

# Alma Mater Studiorum Università di Bologna Archivio istituzionale della ricerca

Energetic fluctuations in amorphous semiconducting polymers: Impact on charge-carrier mobility

This is the final peer-reviewed author's accepted manuscript (postprint) of the following publication:

#### Published Version:

Energetic fluctuations in amorphous semiconducting polymers: Impact on charge-carrier mobility / Gali, Sai Manoj; D'Avino, Gabriele; Aurel, Philippe; Han, Guangchao; Yi, Yuanping; Papadopoulos, Theodoros A.; Coropceanu, Veaceslav; Brédas, Jean-Luc; Hadziioannou, Georges; Zannoni, Claudio; Muccioli, Luca. - In: THE JOURNAL OF CHEMICAL PHYSICS. - ISSN 0021-9606. - ELETTRONICO. - 147:13(2017), pp. 134904.1-134904.10. [10.1063/1.4996969]

Availability:

This version is available at: https://hdl.handle.net/11585/608588 since: 2019-05-10

Published:

DOI: http://doi.org/10.1063/1.4996969

Terms of use:

Some rights reserved. The terms and conditions for the reuse of this version of the manuscript are specified in the publishing policy. For all terms of use and more information see the publisher's website.

This item was downloaded from IRIS Università di Bologna (https://cris.unibo.it/). When citing, please refer to the published version.

(Article begins on next page)

# Energetic fluctuations in amorphous semiconducting polymers: Impact on charge-carrier mobility

Sai Manoj Gali

Institut des Sciences Moléculaires, UMR 5255,
University of Bordeaux, Talence, France and
Laboratoire de Chimie des Polymères Organiques,
UMR 5629, University of Bordeaux, Pessac, France

#### Gabriele D'Avino

Institut Néel, CNRS and Grenoble Alpes University, Grenoble, France

# Philippe Aurel

Institut des Sciences Moléculaires, UMR 5255, University of Bordeaux, Talence, France

Guangchao Han and Yuanping Yi

Institute of Chemistry, Chinese Academy of Sciences, Beijing, China

# Theodoros A. Papadopoulos

Department of Natural Sciences, University of Chester, Chester, United Kingdom

Veaceslav Coropceanu and Jean-Luc Brédas

Center for Organic Photonics and Electronics,

Georgia Institute of Technology Atlanta, Georgia, USA

# Georges Hadziioannou

Laboratoire de Chimie des Polymères Organiques, UMR 5629, University of Bordeaux, Pessac, France

#### Claudio Zannoni

Dipartimento di Chimica Industriale "Toso Montanari", University of Bologna, Bologna, Italy

## Luca Muccioli\*

Institut des Sciences Moléculaires, UMR 5255, University of Bordeaux, France and

Dipartimento di Chimica Industriale "Toso Montanari", University of Bologna, Bologna, Italy
(Dated: September 13, 2017)

# Abstract

We present a computational approach to model hole transport in an amorphous semiconducting fluorene-triphenylamine copolymer (TFB), which is based on the combination of Molecular Dynamics to predict the morphology of the oligomeric system, and Kinetic Monte Carlo (KMC), parameterized with quantum chemistry calculations, to simulate hole transport.

Carrying out a systematic comparison with available experimental results, we discuss the role different transport parameters play in the KMC simulation, and in particular the dynamic nature of positional and energetic disorder on the temperature and electric field dependence of charge mobility. It emerges that a semi-quantitative agreement with experiments is found only when the dynamic nature of the disorder is taken into account. This study establishes a clear link between microscopic quantities and macroscopic hole mobility for TFB, and provides substantial evidence of the importance of incorporating fluctuations, at the molecular level, to obtain results that are in good agreement with temperature and electric field-dependent experimental mobilities. Our work makes a step forward towards the application of nanoscale theoretical schemes as a tool for predictive material screening.

<sup>\*</sup> Luca.Muccioli@unibo.it

#### I. INTRODUCTION

Amorphous semiconducting polymers, due to their proven versatility, offer substantial advantages in terms of ease of processing and low production costs, with promising applications in the field of organic electronics, such as solar cells, light emitting diodes and field effect transistors. Recent investigations suggest that charge transport in conjugated polymers is not simply governed by the degree of crystallinity, but in general by the presence of aggregates with short-range order, that promote inter-chain charge transport.[1] Experimental evidence that aggregation of polymer chains with sufficient short-range order and interconnected domains, in a globally amorphous structure, is sufficient to promote charge transport[2, 3] points towards the refinement of models conventionally employed for describing charge transport in crystalline organic semiconducting systems.

As polymer chains are bound by weak interactions with multiple degrees of freedom, parameters influencing hole mobility that should be included in the modeling of hole transport include (i) the length of polymer chains, (ii) the conformational freedom along the chain subunits, (iii) the conformation space of the polymer, (iv) the number of possible neighboring units, and (v) the orientation of each chain with respect to the direction of external electric field. [4–8] The theoretical frameworks developed to address the role of morphology on electronic transport in semiconducting polymers demonstrate that intra-chain transport dominates at shorter length scales with high mobilities, up to 10-100 cm<sup>2</sup>V<sup>-1</sup>s<sup>-1</sup>[9], while inter-chain transport prevails at larger displacements, with mobilities a few orders of magnitude lower. The slowest inter-chain transport defines the limiting charge carrier mobility attainable at mesoscopic device scale. [5, 10] Such theoretical approaches, in conjunction with Molecular Dynamics (MD) simulations, where reasonably accurate polymer morphologies can be obtained, [11, 12] provide substantial information about the effect of polymer morphology and local structure on charge transport properties of the polymeric system under study. [13]

The main complexity originates from the need of incorporating the whole morphological variability of the macromolecular system into the charge transport modelling. Indeed, a theoretical study at the atomistic level, which considers both intra- and inter-chain transport and explicitly takes into account all possible fluctuations over many timescales and morphologies, is a challenging task, both scientifically and in terms of computational resources.[14]

Such frameworks are particularly unpractical for preliminary material screening, whereas a simplified methodology, able to provide reliable results with respect to experimental observations at low computational cost, would be more appropriate.

In what follows, we describe a step forward the realization of this objective, in which, with a combination of MD and KMC simulations, we systematically compare computational predictions with experimental temperature-dependent mobilities. As a case study, we consider the [(9,9-dioctylfluorenyl-2,7-diyl) (4,4'-(N-(4-sec-butylphenyl))] diphenylamine alternating copolymer, also known as TFB (figure 1). TFB is an amorphous polymer[15] with relatively high hole mobility[16, 17] which has been the subject of several fundamental studies,[18–22] and it is employed in practical applications such as light emitting diodes [23] and solar cells.[24–26] Our approach is based on two main and interdependent assumptions: (i) that the holes are localized along short segments of the polymer chain, and (ii) that hole transport can be described as a hopping process between those oligomeric units. Evidence supporting these hypotheses is provided, and their consequences on calculated charge carrier mobilities are discussed in detail.

#### II. COMPUTATIONAL DETAILS

# Molecular Dynamics simulations

We focused on TFB oligomers composed of five 9,9-dioctylfluorene (FLU) units, alternated with four butyl triphenylamine (TPA) co-monomer units (see figure 1). This specific oligomer length was chosen as it corresponds to the upper limit of the conjugation length of TFB, as estimated on the basis of quantum chemical calculations reported by Sancho-García et al.[27] Our choice of this oligomer approach[11, 28–30] in modeling charge transport in TFB, was also motivated by practical reasons, such as the possibility of dealing with reasonable box sizes and short equilibration times.

