

SUPPORTING INFORMATION

Furfural valorization to γ -valerolactone on Zr/Sn zeolite-supported catalysts in a liquid-phase continuous-flow reactor

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Fresh characterization

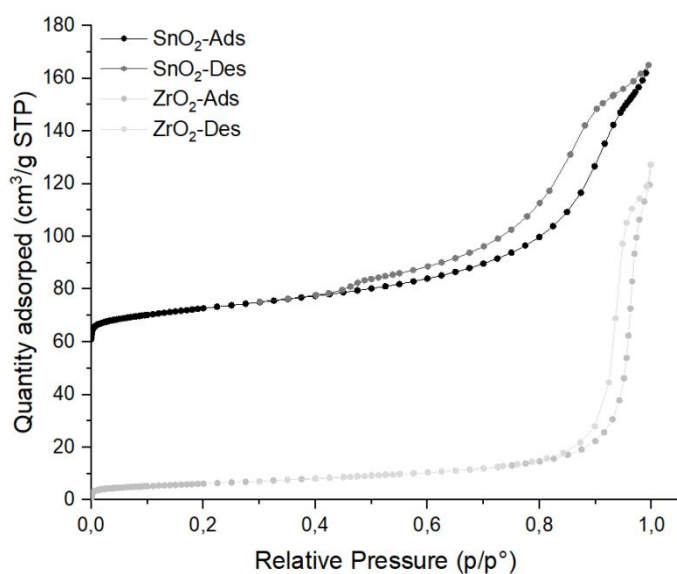


Figure S1. N₂ adsorption and desorption isotherms of SnO₂ and ZrO₂.

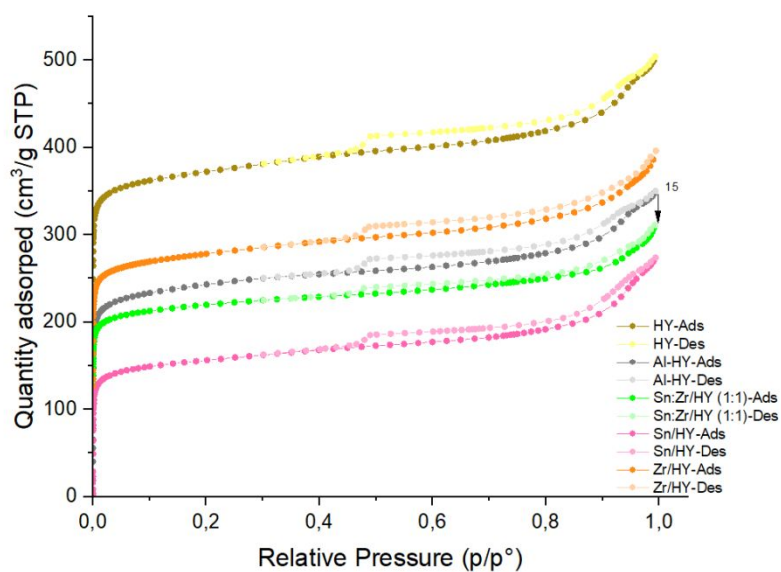


Figure S2. N₂ adsorption and desorption isotherms for Sn and Zr-based catalysts supported on dealuminated zeolite, HY, and for Al-HY and HY zeolites.

Table S1. Porosity characterization for the fresh and spent catalysts.

Catalyst	Pore volume <i>fresh</i> (<i>cm</i> ³ / <i>g</i>)	Pore volume spent (<i>cm</i> ³ / <i>g</i>)	Pore diameter fresh (Å)	Pore diameter spent (Å)
HY	0.48	0.54	60	123
Sn/HY	0.37	0.38	69	125
Sn:Zr/HY (9:1)	0.35	-	72	-
Sn:Zr/HY (1:1)	0.34	0.41	61	138
Sn:Zr/HY (1:9)	0.39	-	61	-

Zr/HY	0.39	0.46	62	128
SnO ₂	0.06	0.15	229	204
ZrO ₂	0.14	0.16	85	85

Catalytic tests

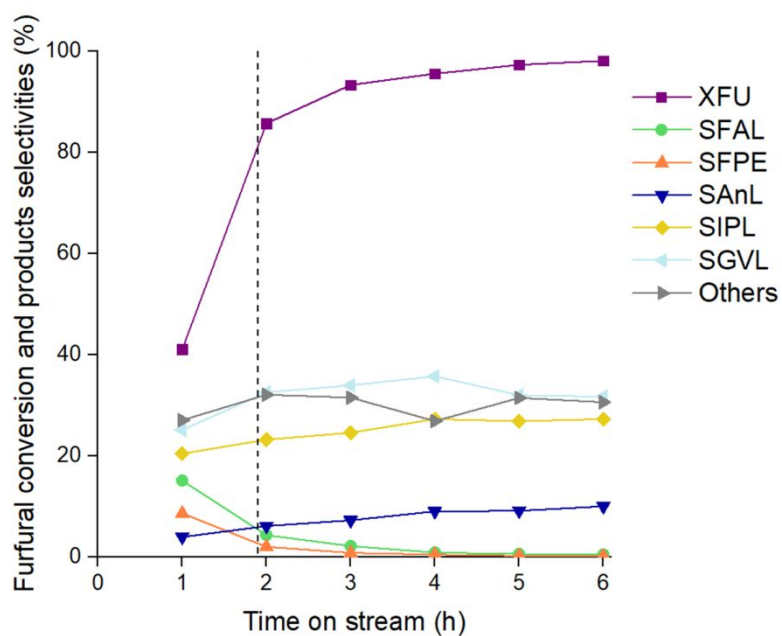


Figure S3. FU conversion and selectivity of products as a function of time (h) on Sn/HY. Reaction conditions: [FU]=67

