for 6 hours using microwave-assisted heating. Then, the solvent was removed under reduced pressure to leave a solid residue, which was subjected to flash chromatography  $(SiO_2)$  using two different eluent phases: CH<sub>3</sub>CN/NaCl (0.6 M)/MeOH 4/1/1) to remove impurities and  $CH_3CN/KPF_6$  (0.6 M)/MeOH 4/1/1) to elute the compound. The product-containing fractions were combined and the solvents evaporated. The obtained residue was then suspended in 100 mL H<sub>2</sub>O and filtered off to remove excess KPF<sub>6</sub> and dissolved in CH<sub>3</sub>CN. Finally, the CH<sub>3</sub>CN was removed under reduced pressure to leave 3·4PF<sub>6</sub> as a brown oil (42 mg, 27%). <sup>1</sup>H NMR (500 MHz, CD<sub>3</sub>CN)  $\delta$ : 9.00 (d, J = 7.0 Hz, 2H), 8.93 (t, J = 6.8 Hz, 4H), 8.84 (d, J = 6.9 Hz, 2H), 8.42 - 8.36 (m, 8H), 7.80 (dd, J = 8.4, 1.2 Hz, 2H), 7.75 (dd, J = 8.3, 1.0 Hz, 1H), 7.64 (d, J = 8.3 Hz, 2H), 7.55 (d, J = 8.3 Hz, 2H), 7.46 (ddd, J = 8.2, 6.8, 1.3 Hz, 1H), 7.40 – 7.32 (m, 2H), 7.21 (dd, J = 9.0, 2.6 Hz, 1H), 5.84 (s, 2H), 5.25 (s, 2H), 4.79 – 4.72 (m, 4H), 4.39 (s, 3H), 3.95 (q, J = 4.9 Hz, 4H), 3.63 – 3.56 (m, 4H), 3.53 – 3.44 (m, 20H) ppm. <sup>13</sup>C NMR (125 MHz, CD<sub>3</sub>CN) δ: 156.19 (C), 150.20 (C), 149.72 (C), 149.70 (C), 149.31 (C), 146.23 (CH), 146.04 (CH), 146.01 (CH), 145.37 (CH), 139.19 (C), 134.30 (C), 132.12 (C), 129.32 (CH), 129.30 (CH), 128.79 (C), 128.51 (CH), 127.35 (CH), 127.30 (CH), 126.56 (CH), 126.50 (CH), 126.46 (CH), 126.44 (CH), 126.40 (CH), 123.70 (CH), 118.53 (CH), 106.97 (CH), 69.99 (CH<sub>2</sub>), 69.75 (CH<sub>2</sub>), 69.73 (CH<sub>2</sub>), 69.56 (CH<sub>2</sub>), 68.71 (CH<sub>2</sub>), 68.34 (CH<sub>2</sub>), 64.18 (CH<sub>2</sub>), 61.35 (CH<sub>2</sub>), 48.36 (CH<sub>3</sub>) ppm. *HRMS-ESI (m/z*): calcd.  $[C_{55}H_{66}N_4O_8F_{12}P_2]^{+2}$ : 600.2076, found 600.2094; calcd.  $[C_{55}H_{66}N_4O_8F_6P]^{+3}$ : 351.8168, found 351.8172.

Synthesis and characterization of **1**⊂CB[8]



