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Tuning polymorphism in 2,3-thienoimide capped oligothiophene based field-effect transistors by implementing vacuum and solution deposition methods

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We report on the investigation of the influence of the molecular packing and film morphology on the field-effect charge mobility in 2,3-thienoimide-based oligothiophenes semiconductors (C_n-NT4N). Organic field-effect transistors are realized by implementing both vacuum and solution methods in order to control the solid-state phase of the active layer. Thermal sublimation in a high vacuum chamber and supersonic molecular beam deposition were used as vacuum-based fabrication approaches for preparing thin films, while lithographically controlled wetting was used, as a solution-deposition technique, for the fabrication of the microstructured films. Thermal sublimation leads to thin films with a phase packing showing ambipolar behaviour, while supersonic molecular beam deposition enables, by varying the deposition rate, the formation of two different crystal phases, showing ambipolar and unipolar field-effect behaviours. On the other hand, lithographically controlled wetting enables the formation of C_n-NT4N microstructured active layers and their implementation in field-effect transistors.

Introduction

Polymorphism is emerging in organic optoelectronics as a powerful tool, alternate to molecular design, for controlling

the packing dependent functionalities of organic materials, such as optical, charge transport and optoelectronic properties.¹ Polymorphic pentacenes,^{2,3} tetrathiafulvalenes^{4,5} and rubrene⁶ materials have been widely investigated to study the influence of molecular packing on the charge transport capability in organic semiconductors. Recently, a clear relationship between the polymorph type and p-type mobility was established for fluorinated 5,11-bis(triethylsilyl)ethynylanthradithiophene.⁷ Similarly, Kim *et al.* demonstrated polymorphism as a tool to control the optoelectronic and charge transport functionalities in two fluorene-based π -conjugated materials,⁸ while Wang *et al.* suggested the possibility of lasing in different colors by exploiting a single polymorphic (4,4'-(thiazolo[5,4-*d*]-thiazole-2,5-diyl)bis-(*N,N*-diphenylaniline)) compound.⁹

Tuning the optical, electrical and optoelectronic properties of organic semiconductors through the control of their polymorphic phases is particularly important for organic light emitting transistors (OLETs),¹⁰ whose electronic, optoelectronic and photonic multifunctionality strictly depends on the active material properties under operational conditions.

Thiophene molecular semiconductors are highly versatile organic materials,^{11,12} whose properties are representative of a broad range of organic semiconductor materials. 2,3-Thienoimide

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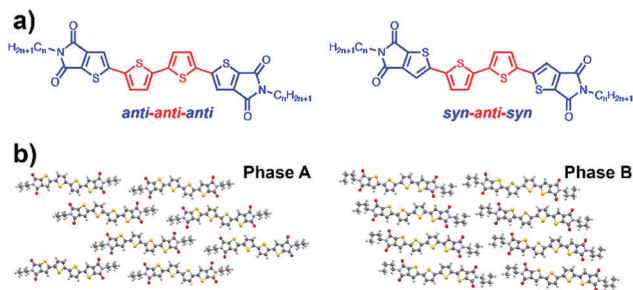


Fig. 1 (a) Molecular structure and conformational polymorphism types of the C_n -NT4N semiconductors herein studied; (b) molecular packing.

end-capped oligothiophenes, also known as thienopyrrolyldione-based oligothiophenes (C_n -NT4N), show polymorphism¹³ and have the peculiarity of combining ambipolar semiconducting capability and electroluminescence, making them suitable materials for the fabrication of both ambipolar organic field-effect transistors (OFETs)¹⁴ and single-layer OLETs.¹⁵ A wide library of C_n -NT4N compounds has been synthesized and characterized to study the effect of the N -alkyl end shape and size on the optical, self-assembly and charge transport properties.¹⁶ We have recently reported on the relationship between the molecular structures and polymorphism properties of C_n -NT4N derivatives.¹⁷ Interestingly, C_n -NT4N derivatives with an even number of carbon atoms in the alkyl end chains display at least two conformational polymorphs that adopt different molecular packings, namely phase A and phase B, as a consequence of the different conformational arrangement of the 2,3-thienoimide end-moieties with respect to the adjacent thiophene core, *i.e.* *anti* or *syn* conformation (Fig. 1a). Consequently, while form A shows a columnar π -stacking packing, phase B shows a brick wall-like packing (Fig. 1b).¹⁷ The occurrence of different molecular packings may limit the reproducibility, stability, and reliability of devices based on these materials, therefore, the control of polymorph formation during film deposition is crucial for their applicability.

Herein, we report on the impact of the deposition technique on the charge transport properties of thin and microstructured films of 2,3-thienoimide-based molecular semiconductors and investigate their molecular packing, morphology and optical properties. We have chosen the butyl (C_4 -NT4N) and hexyl-ended (C_6 -NT4N) oligomers because they are vacuum- and solution-processable semiconductors, respectively, and both exhibit polymorphism. Thermal sublimation in a high vacuum chamber and Supersonic Molecular Beam Deposition (SuMBD) were used as vacuum-based fabrication approaches for preparing thin films, while Lithographically Controlled Wetting (LCW) was used, as the solution-deposition technique, for the fabrication of microstructured films. We observed that thermal sublimation leads to thin films exhibiting ambipolar behaviour, while SuMBD enables the formation of both ambipolar and unipolar thin films. Ambipolar thin films are characterized by a polymorph B with brick-wall like packing, whereas unipolar thin films exhibit a new molecular packing that hampers the hole transport. We further demonstrated that spatial confinement of solution-deposition of

2,3-thienoimide capped oligothiophene achieved by LCW is an effective strategy for fabricating microstructured films with controlled features and suitable for their implementation in devices, overcoming the shortcomings of inhomogeneous films that are achieved by drop casting and spin coating. Solution-processed films of C_6 -NT4N were deposited from both anisole solution at 80 °C and 1,2-dichlorobenzene (1,2-DCB) solution at room temperature. The microstructured films exhibit ambipolar behaviour and are characterized by phase B regardless of the conditions used, while films deposited by drop casting at room temperature adopt the phase A structure, highlighting the key role of spatial confinement in the polymorphism control of wet-deposited C_6 -NT4N films.