mM, 1 eq H₂O, τ =10 min, T=180 °C, mcat= 0.5194 g.

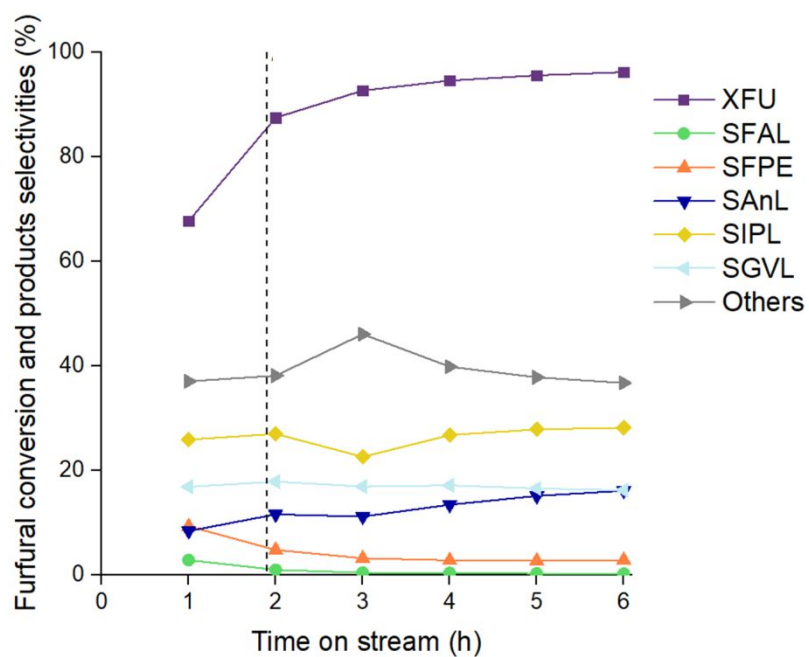


Figure S4. FU conversion and selectivity of products as a function of time (h) on Zr/HY. Reaction conditions: [FU]=67 mM, 1 eq H₂O, τ =10 min, T=180 °C, mcat= 0.4792 g.

Comparison between fresh and spent characterization

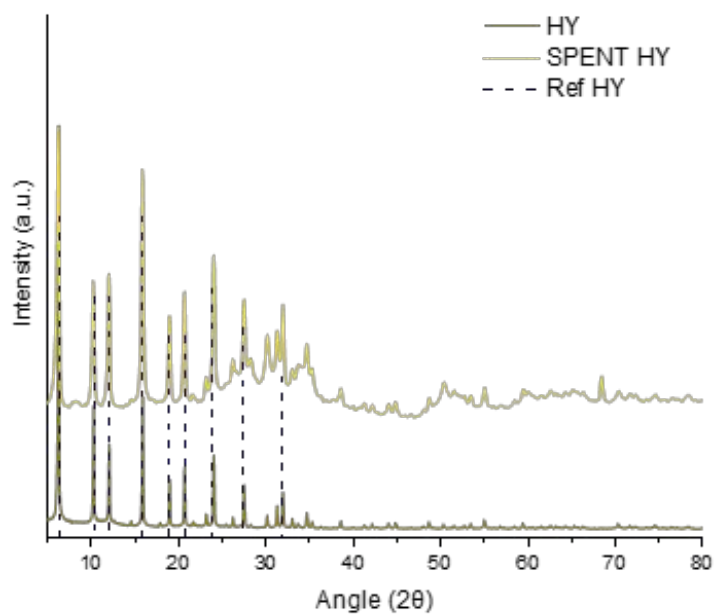


Figure S5. XRD patterns for HY fresh and spent catalysts.

