

Phase Behavior and Proton Conductivity in Crown Ether-Based Supramolecular Sodium Hydrogensulfate Complexes

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CRYSTAL DATA AND REFINEMENT DETAILS

Table SI1. Crystal data and refinement details for: [15-crown-5·Na]HSO₄ (**1**) at RT and [benzo-15-crown-5·Na]HSO₄ (**2**) at 100 K and RT.

	1 (RT)	2 (100K)	2 (RT)
Formula	C ₁₀ H ₂₁ NaO ₉ S	C ₁₄ H ₂₁ NaO ₉ S	C ₁₄ H ₂₁ NaO ₉ S
FW (g/mol)	340.33	388.36	388.36
Cryst. Sys.	monoclinic	monoclinic	monoclinic
Space Group	P2 ₁ /n	P2 ₁ /n	P2 ₁ /n
a/Å	9.9522(7)	11.9217(4)	12.1259(2)
b/Å	10.1250(5)	12.0540(4)	12.2293(2)
c/Å	16.270(1)	12.1174(4)	12.1380(2)
α/°	90	90	90
β/°	103.385(7)	100.094(4)	99.854(2)
γ/°	90	90	90
Volume/Å³	1594.93(17)	1714.37(11)	1773.40(5)
Z	4	4	4
ρ_{calc} g/cm³	1.417	1.505	1.455
μ/mm⁻¹	0.268	0.260	0.251
measd rflns	6735	13167	26133
indep rflns	3608	4091	4488
R₁	0.0798	0.0456	0.0775
wR₂	0.2066	0.0930	0.2183

