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1 **CONCENTRATE REDUCTION IN NF AND RO DESALINATION**
2 **SYSTEMS BY MEMBRANE-IN-SERIES CONFIGURATIONS-**
3 **EVALUATION OF PRODUCT WATER FOR REUSE IN IRRIGATION**

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11
12 **Abstract**

13 Wastewater reclamation and reuse became essential to meet the more restrictive discharge
14 limits and to overcome the water scarcity issue. Industrial wastewaters treated by membrane
15 bioreactor (MBR) systems generally include high amounts of salinity, nutrients as nitrogen and
16 phosphorus, heavy metals, hardness, etc. which in most cases, do not fit with discharge and
17 reuse limits. Membrane processes like nanofiltration (NF) and reverse osmosis (RO) are widely
18 adopted for secondary effluent treatment producing high-quality water. Since untreated brine
19 can damage the environment, this work focused on reduction of amount of concentrate by
20 working in membrane-in-series configuration. Tests with two membrane-in-series
21 (NF90+NF90, BW30+BW30, BW30+NF270 combinations) were performed as the concentrate
22 stream of the first one was fed to the second one. Water recoveries were over 80% in each
23 experiment. Water fluxes of 44.0, 41.4 and 73.4 L/m².h were obtained with NF90+NF90,
24 BW30+BW30, BW30+NF270 combinations, respectively. Also, salinity rejections represented
25 by electrical conductivity were 93.0%, 97.4% and 42.4% for NF90+NF90, BW30+BW30,
26 BW30+NF270 configurations, respectively. Permeates of BW30+BW30 and NF90+NF90
27 combinations seem to be useable for agricultural irrigation although soil permeability is an
28 important issue, not only ion concentrations and salinity of water used.

29
30 **Keywords:** Desalination, concentrate reduction, membrane processes, nanofiltration (NF), reverse
31 osmosis (RO), water reuse.

32 *****

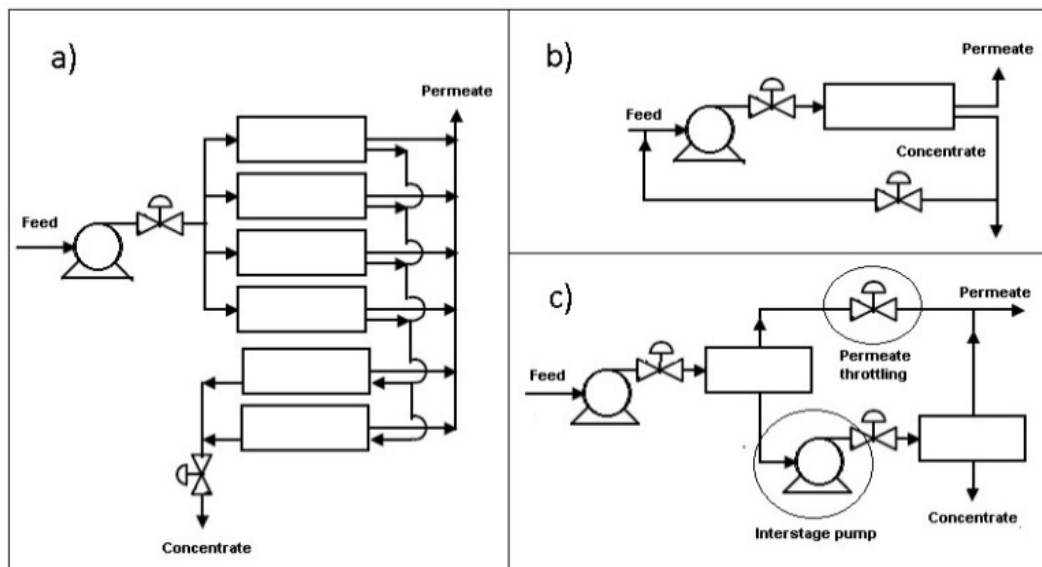
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35 **1.0 INTRODUCTION**

36 Urbanization, agricultural practices and industrialization increases both the water demand
 37 and the volume of wastewater produced. Also, researchers draws attention to increasing
 38 restriction on discharge limits due to environmental issues and the need to reclaim and reuse
 39 wastewaters^[1-2]. Membrane bioreactor (MBR) process is replacing conventional activated
 40 sludge process as a secondary treatment unit, especially for domestic wastewater treatment,
 41 nowadays. Stability of MBR effluent quality represents a favorable approach for water reuse^[3].
 42 However, high amounts of salinity, nutrients and hardness make industrial secondary effluents
 43 from MBR process not reusable and sometimes not even dischargeable^[4]. MBR effluent should
 44 be demineralized via one of the desalination methods such as nanofiltration (NF) and reverse
 45 osmosis (RO) when the salinity is high prior to reuse of MBR effluent for various purposes. On
 46 the other hand, concentrate streams generated from NF and RO processes are of important
 47 concern. Indeed, untreated or improperly managed brine can lead to serious negative effects on
 48 the environment^[5]. Both the need to reclaim water from industrial or domestic effluents and to
 49 decrease discharge amount let researchers use membranes for tertiary treatment.

50 Several flow configurations of NF and RO systems are used to reduce brine generation
 51 (Figure 1). Among them, membrane modules can be connected in series, in which the
 52 concentrate stream of the first membrane is fed to the second one ^[6].