We employed a united-atom[31] force field complemented with quantum chemical calculations for atomic charges and torsional potentials, as detailed in the supplemental material. All molecular dynamics simulations were performed with the NAMD software [32]. In order to simulate amorphous TFB samples, the starting geometries for the simulation were built by introducing a certain degree of randomness, according to the following procedure: (i) the

starting geometry was generated by placing the oligomers with random orientations on the nodes of a cubic mesh of  $5\times5\times5$  sites, with a lattice constant of 80 Å; (ii) the low-density sample was compressed at 1000 K and 100 atm up to a rough volume stabilization, that occurs in about 1 ns; (iii) high-temperature annealing was performed for 10 ns of simulation at 1000 K and 1 atm; (iv) equilibration was carried out for 20 ns of simulation at  $T_{MD}$ =300 K and 1 atm; (v) this procedure was applied to simulate four different and uncorrelated samples composed of 125 oligomers (33380 united atom centers). A snapshot of one of the four samples after equilibration at 300 K is shown in the right panel of figure 1. The final density of the four samples is  $0.987 \pm 0.006$  g/cm<sup>3</sup>. It is worth noting that since the equilibration temperature is well below the experimental glass transition, i.e. 393-415 K depending on film thickness[15], significant changes on the morphology are not expected at lower temperatures. For this reason, and for keeping the approach computationally efficient, all charge transport simulations were conducted on the 300 K morphologies, neglecting the density variation with temperature. It is worth noting that this preparation scheme does not take into account any solvent inclusion in the polymer, and therefore leads to ideal, completely amorphous morphologies in which the number of contacts between the chains is maximized.

#### Charge transport model

Charge transport in amorphous TFB was described with a hopping model. Specifically, we resort to Marcus formula for hopping rates,[33] describing non-adiabatic hole transfer in the weak coupling regime, which has been widely employed to compute the charge transfer rates between adjacent molecules in organic semiconductors.[34] The charge transfer rate in the semi-classical Marcus formalism reads:

$$k_{ij} = \frac{2\pi}{\hbar} \frac{J_{ij}^2}{\sqrt{4\pi\lambda k_B T}} \exp\left[-\frac{(\Delta G_{ij} + \lambda)^2}{4\lambda k_B T}\right]$$
(1)

with 
$$\Delta G_{ij} = \epsilon_j - \epsilon_i + e\vec{E} \cdot \vec{R}_{ij}$$
 (2)

where e is the elementary charge,  $\epsilon_i$  and  $\epsilon_j$  are the energies sites i and j involved in the charge transfer,  $\Delta G_{ij}$  is the energy difference between the initial and final states,  $J_{ij}$  is the electronic coupling (or transfer integral)  $\lambda = \lambda_i + \lambda_e$  is the reorganization energy,  $k_B$  is the Boltzmann constant and T is the temperature. In the presence of an applied electric field  $(\vec{E})$ ,  $\Delta G_{ij}$ 

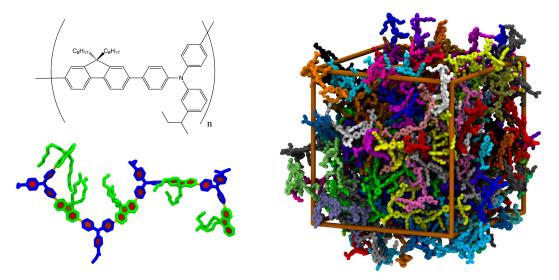


FIG. 1. Left: Chemical structure of poly[(9,9-dioctylfluorenyl-2,7-diyl)-co-(4,4'-(N-(4-sec-butylphenyl)diphenylamine)] (TFB, top), and example of an individual oligomer conformation as produced by MD simulations at united atom level; FLU and TPA units are shown in green and blue, respectively while aromatic rings are highlighted in red (bottom). Right: Snapshot of a single sample after equilibration at 300 K and 1 atm, under periodic boundary conditions.

also includes the contribution  $e\vec{E} \cdot \vec{R}_{ij}$ ,  $\vec{R}_{ij}$  being the site-site distance vector. As positions of the sites, we used the center of the Mulliken charges evaluated from single point AM1 calculations[35] for positively charged oligomers at their instantaneous MD geometries. The reorganization energy consists of (i) an internal contribution  $(\lambda_i)$  associated to the change in geometry upon charge transfer and (ii) an external contribution  $(\lambda_e)$  which reflects the structural and electronic changes in the surrounding medium. The intramolecular part was set  $\lambda_i = 0.1$  eV on the basis of DFT calculations carried out on oligomers of increasing length (see supplemental material), a value very similar to the one reported for the well-known poly(3-hexylthiophene (P3HT),[36] and for the TPA moiety.[37] The external contribution, a challenging quantity to calculate exactly,[38] was set to  $\lambda_e = 0.2$  eV in the majority of the calculations, a value expected for a low dielectric constant medium according to classical continuum model calculations[39], and in line with theoretical estimates for oligoacenes crystals.[38]

For all molecules of the four MD samples the following conditions were applied: (i) for quantum chemistry calculations the hydrogen atoms, absent in the MD model, were added to simulated molecular structures using geometrical criteria; (ii) hole transfer integrals  $J_{ij}$  were

calculated as one-electron HOMO-HOMO couplings using the dimer projection method [40] at the ZINDO level of theory with the Mataga-Nishimoto potential [41, 42]; (iii) periodic boundary conditions were employed to select the pairs of molecules in contact, using 6 Å as the interatomic cutoff distance between atoms belonging to different molecules; (iv) site energies  $\epsilon_i$ ,  $\epsilon_j$  were approximated by ZINDO HOMO energies, including the effect of distortion of the molecular geometry in realistic conformations.

Although the energetic disorder introduced by intermolecular electrostatic interactions, due to permanent and induced dipoles, can play a key role in improving charge separation at heterojunctions, [43–45] and charge-carrier localization [46], it was not included in this work because of the nonpolar nature of TFB, which implies low energetic disorder as confirmed by Malliaras et al. [18].