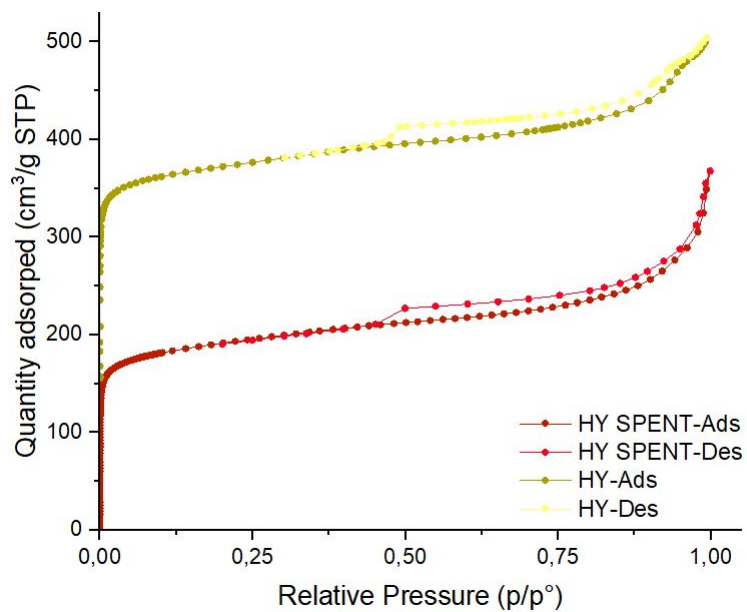


Figure S6. N₂ adsorption and desorption isotherms for HY fresh and spent catalysts.

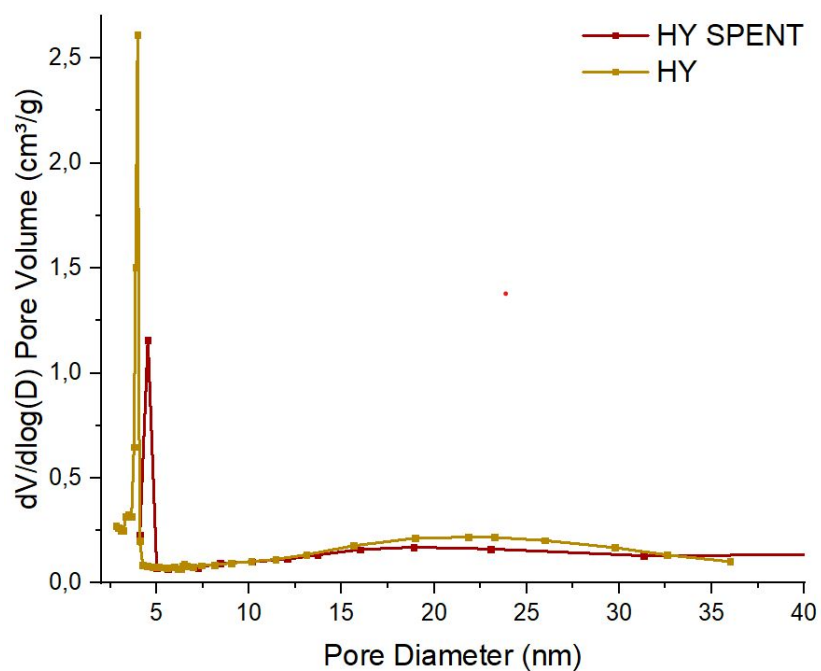


Figure S7. Pore size distribution for HY fresh and spent catalysts.

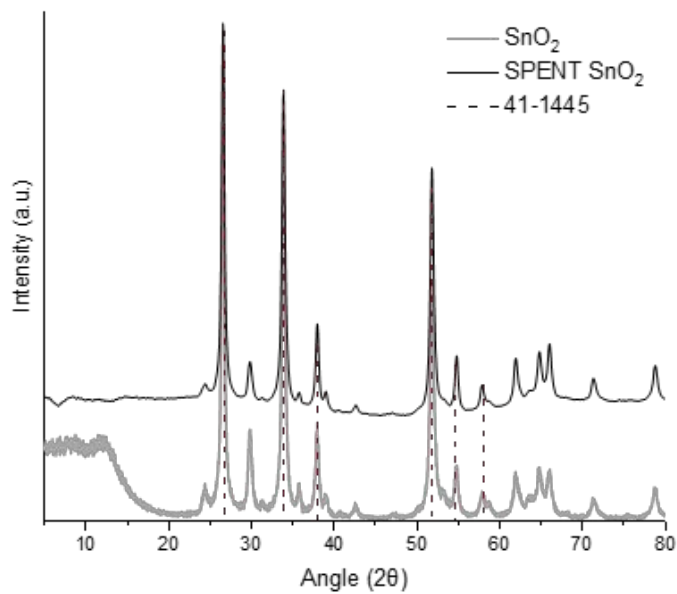


Figure S8. XRD patterns for SnO₂ fresh and spent catalysts.