61 **Figure 1.** System flow configurations for concentrate reduction: a) pyramidal design
 62 b) concentrate recirculation c) membrane-in-series (modified from Hydranautics)^[6]
 63
 64

65 Reuse of wastewater for irrigation is widely used in countries where fresh water resources
 66 are becoming insufficient. Nutrient content of secondary effluents and the relative cost of

67 reclaiming water for irrigation unlike reclaiming for drinking purposes make this process
68 favorable [7]. Also, some studies showed that the amount of irrigation water is the first to be
69 decreased rather than drinking or process water when the water supplies started to be
70 insufficient. That being said, water reclamation for agricultural irrigation gains even more
71 importance[8]. However, reclaimed water should obey to water quality guidelines for
72 agricultural irrigation which requires development of a proper water management strategy. For
73 this purpose, some concepts such as soil permeability and salinity of irrigation water should be
74 checked carefully.

75 High sodium concentration disperses the soil clay and causes the soil to become hard and
76 compact when the soil is dry and reduces the rate of water penetration when the soil is wet. The
77 low infiltration rate can cause deficiencies of several nutrients. In contrast, the effect of calcium
78 and magnesium is to cause flocculation of soil clay, thus promoting and maintaining good soil
79 structure. Thus, the Sodium Adsorption Ratio (SAR) is a standart indicator of the flocculation
80 or dispersion of aggregates in soil, which is directly influencing the structure of the soil and
81 hence the permeability. SAR is defined in Equation 1 as an expression of the sodium hazard of
82 irrigation water [8].

83

84

$$\text{SAR} = \frac{[\text{Na}^+]}{\sqrt{\frac{[\text{Ca}^{2+}] + [\text{Mg}^{2+}]}{2}}} \quad (\text{Equation 1})$$

85 where the concentrations are in meq/L. Infiltration of irrigation water to the soil is not only
86 connected to SAR but also to electrical conductivity (EC). FAO and WHO guidelines offer a
87 relation between SAR and EC to assess the suitability of irrigation water.

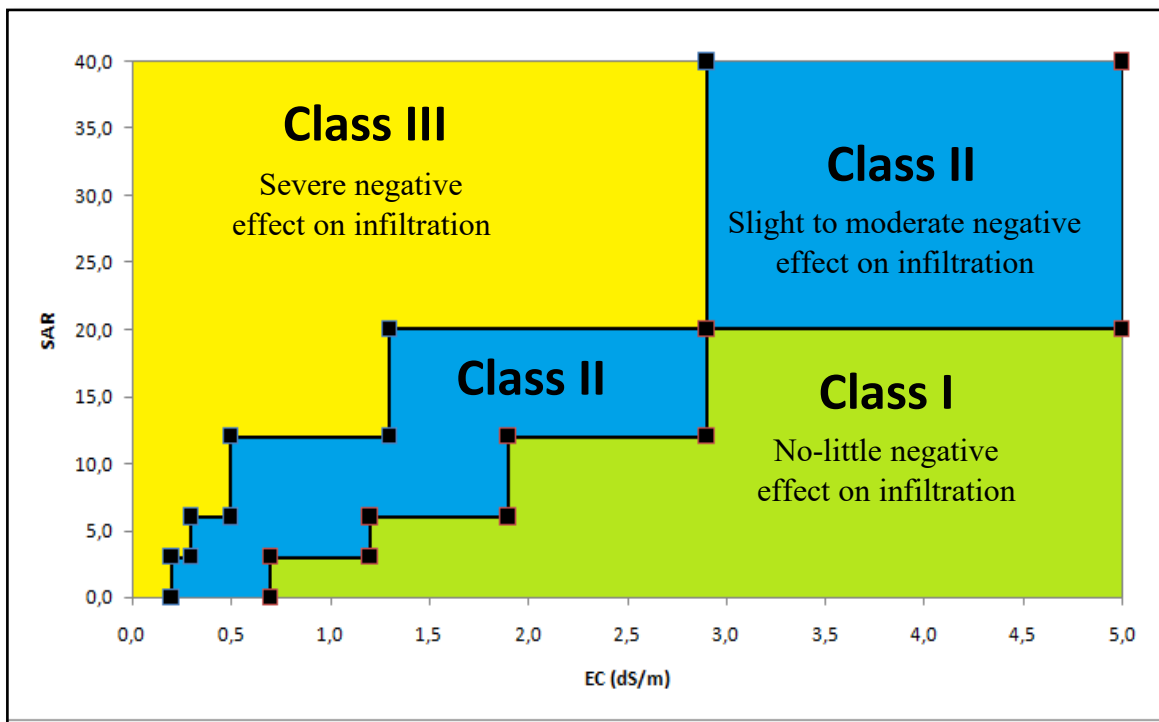
88 The values given in Table 1 were illustrated in Figure 2 for better understanding of the effect
89 of irrigation water on soil permeability. Classes I, II and III correspond to degrees of restriction
90 on use as none, slight to moderate and severe, respectively. It must be noted that Class I water
91 means the water will not have negative effects on soil permeability but does not say anything
92 about salinity. While high EC can toxicate plants, low EC will mean low amount of minerals
93 for plant to sustain its life. Therefore, a suitable irrigation water should obey to both SAR-EC
94 classification and salinity criteria.

95 Potassium adsorption ratio (PAR) is another parameter to be investigated to assess the effect
96 of potassium ion to soil clay dispersion, which is calculated by Equation 2[11].