#### **Charge Propagation**

The propagation of charges is simulated with the Kinetic Monte Carlo (KMC) technique. The charge is initially located at a randomly chosen site i. The hopping rates  $k_{ij}$  from donor site i to any potential acceptor site j, computed using equation 1, are employed to generate instantaneous hopping times  $\tau_{ij}$  in the framework of the "first reaction" method [47]:

$$\tau_{ij} = \frac{1}{k_{ij}} \ln \frac{1}{\xi} \tag{3}$$

where  $\xi$  is a random number drawn from a uniform distribution within the interval 0 to 1. The destination site i' is selected from the set of available neighbor sites j as the one having the smallest reaction time  $\tau_{ij}^{min}$ . The simulation progresses by following the subsequent hopping steps during charge propagation and the simulated elapsed time is increased by  $\tau_{ii'}$  at each hopping step. All the KMC simulations were run assuming a linear voltage profile across the sample, by varying the magnitude of the electric field from  $E = 1 \cdot 10^4$  V/cm to  $30 \cdot 10^4$  V/cm. In each simulation, a single charge is propagated, using 3D periodic boundary conditions, until it covers a fixed distance  $d = 4 \mu m$  along the direction of the electric field vector. Since only one charge is present at a time, the simulation targets the limit of very low charge densities following experimental evidence of charge density-independent mobility for a polymer very similar to TFB.[48] For each charge propagation run k, mobility is then

estimated as:

$$\mu_k = \frac{d}{E} \frac{1}{\tau_k} \tag{4}$$

where  $\tau_k$  is the sum of all hopping times  $\tau_{ii'}$ , i.e. the time needed for a charge propagating in the field direction to cover the distance d. Unless explicitly stated, mobility values reported in this work are averages over 500 up to 750 KMC runs performed with the electric field vector parallel to x, y and z Cartesian directions, and over the four MD samples. Average values are reported either as arithmetic means,  $\mu = (\sum_k \mu_k)/N$ , or as logarithmic means  $\mu_{ln} = \exp(\sum_k \ln \mu_k/N)$ , as the latter is a better estimator of the mobility in small samples.[49] We noticed that for our sample sizes and propagation distances the two means are almost coincident, indicating a satisfactory convergence of the KMC simulations.

# Analysis of charge mobilities

According to Bässler [50], at intermediate fields and in conditions of thermally-assisted hopping, the dependence of mobility on electric field modulus can be described by the following empirical relation:

$$\mu(E) = \mu_0 \exp\left(\beta \sqrt{E}\right) \tag{5}$$

where  $\beta > 0$  is the Poole-Frenkel factor. Using zero field, temperature-dependent mobilities  $\mu_0(T)$  derived from equation 5, the "apparent" diagonal energetic disorder  $\sigma_A$  and the infinite temperature mobility  $(\mu_{\infty})$  may in turn be obtained by extrapolating the temperature dependence of  $\mu_0$  with:

$$\mu_0(T) = \mu_\infty \exp\left[-\frac{4}{9} \left(\frac{\sigma_A}{k_B T}\right)^2\right] \tag{6}$$

The Poole-Frenkel factor depends on both apparent diagonal  $(\sigma_A)$  and off-diagonal  $(\Sigma_A)$  disorders:

$$\beta(T) = C \left[ \left( \frac{\sigma_A}{k_B T} \right)^2 - \Sigma_A^2 \right] \tag{7}$$

with C being an empirical proportionality factor. The combination of equations 5, 6, and 7 provides an approximate [51] but universal equation for the dependence of mobility on temperature, field, and energetic disorder. We use this theoretical framework to compare our data with experimental results obtained by Malliaras and coworkers [18, 52].

In the original Gaussian disorder lattice model (GDM)[50]  $\sigma_A$  and  $\Sigma_A$  correspond to the standard deviations of the distributions of site energies (the density of states, DOS) and transfer integrals. In the present work we label these two parameters as "apparent", since they are effective quantities extracted by fitting the mobility obtained from our off-lattice KMC simulations with the equations above, which are strictly valid only for the Gaussian disorder model. The apparent tag serves to distinguish the phenomenological disorder parameters obtained by such a fit from the microscopic parameters employed in our KMC simulations. Although the GDM model prescribes a  $1/T^2$  dependence of the logarithm of the mobility, it has been shown experimentally that often the temperature dependence of the mobility of organic semiconductors at low fields,[53] and for exponential densities of states,[54] can be well reproduced with an Arrhenius law, a behaviour also well established for inorganic disordered semiconductors[51]:

$$\mu_0(T) = \mu_\infty' \exp\left[-\frac{\Delta}{k_B T}\right] \tag{8}$$

We used the latter equations to fit the temperature dependence of the zero field mobility, obtained by extrapolating simulation and experimental data with equation 6. The infinite temperature mobility in equation 8 is primed because it is different (higher) with respect to the one obtained by fitting with equation 6.

#### III. RESULTS AND DISCUSSIONS

# A. Charge transport parameters

We initially analyzed the fluctuations of site energies in the samples, which generate the so-called diagonal (energetic) disorder, known to strongly influence charge transport [46, 50, 55–58]. The distribution of site energies (HOMO levels) calculated at MD geometries has an approximately Gaussian shape with standard deviation  $\sigma_{\epsilon} = 50.2$  meV (Figure S5), consistent with the assumptions of the Gaussian disorder model. This value is also close to the effective disorder  $\sigma_A = 65.9$  meV reported by Fong et al.[18], and was hence used in our KMC simulations.

Other fundamental parameters in determining the transfer rate, besides the reorganization energy discussed previously, are the magnitude of the electronic couplings between the different sites (the nodes of the charge percolation network) and the average number of pathways linking one node to the others, i.e. the number of neighboring sites with non-zero electronic couplings. When examining the calculated transfer integrals it is interesting to notice (Figure S6) the very low values of the couplings, with an average value of 0.40 meV, when for instance values of the order of  $\sim 10$  meV are found for crystalline and amorphous fullerenes at short intermolecular distances [46]. In addition, as it happens also for crystalline systems along the directions with small couplings[59], the standard deviation  $\sigma_J = 1.43 \text{ meV}$ is higher than the mean value, suggesting an important contribution of dynamic fluctuations on the mobility values. [60] Despite the low coupling values, the number of neighbors with non zero coupling (J > 0.1 meV) ranges between 10 and 20 per oligomer, ensuring a high number of percolation pathways for charge transport. Such couplings most often correspond to triphenylamine-triphenylamine contacts, while fluorene-fluorene contacts are less frequent because of the presence of the bulky dioctyl substituent on one side of the indeno group (cf figure 1). The chemical design principle of attaching the solubilizing side chains on the fluorene unit is of course beneficial for charge transport, since the hole is mostly localized on a triphenylamine unit (87% according to AM1 calculations), and then the HOMO-HOMO contacts are maximized if this region of the backbone is not hidden by the alkyl chains.

#### B. Mobility calculated for static networks

In the following sections we address in detail the relation between charge transport simulation parameters and calculated mobility as a function of temperature and electric field, utilizing the comparison with available experimental data as an assessment of the simulation predictions. In doing this, we implicitly assume that an oligomeric representation of the polymer is sufficient to describe the charge transport for longer chains; this means that the conjugation length of the charge is assumed to be shorter than the oligomer length, and that inter and intrachain charge transport mechanisms are equally well described by oligomeroligomer hopping and have identical characteristic times.[5] A low conjugation length is expected for an amorphous and relatively flexible polymer like TFB, which was verified at the AM1 level by the observation that for our structures the extra charge is indeed localized on average within 1.5 units only. A similarly short localization length was reported by Sancho-García et al. for polyfluorene and TFB.[27] The molecular origin of the short conjugation length are the equilibrium values of the torsional angles along the chain, which

are far away from the planar conformation, see also figures S3 and S5. Working at the low molecular weight limit implies that trapping at the chain ends is maximized,[7] hence we would expect to somewhat underestimate experimental mobilities for long polymer chains, which for TFB are on the order of  $10^{-2}$  cm<sup>2</sup>/(Vs) at room temperature.[18]

We started with running KMC simulations on the original MD samples (denoted as  $\mathbb{R}^1$  in Table I) with site energies and transfer integrals calculated for each sample at the ZINDO level  $(\epsilon_Q, J_Q)$ . In these simulations charges are therefore propagating in a transport network whose properties are static in time, corresponding to single snapshots of the samples. As discussed in the literature, [36, 61, 62] transfer integrals fluctuate over timescales often shorter than the typical hopping times and taking a static picture eventually leads to an underestimation of materials mobility. The KMC output at different electric fields are reported in figure 2. Apart the underestimation of the experimental mobilities of one or two orders of magnitude, what is striking in figure 2 is the absence, at least at the higher temperatures, of the Poole-Frenkel behaviour ( $\ln \mu \propto E^{1/2}$ ) reported experimentally. To demonstrate that this offset is not originated by an unrealistically high value of the reorganization energy, we report in the same panels also mobilities calculated with  $\lambda=100$  meV (filled symbols), which can be considered a lower limit for this parameter [63]. The calculated mobilities for  $\lambda=100$  meV are of course larger than the ones obtained at  $\lambda=300$  meV (empty symbols), consistently with equation 1, but they are still one order of magnitude lower than the experimental values and show a decreasing trend of mobility with field at high temperature.