A solution of  $1.4PF_6$  (6.5 mg, 0.005 mmol) in 2.5 mL of a mixture of  $H_2O/CH_3CN$  (3/2 v/v) was prepared and 1 equivalent of CB[8] was added. The mixture of the reaction was stirred for 10 minutes. A portion of 0.6 mL was taken from the resulting mixture and the solvent was evaporated under reduced pressure to leave the crude product. The solid was dissolved in  $D_2O/CD_3CN$  (3/2 v/v) (0.6 mL, 2 mM with respect to  $1.4PF_6$ ). <sup>1</sup>H NMR (500 MHz,  $D_2O/CD_3CN$  (3/2)):  $\delta$  9.46 (d, J = 6.3 Hz, 2H), 9.35 (d, J = 6.3 Hz, 2H), 9.06 (d, J = 6.4 Hz, 2H), 9.01 (t, J = 6.6 Hz, 4H), 8.94 (d, J = 6.3 Hz, 2H), 8.86 (d, J = 6.3 Hz, 2H), 8.83 – 8.77 (m, 8H), 8.18 (dd, J = 9.0, 4.2 Hz, 2H), 8.11 (d, J = 8.3 Hz, 1H), 7.90 (m, 4H), 7.82 (t, J = 7.6 Hz, 1H), 7.73 (t, J = 7.5 Hz, 1H), 7.68 (s, 1H), 7.58 (dd, J = 9.0, 2.5 Hz, 1H), 6.21 (s, 2H), 6.01 (dd, J = 25.3, 15.2 Hz, 16H), 5.70 (s, 16H), 5.58 (s, 1H), 4.45 (t, J = 14.9 Hz, 16H), 3.93 (s, 3H), 3.36 (bs, 4H), 2.54 (bs, 2H), 2.47 – 2.36 (bs, 2H), 2.19 – 2.14 (bs, 2H), 2.11 (bs, 2H) ppm. <sup>13</sup>C NMR (125 MHz, CD<sub>3</sub>CN): δ 155.77 (C), 149.85 (C), 148.91 (C), 147.55 (C), 147.13 (CH), 146.92 (CH), 146.16 (CH), 145.97 (CH), 145.88 (CH), 145.46 (C), 137.76 (C), 134.99 (C), 132.29 (C), 130.41 (CH), 129.30 (CH), 129.27 (C), 128.50 (C), 127.80 (CH), 127.26 (CH), 127.16 (CH), 126.68 (CH), 126.46 (CH), 126.34 (CH), 125.98 (CH), 125.81 (CH), 123.82 (CH), 107.06 (CH), 102.04 (CH), 72.84 (CH<sub>2</sub>), 71.25 (CH), 69.43 (CH<sub>2</sub>), 69.30 (CH<sub>2</sub>), 69.21 (CH<sub>2</sub>), 68.64 (CH<sub>2</sub>), 66.92 (CH<sub>2</sub>), 66.64 (CH<sub>2</sub>), 64.14 (CH<sub>2</sub>), 61.11 (CH<sub>2</sub>), 53.53 (CH<sub>2</sub>), 47.80 (CH<sub>3</sub>).

The solid was dissolved in CD<sub>3</sub>CN (0.6 mL, 2 mM with respect to  $1 \cdot 4PF_6$ ). <sup>1</sup>*H NMR* (500 *MHz*, *CD*<sub>3</sub>*CN*)  $\delta$ : 9.01 (d, *J* = 7.0 Hz, 2H), 8.92 (d, *J* = 7.0 Hz, 2H), 8.90 – 8.86 (m, 4H), 8.70 (d, *J* = 7.0 Hz, 2H), 8.67 (d, *J* = 7.1 Hz, 2H), 8.62 (d, *J* = 7.2 Hz, 2H), 8.58 (d, *J* = 7.2 Hz, 2H), 7.84 (d, *J* = 8.9 Hz, 2H), 7.79 (d, *J* = 8.3 Hz, 1H), 7.59 (s, 4H), 7.48 (ddd, *J* = 8.2, 6.8, 1.3 Hz, 1H), 7.42 – 7.35 (m, 2H), 7.25 (dd, *J* = 9.0, 2.5 Hz, 1H), 5.84 (s, 2H), 5.77 (dd, *J* = 25.2, 14.9 Hz, 16H), 5.29 (s, 16H), 5.25 (s, 2H), 4.45– 4.55 (s, 2H), 4.40 (s, 3H), 4.08 (dd, *J* = 15.0, 10.7 Hz, 16H), 3.11 – 2.91 (m, 2H), 1.75 (s, 4H) ppm. <sup>13</sup>*C NMR* (125 *MHz*, *CD*<sub>3</sub>*CN*)  $\delta$ : 156.36 (C), 155.64 (C), 155.57 (C), 148.46 (C), 148.26 (C), 147.05 (C), 146.76 (CH), 146.71 (CH), 145.91 (CH), 145.87 (CH), 138.95 (C), 134.41 (C), 132.39 (C), 129.74 (CH), 129.40 (CH), 128.89 (C), 128.57 (C), 128.20 (CH), 127.47 (CH), 127.23 (CH), 126.54 (CH), 126.54 (CH), 126.52 (CH), 126.09 (CH), 125.83 (CH), 123.84 (CH), 118.65 (CH), 107.16 (CH), 72.19 (CH<sub>2</sub>), 71.31 (CH), 70.14 (CH<sub>2</sub>), 70.11 (CH<sub>2</sub>), 68.93 (CH<sub>2</sub>), 67.26 (CH<sub>2</sub>), 67.21 (CH<sub>2</sub>), 63.82 (CH<sub>2</sub>), 63.82 (CH<sub>2</sub>), 60.85 (CH<sub>2</sub>), 53.08