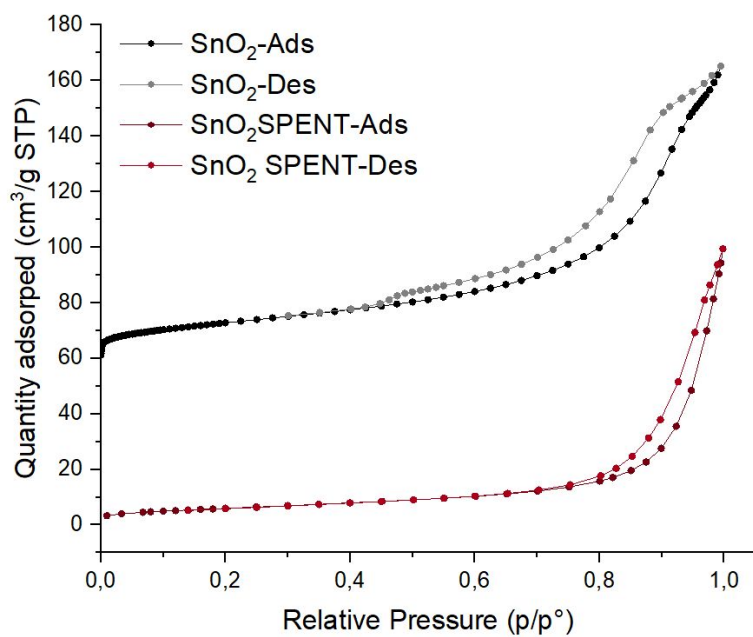


Figure S9. N₂ adsorption and desorption isotherms for SnO₂ fresh and spent catalysts.

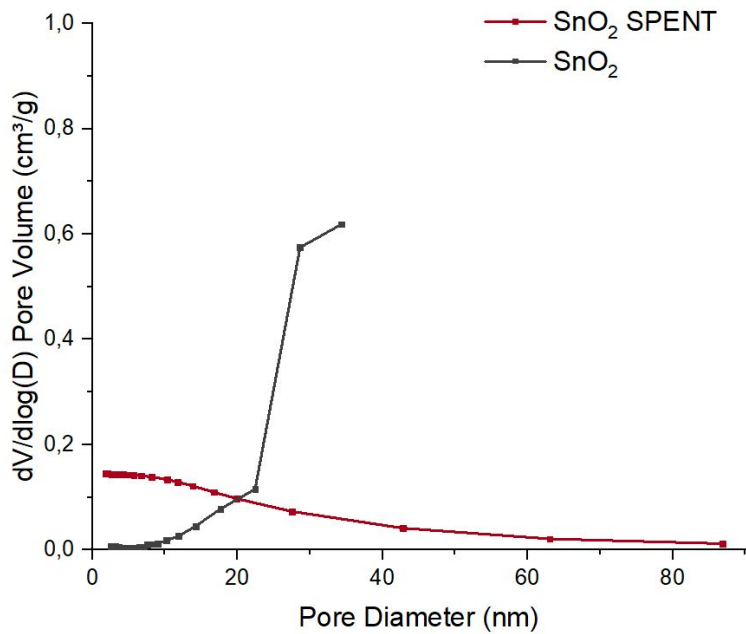


Figure S10. Pore size distribution for SnO₂ fresh and spent catalysts.

ZrO₂

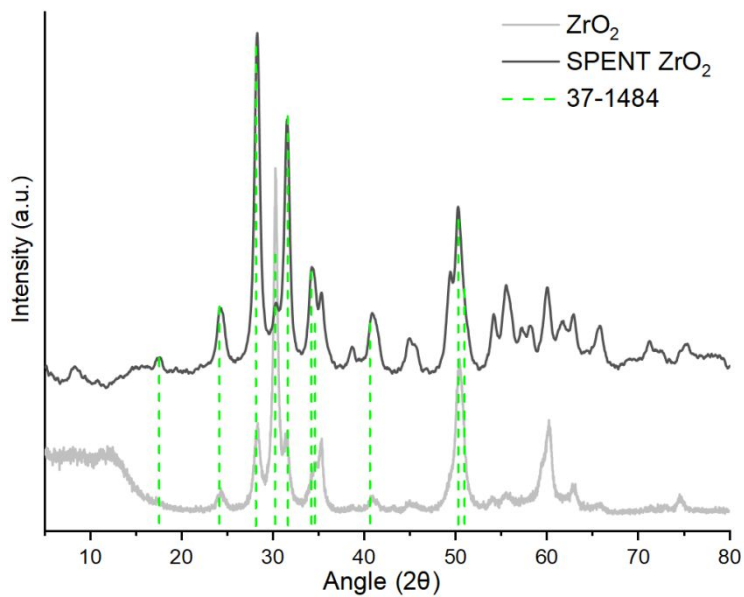


Figure S11. XRD patterns for ZrO₂ fresh and spent catalysts.