Results and discussion

Vacuum deposition

Thin films of C_4 -NT4N were deposited by both vacuum thermal deposition and SuMBD and used as active layer of OFETs. OFETs were built in a bottom gate/bottom contact architecture using hexamethyldisilazane (HMDS)-treated Si/SiO₂ substrates to minimize charge trapping at the interface between the organic semiconductor and the dielectric¹⁸ and for guaranteeing a more hydrophobic interface suitable for C_n -NT4N growth.

As demonstrated by the transfer characteristics in the saturation regime (Fig. 2a) and summarized in Table 1, thin films of C_4 -NT4N deposited at 0.1 Å s⁻¹ by thermal sublimation exhibit ambipolar field-effect transport with saturated electron and hole mobilities up to 4.5×10^{-2} cm² V⁻¹ s⁻¹ and 6.7×10^{-4} cm² V⁻¹ s⁻¹, respectively. In agreement with a previous report on C_4 -NT4N deposited on

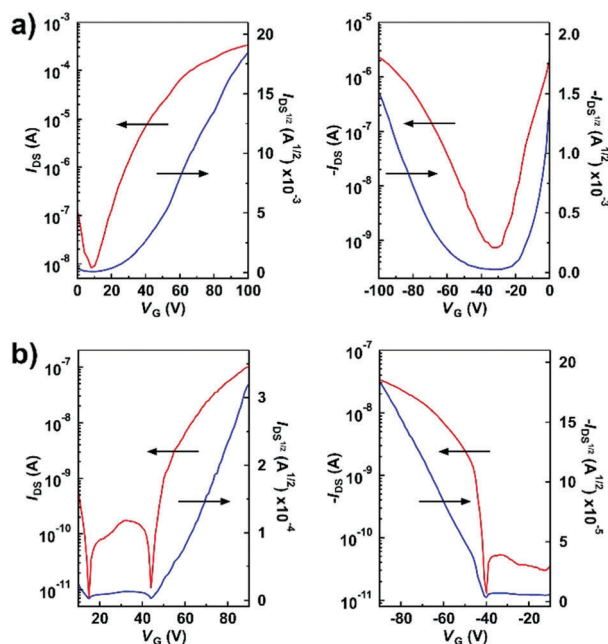


Fig. 2 Transfer characteristics measured at positive and negative source-drain bias of representative field-effect transistors based on C_4 -NT4N films deposited by thermal sublimation (a) and SuMBD (b) at 0.1 Å s⁻¹ (drain voltage $V_{SD} = \pm 100$ V).

Table 1 Electrical data of bottom gate/bottom contact OFETs based on vapour-deposited films of C₄-NT4N

	Rate (Å s ⁻¹)	Phase	μ_e (cm ² V ⁻¹ s ⁻¹)	V_{th}^e (V)	I_{on}^e/I_{off}^e	μ_h (cm ² V ⁻¹ s ⁻¹)	V_{th}^h (V)	I_{on}^h/I_{off}^h
Sublimation	0.1	B	4.5×10^{-2}	58	10^5	6.7×10^{-4}	-72	10^3
Sublimation	0.017	B	1.2×10^{-2}	61	10^4	1.3×10^{-4}	-67	10^3
SuMBD	0.1	B	1.2×10^{-5}	50	10^4	2.2×10^{-6}	-35	10^3
SuMBD	0.017	C	1.3×10^{-4}	53	10^4	—	—	—

PMMA films,¹⁶ XRD analysis reveals that thermal sublimation of C₄-NT4N on HMDS-functionalized silicon dioxide yielded the phase B (Fig. S1, ESI†).