(CH<sub>2</sub>), 53.03 (CH<sub>2</sub>), 47.93 (CH<sub>3</sub>) ppm. *HRMS-ESI (m/z*): calcd.  $[C_{95}H_{98}N_{36}O_{20}F_{12}P_2]^{+2}$ : 1176.3515, found 1176.3417; calcd.  $[C_{95}H_{98}N_{36}O_{20}F_6P]^{+3}$ : 735.9127, found 735.9092.

Synthesis and characterization of 2⊂CB[8]



A solution of  $2.4PF_6$  (7.0 mg, 0.005 mmol) in 2.5 mL of a mixture of  $H_2O/CH_3CN$  (3/2 v/v) was prepared and 1 equivalent of CB[8] was added. The mixture of the reaction was stirred for 10 minutes. A portion of 0.6 mL was taken from the resulting mixture and the solvent was evaporated under reduced pressure to leave the crude product. The solid was dissolved in CD<sub>3</sub>CN (0.6 mL, 2 mM with respect to  $2.4PF_6$ ). <sup>1</sup>H NMR (500 MHz, CD<sub>3</sub>CN):  $\delta$  8.99 (d, J = 6.5 Hz, 2H), 8.82 (m, 8H), 8.74 (d, J = 6.7 Hz, 2H), 8.57 (d, J = 6.5 Hz, 2H), 8.53 (d, J = 6.3 Hz, 2H), 7.83 (d, J = 8.3 Hz, 2H), 7.78 (d, J = 8.2 Hz, 1H), 7.58 (s, 5H), 7.47 (t, J = 7.5 Hz, 1H), 7.41 – 7.34 (m, 2H), 7.24 (d, J = 9.0 Hz, 1H), 5.81 (d, J = 6.4 Hz, 2H), 5.76 (d, J = 15.0 Hz, 16H), 5.28 (d, J = 3.0 Hz, 2H), 5.25 (s, 16H), 4.72 (s, 3H), 4.67 (m, 2H), 4.38 (m, 2H), 4.02 (d, J = 15.0 Hz, 16H), 3.68 (m, 2H), 3.41 (m, 2H), 3.27 (m, 2H), 3.15 (m, 2H), 2.93 (m, 2H), 2.90 (m, 2H), 2.66 (m, 2H), 2.63 (m, 2H), 2.36 (m, 2H), 2.29 (overlap) ppm. <sup>13</sup>C NMR (125 *MHz, CD*<sub>3</sub>*CN*): δ 156.63 (C), 148.01 (CH), 147.27 (CH), 147.14 (CH), 146.39 (CH), 133.15 (CH), 131.58 (CH), 130.56 (CH), 129.87 (CH), 129.40 (CH), 128.87 (CH), 128.30 (CH), 127.70 (CH), 127.24 (CH), 125.00 (CH), 119.78 (CH), 108.34 (CH), 72.46 (CH), 72.20 (CH<sub>2</sub>), 71.62 (CH<sub>2</sub>), 71.29 (CH<sub>2</sub>), 70.89 (CH<sub>2</sub>), 70.49 (CH<sub>2</sub>), 70.10 (CH<sub>2</sub>), 69.32 (CH<sub>2</sub>), 65.04 (CH<sub>2</sub>), 64.24 (CH<sub>2</sub>), 63.62 (CH<sub>2</sub>), 55.22 (CH<sub>2</sub>), 49.66 (CH<sub>3</sub>). HRMS-ESI (*m/z*):  $[C_{99}H_{106}N_{36}O_{22}F_{12}P_2]^{+2}$ : 1220.3777, found calcd. 1220.3758; calcd. [C<sub>99</sub>H<sub>106</sub>N<sub>36</sub>O<sub>22</sub>F<sub>6</sub>P]<sup>+3</sup>: 765.2635, found 765.2657.