97 **Table 1.** Classification of irrigation water regarding soil permeability [9]

Parameter	Degree of Restriction on Use		
	None	Slight to Moderate	Severe
SAR	EC (mS/cm)		
0-3	>0.7	0.2-0.7	<0.2
3-6	>1.2	0.3-1.2	<0.3
6-12	>1.9	0.5-1.9	<0.5
12-20	>2.9	1.3-2.9	<1.3
20-40	>5.0	2.9-5.0	<2.9

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Figure 2. Illustration of SAR-EC classification for irrigation water ^[10]

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$$PAR = \frac{[K^+]}{\sqrt{\frac{[Ca^{2+}] + [Mg^{2+}]}{2}}} \quad (\text{Equation 2})$$

107 FAO, WHO and USEPA guidelines and regulations offers also other parameters for
 108 assessing suitability of water for agricultural irrigation. Three documents are cross-checked and
 109 Table 2 contains standards that fit with all of them.

110

111 **Table 2.** Standard limits for agricultural irrigation ^[12-14]

Parameter	Unit	Degree of Restriction on Use		
		None	Slight to Moderate	Severe
EC	mS/cm	<0.7	0.7-3	>3
SAR	-	Check Table 1		
TDS	g/L	<0.45	0.45-2	>2
NO ₃ -N	mg/L	<5	5-30	>30
TN	mg/L	<5	5-30	>30
PO ₄ -P	mg/L	-	<2 or <5	-
PAR	-	<5	5-10	>10
HCO ₃	mg/L	<90	90-500	>500
Boron	mg/L	<0.7	0.7-3	>3
Chloride	mg/L	<106.5	>106.5	-
Sodium	mg/L	<69	>69	-
Free Cl	mg/L	<1	1-5	>5
TSS	mg/L	<50	50-100	>100
pH	-	6-9		
Turbidity	NTU	<2		

112

113 It must be noted that chloride and sodium standards are taken for sprinkler irrigation since it
 114 is already more efficient than surface irrigation.

115 Although different plants have different tolerance levels to various components of water
 116 (fruit plants have low tolerance to ion toxicity and boron while harsh-climate plants like date
 117 palm are highly resistant), Table 2 can be used as a preliminary information for the evaluation.

118 Previously, we checked the desalination performances of different NF and RO membranes
 119 by using a cross-flow flat-sheet membrane test unit. Permeates of RO membrane (GE-Osmonics
 120 AG membrane) and NF membranes (GE-Osmonics HL, DL membranes) were compared with

121 irrigation water standards for its reuse in irrigation ^[15]. Elsewhere, application NF and RO
122 processes for reclamation and reuse of industrial wastewater treated with MBR process was
123 investigated using a mini-pilot scale membrane test system. The qualities of permeates obtained
124 by BW30-RO and NF-90 membranes were evaluated according to irrigation water standards
125 ^[16]. At the same time, effect of membrane type on product water quality for industrial usage as
126 cooling and boiling feed water as well as process water in paper and textile industries was
127 studied ^[17]. Also, effect of pressure on desalination of MBR effluents with high salinity by using
128 NF and RO processes for reuse in irrigation was reported ^[18].

129 Concentrate management in pressure driven membrane processes is an important and
130 challenging task to be solved. Besides the product water, the brine generation from the
131 membrane filtration processes is of important environmental concern. Indeed, an appropriate
132 concentrate management is required to avoid serious negative effects on environment ^[5].

133 Ocean concentrate disposal is possible if specific discharge regulation are observed ^[19].
134 Another method to eliminate the generated brine is leaving it to natural evaporation process
135 creating a solid waste ^[20]. However, this process requires high time and area and can be not
136 feasible when high amounts of concentrate are to be treated ^[21]. Therefore, is necessary to
137 reduce, as much as possible, the concentrate amount before further treatments or discharge.

138 Many different flow configurations of RO systems were utilized to reduce the brine
139 generation. Membrane modules can be connected in series with or without an interstage pump
140 or permeate throttling, connected in a pyramidal design, etc. Also the concentrate recirculation
141 configuration allow to increase the water recovery recycling part of the concentrate stream ^[6].
142 These configurations are shown in Figure 1.

143 In our previous study, effect of concentrate recirculation on the product water quality of
144 integrated MBR+ NF processes in sequential mode was investigated for wastewater reclamation
145 and industrial reuse ^[22]. Using the same membrane tests system, MBR-treated industrial
146 wastewater was further processed by BW30, a brackish water RO membrane in concentrate
147 recirculation configuration. Membrane and system performances were assessed, including
148 product water quality and the concentrate retention ratio ^[23]. The aim of this study is to reduce
149 the amount of concentrate stream of RO and NF processes while assessing alternative use of
150 the permeate stream for agricultural irrigation and hence to establish a two-way solution for
151 problems of industrial-scale operations. For this, various combinations of different membrane
152 pairs were employed in batch mode of operation by feeding concentrate stream of the first
153 membrane to the second membrane in membrane-in series mode.

154

155 **2. MATERIAL AND METHODS**

156 In this study, three different membranes (NF90, NF270 and BW30) were used in such two
 157 membrane-in series configurations as NF90+NF90, BW30+BW30, BW30+NF270
 158 combinations using the same mini pilot system for water recovery from MBR effluent at
 159 wastewater treatment plant of ITOB Organized Industrial Zone, Menderes-Izmir, Turkey. The
 160 concentrate stream of the second membrane and the total permeate were recycled to the feed
 161 tank to operate in batch mode. Concentrate stream of the first membrane was fed to the second
 162 membrane in this membrane-in-series configuration to increase the water recovery (as shown
 163 in Figure 1c without throttling or interstage pump). Properties and informations about the
 164 membranes employed are given in Table 3.