Moreover, considering that the bottom contact configuration is usually characterized by less effective charge injection than that of the top contact one,¹⁹ also the hole and electron carrier mobilities are in good agreement with what was previously reported.¹⁶ On the other hand, the threshold for electron transport is higher when HMDS-treated SiO₂ was used instead of PMMA (*i.e.* a material with a low surface trapping density) as a dielectric layer, confirming that the electron traps cannot be completely passivated through the functionalization with HMDS.¹⁸ As reported in Table 1, the field-effect properties are not substantially affected by the decrease of the deposition rate, thin films of C₄-NT4N grown by thermal deposition at 0.017 Å s⁻¹ exhibit ambipolar field-effect transport with mobilities and threshold voltages close to those obtained at higher rate. As revealed by XRD analysis (Fig. S1, ESI†), these results are due to the fact that also at a low rate the films are composed of phase B. Beside vacuum thermal deposition, we have also investigated the deposition of C₄-NT4N by SuMBD, which is an effective strategy to activate specific film growth²⁰ since it allows a good control of the kinetic energy, E_k , of the molecules impinging the growth surface. High performing n-type²¹ and p-type²² OFETs have been realized by this method as well as the controlled deposition of different polymorphic phases in titanyl phthalocyanine layers.²³ C₄-NT4N was deposited with a kinetic energy $E_k = 12.4$ eV, this being close to the saturation value that is theoretically predicted to be 13 eV. This feature induces a growth regime very far from equilibrium, where the roto-vibrational degrees of freedom of the molecules are frozen. C₄-NT4N films deposited at 0.1 Å s⁻¹ by SuMBD exhibit ambipolar behaviour, but yielded field-effect electron and hole mobilities in the saturation regime of 1.2×10^{-5} cm² V⁻¹ s⁻¹ and 2.2×10^{-6} cm² V⁻¹ s⁻¹, respectively, thus two orders of magnitude lower than what is observed for films deposited at the same deposition rate by thermal sublimation (Fig. 2b and Table 1). Nevertheless, XRD measurements reported in Fig. 3b show that, in agreement with what was obtained with thermal deposition, the thin films of C₄-NT4N deposited by SuMBD at 0.1 Å s⁻¹ consist of phase B. Therefore, the difference in electrical performance between films deposited by thermal sublimation and SuMBD can be explained in terms of the film morphology. Indeed, the film morphology assessed by AFM reveals that thin films deposited by thermal sublimation are characterized by growth of tridimensional crystals on continuous layers independent of deposition rate (Fig. 4a and b). In contrast, the morphology of thin films deposited by SuMBD is characterized by the coexistence of small grains with a lateral

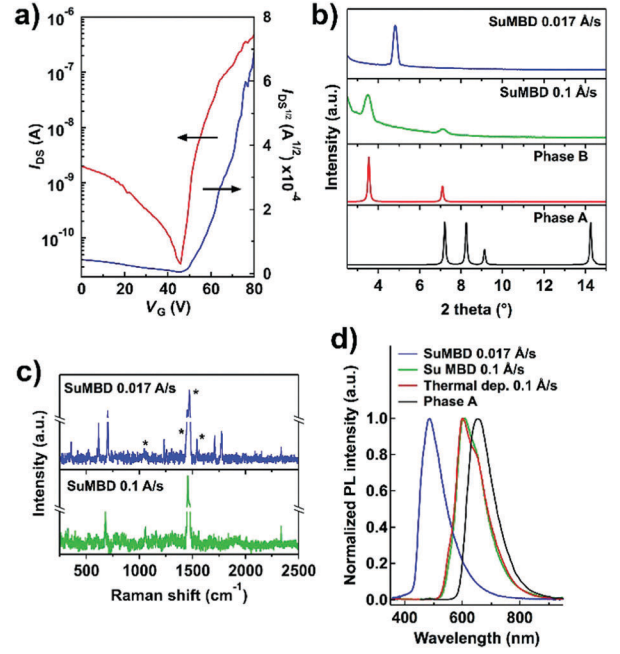


Fig. 3 (a) Transfer characteristics measured at positive source-drain bias of a representative field-effect transistor based on a C₄-NT4N film deposited by SuMBD at 0.017 Å s⁻¹ (drain voltage $V_{SD} = 100$ V). (b) XRD patterns of the C₄-NT4N thin films deposited on HMDS-treated Si/SiO₂ by SuMBD at 0.1 (green line) and 0.017 Å s⁻¹ (blue line), and the calculated diffractograms of A (black line) and B (red line) phases. (c) Raman spectra of C₄-NT4N thin films deposited on HMDS-treated Si/SiO₂ by SuMBD at 0.1 (green line) and 0.017 Å s⁻¹ (blue line). (d) PL spectra of C₄-NT4N thin films deposited on quartz by SuMBD at 0.017 Å s⁻¹ (blue line) and at 0.1 Å s⁻¹ (green line), and by thermal sublimation at 0.1 Å s⁻¹ (red line), and C₄-NT4N (phase A) powder (black line).

size lower than 100 nm and tridimensional structures (Fig. 4c). To investigate the effect of the deposition rate, we have also studied the field-effect properties of thin films deposited by SuMBD at a lower deposition rate, *i.e.* 0.017 Å s⁻¹. In sharp contrast to the films deposited at 0.1 Å s⁻¹, thin films of C₄-NT4N grown at a low deposition rate show only an n-type field-effect response with a maximum mobility of about 10^{-4} cm² V⁻¹ s⁻¹ (Fig. 3a and Table 1) and are characterized by rounded and well-defined terraced grains with steps close to 2 nm, *i.e.* comparable with the length of C₄-NT4N molecule (Fig. 4d). As shown in Fig. 3b, the XRD pattern of thin films deposited at 0.017 Å s⁻¹ displays a unique peak at low angles (about 4.8°) that is not compatible with either phase A or B, revealing the presence of a polymorph with a different crystal structure, namely phase C. Therefore, as previously observed for other organic semiconductors,^{24,25} the absence of p-type