#### Synthesis and characterization of **3**⊂CB[8]



A solution of  $3.4PF_6$  (7.5 mg, 0.005 mmol) in 2.5 mL of a mixture of  $H_2O/CH_3CN$  (3/2 v/v) was prepared and 1 equivalent of CB[8] was added. The mixture of the reaction was stirred for 10 minutes. A portion of 0.6 mL was taken from the resulting mixture and the solvent was evaporated under reduced pressure to leave the crude product. The solid was dissolved in CD<sub>3</sub>CN (0.6 mL, 2 mM with respect to  $3.4PF_6$ ).<sup>1</sup>H NMR (500 MHz, CD<sub>3</sub>CN):  $\delta$  9.03 (d, J = 6.4 Hz, 2H), 8.98 (d, J = 6.4 Hz, 2H), 8.89 (d, J = 6.5 Hz, 2H), 8.86 (d, J = 6.5 Hz, 2H), 8.71 (d, J = 6.3 Hz, 2H), 8.59 (m, 2H), 8.48 (d, J = 6.3 Hz, 2H), 8.59 (m, 2H), 8.48 (d, J = 6.3 Hz, 2H), 8.59 (m, 2H), 8.48 (d, J = 6.3 Hz, 2H), 8.59 (m, 2H), 2H), 8.44 (m, 2H), 7.83 (d, J = 8.4 Hz, 2H), 7.78 (d, J = 8.3 Hz, 1H), 7.58 (s, 4H), 7.48 (ddd, J = 8.2, 6.8, 1.3 Hz, 1H), 7.40 – 7.36 (m, 1H), 7.35 (d, J = 2.5 Hz, 1H), 7.23 (dd, J = 8.9, 2.6 Hz, 1H), 5.83 (s, 2H), 5.76 (d, J = 14.9 Hz, 16H), 5.24 (s, 16H), 4.74 (m, 2H), 4.63 (m, 2H), 4.37 (s, 3H), 4.01 (d, J = 15.0 Hz, 16H), 3.78 (m, 2H), 3.52 (m, 2H), 3.45 (m, 2H), 3.29 (m, 2H), 3.23 (m, 2H), 3.13 (m, 2H), 3.09 (m, 2H), 2.92 (m, 4H), 2.82 (m, 6H), 2.74 (m, 2H), 2.55 (m, 2H).<sup>13</sup>C NMR (125 MHz, CD<sub>3</sub>CN): δ 156.09 (C), 155.25 (C), 149.14 (C), 148.70 (C), 148.53 (C), 148.24 (CH), 146.76 (CH), 146.25 (CH), 145.99 (CH), 145.17 (CH), 138.67 (CH), 134.17 (CH), 132.39 (CH), 129.40 (CH), 129.16 (CH), 128.64 (CH), 128.06 (CH), 127.40 (CH), 127.25 (CH), 126.72 (CH), 126.33 (CH), 123.62 (CH), 71.96 (CH), 69.92 (CH<sub>2</sub>), 69.53 (CH<sub>2</sub>), 69.32 (CH<sub>2</sub>), 69.25 (CH<sub>2</sub>), 69.15 (CH<sub>2</sub>), 69.10 (CH<sub>2</sub>), 68.96 (CH<sub>2</sub>), 68.71 (CH<sub>2</sub>), 68.69 (CH<sub>2</sub>), 68.41 (CH<sub>2</sub>), 68.18 (CH<sub>2</sub>), 68.00 (CH<sub>2</sub>), 66.26 (CH<sub>2</sub>), 63.59 (CH<sub>2</sub>), 61.83 (CH<sub>2</sub>), 61.16 (CH<sub>2</sub>), 54.15 (CH<sub>2</sub>), 52.24 (CH<sub>2</sub>), 47.82 (CH<sub>3</sub>) ppm. *HRMS-ESI* (*m*/*z*): calcd. [C<sub>103</sub>H<sub>114</sub>N<sub>36</sub>O<sub>24</sub>F<sub>12</sub>P<sub>2</sub>]<sup>+2</sup>: 1264.4039, found 1264.4003; calcd. [C<sub>103</sub>H<sub>114</sub>N<sub>36</sub>O<sub>24</sub>F<sub>6</sub>P]<sup>+3</sup>: 794.6144, found 794.6138.