#### C. Effect of sample size and of dynamic diagonal disorder

In order to understand the factors that contribute to the large offset between simulations and experiment, KMC simulations were carried out by considering different conditions: (i) sample size, varied by replicating in space the original MD box ( $\mathbb{R}^1$ ) two or three times along each Cartesian direction ( $\mathbb{R}^2$  and  $\mathbb{R}^3$  respectively); (ii) diagonal disorder, with site energies assigned from quantum chemistry calculations ( $\epsilon_Q$ ), or randomized by extracting site energy values from a Gaussian distribution having the same standard deviation of the calculated HOMO energies, either at the beginning of each KMC simulation (static,  $\epsilon_G^S$ ) or at each KMC step (dynamic,  $\epsilon_G$ ); (iii) fixed ( $J_Q$ ) or dynamic off-diagonal disorder ( $J_G$ ). While the

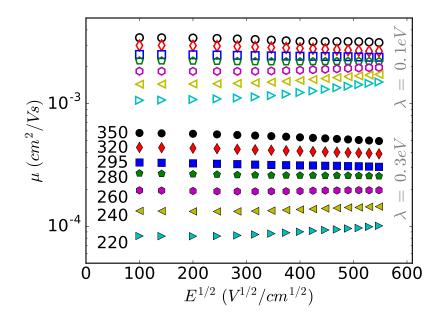


FIG. 2. Average mobility at two different reorganization energies, as a function of electric field and temperature (given in units of K on the left side of the graph):  $\lambda = 0.1$  eV (empty symbols) and 0.3 eV (filled symbols), obtained from  $R^1 | \epsilon_Q | J_Q$  KMC simulations (see table I for details). Mobilities are in quantitative and qualitative disagreement with experimental data.

fixed case for transfer integrals has an exact correspondence with the fixed site energies, in the dynamic case disorder is added on top of the quantum chemistry values, using random numbers drawn from a Gaussian distribution with  $\sigma_J$ =1.43 meV, corresponding to the standard deviation of ZINDO values. Finally, for (iv) we ran KMC simulations in which off-diagonal disorder, that we estimated at  $T_{MD}$  =300 K, is assumed to be completely dynamic and is rescaled to the simulation temperature T using a factor  $(T/T_{MD})^{1/2}$ , according to the electron-phonon coupling model[42, 64]. Furthermore, the effect of increasing the mean values of the transfer integral is considered in an average way, by adding the variance to the square of the QM transfer integrals [65, 66]  $(J_G^T$  conditions). The meaning of the symbols corresponding to different simulations conditions is detailed in table I.

The limitations of calculating mobilities from small-sized systems, yielding to unrealistically high values and excessively broad distributions of single charge velocities, have been well discussed in the recent literature. [28, 49, 56, 67] Among others, Andrienko and coworkers[56, 68] underlined the relationship with the calculated mobility dispersion, ener-

TABLE I. List of symbols utilized for defining the KMC simulations conditions.

Symbol	Definition						
$R^1$	Original four MD samples (125 sites each)						
$R^2$	$2\times2\times2$ replica of MD samples (1000 sites)						
$R^3$	$3\times3\times3$ replica of MD samples (3375 sites)						
$\epsilon_Q$	HOMO energies calculated at ZINDO level at the MD geometry,						
	static in time with $\sigma_{\epsilon}$ =50.2 meV						
$\epsilon_G$	HOMO energies drawn at every KMC step from a Gaussian distribution						
	with $\sigma_{\epsilon} = 50.2 \text{ meV}$						
$\epsilon_G^S$	HOMO energies drawn from a Gaussian distribution at the beginning						
	of each KMC simulation, with $\sigma_{\epsilon} = 50.2 \text{ meV}$						
$J_Q$	Electronic couplings fixed at their quantum mechanics value.						
$J_G$	Electronic couplings fixed at their quantum mechanics value,						
	+ random fluctuations $x$ extracted at every KMC step from a						
	Gaussian distribution with $\sigma_J$ =1.43 meV: $J = J_Q + x$						
$J_Q^T$	Electronic couplings augmented according to a temperature-dependent						
	disorder: $J^2 = J_Q^2 + \sigma_J^2 \cdot (T/T_{MD})$						

getic disorder and sample size in amorphous systems, and showed that for large disorder values ( $\sigma_{\epsilon} > 3k_BT$ ), only samples containing millions of sites can lead to correct results. The problem originates from the fact that the thermally accessible, low energy tail of the density of states, which becomes populated if the charge propagation experiments are adequately long, is not well represented by a discrete distribution of site energies (Figure S5), in particular if the site energies are fixed in time and the distribution is broad (i.e. the disorder is high).

In order to alleviate sample size effects and test their importance as a possible cause of the mismatch with experimental mobilities, we (i) introduced a static energetic disorder by randomizing the site energies at the beginning of each KMC run, (ii) we performed the simulations for periodic replicas of the original MD boxes, so as to have a larger number of sites and then a smoother density of states, and (iii) we calculated mean logarithmic mobilities as suggested by Bobbert and coworkers.[49] By comparing the simulated mobilities for replicated samples  $R^2$  and  $R^3$  we first notice that the differences between the two systems are minimal. In fact for TFB, sample size is not expected to have significant impact on mobility due to the small energetic disorder present in these systems ( $\sigma_{\epsilon} < 3k_BT$ ) at every temperature studied. The convergence with system size is confirmed by the similarity between arithmetic and logarithmic averages (points and lines in figure 3, respectively), and consequently in the remainder of the article only logarithmic mobilities are reported. More

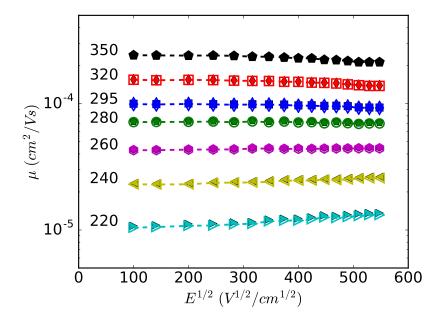


FIG. 3. Simulated mobility of replicated samples as a function of electric field, with randomized static energetic disorder with  $\sigma = 50.2$  meV and fixed transfer integrals, for  $3 \times 3$  replicas of the MD samples  $(R^3|\epsilon_G^S|J_Q)$  conditions. Points correspond to arithmetic averages, dashed lines to logarithmic averages of mobility.

interestingly, by comparing figures 2 and 3, it turns out that the static randomization of site energies scales down the mobility by a factor of about five. This reduction can be explained by the removal of (spurious) static correlations between the site energies which is instead present if  $\epsilon_Q$  values are used; in fact, as shown for instance by Kordt and Andrienko,[69] uncorrelated disorder generally leads to lower mobilities.