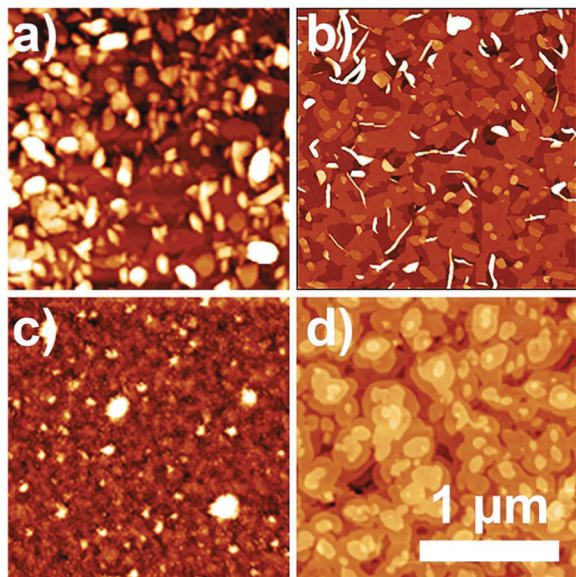


Fig. 4 AFM images of C_4 -NT4N films deposited on HMDS-treated Si/SiO₂ by thermal sublimation at (a) 0.1 Å s^{-1} (z-scale: 0–60 nm) and (b) 0.017 Å s^{-1} (z-scale: 0–30 nm), and by SuMBD at (c) 0.1 Å s^{-1} (z-scale: 0–15 nm) and (d) 0.017 Å s^{-1} (z-scale: 0–15 nm).

behaviour can be ascribed to skeletal distortions of C_4 -NT4N molecules in the molecular packing of phase C that disrupt π -conjugation, as also supported by Raman and photophysical characterization (see below), and hinder the hole transport.

As shown in Fig. 3c, the Raman spectra of the thin films deposited by SuMBD with different deposition rates show different bands. The films grown at 0.1 Å s^{-1} are characterized by the specific Raman pattern, ranging between 1100 and 1400 cm^{-1} , previously assigned to the phase B packing of the C_n -NT4N molecules, whereas the films deposited at 0.017 Å s^{-1} do not show in the same spectral region the Raman features of either phase A or B.²⁶ However, their spectrum displays the bands characteristic of the C_n -NT4N molecules (see marks in the Fig. 3), revealing that the C_n -NT4N chemical identity after the SuMBD process is fully preserved and thus confirming that the films consist of a new polymorph with a different molecular packing, *i.e.* the phase C packing.²⁶ Interestingly, the Raman band of the phase C corresponding to the out of phase C=C stretching mode ($\sim 1500 \text{ cm}^{-1}$),²⁷ which mainly involves displacements in the outermost rings of the oligothiophene skeleton, is shifted towards a higher frequency than that of the other phases. This is likely the indication of a lowering in the conjugation length that can be ascribed to a reduced planarity of the molecular structure. Unfortunately, attempts to obtain pure phase C in bulk were unsuccessful, suggesting that, as previously observed for other organic semiconductors,²⁸ this polymorph might exist only in thin films.

The optical properties of the C_4 -NT4N films were investigated by photoluminescence spectroscopy. The PL spectra of the thin films of C_4 -NT4N deposited at 0.1 Å s^{-1} by both vacuum thermal deposition and SuMBD reveal the characteristic emission features expected for phase B,¹⁷ consisting of an emission band centered

at about 600 nm (Fig. 3c, red and green lines). In sharp contrast, the films of C_4 -NT4N deposited at 0.017 Å s^{-1} by SuMBD show an emission band centered at 488 nm and are thus blue-shifted compared to the characteristic PL emission of A and B polymorphs of C_4 -NT4N (Fig. 3c, blue line).¹⁷ Therefore, in good agreement with the XRD and Raman measurements, the photophysical properties indicate that the thin films of C_4 -NT4N obtained at 0.017 Å s^{-1} by SuMBD are composed of a new phase in which, probably due to the energetics of the supersonic beam, the molecular moieties are arranged in a sterically hindered configuration so that the π - π delocalization in the crystalline domains is reduced. However, as demonstrated by the electrical characterization, the molecular moieties are arranged according to a solid-state configuration that preserves the efficient overlapping of π orbitals²⁹ to allow lateral electron transport in FET devices.

Solution deposition

Previous studies carried out by our group demonstrated that both the drop casting and spin coating of solution of C_n -NT4N molecules lead to the formation of needle shaped microcrystals unsuitable for the fabrication of OFETs.¹⁷ In light of that, we have exploited an unconventional wet lithographic technique, *i.e.* Lithographically Controlled Wetting (LCW), to pattern and study the electrical properties of C_6 -NT4N, as it is the most soluble of the C_n -NT4N series.¹⁶

LCW is a stamp-assisted technique that exploits the self-organizing properties of the materials in a spatially confined system for creating micro and nanostructured thin films replicating the stamp features. LCW has been successfully employed for the integration of soluble organic and metal-organic compounds in OFETs.^{30,31} Since attempts to implement microstructured films in bottom gate/bottom contact transistors have been unsuccessful, OFETs based on microstripes of C_6 -NT4N were fabricated by LCW in a bottom gate/top contact configuration, *i.e.* the best performing configuration for a C_n -NT4N based device,¹⁶ using a perfluorinated polymer (Cytop) as dielectric layer to guarantee the solvent orthogonality (Fig. 5a). All the steps of fabrication were performed under a controlled atmosphere in order to prevent the degradation phenomena activated by oxygen and water. As shown in Fig. 5b, C_6 -NT4N microstripes ($\sim 80 \text{ nm}$ high) with a controlled size, shape, and orientation were successfully fabricated by LCW using anisole solution at 80°C and 1,2-dichlorobenzene (1,2-DCB) as solvent at room temperature. The temperature was optimized in order to enhance the solubility and allow the formation of long-range molecular ordered stripes.

Fig. 5c, d and Fig. S2, ESI† show the transfer and output characteristics of the C_6 -NT4N based devices fabricated by LCW. The output characteristics curves exhibit the superimposition of standard saturation behaviour for one charge carrier and a superlinear current increase due to injection of the other charge carrier, indicative of the typical ambipolar transport behaviour expected for C_6 -NT4N.