Figure S1 UV-VIS Titration data of  $1.4PF_6$  with increasing amounts of CB[8] in H<sub>2</sub>O/CH<sub>3</sub>CN (3/2 v/v).



Figure S2 UV-VIS Representation of titration  $1 \cdot 4PF_6$  and CB[8] at  $\lambda_{obs}$ = 261 nm in H<sub>2</sub>O/CH<sub>3</sub>CN (3/2 v/v).



Figure S3 Fitting of titration of  $1.4PF_6$  and CB[8] at  $\lambda_{obs}$ = 261 nm using supramolecular.org in H<sub>2</sub>O/CH<sub>3</sub>CN (3/2 v/v).

x1: Guest concentration / M	x2:Host concentration / M	x3: H/G equivalent total	y1: 261 nm
0.000014	0	0	0.51585
0.000014	0.0000014	0.1	0.51422
0.000014	0.000028	0.2	0.51189
0.000014	0.0000042	0.3	0.50946
0.000014	0.000056	0.4	0.50663
0.000014	0.000007	0.5	0.50377
0.000014	0.000084	0.6	0.49849
0.000014	0.000098	0.7	0.49584
0.000014	0.0000112	0.8	0.49319
0.000014	0.0000126	0.9	0.49303
0.000014	0.000014	1	0.49403
0.000014	0.0000154	1.1	0.49509
0.000014	0.0000168	1.2	0.49571
0.000014	0.0000196	1.4	0.49705
0.000014	0.0000224	1.6	0.49851
0.000014	0.000028	2	0.49949
0.000014	3.30103E-05	2.4	0.5012
0.000014	3.76667E-05	2.7	0.50163

K <sub>11</sub>	K <sub>21</sub>	K <sub>11</sub> error (%)	K <sub>21</sub> error (%)	SSR	Datapoints fitted	Params fitted
197160.507	۔ 35726.9294	10.5280012	۔ 0.14815273	1.5125E-05	18	4

H coeffs	HG coeffs	H2G coeffs	Raw coeffs 1	Raw coeffs 2	Raw coeffs 3
36846.4286	35955.7143	37620.1208	36846.4286	35955.7143	37620.1208



**Figure S4** UV-visible spectrum of  $2 \cdot 4PF_6$  and  $2 \subset CB[8]$  in  $H_2O/CH_3CN$  (3/2 v/v) before and after addition of an excess of  $Na_2S_2O_4$  under  $N_2$  atmosphere.



**Figure S5** UV-visible spectrum of  $2.4PF_6$  and  $2 \subset CB[8]$  (5mM) in  $H_2O/CH_3CN$  (3/2 v/v) showing the increasing of the absorbance because of the charge-transfer band.



**Figure S6** Room temperature EPR spectrum of a  $H_2O/CH_3CN$  (3/2, v/v) solution containing compound  $3^{4+}$  and  $Na_2S_2O_4$ . The asterisk indicates the signal due to  $SO_2^{\bullet-}$  radical anion.



Figure S7 EPR spectroscopy of  $1.4PF_6$  and  $1 \subset CB[8]$  after addition of an excess of  $Na_2S_2O_4$  under  $N_2$  atmosphere.



**Figure S8** EPR spectroscopy of  $2 \cdot 4PF_6$  and  $2 \subset CB[8]$  after addition of an excess of  $Na_2S_2O_4$  under  $N_2$  atmosphere.



**Figure S9** EPR spectroscopy of  $2 \cdot 4PF_6$  and  $2 \subset CB[8]$  after addition of an excess of  $Na_2S_2O_4$  under  $N_2$  atmosphere.











Figure S12 Cyclic voltammetric response on a glassy carbon electrode of 1.0 mM 1·4PF<sub>6</sub> in H<sub>2</sub>O/CH<sub>3</sub>CN (3:2, v/v). Supporting electrolyte: 0.1 M KCl. Scan rate: 50 mV/s.