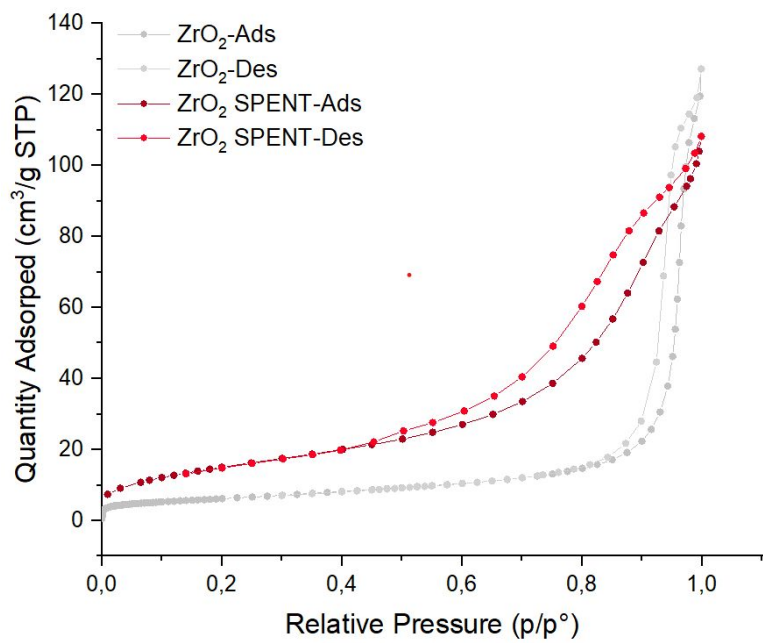


Figure S12. N₂ adsorption and desorption isotherms for ZrO₂ fresh and spent catalysts.

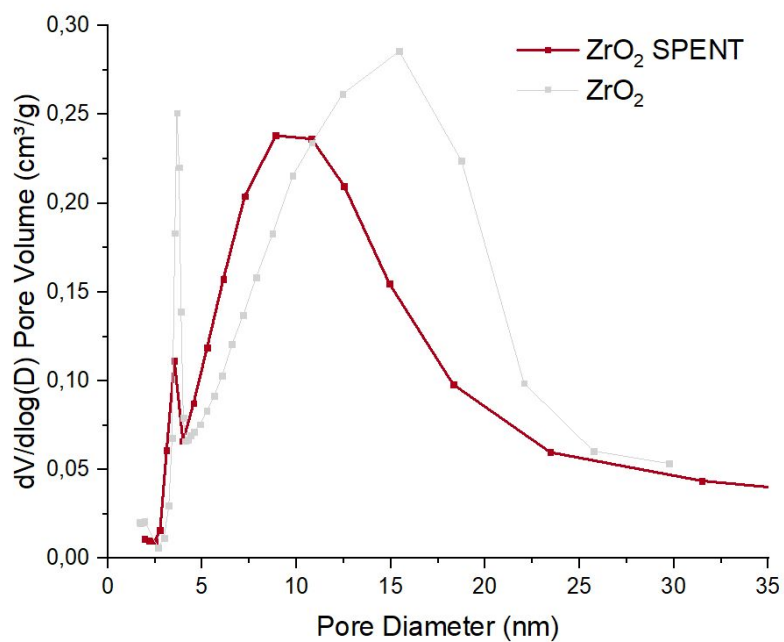


Figure S13. Pore size distribution for SnO₂ fresh and spent catalysts.

Sn/HY

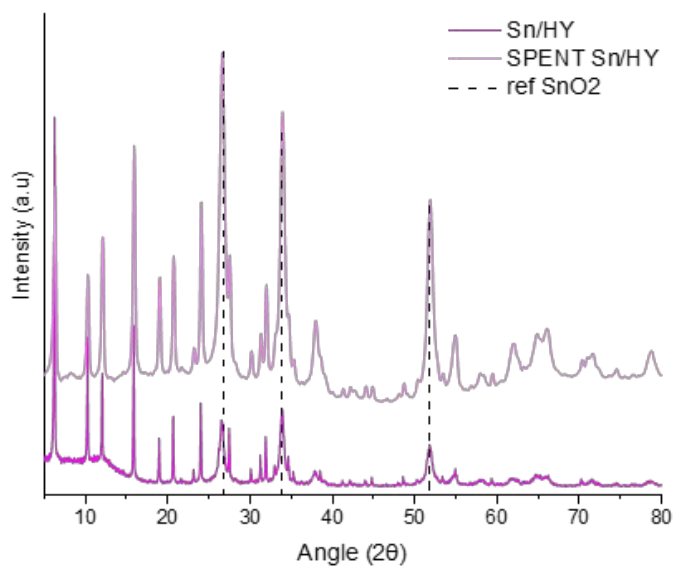


Figure S14. XRD patterns for Sn/HY fresh and spent catalysts.