In Table 2 the electrical performance of the devices fabricated both by LCW and, for comparison, by thermal sublimation of C_6 -NT4N (Fig. S3, ESI†) are summarized. All the devices exhibit

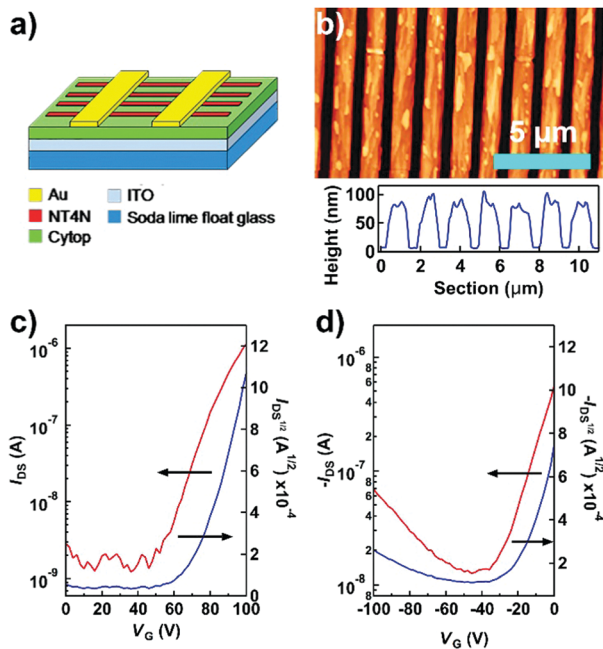


Fig. 5 (a) Schematic representation of a bottom gate/top contact OFET based on C₆-NT4N microstrips. (b) AFM image of C₆-NT4N deposited on Cytop by LCW from anisole solution (z-scale: 0–150 nm) and the corresponding profile. Transfer characteristics measured at a positive or negative source–drain bias of a representative field-effect transistor based on C₆-NT4N deposited by LCW from (c) anisole and (d) 1,2-DCB solution (drain voltage $V_{SD} = \pm 100$ V).

unbalanced field-effect charge transport, but the solvent-processed microstructured C₆-NT4N films yielded lower field-effect performance than the vapor-deposited films. Microstrips of C₆-NT4N deposited from anisole solution exhibit electron and hole mobilities in the saturation regime of $2.4 \times 10^{-3} \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ and $1.0 \times 10^{-5} \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$, respectively, but using 1,2-DCB solution the electron mobility decreases by one order of magnitude. Noteworthy, thin films of C₆-NT4N deposited on Cytop by thermal sublimation showed an electrical performance (Table 2) similar to those that have been reported using PMMA as a dielectric layer.¹⁶ XRD analysis shows that microstructured films of C₆-NT4N deposited from anisole solution are characterized by only polymorph B, while the XRD pattern of films deposited from 1,2-DCB solution displays a profile with a doublet for the peak at $2\theta \sim 6.1^\circ$ revealing the coexistence of both A and B phases (Fig. 6a). In addition, it shows two not identified and correlated peaks at $2\theta = 2.2^\circ$ and 4.4° (respectively $d = 44.6$ and 22.3 \AA) that possibly refer to the same family of crystalline planes. Noteworthy, the similarity between the peak at $2\theta = 4.4^\circ$ with the peak at $2\theta = 4.8^\circ$ of

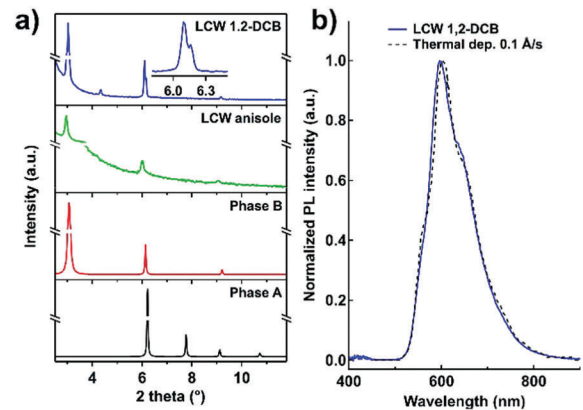


Fig. 6 (a) XRD patterns of the C₆-NT4N microstrips deposited on Cytop by LCW from 1,2-DCB (blue line) and anisole (green line) solution, and calculated diffractograms of A (black line) and B (red line) phases. (b) PL spectra of C₆-NT4N microstrips deposited on Cytop by LCW from 1,2-DCB (blue line) and C₆-NT4N thin film deposited on Cytop by thermal sublimation at 0.1 \AA s^{-1} (black dashed line).

the phase C of C₄-NT4N (Fig. 3b) suggests the presence of a polymorph of C₆-NT4N with a similar molecular packing. Therefore, this mixed composition of phases may explain the worse performance of microstructured films of C₆-NT4N deposited from 1,2-DCB solution.

To investigate the role of spatial confinement on the selection of phase packing, XRD analysis of C₆-NT4N solution-cast films was performed. As expected, after solvent evaporation C₆-NT4N forms needle-shaped microcrystals, but while the crystals deposited from the anisole solution at 80°C exhibit a XRD profile compatible with phase B, the crystals achieved from 1,2-DCB solution at room temperature exhibit phase A (Fig. S4, ESI†). Similarly, it has been previously reported that the crystals of C₆-NT4N prepared from toluene solution at room temperature are characterized by phase A, but after thermal treatment, the conversion to phase B occurs.¹⁷ Therefore, considering that the two phases are enantiotropically related and, at room temperature, the stable phase is A,¹⁷ our results suggest that the formation of phase B in microstructured films of C₆-NT4N deposited using anisole at 80°C can be mainly ascribed to the thermal treatment; while differently, in the case of microstructured films deposited at room temperature from 1,2-DCB solution, the preferential growth of phase B is induced by the spatial confinement.³²

Noteworthy, photoluminescence spectroscopy showed that C₆-NT4N microstrips deposited on Cytop by LCW from 1,2-DCB solution exhibit the emission profile of the phase B (Fig. 6b) as also confirmed by the localized emission spectra collected in different areas of the microstructured film (Fig. S5, ESI†).