To investigate the existence of spatial correlation in  $R^1|\epsilon_Q|J_Q$  simulations, we take a closer look into the mobility anisotropy for one of the four original MD samples (figure 4, filled symbols). Quite strikingly, there are marked differences among the three directions of application of the electric field, which in principle is not expected for an amorphous material. This behavior can be attributed to the relatively small size of the simulation samples (with respect to the oligomer size), since a few hopping events are sufficient for a charge to travel across the whole box. The randomization of site energies (grey-shaded and empty symbols) seems to fix the issue and improve the validity of transport landscape by yielding to more isotropic mobility values, albeit two orders of magnitudes lower than the experimental ones

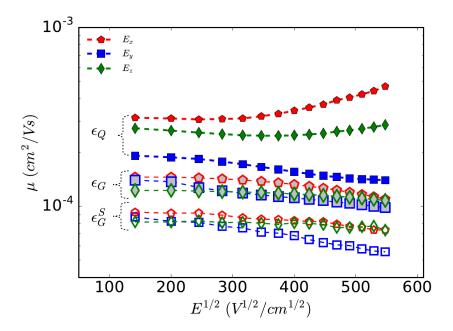


FIG. 4. Mobility as a function of the magnitude and direction (x, y, z) of the electric field for a single MD sample, obtained with fixed site energies (filled symbols,  $R^1|\epsilon_Q|J_Q$  conditions), with site energies randomized at the beginning of every KMC simulation (empty symbols,  $R^3|\epsilon_G^S|J_Q$ ), or randomized at every KMC step (gray-shaded symbols,  $R^3|\epsilon_G|J_Q$ ).

and decreasing with increasing electric field. The mobilities obtained with frequent randomization of site energies ( $\epsilon_G$ ) are as expected slightly larger (and more isotropic) than the ones with static random site energies ( $\epsilon_G^S$ ),

# D. Mobility for dynamic networks

We showed so far that, despite adopting reasonable approximations and parameters, calculated mobilities for static networks substantially differ from the experimental time-of-flight values. One possible source of error could be the adoption of the semiempirical ZINDO Hamiltonian in the calculation of the electronic couplings. ZINDO is in fact known to underestimate the couplings with respect to DFT calculations (which in turn are functional-dependent[70]), however the scaling factor ranges approximately from one to two depending on the material[46, 71], which is not sufficient to recover two orders of magnitude in predicted mobilities. The other possible source of discrepancy with respect to experimental

observations is the neglect of time fluctuations of the couplings, i.e. the static description of the percolation network. Cornil and coworkers elegantly showed how including fluctuations in KMC simulations increases the calculated mobility when the average value of the couplings is smaller than their standard deviation ( $\langle J \rangle / \sigma_J < 1$ ) [60]. The existence of these fluctuations, essentially generated by low frequency vibrations and with typical correlation times in the subpicosecond range, were observed by many authors. [36, 61, 62, 72–74] It then appears that a sensible approach to introduce dynamic disorder on transfer integrals, as an effective way of considering small fluctuations of atomic positions around their equilibrium value, the so-called positional disorder [50]. The electronic couplings calculated from the four MD samples indeed provide an instantaneous photograph of the network which magnifies the hindering effects on mobility due to trap sites. In order to improve the realism of our model, we accordingly reshuffled the transfer integrals at each KMC step by adding to each ZINDO-calculated coupling a random number extracted from a Gaussian distribution with  $\sigma_J = 1.43$  meV. It is worth noting that this scheme  $(J_G)$  (i) assumes that the correlation time of transfer integral fluctuations is faster than the average hopping time; (ii) preserves the topology of the network, but allows for a time-dependent probability of the different pathways; (iii) accounts for the distance dependence of the transfer integrals; (iv) presumes that spatial fluctuations are a measure of the fluctuations in time, as is expected for a crystal, and (v) neglects the temperature dependence of the disorder. A more rigorous way of accounting for dynamic off-diagonal disorder, at least in the framework of Marcus theory, is to augment all the squared transfer integrals of the dynamic variance [65], and to consider that this variance is proportional to temperature [42] ( $J_Q^T$  conditions).

In figure 5, left panel, we plot the results of KMC simulations performed on replicated sample  $R^3$ , and compare them with the experimental data on the right panel. Clearly, introducing dynamic fluctuations on the electronic coupling increases significantly the simulated mobility and is key to improve the agreement with experimental data, which confirms the importance of including dynamic fluctuations in charge transport studies of amorphous systems characterized by small values of electronic couplings.

The last step in this systematic review of the KMC simulation parameters concerns the temperature dependence of the positional (off-diagonal) disorder. So far we implicitly made the assumption of a disorder independent on temperature, but it is worth testing the opposite case and how much this would affect our conclusions ( $\epsilon_G^S|J_Q^T$  conditions). We see in figure 5

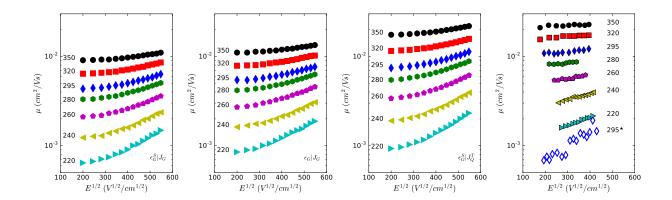


FIG. 5. Calculated and experimental mobilities as a function of the electric field and at different temperatures (K units). From the left to the right: static diagonal disorder and dynamic off-diagonal disorder  $(R^3|\epsilon_G^S|J_G)$ ; dynamic diagonal disorder and off-diagonal disorder  $(R^3|\epsilon_G^S|J_G)$ ; temperature-dependent off-diagonal dynamic disorder  $(R^3|\epsilon_G^S|J_Q^T)$ . On the rightmost panel we show mobilities from experimental time-of-flight measurements (adapted from references [18] and [16]). The second set of measurements at 295 K, denoted by a star, is reported for completeness; however it comes from experiments where an excessive dispersion of the signal was registered, attributed by the authors to the presence of impurities acting as traps, hypothesis consistent with the large value of  $\beta$  in table II.[16] The differences between the two measurement could also be originated by the use of different solvents (toluene[16] vs. trichloroethane[18]).

that the temperature-dependent disorder on transfer integrals does not produce qualitative changes, since it marginally increases the mobility and the separation between the curves at different temperatures, improving the agreement with the experimental data from Malliaras and coworkers.[18] The key factor for recovering the experimental order of magnitude of mobility and the Poole-Frenkel behavior is indeed the introduction of the fluctuations in the electronic couplings.