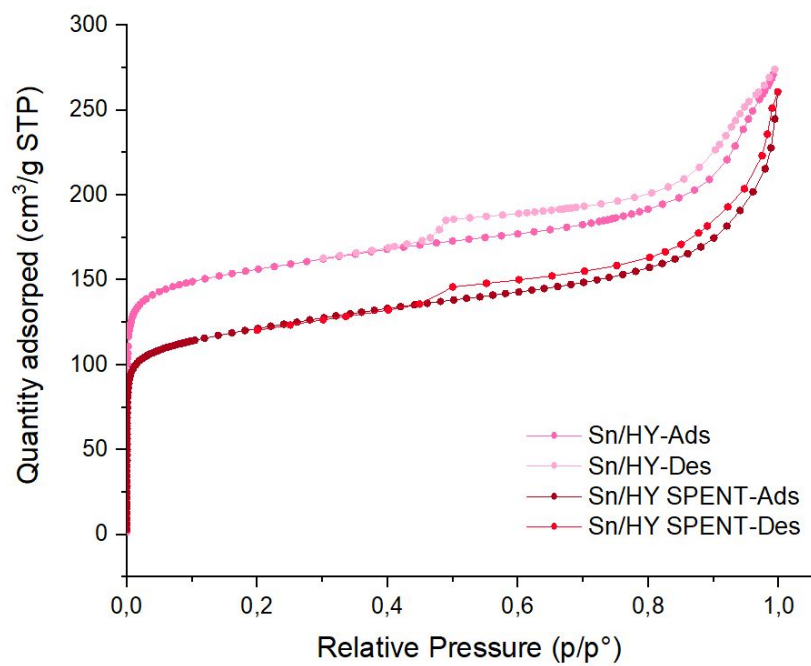


Figure S15. N₂ adsorption and desorption isotherms for Sn/HY fresh and spent catalysts.

Zr/HY

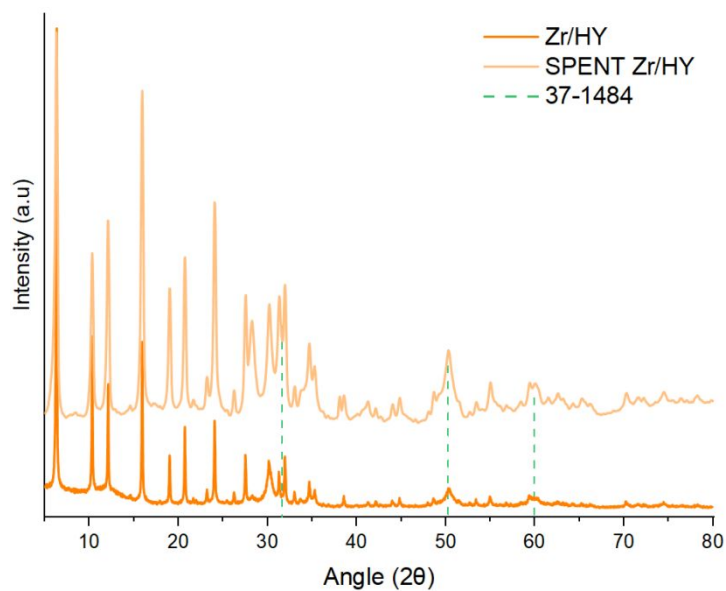


Figure S16. XRD patterns for Zr/HY fresh and spent catalysts.

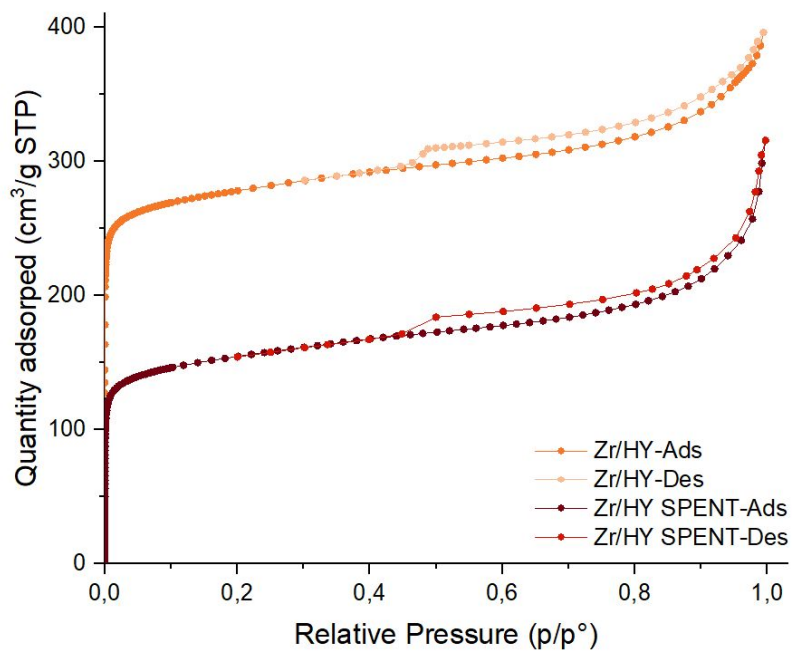


Figure S17. N₂ adsorption and desorption isotherms for Zr/HY fresh and spent catalysts.

Sn:Zr/HY (9:1)