165

166 **Table 3.** Membrane properties ^[24]

Membrane	Manufacturer	Material	NaCl rejection (%)	pH Interval	Max Temperature (°C)	Max Pressure (bar)
BW30-RO	Dow-Filmtec	Polyamide Thin-Film Composite	99.5	2-11	50	41
NF90-NF	Dow-Filmtec	Polyamide Thin-Film Composite	95.0	2-11	45	41
NF270-NF	Dow-Filmtec	Polyamide Thin-Film Composite	50.0	2-11	45	41

167

168

169 Experiments were carried out using a mini pilot system installed at wastewater treatment
 170 plant. Photographs of the mini pilot membrane test system are given in Figure 3.

171 MBR effluent discharged from wastewater treatment plant was used as feed stream for the
 172 membrane tests using mini pilot membrane test system. Properties of MBR effluent used as
 173 feed are given in Table 4.

174 The flow configuration used during membrane tests is depicted in Figure 4. It must be noted
 175 that while this pilot test system has six membranes, only five of them (two NF 90, two BW 30
 176 and one NF 270) were used for tests.

177 The MBR effluent utilized as feed was collected in a 500 L of feed tank and pumped to the
178 membrane test system. Average feed flow rate supplied by the pump was 4.33 L/min varied by
179 6% margin. Permeate and concentrate streams were fed back to the feed tank and experiments
180 were carried out in batch mode for 6 h. The applied pressure at the inlet of the first membrane
181 of the pair is 20 bar for all experiments.

182 Samples of feed, permeate and concentrate streams were taken at each hour for lab analysis
183 while the field analysis (EC, TDS, salinity, pH) were performed online by means of a
184 conductometer and a pH-meter for all samples at each 15 min. The quality parameters such
185 TSS, color, turbidity, COD, SiO₂, total nitrogen, nitrite-N, phosphate-P and ammonium-N
186 concentrations were measured by using Hach-Lange chemical measurement kits by means of
187 DR 3900 benchtop VIS model spectrophotometer. Most cation (Ca²⁺, Mg²⁺, K⁺ and Na⁺)
188 concentrations were measured by an atomic absorption spectrometer (Shimadzu AA-7000)
189 while anion concentrations (NO₃⁻, Cl⁻ and SO₄²⁻) were measured by an ion chromatography
190 (Shimadzu LC-10Ai). Bicarbonate concentration was measured by titrating samples with 0.05
191 M HCl solution using methyl orange indicator. Permanent hardness were measured by titration
192 of samples with pre-prepared 0.01 M Na₂EDTA solution while using eriochrome black T as
193 indicator.

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195

196 **Figure 3.** Photographs of mini pilot membrane test system

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209 **Table 4.** Properties of the MBR effluent used as feed for membrane tests

MBR Effluent Parameter	Two membrane-in series tests		
	BW30 to BW30	NF90 to NF90	BW30 to NF270
TSS (mg/L)	1.3 ± 0.2	2.0 ± 0.3	1.0 ± 0.2
Color (mg/L Pt-Co)	12.0 ± 1.4	25.5 ± 3.2	9.0 ± 1.5
Turbidity (NTU)	0.2 ± 0	1.1 ± 0.2	0.5 ± 0.2
COD (mg/L)	19.6 ± 1.5	17.5 ± 1.3	20.4 ± 1.5
SiO ₂ (mg/L)	5.9 ± 1.9	7.2 ± 2.5	6.8 ± 2.3
HCO ₃ (mg/L)	16.8 ± 0.9	15.0 ± 1.0	15.3 ± 0.9
Na (mg/L)	395 ± 17	399 ± 21	462 ± 23
Ca (mg/L)	101 ± 5	100 ± 5	113 ± 5
K (mg/L)	130 ± 12	139 ± 13	92.0 ± 12
Mg (mg/L)	22.8 ± 0.9	25.5 ± 1.1	34.7 ± 1.4
SO ₄ (mg/L)	339 ± 41	406 ± 43	442 ± 37
Cl (mg/L)	488 ± 35	495 ± 34	497 ± 39
Hardness (mg/L CaCO ₃)	367 ± 9	358 ± 11	448 ± 12
NO ₃ -N (mg/L)	35.0 ± 1.2	38.0 ± 1.1	35.6 ± 1.2
Total Nitrogen (TN) (mg/L)	36.8 ± 3.4	38.5 ± 4.1	38.5 ± 3.6
NO ₂ -N (mg/L)	0.03 ± 0	<0.02	0.03 ± 0

PO₄-P (mg/L)	<0.5	<0.5	<0.5
NH₄-N (mg/L)	<0.2	<0.2	<0.2
EC (μS/cm)	4200 ± 75	4300 ± 78	3800 ± 74
TDS (mg/L)	1400 ± 21	1400 ± 19	1500 ± 18
Salinity (‰)	1.4 ± 0	1.4 ± 0	1.5 ± 0
pH	7.7 ± 0.3	7.6 ± 0.3	7.7 ± 0.3