#### E. Analysis of field and temperature dependence of mobilities

Analytical equations providing relatively simple formulae to fit the variation of mobility with field and temperature [50], as well as charge density, [51, 75] are often used in the experimental literature, with the double purpose of validating theoretical models and extracting intrinsic material parameters under the tacit assumption of the validity of the models. Here

we perform the same exercise using simulated mobilities, in order to compare directly with the parameters extracted from experiment, and to verify the internal consistency of the various theories developed to describe charge transport, since some of the microscopic parameters, and notably the magnitude of the energetic disorder, are known beforehand.

In figure 6 we analyze the temperature dependence of zero field mobilities utilizing either equation 6, which is appropriate for hopping in a Gaussian density of states (left panel), or equation 8, typical of an exponential DOS, or of a Gaussian DOS in the presence of extrinsic charges (right panel).[53] Even if the fitting with the Gaussian disorder model is more accurate as expected, also the Arrhenius equation gives satisfactory results, although not in the whole temperature range. The two fits produce very different infinite temperature mobilities ( $\mu_{\infty} \ll \mu'_{\infty}$  in table II) as a consequence of the different temperature dependence; however, the apparent energetic disorders  $\sigma_A$  and activation energies  $\Delta$  are rather consistent, with  $\Delta \simeq 2\sigma_A$  for all KMC setups and experiments. Changing the reorganization energy from  $\lambda = 200$  meV to 300 meV does not alter significantly the activation energy, consistent with what was reported in reference [76] for oriented polymers and fields perpendicular to the direction of chain alignment, i.e. inter-chain transport.

The variations of a few meV of  $\sigma_A$  upon changing the simulation parameters are too small for drawing conclusions about their origin, but it is interesting to notice that all simulations produce similar apparent energetic disorder  $\sigma_A$ , in line with the experimental value of 66 meV, and only slightly larger than the intrinsic energetic disorder  $\sigma_{\epsilon} = 50.2$  meV. The most significant change with KMC conditions, as shown already in figures 2, 4, and 5, occurs to the infinite temperature mobilities  $\mu_{\infty}$ ; in particular the best match with the experiments of Malliaras et al.[18] is achieved with static diagonal disorder and temperature dependent off-diagonal disorder  $(R^3|\epsilon_G^S|J_Q^T)$ . In figures 6 we also plot the zero field mobilities as obtained with  $R^3|\epsilon_G|J_G$  conditions but with a lower reorganization energy ( $\lambda = 0.2$  eV, green pentagons), with the dual objective of demonstrating the robustness of the results versus the partly arbitrary choice of this parameter, and of showing how the experimental data are well bracketed by the KMC results with  $\lambda = 0.2$  and  $\lambda = 0.3$  eV (red squares).

To conclude our discussion, we focused on the material parameters that are extracted from fitting the temperature dependence of the Poole-Frenkel factors  $\beta$  with equation 7 (Figure 7). The analysis could be performed only for some simulation setups because not all of them showed a Poole-Frenkel behaviour (equation 5); the approach consists in utilizing

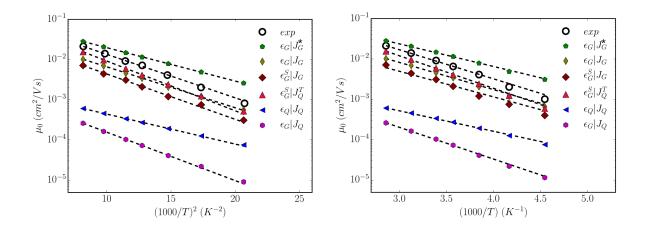


FIG. 6. Temperature dependence of simulated ( $R^3$  supercell) and experimental zero field mobilities, extrapolated with equation 5 for fields ranging from 150<sup>2</sup> to 450<sup>2</sup> V/cm, and corresponding fitting lines with equations 6 (left panel) and 8 (right panel). Fitting parameters are reported in table II. An asterisk\* indicates simulation results for  $\lambda = 0.2$  eV;  $\lambda = 0.3$  eV otherwise.

previously derived  $\sigma_A$  values in equation 7, obtaining the parameter C from the slope of  $\log \beta$  vs  $(\sigma_A/k_BT)^2$ , and finally obtaining the square of the positional disorder  $\Sigma_A$  as the intercept of the fitting line with the x-axis. Figure 7 shows that experiment and simulations follow closely the behavior predicted by Bässler[50], however the angular coefficient obtained from simulations is about half the experimental one when site energies are dynamically changed during KMC (see also Table II). If the diagonal disorder is instead assumed as completely static, we see indeed in figure 7 and table II that larger values of C are obtained, very close to the experimental one. The positional disorder parameter  $\Sigma_A$ , which is a measure of the spread of the transfer integrals, shows instead a better overall agreement between experiment and simulations, independently from the details of the KMC algorithm, indicating that our model adequately captures the physical nature of the actual system.

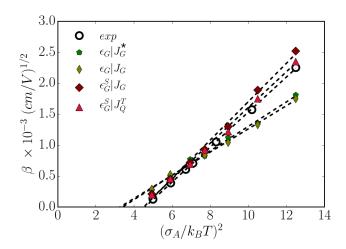


FIG. 7. Calculated ( $R^3$  supercell) and experimental[18] Poole-Frenkel factors plotted against the square of reduced diagonal disorder. Dashed lines correspond to least-square fits with equation 7. Fitting parameters are reported in table II. An asterisk\* indicates simulation results for  $\lambda = 0.2$  eV;  $\lambda = 0.3$  eV otherwise.

TABLE II. Physical parameters extracted by fitting experimental and simulated mobilities with equations 5, 6, and 7. Mobilities are given in units of cm<sup>2</sup>/(Vs),  $\sigma_A$  and  $\Delta$  in meV,  $\beta$  and C in (cm/V)<sup>1/2</sup>.  $\Sigma_A$  is a dimensionless quantity. <sup>(a)</sup> Quantity determined at T=295 K. <sup>(b)</sup> Simulation results for  $\lambda = 0.2$  eV;  $\lambda = 0.3$  eV otherwise.

Quantity	exp[18]	$\exp[16]$	$R^1  \epsilon_Q  J_Q$	$R^3  \epsilon_G  J_Q$	$R^3  \epsilon_G^S  J_G$	$R^3  \epsilon_G  J_G$	$R^3  \epsilon_G  J_G^{(b)}$	$R^3  \epsilon_G^S  J_Q^T$
$\mu_0 \cdot 10^{3} \ ^{(a)}$	9.6	0.31	0.33	0.14	3.1	4.5	14.7	5.9
$\beta \cdot 10^3 \ ^{(a)}$	0.61	3.8	-0.18	-0.62	0.79	0.75	0.77	0.59
$\mu_{\infty}' \cdot 10^3$	2200		17	46	920	870	1130	1990
Δ	140	-	100	155	149	135	111	154
$\mu_{\infty} \cdot 10^3$	190	-	2.4	2.5	77	71	130	160
$\sigma_A$	66	-	54	67	71	64	62	69
$C \cdot 10^4$	2.8	-	-	-	3.1	1.3	1.4	3.3
$\Sigma_A$	2.2	-	_	-	2.1	1.9	1.9	2.1

#### IV. CONCLUSIONS

We presented a theoretical study of the charge transport properties of TFB fluorenetriphenylamine copolymer, a system with remarkably high hole mobility among amorphous organic semiconductors. KMC simulations based on atomistic input, i.e. atomistic molecular dynamics to build the morphologies and quantum chemistry to extract charge hopping rates, are employed to calculate the temperature and electric-field dependence of the hole mobility. Our analysis allows rationalizing why amorphous semi-conducting systems, characterized by poor intermolecular packing, relatively low energetic disorder, and small charge transfer integrals, can have reasonably high hole mobility, highlighting the crucial role of dynamic fluctuations on charge transport. The introduction in the model of dynamic off-diagonal disorder in fact boosts mobility by two orders of magnitude, allowing the achievement of an almost quantitative agreement with experimental mobility values, and the recovery of the so-called Poole-Frenkel behavior of the mobility against the electric field. These results provide a strong evidence of the importance of dynamic energetic disorder for the effective simulation of charge transport in amorphous semiconductors, and introduce a relatively simple and cheap protocol for the computational screening of this class of materials.