Table 2 Electrical data of bottom gate/top contact OFETs based on solution-deposited films of C₆-NT4N

	Solvent	Phase	$\mu_e (\text{cm}^2 \text{ V}^{-1} \text{ s}^{-1})$	I_{on}^e/I_{off}^e	$V_{th}^e (\text{V})$	$\mu_h (\text{cm}^2 \text{ V}^{-1} \text{ s}^{-1})$	$V_{th}^h (\text{V})$	I_{on}^h/I_{off}^h
LCW	Anisole	B	2.4×10^{-3}	10^3	62	1.0×10^{-5}	−56	10^1
LCW	1,2-DCB	B (A)	4.4×10^{-4}	10^2	60	2.7×10^{-5}	−64	10^1
Sublimation	—	B	2.0×10^{-1}	10^4	57	2.2×10^{-3}	−68	10^3

We point out that the intensity of the PL spectra is correlated not only to the amount of emissive material present in the film, but also to the PL quantum yield of each phase, which is likely the basis of the difficulty to optically unveil phase A in the microstructured material.

Experimental

Materials

C_n -NT4N compounds were synthesized according to the literature.¹⁷ Hexamethyldisilazane (HMDS) and spectroscopic grade quality solvents were purchased from Sigma-Aldrich. Sylgard 184 (polydimethylsiloxane, PDMS) silicone elastomer base and the curing agent were purchased from Dow Corning. CYTOP CTL-809M was purchased from AGC chemicals.

Device fabrication

Bottom gate/bottom contact transistors were fabricated on heavily n-type doped silicon (100) wafers with a 200 nm thermally grown silicon dioxide dielectric layer ($C_i = 17.25 \text{ nF cm}^{-2}$). Gold source-drain electrodes (100 nm thick) with a width/length ratio of 550 were deposited by vacuum evaporation and patterned by photolithography. Before the deposition of the semiconductor, the substrates were cleaned by sonication in acetone for 10 min, then in 2-propanol for 10 min, and functionalized with HMDS.³³

Bottom gate/top contact transistors were fabricated on glass/ITO/Cytop substrates where ITO and Cytop (450 nm) were used as the gate electrode and gate dielectric, respectively. Before the deposition of Cytop, the substrates were cleaned by ultrasonic baths in acetone, then 2-propanol and dried under a nitrogen flow. Gold source-drain electrodes (70 nm thick) were deposited by thermal evaporation onto the semiconductor films through a shadow mask at room temperature and 10^{-6} mbar. Devices were fabricated with 12 mm channel width and 70 μm channel length.

Vacuum deposition of ~ 30 nm thick layers of C_4 -NT4N was performed by thermal deposition and SuMBD keeping the substrate at room temperature. The organic precursor vapor for SuMBD was produced in a quartz tube by thermal evaporation with a stagnation pressure of He gas of about 1300 mbar. Molecular vapor was seeded in a supersonic jet of the He carrier gas, resulting in a kinetic energy (E_k) value of the emerging molecular beam of approximately 12.4 eV per molecule.

Solution deposition of C_6 -NT4N was performed by LCW.³² Elastomeric polydimethylsiloxane (PDMS) molds were prepared by replica molding of Compact Disk supports or silicon masters. The stamp motifs consist of parallel lines 1 μm wide, 230 ± 15 nm deep and 1.7 μm periodicity. After curing for 6 h at 60 °C, PDMS stamps were peeled off, cleaned by sonication in ethanol for 10 min, and dried under nitrogen flow. 5 μL of saturated solution of C_6 -NT4N in anisole at 80 °C or 1,2-dichlorobenzene at room temperature were drop cast on the substrate, then a PDMS stamp was placed on the top, paying attention to the stripe orientation, and the solvent was allowed to evaporate under nitrogen for 24 h. For anisole solution, the substrates were

heated on a hot plate at 80 °C during film deposition. Afterwards, the stamp was removed and, before characterization, the samples were aged under a nitrogen atmosphere for 2 hours.

Electrical characterization

The electrical measurements were carried out in a MBraun nitrogen glovebox using a standard SUSS probe station coupled to a B1500A Agilent semiconductor device analyser. The field-effect mobility in the saturation regime (μ_{sat}) was calculated using the equation $I_{\text{DS}} = (W/2L)C_i\mu_{\text{sat}}(V_G - V_T)^2$, where W and L are the channel widths and lengths, respectively, C_i is the capacitance per unit area of the insulating layer, and V_T is the threshold voltage extracted from the square root of the drain current ($I_{\text{DS}}^{1/2}$) versus the gate voltage (V_G) characteristics for a fixed drain voltage (V_{DS}). For solution-processed devices, the effective channel widths were estimated on the basis of the statistical analysis of optical microscopy images.