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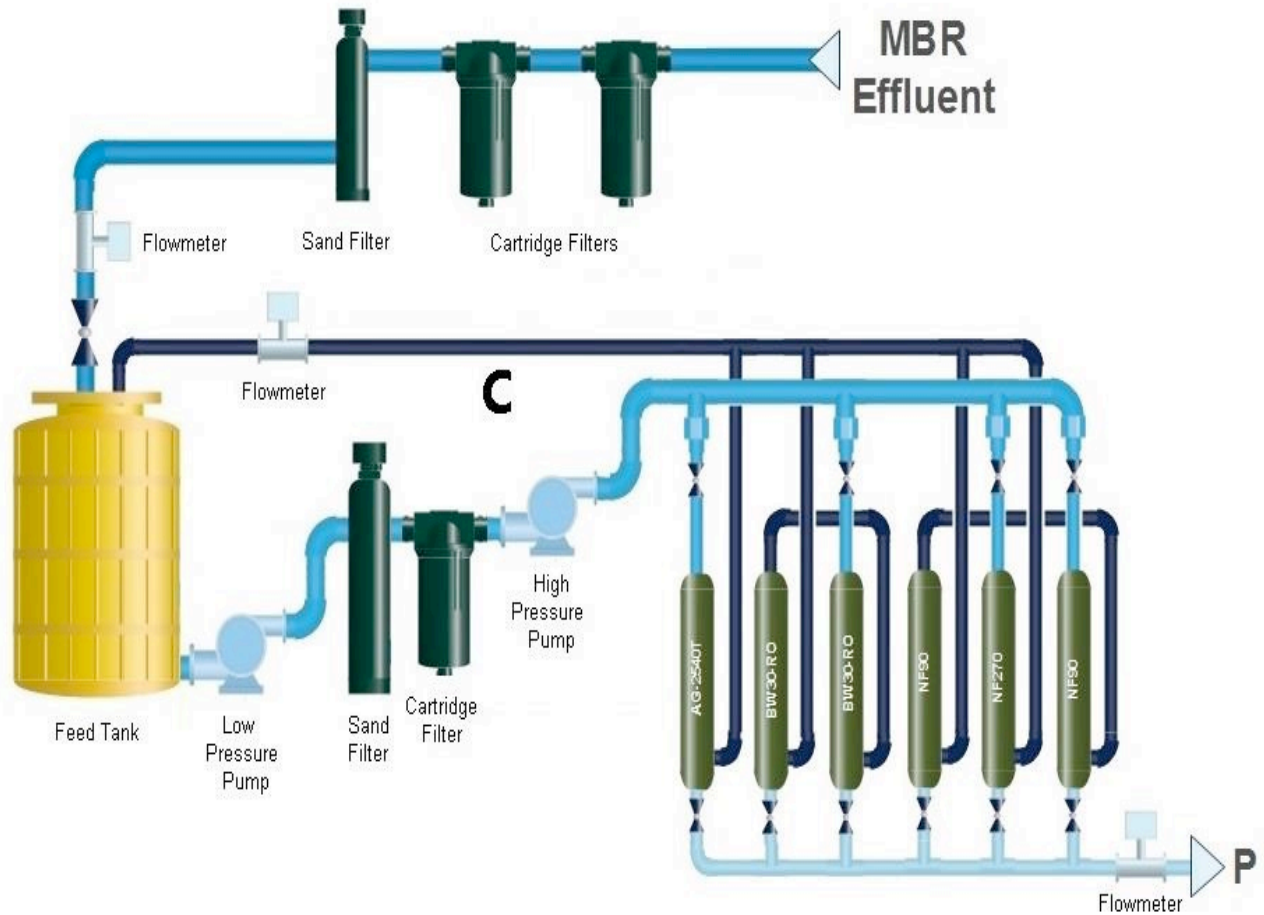
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220 **Figure 4.** Flow configuration in mini-pilot membrane test system

221 (F: Feed, C: Concentrate, P: Permeate)

222

223

224 Water recovery is calculated by Equation 3:

225
$$\text{Water recovery (\%)} = \frac{P}{M} \quad \text{(Equation 3)}$$

226 Permeate flux is calculated by Equation 4. Permeate fluxes are normalized according to a
 227 standart temperature via Equation 5:

228
$$J_P \left(\frac{L}{m^2 \cdot h} \right) = \frac{P}{A} \quad \text{(Equation 4)}$$

229
$$J_{P_{adj}} = \frac{J_P}{1.03^{(T-25)}} \quad \text{(Equation 5)}$$

230 Rejection for any parameter is calculated by Equation 6

231
$$R (\%) = \frac{C_f - C_p}{C_f} \quad (\text{Equation 6})$$

232 where P is permeate flow rate, M is MBR effluent flow rate, J_p is the permeate flux, A is the
233 active membrane surface area, T is temperature ($^{\circ}\text{C}$), R is the rejection of a
234 component/parameter, C_f and C_p are concentrations or values of any parameter in feed and
235 permeate, respectively.

236 Water recovery of individual membranes were calculated by manual flow readings of
237 permeate streams and concentrate stream of the first membrane performed in each 10 min with
238 30 s of sampling time. Manually calculated permeate flow rates are then divided with the feed
239 amount to each membrane (feed flow rate to the first membrane, concentrate flow rate produced
240 by the first membrane to the second membrane) to calculate water recoveries.

241 MBR effluent and product waters obtained from various two membrane-in series
242 configurations (NF90+NF90, BW30+BW30, BW30+NF270) were also evaluated for their
243 reuse in agricultural irrigation according to FAO, WHO and USEPA guidelines and regulations.