#### SUPPLEMENTARY MATERIAL

See Supplementary Material for a detailed description of the molecular mechanics force field parametrization; radial and dihedral distributions from MD simulations; distributions of transfer integrals and site energies; additional mobility vs electric field plots.

#### ACKNOWLEDGMENTS

The work at University of Bordeaux was financed by the French national grant ANR-10-LABX-0042-AMADEus managed by the National Research Agency under the initiative of excellence IdEx Bordeaux programme (reference ANR-10-IDEX-0003-02). Computer time for this study was provided by "Mésocentre de Calcul Intensif Aquitain" computing facilities at Université de Bordeaux and Université de Pau et des Pays de l'Adour. The work at Georgia Tech was supported by the Office of Naval Research (Award No. N00014-17-1-2208). Y.Y. acknowledges financial support from the National Natural Science Foundation of China

(21373229). This project has also received funding from the European Union Horizon 2020 research and innovation programme under grant agreement No. 646176 (EXTMOS).

- R. Noriega, J. Rivnay, K. Vandewal, F. P. V. Koch, N. Stingelin, P. Smith, M. F. Toney, and A. Salleo, Nat. Mater. 12, 1038 (2013).
- [2] S. Wang, S. Fabiano, S. Himmelberger, S. Puzinas, X. Crispin, A. Salleo, and M. Berggren, Proc. Natl. Acad. Sci. USA 112, 10599 (2015).
- [3] S. Y. Son, Y. Kim, J. Lee, G.-Y. Lee, W.-T. Park, Y.-Y. Noh, C. E. Park, and T. Park, J. Am. Chem. Soc. 138, 8096 (2016).
- [4] R. Noriega, A. Salleo, and A. J. Spakowitz, Proc. Natl. Acad. Sci. USA 110, 16315 (2013).
- [5] P. Carbone and A. Troisi, J. Phys. Chem. Lett. 5, 2637 (2014).
- [6] D. Mendels and N. Tessler, Sci. Rep. 6 (2016).
- [7] S. A. Mollinger, A. Salleo, and A. J. Spakowitz, ACS Cent. Sci. 2, 910 (2016).
- [8] R. P. Fornari, P. W. M. Blom, and A. Troisi, Phys. Rev. Lett. 118, 086601 (2017).
- [9] F. C. Grozema and L. D. A. Siebbeles, J. Phys. Chem. Lett. 2, 2951 (2011).
- [10] S. Athanasopoulos, J. Kirkpatrick, D. Martínez, J. M. Frost, C. M. Foden, A. B. Walker, and J. Nelson, Nano Lett. 7, 1785 (2007).
- [11] V. Rühle, J. Kirkpatrick, and D. Andrienko, J. Chem. Phys. 132, 134103 (2010).
- [12] N. E. Jackson, K. L. Kohlstedt, B. M. Savoie, M. Olvera de la Cruz, G. C. Schatz, L. X. Chen, and M. A. Ratner, J. Am. Chem. Soc. 137, 6254 (2015).
- [13] L. Muccioli, G. D'Avino, R. Berardi, S. Orlandi, A. Pizzirusso, M. Ricci, O. M. Roscioni, and C. Zannoni, "Supramolecular organization of functional organic materials in the bulk and at organic/organic interfaces: A modeling and computer simulation approach," in <u>Multiscale Modelling of Organic and Hybrid Photovoltaics</u>, Top. Curr. Chem. (Springer Berlin Heidelberg, Berlin, Heidelberg, 2014) pp. 39–101.
- [14] P. Kordt, J. J. M. van der Holst, M. Al Helwi, W. Kowalsky, F. May, A. Badinski, C. Lennartz, and D. Andrienko, Adv. Funct. Mater. 25, 1955 (2015).
- [15] D. Liu, R. Osuna Orozco, and T. Wang, Phys. Rev. E 88, 022601 (2013).
- [16] M. Redecker, D. D. C. Bradley, M. Inbasekaran, W. W. Wu, and E. P. Woo, Adv. Mater. 11, 241 (1999).
- [17] T. Fukumatsu, A. Saeki, and S. Seki, Appl. Phys Express 5, 061701 (2012).
- [18] H. H. Fong, A. Papadimitratos, and G. G. Malliaras, Appl. Phys. Lett. 89, 172116 (2006).

- [19] J. Hwang, E.-G. Kim, J. Liu, J.-L. Brédas, A. Duggal, and A. Kahn, J. Phys. Chem. C 111, 1378 (2007).
- [20] R. U. A. Khan, D. Poplavskyy, T. Kreouzis, and D. D. C. Bradley, Phys. Rev. B 75, 035215 (2007).
- [21] J.-S. Kim, L. Lu, P. Sreearunothai, A. Seeley, K.-H. Yim, A. Petrozza, C. E. Murphy, D. Beljonne, J. Cornil, and R. H. Friend, J. Am. Chem. Soc. 130, 13120 (2008).
- [22] A. Bruno, L. X. Reynolds, C. Dyer-Smith, J. Nelson, and S. A. Haque, J. Phys. Chem. C 117, 19832 (2013).
- [23] L.-P. Lu, D. Kabra, K. Johnson, and R. H. Friend, Adv. Funct. Mater. 22, 144 (2012).
- [24] A. W. Hains and T. J. Marks, Appl. Phys. Lett. 92, 023504 (2008).
- [25] Z. Zhu, Y. Bai, H. K. H. Lee, C. Mu, T. Zhang, L. Zhang, J. Wang, H. Yan, S. K. So, and S. Yang, Adv. Funct. Mater. 24, 7357 (2014).
- [26] P. Agarwala and D. Kabra, J. Mater. Chem. A 5, 1348 (2017).
- [27] J. C. Sancho-García, C. L. Foden, I. Grizzi, G. Greczynski, M. P. de Jong, W. R. Salaneck, J. L. Brédas, and J. Cornil, J. Phys. Chem. B 108, 5594 (2004).
- [28] M. Jakobsson, M. Linares, and S. Stafström, J. Chem. Phys. 137, 114901 (2012).
- [29] G. D'Avino, S. Mothy, L. Muccioli, C. Zannoni, L. Wang, J. Cornil, D. Beljonne, and F. Castet, J. Phys. Chem. C 117, 12981 (2013).
- [30] R. Volpi, S. Stafström, and M. Linares, J. Chem. Phys. 142, 094503 (2015).
- [31] M. Moral, W.-J. Son, J. C. Sancho-García, Y. Olivier, and L. Muccioli, J. Chem. Theory Comput. 11, 3383 (2015).
- [32] J. C. Phillips, R. Braun, W. Wang, J. s. Gumbart, E. Tajkhorshid, E. Villa, C. Chipot, R. D. Skeel, L. Kalé, and K. Schulten, J. Comput. Chem. 26, 1781 (2005).
- [33] R. A. Marcus, Rev. Mod. Phys **65,3**, 599 (1993).
- [34] V. Coropceanu, J. Cornil, D. A. da Silva Filho, Y. Olivier, R. Silbey, and J.-L. Brédas, Chem. Rev. 107, 926 (2007).
- [35] M. J. S. Dewar, E. G. Zoebisch, E. F. Healy, and J. J. P. Stewart, J. Am. Chem. Soc. 107, 3902 (1985).
- [36] C. Poelking and D. Andrienko, Macromolecules 46, 8941 (2013).
- [37] M. Malagoli and J. L. Brédas, Chem. Phys. Lett. **327**, 13 (2000).
- [38] D. P. McMahon and A. Troisi, J. Phys. Chem. Lett. 1, 941 (2010).