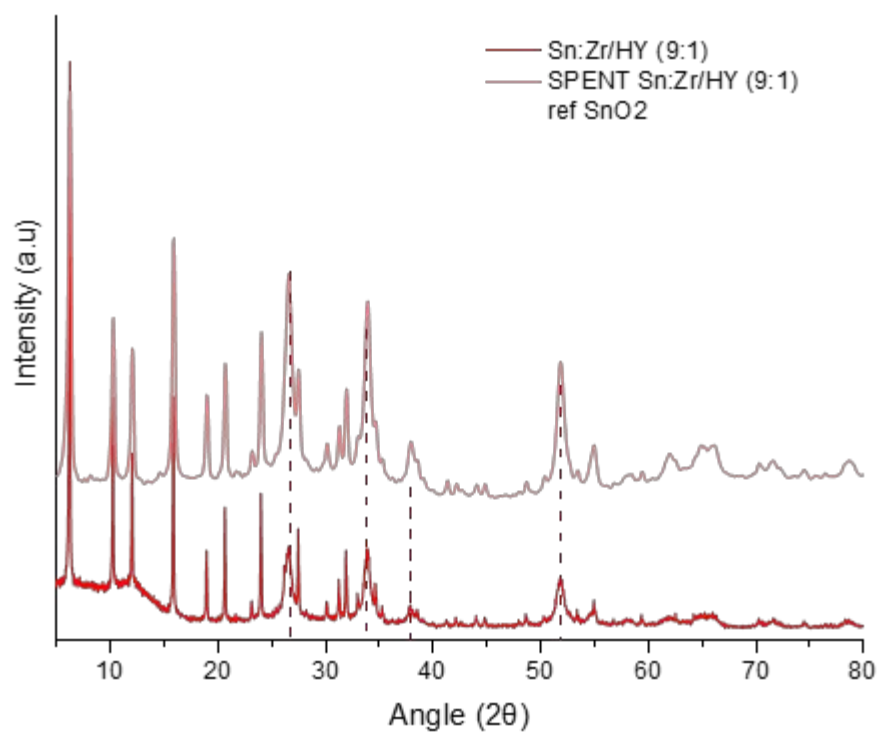


Figure S18. XRD patterns for Sn:Zr/HY (9:1) fresh and spent catalysts.

Sn:Zr/HY (1:9)

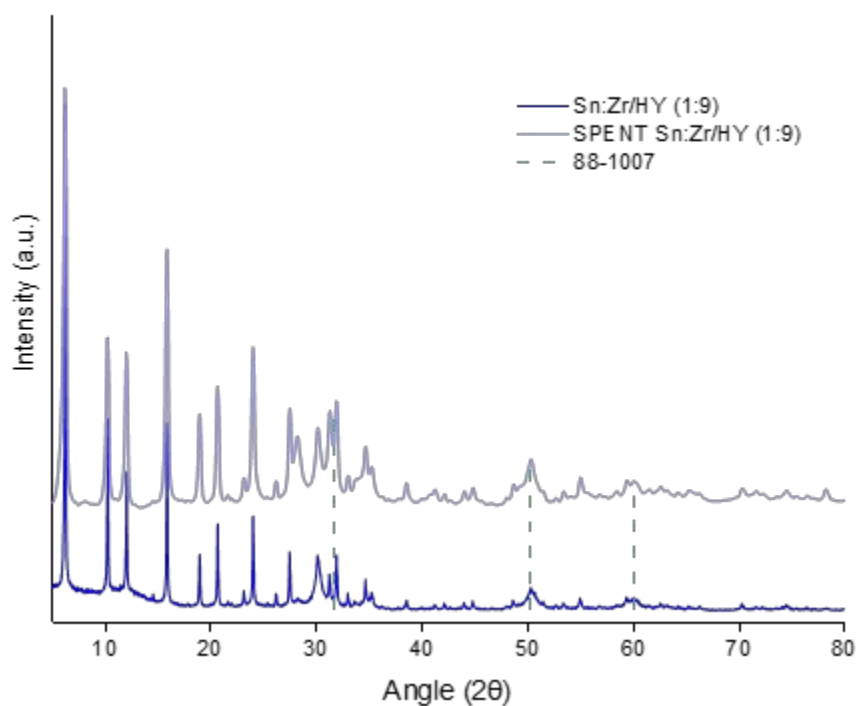


Figure S19. XRD patterns for Sn:Zr/HY (1:9) fresh and spent catalysts.

Study of the reaction mechanism

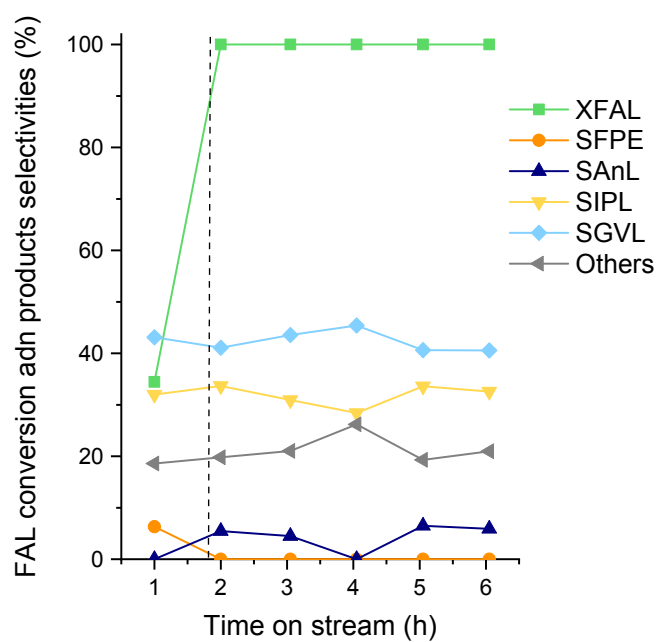


Figure S20. FAL conversion and selectivity of products as a function of time (h) on Sn:Zr/HY (1:1). Reaction

conditions: [FAL]=67 mM, 1 eq H₂O, τ =10 min, T=180 °C, mcat = 0.47 g.