244

245 **3.0 RESULTS AND DISCUSSION**

246 The system performances for different two membrane-in series configurations were
247 compared in terms of permeate flux, solute rejections and permeate quality. System water
248 recoveries were kept almost constant as 81.5-81.6%. Average permeate flux was only slightly
249 higher for the NF90+NF90 combination compared to the BW30+BW30 system, while the
250 salinity removal was 92.9% in the first case and 97.4% in the second one. This was due to the
251 high performances of NF90 membrane, which are close to typical performances of brackish
252 water RO membranes. Significant increase in the permeate flux was obtained with a
253 BW30+NF270 system. However, the permeate quality was severely affected, obtaining 42.9%
254 of salinity rejection.

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Table 5. System performances in the different configurations

System configuration		BW30 to BW30	NF90 to NF90	BW30 to NF270
Average water recovery (%)	1 st Membrane	48.5%	39.4%	28.4%
	2 nd Membrane	64.0%	69.6%	74.3%
	TOTAL	81.5%	81.6%	81.6%
Average normalized permeate flux (L/m ² .h)	1 st Membrane	51.5 ± 0.5	60.5 ± 0.6	43.2 ± 0.4
	2 nd Membrane	35.1 ± 0.4	65.0 ± 0.7	80.9 ± 0.8
	TOTAL	86.6	125.6	124.1
Salinity rejection (%)*		97.4	93.7	42.9

261 *Individual salinity rejections of the membranes are unknown because the system configuration does not allow
262 to take samples between membranes but only to measure flux.

263

264 Since by definition, water permeability is the inverse of the resistance of the membrane to
265 the permeate flow (closely related to pore diameters), permeate flux of NF270 is much bigger
266 than NF90 and BW30. While the salt rejection of NF90 is defined close to reverse osmosis
267 membranes with its nearly 200 Da; NF270 is loosest NF membrane produced by Dow with its
268 400 Da, hence lower salt rejection was already expected [25].

269 Although the fluxes of NF90+NF90 and BW30+BW30 combinations are nearly equal and
270 salinity rejection performance of BW30+BW30 pair is higher, it must also be noted that NF
271 membranes require lower applied pressure than RO membranes. In this case, 20 bar of operating
272 pressure was applied for effective comparison between membranes. However, 10-15 bar of
273 operating pressure is mostly used in real NF applications.

274 Average rejections of some selected parameters from MBR effluent using various two
275 membrane-in series configurations are given in Table 6.

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287 **Table 6.** Percent average rejections of some selected quality parameters

Parameter	BW30 to BW30	NF90 to NF90	BW30 to NF270
TSS	~100	~100	~100
Color	~100	~100	88.9 ± 0.8
Turbidity	64.2 ± 3.4	74.8 ± 3.5	49.3 ± 2.4
COD	29.8 ± 3.5	49.7 ± 7.1	75.5 ± 9.1
SiO₂	87.8 ± 2.5	90.3 ± 2.3	13.2 ± 0.4
HCO₃	84.2 ± 1.1	87.1 ± 1.5	47.9 ± 0.9
Na	94.7 ± 0.6	91.2 ± 1.2	50.4 ± 0.8
Ca	98.5 ± 1.1	98.7 ± 1.2	62.1 ± 1.0
K	96.1 ± 2.0	92.2 ± 2.3	40.7 ± 1.5
Mg	>99.9	>99.9	83.9 ± 1.9
SO₄	>99.5	>99.5	>99.5
Cl	97.3 ± 1.1	94.0 ± 1.4	21.7 ± 0.7
Hardness	>99.7	97.5 ± 0.5	76.1 ± 0.9
NO₃-N	89.1 ± 1.3	77.1 ± 1.1	13.0 ± 0.4
TN	86.2 ± 7.4	71.7 ± 6.7	18.6 ± 3.1
Conductivity	97.0 ± 0.4	92.9 ± 0.3	41.4 ± 0.3
TDS	97.3 ± 0.4	93.4 ± 0.3	42.6 ± 0.3
Salinity	97.4 ± 0.5	93.7 ± 0.3	42.9 ± 0.3

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289 Feed and permeate compositions along with their evaluations for degrees of restriction on
 290 agricultural irrigation were given in Table 7.