- [39] V. Lemaur, M. Steel, D. Beljonne, J.-L. Brédas, and J. Cornil, J. Am. Chem. Soc. 127, 6077 (2005).
- [40] E. F. Valeev, V. Coropceanu, D. A. da Silva Filho, S. Salman, and J.-L. Brédas, J. Am. Chem. Soc. 128, 9882 (2006).
- [41] J. Ridley and M. Zerner, Theoret. Chim. Acta (Berl.) 32, 111 (1973).
- [42] Y. Yi, V. Coropceanu, and J.-L. Brédas, J. Chem. Phys. 137, 164303 (2012).
- [43] F. Castet, G. D'Avino, L. Muccioli, J. Cornil, and D. Beljonne, Phys. Chem. Chem. Phys. 16, 20279 (2014).
- [44] R. Volpi and M. Linares, J. Comput. Chem. 38, 1039 (2017).
- [45] G. D'Avino, L. Muccioli, F. Castet, C. Poelking, D. Andrienko, Z. G. Soos, J. Cornil, and D. Beljonne, J. Phys.: Condens. Matter 28, 433002 (2016).
- [46] G. D'Avino, Y. Olivier, L. Muccioli, and D. Beljonne, J. Mater. Chem. C 4, 3747 (2016).
- [47] D. T. Gillespie, Annu. Rev. Phys. Chem. 58, 35 (2007).
- [48] A. J. Campbell, R. Rawcliffe, A. Guite, J. C. D. Faria, A. Mukherjee, M. A. McLachlan, M. Shkunov, and D. D. C. Bradley, Adv. Funct. Mater. 26, 3720 (2016).
- [49] A. Massé, R. Coehoorn, and P. A. Bobbert, Phys. Rev. Lett. 113, 116604 (2014).
- [50] H. Bässler, Physica Status Solidi (b) **175**, 15 (1993).
- [51] S. D. Baranovskii, Phys. Status Solidi B 251, 487 (2014).
- [52] H. H. Fong, A. Papadimitratos, J. Hwang, A. Kahn, and G. G. Malliaras, Adv. Funct. Mater. 19, 304 (2009).
- [53] N. I. Craciun, J. Wildeman, and P. W. M. Blom, Phys. Rev. Lett. 100, 056601 (2008).
- [54] G. Horowitz, R. Hajlaoui, and P. Delannoy, J. Phys. III France 5, 355 (1995).
- [55] N. G. Martinelli, M. Savini, L. Muccioli, Y. Olivier, F. Castet, C. Zannoni, D. Beljonne, and J. Cornil, Adv. Funct. Mater. 19, 3254 (2009).
- [56] A. Lukyanov and D. Andrienko, Phys. Rev. B 82, 193202 (2010).
- [57] J. Cornil, S. Verlaak, N. Martinelli, A. Mityashin, Y. Olivier, T. Van Regemorter, G. D'Avino, L. Muccioli, C. Zannoni, F. Castet, D. Beljonne, and P. Heremans, Acc. Chem. Res. 46, 434 (2013).
- [58] A. Mityashin, O. M. Roscioni, L. Muccioli, C. Zannoni, V. Geskin, J. Cornil, D. Janssen, S. Steudel, J. Genoe, and P. Heremans, ACS Appl. Mater. Interfaces 6, 15372 (2014).

- [59] R. S. Sánchez-Carrera, P. Paramonov, G. M. Day, V. Coropceanu, and J.-L. Brédas, J. Am. Chem. Soc. 132, 14437 (2010).
- [60] N. G. Martinelli, Y. Olivier, S. Athanasopoulos, M.-C. R. Delgado, K. R. Pigg, D. A. da Silva Filho, R. S. Sanchez-Carrera, E. Venuti, R. G. Della Valle, J.-L. Bredas, D. Beljonne, and J. Cornil, ChemPhysChem 10, 2265 (2009).
- [61] Y. Olivier, L. Muccioli, V. Lemaur, Y. H. Geerts, C. Zannoni, and J. Cornil, J. Phys. Chem. B 113, 14102 (2009).
- [62] N. E. Jackson, L. X. Chen, and M. A. Ratner, Proc. Natl. Acad. Sci. USA 113, 8595 (2016).
- [63] J.-L. Brédas, D. Beljonne, V. Coropceanu, and J. Cornil, Chem. Rev. 104, 4971 (2004).
- [64] N. R. Tummala, Z. Zheng, S. G. Aziz, V. Coropceanu, and J.-L. Brédas, J. Phys. Chem. Lett. 6, 3657 (2015).
- [65] I. A. Balabin and J. N. Onuchic, Science **290**, 114 (2000).
- [66] D. N. Beratan, S. S. Skourtis, I. A. Balabin, A. Balaeff, S. Keinan, R. Venkatramani, and D. Xiao, Acc. Chem. Res. 42, 1669 (2009).
- [67] I. Yavuz and S. A. Lopez, Phys. Chem. Chem. Phys. 19, 231 (2017).
- [68] P. Kordt, T. Speck, and D. Andrienko, Phys. Rev. B 94, 014208 (2016).
- [69] P. Kordt and D. Andrienko, J. Chem. Theory Comput. 12, 36 (2016).
- [70] C. Sutton, J. S. Sears, V. Coropceanu, and J.-L. Brédas, J. Phys. Chem. Lett. 4, 919 (2013).
- [71] J. Idé, D. Fazzi, M. Casalegno, S. V. Meille, and G. Raos, J. Mater. Chem. C 2, 7313 (2014).
- [72] A. Troisi and G. Orlandi, J. Phys. Chem. A 110, 4065 (2006).
- [73] A. Troisi, D. L. Cheung, and D. Andrienko, Phys. Rev. Lett. 102, 116602 (2009).
- [74] H. Geng, Q. Peng, L. Wang, H. Li, Y. Liao, Z. Ma, and Z. Shuai, Adv. Mater. 24, 3568 (2012).
- [75] W. F. Pasveer, J. Cottaar, C. Tanase, R. Coehoorn, P. A. Bobbert, P. W. M. Blom, D. M. de Leeuw, and M. A. J. Michels, Phys. Rev. Lett. 94, 206601 (2005).
- [76] M. Jakobsson and S. Stafström, J. Chem. Phys. 135, 134902 (2011).