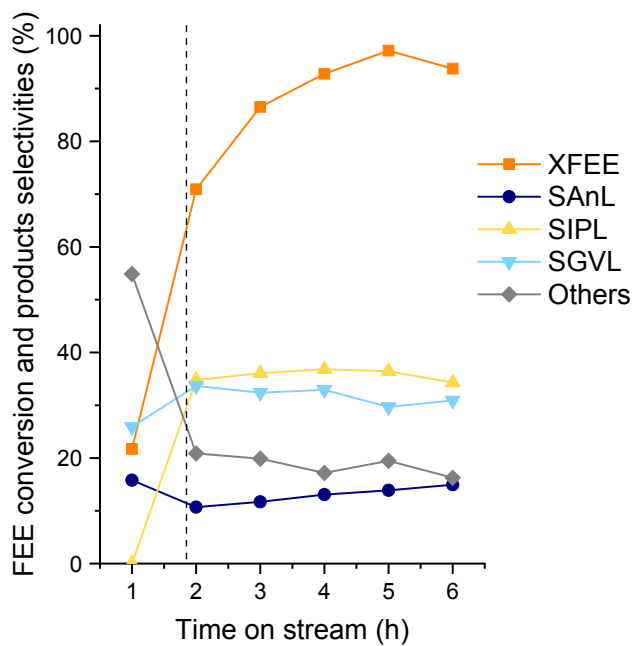


Figure S21. FEE conversion and selectivity of products as a function of time (h) on Sn:Zr/HY (1:1). Reaction

conditions: [FEE]=67 mM, 1 eq H₂O, τ =10 min, T=180 °C, mcat = 0.47.

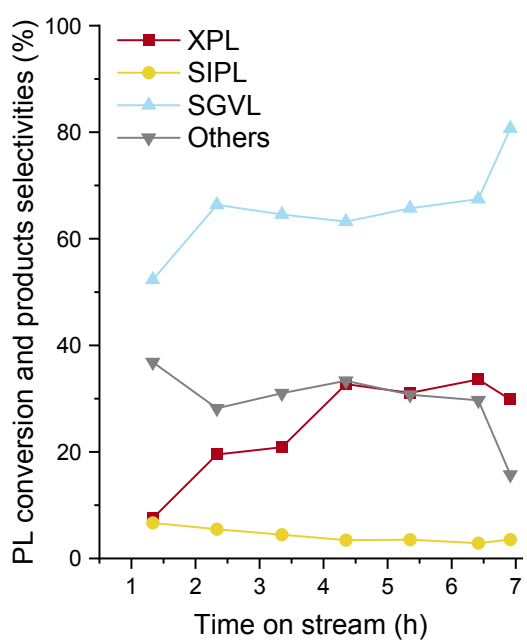


Figure S22. IPL conversion and selectivity of products as a function of time (h) on Sn:Zr/HY (1:1). Reaction conditions:

[IPL]=67 mM, 1 eq H₂O, τ =10 min, T=180 °C, mcat = 0.47 g.

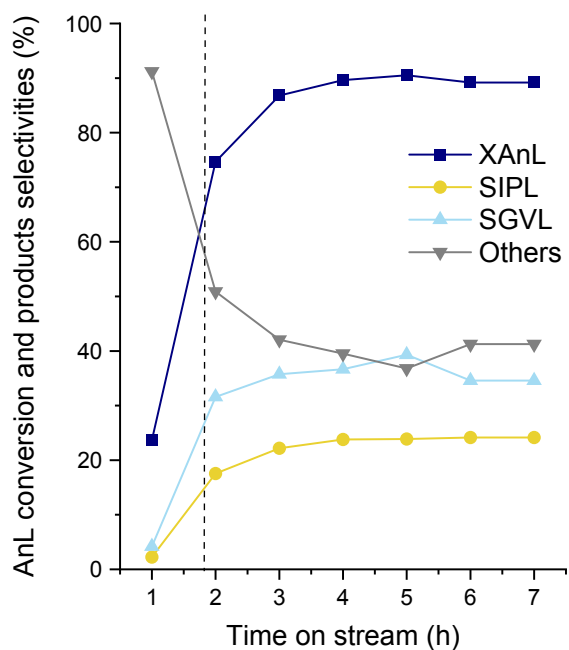


Figure S23. Trends of AnL conversion and selectivity of products as a function of time (h) on Sn:Zr/HY (1:1). Reaction

conditions: [AnL]=67 mM, 1 eq H₂O, τ =10 min, T=180 °C, mcat = 0.47.

ASSOCIATED CONTENT

Data availability

The data underlying this study are openly available in “*Data on the valorization of furfural to γ -valerolactone in liquid-phase continuous-flow over Zr/Sn zeolite-supported catalysts*” at <https://doi.org/10.6092/unibo/amsacta/8555>