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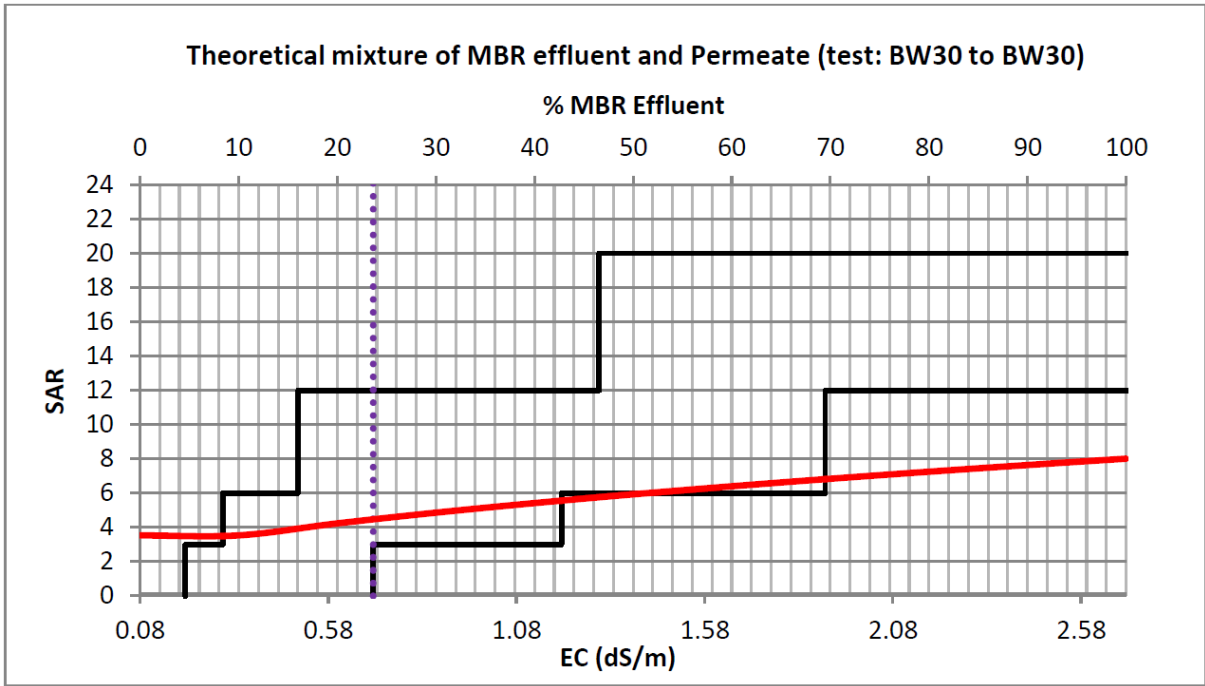
300 **Table 7.** Average feed and permeate compositions compared with agricultural irrigation
 301 standards

Parameter	Unit	Average Feed	Permeate			Degree of Restriction on Use		
			BW30 to BW30	NF90 to NF90	BW30 to NF270	None	Slight to Moderate	Severe
EC	mS/cm	4.1	0.1	0.2	1.7	<0.7	0.7-3	>3
SAR	-	8-10	3.8	7.7	8.6			
TDS	g/L	1.4	<0.1	0.1	0.9	<0.45	0.45-2	>2
NO ₃ -N	mg/L	36.2	3.8	8.7	31.0	<5	5-30	>30
T-N	mg/L	37.9	5.1	10.9	31.3	<5	5-30	>30
PO ₄ -P	mg/L	<0.1	<0.1	<0.1	<0.1	-	<2 or <5	-
PAR	-	1.0	0.3	0.8	1.2	<5	5-10	>10
HCO ₃	mg/L	15.7	2.7	1.9	7.9	<90	90-500	>500
B	mg/L	2.3	<0.5	1.1	2.0	<0.7	0.7-3	>3
Cl	mg/L	493.3	13.4	29.6	389	<106.5	>106.5	-
Na	mg/L	418.7	21.0	35.2	229	<69	>69	-
Free chlorine	mg/L	<0.1	<0.1	<0.1	<0.1	<1	1-5	>5
TSS	mg/L	1.4	0	0	0	<50	50-100	>100
pH	-	7.7	6.4	6.8	7.2	6-9		
Turbidity	NTU	0.6	0.1	0.3	0.2	<2		

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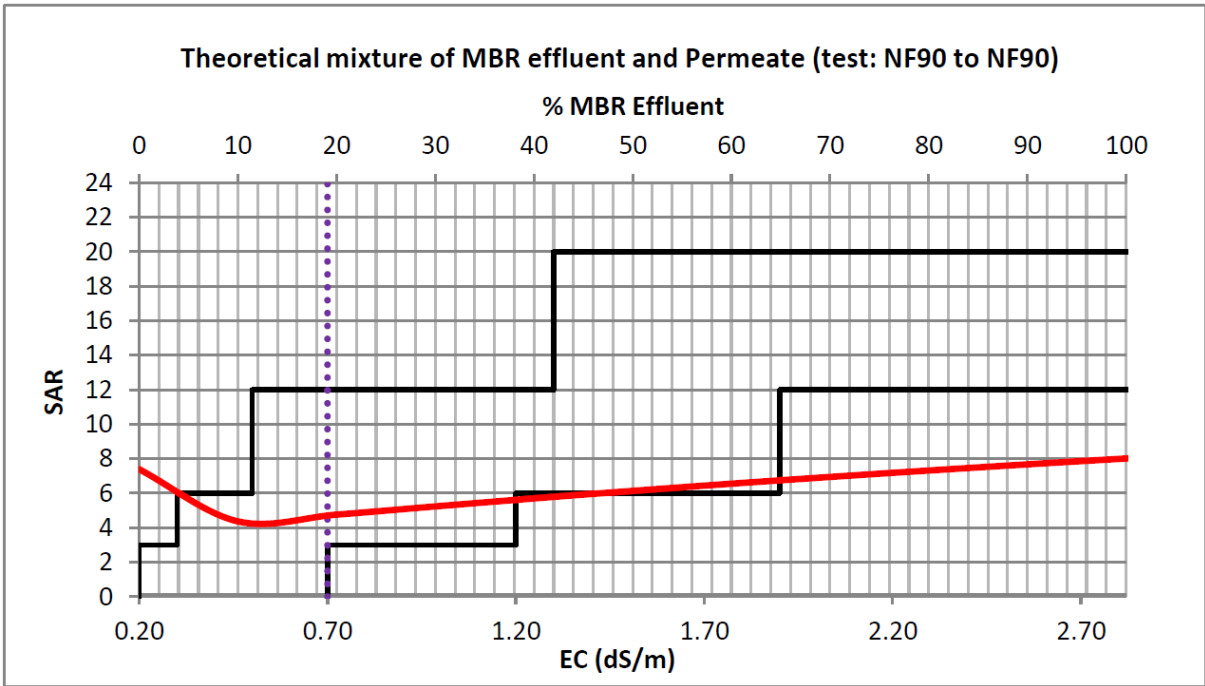
303 It is obvious that MBR effluent is not suitable for agricultural irrigation due to excessive
 304 amounts of sodium and chloride ions (causing ion toxicity), salinity and nitrate. BW30+NF270
 305 combination produced water with excess sodium, chloride and nitrate ions although their
 306 concentrations are not as much as in MBR effluent. The permeates of BW30+BW30 and
 307 NF90+NF90 pairs are good irrigation waters for boron-sensitive and salinity-sensitive plants.