Figure S13 Cyclic voltammetric response on a glassy carbon electrode of 1.0 mM 2·4PF<sub>6</sub> in H<sub>2</sub>O/CH<sub>3</sub>CN (3:2, v/v). Supporting electrolyte: 0.1 M KCl. Scan rate: 50 mV/s.



Figure S14 Cyclic voltammetric response on a glassy carbon electrode of 1.0 mM 3·4PF<sub>6</sub> in H<sub>2</sub>O/CH<sub>3</sub>CN (3:2, v/v). Supporting electrolyte: 0.1 M KCl. Scan rate: 50 mV/s.



Figure S15 Cyclic voltammetric response on a glassy carbon electrode of 1.0 mM 2<sup>4+</sup> CCB[8] in H<sub>2</sub>O/CH<sub>3</sub>CN (3:2, v/v) at different scan rate. Supporting electrolyte: 0.1 M KCl.



Figure S16 Cyclic voltammetric response on a glassy carbon electrode of 1.0 mM 3<sup>4+</sup> CCB[8] in H<sub>2</sub>O/CH<sub>3</sub>CN (3:2, v/v) at different scan rate. Supporting electrolyte: 0.1 M KCl.







Figure S18  $^{13}\text{C}$  NMR (125 MHz, CDCl\_3) spectrum of 4.










































Figure S30  $^{\rm 13}\text{C}$  NMR (125 MHz, CDCl $_3$ ) spectrum of 7.

<u> 2ε.0ε</u> —

61.17 20.65 20.65



























Figure S37  $^{13}\text{C}$  NMR and DEPT (125 MHz, CD\_3CN) spectrum of  $9\cdot\text{2PF}_6$ 













Figure S40  $^{13}\text{C}$  NMR and DEPT (125 MHz, CD $_3\text{CN})$  spectrum of 1.4PF $_6.$ 



















Figure S45  $^{\rm 13}{\rm C}$  NMR (125 MHz, CDCl $_{\rm 3})$  spectrum of 10.





Figure S46  $^{13}\text{C}$  NMR and DEPT (125 MHz, CDCl\_3) spectrum of 10.







86.06 ----

29:02 89:02 99:02 61:12

Figure S48 <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) spectrum of 11.

-0











Figure S51  $^{13}\text{C}$  NMR (125 MHz, CD $_3\text{CN}$ ) spectrum of 12·2PF $_6.$ 

-0

−₽

-20

- 6

-6

- 22

- 09

- 2

- 8

- 6

100 11 (ppm)

110

120

130

140

150

160

-170

- 81

190





- 6

Figure S52  $^{13}\text{C}$  NMR and DEPT (125 MHz, CD $_3\text{CN})$  spectrum of 12·2PF $_6.$ 











Figure S55  $^{13}\text{C}$  NMR and DEPT (125 MHz, CD\_3CN) spectrum of  $2\cdot 4\text{PF}_6.$ 











Figure S58 HMBC (500 MHz, CD $_3$ CN) spectrum of 2·4PF $_6$ 












- 22

- 99

- 9

- 2

75

- 8

- 85

- 6

95

100 11 (ppm)

105

-11

115

120

125

130

135

140









28.05 <del>—</del>

10.63 70.65 70.65 70.53



















Figure S67  $^{13}\text{C}$  and DEPT NMR (125 MHz, CD $_3\text{CN}$ ) spectrum of 15·2PF $_6.$ 



Figure S68  $^1\text{H}$  NMR (500 MHz, CD $_3\text{CN})$  spectrum of  $3\cdot 4\text{PF}_6.$ 









Figure S70  $^{13}\text{C}$  and DEPT NMR (125 MHz, CD\_3CN) spectrum of  $3\cdot4\text{PF}_6.$ 



















Figure S75 <sup>13</sup>C NMR (125 MHz, CD<sub>3</sub>CN) spectrum of 1CCB[8].





35 -

- 6

- 45

- 23

- 22

- 09

- 29

- 2

-75

- 8

- 58

- 6

1 - 1 100 95 f1 (ppm)

105

-11

115

120

125

130

135

140

145

150

155





















Figure S81 <sup>13</sup>C NMR (500 MHz, CD<sub>3</sub>CN) spectrum of 2CCB[8].

























Figure S88  $^{13}\text{C}$  and DEPT NMR (125 MHz, CD\_3CN) spectrum of 3CCB[8].