Optical characterization

Steady-state photo-luminescence (PL) was excited using a CW He-Cd laser at a wavelength of 325 nm and collected under transmission conditions (no color filter was necessary for truncating laser excitation). PL emission was collected using a calibrated optical multichannel analyzer (PMA-11, Hamamatsu). CLSM images of the microstructured active layer and localized photoluminescence measurements were obtained with a Nikon Eclipse 2000-E laser scanning confocal microscope using an excitation wavelength of 405 nm.

XRD characterization

X-ray data were collected using a PANalyticalX'Pert PRO diffractometer in Bragg-Brentano configuration equipped with a X'Celerator detector with Cu K α ($\lambda = 1.5418 \text{ \AA}$) as the incident radiation. Soller 0.04 rad, Anti-scatter slit 1/8, divergence slit 1/16, step size 0.017°.

Raman characterization

Raman spectra were recorded with the Renishaw System InVia spectrometer equipped with a Peltier cooled CCD. The spectrometer was coupled to a Leica DLML microscope equipped with 50 \times , 20 \times and 5 \times objectives which allowed for a spatial resolution of about 1 μm with the 50 \times objective and a nominal field depth ranging from about 25 to 450 μm . The excitation was from a diode laser tuned at 785 nm with a nominal power of 500 mW. The power was reduced by neutral density filters to avoid sample damage.

Morphology measurements

AFM images were collected at room temperature in air on a Multimode 8 microscope operated in PeakForce mode and equipped with a type J scanner (Bruker Nano Inc. GmbH) and SNL-A probes with a nominal spring constant of 0.35 N m $^{-1}$ (Bruker AFM Probes, Camarillo, CA), or on an XE-100 microscope operated in non-contact mode (Park systems) and equipped with a silicon nitride cantilever (Nanosensors). Background interpolation was performed with Gwyddion 2.48 (<http://gwyddion.net/>).

Conclusions

In this study, we have investigated the impact of the processing method in controlling the polymorphism of two 2,3-thienoimide-based oligothiophenes semiconductors (C_4 - and C_6 -NT4N) that are typically implemented in the fabrication of multifunctional field-effect transistors. Giving the possibility of C_n -NT4N derivatives for being processed by both vacuum and solution methods, the polymorphism was probed and revealed by choosing different deposition techniques, *i.e.* thermal sublimation, SuMBD and LCW. While thermal sublimation leads to phase B showing an ambipolar behaviour, SuMBD enables, by varying the deposition rate, the formation of both phase B and a new form, named C, showing unipolar field-effect behaviour. Moreover, comparing thin films with the same molecular packing, but deposited by different techniques, we highlighted the fundamental role of film morphology on their electrical performance.

On the other hand, the use of a stamp-assisted technique such as LCW was fundamental in the fabrication of the device since it enables the realization of a microstructured active layer, thus overcoming the inhomogeneity achieved by drop casting and spin coating. In particular, microstructured thin films were deposited from anisole at 80 °C and from 1,2 DCB at room temperature. While drop casting at room temperature leads to phase A, the LCW preferentially leads to phase B, showing the importance of the spatial confinement in controlling the polymorphism. For the first time, we succeeded in demonstrating field-effect charge transport for the C_n -NT4N derivative deposited from solution.

In conclusion, our study confirms that the process of deposition, with control of the polymorph type and the morphological features, may represent a straightforward and effective way to tune the electrical properties of 2,3-thienoimide-ended materials.

Conflicts of interest

There are no conflicts to declare.