308 Previous studies ^[10, 26, 27] suggested to mix permeate streams with MBR effluent in certain
 309 amounts to increase its soil permeability and to remineralize it. Therefore, theoretical mixtures
 310 are calculated for both permeate of NF90+NF90 combination with MBR effluent and permeate
 311 of BW30+BW30 combination with MBR effluent. Soil permeability (SAR-EC classification)
 312 graphs were shown in Figures 5 and 6.



313

314 **Figure 5.** SAR-EC classification of theoretical mixture of MBR effluent and permeate of
 315 BW30+BW30 combination



316

317 **Figure 6.** SAR-EC classification of theoretical mixture of MBR effluent and permeate of
 318 NF90+NF90 combination

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320 Water with a first class soil-permeability seems to be unachievable without high salinity
321 (vertical dotted lines are salinity criteria), but close to first class soil-permeability together with
322 a fit to other irrigation standards which are achievable. This can be obtained by a theoretical
323 mixtures of 85-95% of NF90+NF90 and BW30+BW30 permeates mixed with 15-5% of MBR
324 effluent. Percentage of MBR effluent can be increased if the resistance of the plant for salinity
325 and boron is high.

326 While MBR effluent is toxic to the plants, especially for salt-sensitive plants, NF90 and
327 BW30 permeates do not include enough minerals to sustain the plant life. Theoretical mixtures
328 using 15-5% of MBR effluent and 85-95% of BW30+BW30 or NF90+NF90 permeates are
329 thought to solve this problem and create useable water for agricultural irrigation.

330 Findings are partially in contrast with the findings of other researchers using theoretical
331 mixtures approach for the agricultural irrigation although their focus was on the treatment of
332 domestic wastewaters, while this study is based on secondary-treated industrial wastewaters.
333 While they thought the sustainability of plant life would be better if the permeate is re-
334 mineralized via addition of some MBR effluent, but there was no any indicator for its effect on
335 soil permeability.

336 SAR values –when combined with EC values- mathematically proved the comments of other
337 researchers and shows the need of demineralization of NF90 and BW30 pair permeates for
338 agricultural irrigation purposes. Figures 5 and 6 (created from placement of research findings
339 on Figure 2) show a plausible way to assess the suitability of use of any type of water in
340 agricultural irrigation.

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353 **Table 8.** Properties of optimum theoretical mixtures of NF90+NF90 and BW30+BW30
 354 permeates with MBR effluent

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Two membranes-in series configuration	Unit	BW30+BW30	NF90+NF90
MBR effluent (%)		12	9
EC	mS/cm	0.4	0.4
SAR	-	3.9	4.3
TDS	g/L	0.2	0.2
NO₃-N	mg/L	7.6	11.3
TN	mg/L	8.9	13.4
PO₄-P	mg/L	0.1	0.1
PAR	-	0.4	0.5
HCO₃	mg/L	8.9	13.4
B	mg/L	<0.5	0.5-1.0
Cl	mg/L	70.3	71.4
Na	mg/L	65.8	68.0
Free chlorine	mg/L	<0.1	<0.1
TSS	mg/L	0.2	0.2
pH	-	6.5	6.9
Turbidity	NTU	0.1	0.3

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357 SAR-EC criteria of those theoretical mixtures can be checked from Figures 5 and 6 showing
 358 that they are in second class of water quality which is close to the first class of water quality.

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368 **4.0 CONCLUSIONS**

369 Water recoveries were obtained as 81.5% for BW30+BW30 pair and 81.6% for other two
370 pairs (NF90+NF90, BW30-NF270) by the application of membrane-in-series configuration
371 which are much higher than water recovery that is usually obtained by one membrane at around
372 40-50% without considerable fouling.

373 Among various two membrane-in series configurations, NF90+NF90 and BW30+BW30
374 combinations are found to be favorable with 93.0% and 97.4% of conductivity rejections,
375 respectively while BW30+NF270 pair was not successful to decrease salinity of MBR effluent
376 with its only 42.4% of conductivity rejection.

377 Even though the concentrate stream is more saline than the feed solution (MBR effluent), its
378 amount is reduced more than 81.5%. The rest (less than one fifth of before) would be easier to
379 be dealt with natural evaporation or any other waste management option.

380 Theoretical mixtures of MBR effluent and BW30+BW30 or NF90+NF90 permeate as 15-
381 5% of MBR effluent and 85-95% of permeate of BW30+BW30 or NF90+NF90 pair were found
382 suitable theoretical mixture for use in agricultural irrigation.

383

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