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Notes and references

- 1 H. Chung and Y. Diao, *J. Mater. Chem. C*, 2016, **4**, 3915–3933.
- 2 A. Troisi and G. Orlandi, *J. Phys. Chem. B*, 2005, **109**, 1849–1856.
- 3 D. J. Gundlach, T. N. Jackson, D. G. Schlom and S. F. Nelson, *Appl. Phys. Lett.*, 1999, **74**, 3302–3304.
- 4 H. Jiang, X. Yang, Z. Cui, Y. Liu, H. Li, W. Hu, Y. Liu and D. Zhu, *Appl. Phys. Lett.*, 2007, **91**, 123505.
- 5 R. Pfattner, M. Mas-Torrent, I. Bilotti, A. Brillante, S. Milita, F. Liscio, F. Biscarini, T. Marszalek, J. Ullanski, A. Nosal, M. Gazicki-Lipman, M. Leufgen, G. Schmidt, W. M. Laurens, V. Laukhin, J. Veciana and C. Rovira, *Adv. Mater.*, 2010, **22**, 4198–4203.
- 6 O. D. Jurchescu, A. Meetsma and T. T. M. Palstra, *Electrochem. Solid-State Lett.*, 2006, **9**, 330–334.
- 7 O. D. Jurchescu, D. a. Mourey, S. Subramanian, S. R. Parkin, B. M. Vogel, J. E. Anthony, T. N. Jackson and D. J. Gundlach, *Phys. Rev. B: Condens. Matter Mater. Phys.*, 2009, **80**, 85201.
- 8 J. Y. Kim, T. Yasuda, Y. S. Yang, N. Matsumoto and C. Adachi, *Chem. Commun.*, 2014, **50**, 1523–1526.
- 9 K. Wang, H. Zhang, S. Chen, G. Yang, J. Zhang, W. Tian, Z. Su and Y. Wang, *Adv. Mater.*, 2014, **26**, 6168–6173.
- 10 M. Muccini, W. Koopman and S. Toffanin, *Laser Photonics Rev.*, 2012, **6**, 258–275.
- 11 I. F. Perepichka and D. F. Perepichka, *Handbook of Thiophene Based Materials*, 2009.
- 12 L. Zhang, N. S. Colella, B. P. Cherniawski, S. C. B. Mannsfeld and A. L. Briseno, *ACS Appl. Mater. Interfaces*, 2014, **6**, 5327–5343.
- 13 M. Zambianchi, L. Favaretto, M. Durso, C. Bettini, A. Zanelli, I. Manet, M. Gazzano, L. Maini, D. Gentili, S. Toffanin, F. Gallino, M. Muccini, M. Cavallini and M. Melucci, *J. Mater. Chem. C*, 2015, **3**, 121–131.
- 14 M. Durso, C. Bettini, A. Zanelli, M. Gazzano, M. G. Lobello, F. De Angelis, V. Biondo, D. Gentili, R. Capelli, M. Cavallini, S. Toffanin, M. Muccini and M. Melucci, *Org. Electron.*, 2013, **14**, 3089–3097.
- 15 M. Natali, S. D. Quiroga, L. Passoni, L. Criante, E. Benvenuti, G. Bolognini, L. Favaretto, M. Melucci, M. Muccini, F. Scotognella, F. Di Fonzo and S. Toffanin, *Adv. Funct. Mater.*, 2017, **27**, 1–8.
- 16 M. Melucci, M. Durso, C. Bettini, M. Gazzano, L. Maini, S. Toffanin, S. Cavallini, M. Cavallini, D. Gentili, V. Biondo, G. Generali, F. Gallino, R. Capelli and M. Muccini, *J. Mater. Chem. C*, 2014, **2**, 3448–3456.
- 17 L. Maini, F. Gallino, M. Zambianchi, M. Durso, M. Gazzano, K. Rubini, D. Gentili, I. Manet, M. Muccini, S. Toffanin, M. Cavallini and M. Melucci, *Chem. Commun.*, 2015, **51**, 2033–2035.
- 18 L.-L. Chua, J. Zaumseil, J.-F. Chang, E. C.-W. Ou, P. K.-H. Ho, H. Sirringhaus and R. H. Friend, *Nature*, 2005, **434**, 194–199.
- 19 K. Noda, Y. Wada and T. Toyabe, *Org. Electron.*, 2014, **15**, 1571–1578.
- 20 K. Walzer, T. Fritz and K. Leo, *Surf. Sci.*, 2006, **600**, 2064–2069.
- 21 F. Chiarella, T. Toccoli, M. Barra, L. Aversa, F. Ciccullo, R. Tatti, R. Verucchi, S. Iannotta and A. Cassinese, *Appl. Phys. Lett.*, 2014, **104**, 143302.
- 22 Y. Wu, T. Toccoli, N. Koch, E. Iacob, A. Pallaoro, P. Rudolf and S. Iannotta, *Phys. Rev. Lett.*, 2007, **98**, 1–4.

- 23 N. Coppedè, E. Bonnini, F. Mezzadri, G. Tarabella, P. Ranzieri, L. Barba, G. Arrighetti, L. Lutterotti and S. Iannotta, *Org. Electron. Physics, Mater. Appl.*, 2016, **32**, 15–20.
- 24 A. Facchetti, T. J. Marks and Y. Zheng, *J. Am. Chem. Soc.*, 2010, **132**, 8440–8452.
- 25 S. Thankaraj Salammal, Z. Zhang, J. Chen, B. Chattopadhyay, J. Wu, L. Fu, C. Fan and H. Chen, *ACS Appl. Mater. Interfaces*, 2016, **8**, 20916–20927.
- 26 C. Cappuccino, P. P. Mazzeo, T. Salzillo, E. Venuti, A. Giunchi, R. G. Della Valle, A. Brillante, C. Bettini, M. Melucci and L. Maini, *Phys. Chem. Chem. Phys.*, 2018, **20**, 3630–3636.
- 27 J. Casado, V. Hernandez, R. Ponce Ortiz, M. C. Ruiz Delgado, J. T. Lopez Navarrete, G. Fuhrmann and P. Bauerle, *J. Raman Spectrosc.*, 2004, **35**, 592–599.
- 28 A. M. Hiszpanski, R. M. Baur, B. Kim, N. J. Tremblay, C. Nuckolls, A. R. Woll and Y. L. Loo, *J. Am. Chem. Soc.*, 2014, **136**, 15749–15756.
- 29 M. Muccini and S. Toffanin, *Organic Light-Emitting Transistors: Towards the Next Generation Display Technology*, Wiley-Science, Wise Co-Publication, Wiley, 2016.
- 30 D. Gentili, F. Di Maria, F. Liscio, L. Ferlauto, F. Leonardi, L. Maini, M. Gazzano, S. Milita, G. Barbarella and M. Cavallini, *J. Mater. Chem.*, 2012, **22**, 20852–20856.
- 31 A. de la Peña Ruigómez, D. Rodríguez-San-Miguel, K. C. Stylianou, M. Cavallini, D. Gentili, F. Liscio, S. Milita, O. M. Roscioni, M. L. Ruiz-González, C. Carbonell, D. Maspoch, R. Mas-Ballesté, J. L. Segura and F. Zamora, *Chem. – Eur. J.*, 2015, **21**, 10666–10670.
- 32 D. Gentili, F. Valle, C. Albonetti, F. Liscio and M. Cavallini, *Acc. Chem. Res.*, 2014, **47**, 2692–2699.
- 33 M. Barra, D. Viggiano, P. Ambrosino, F. Bloisi, F. V. Di Girolamo, M. V. Soldovieri, M. Tagliatela and A. Cassinese, *Biochim. Biophys. Acta, Gen. Subj.*, 2013, **1830**, 4365–4373.