POWDER XRD PATTERN COMPARISON

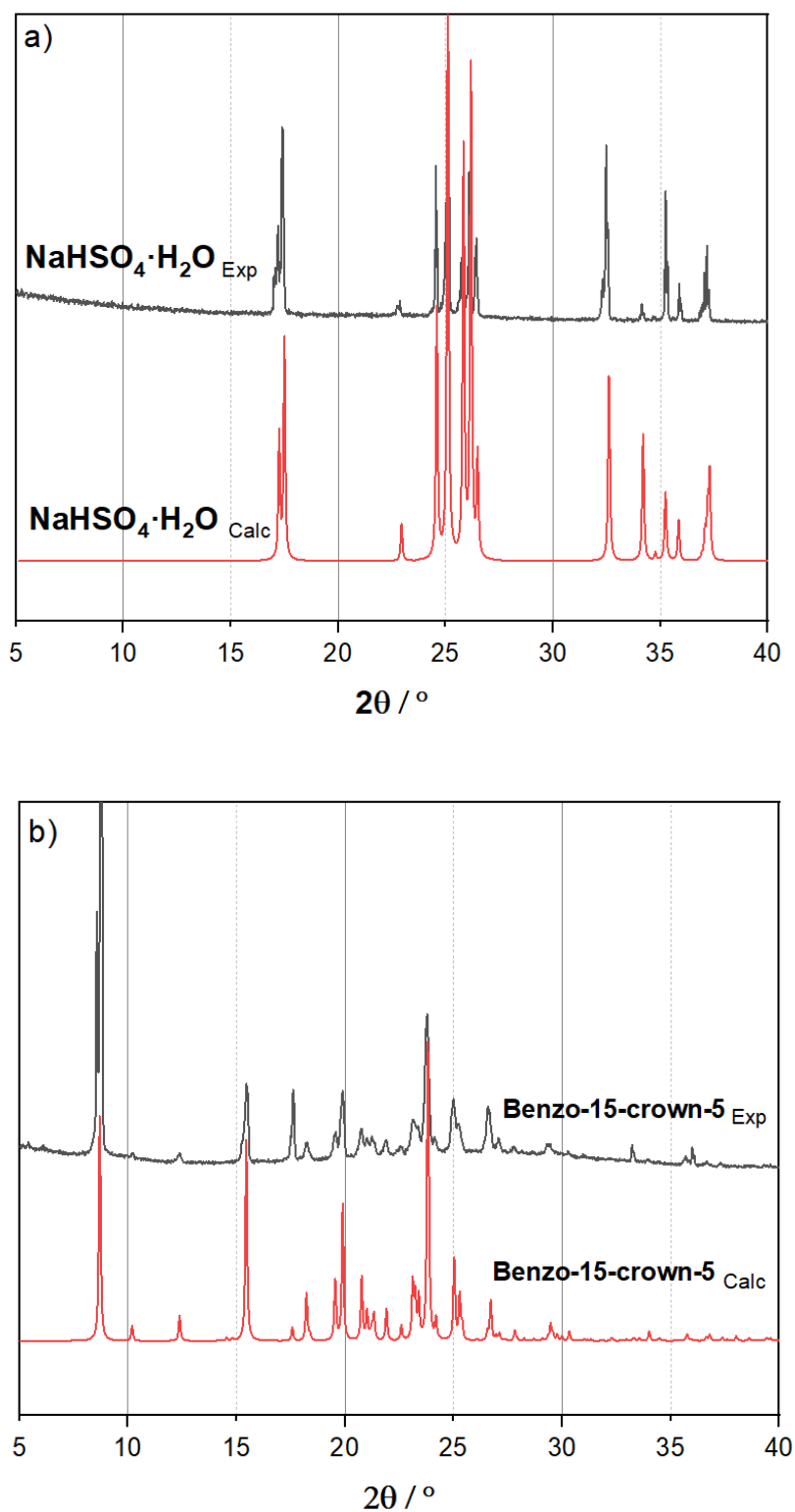


Figure S11. Comparison between the calculated powder patterns obtained from the databases and the experimental ones of: (a) $\text{NaHSO}_4 \cdot \text{H}_2\text{O}$ (ICSD Reference code: 15066) and (b) Benzo-15-crown-5 (CCDC Reference code: OHBOCP).

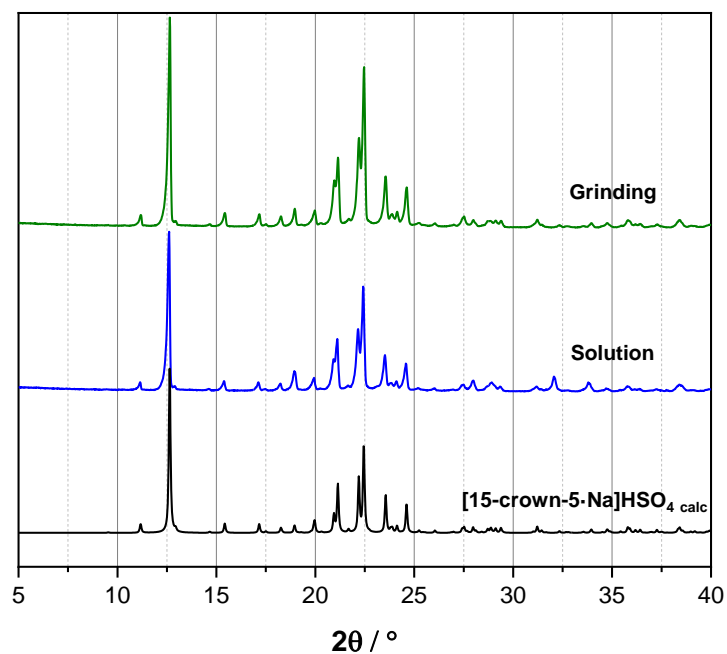


Figure SI2. Comparison between the calculated powder patterns based on the single-crystal structures of [15-crown-5-Na]HSO₄ (1) and the corresponding experimental ones obtained from grinding and crystallization from solution.

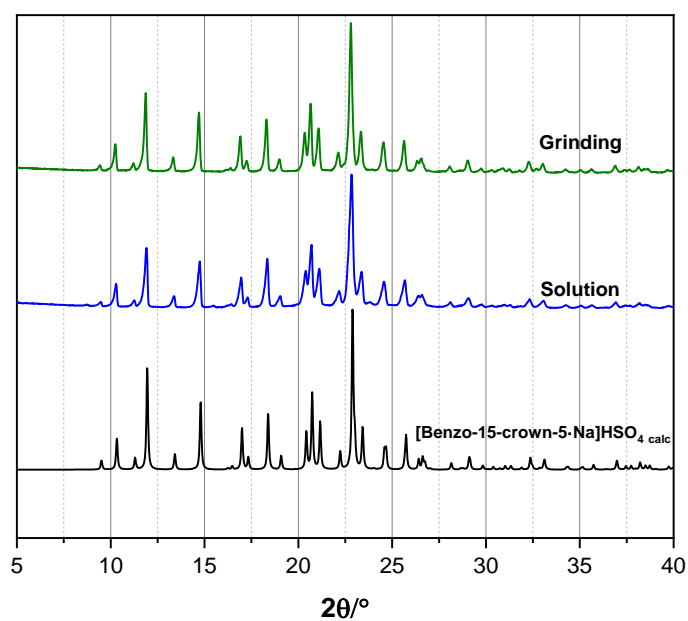


Figure SI3. Comparison between the calculated powder patterns based on the single-crystal structures of [benzo-15-crown-5-Na]HSO₄ (2) and the corresponding experimental ones obtained from grinding and crystallization from solution.

ATR-FTIR SPECTRA

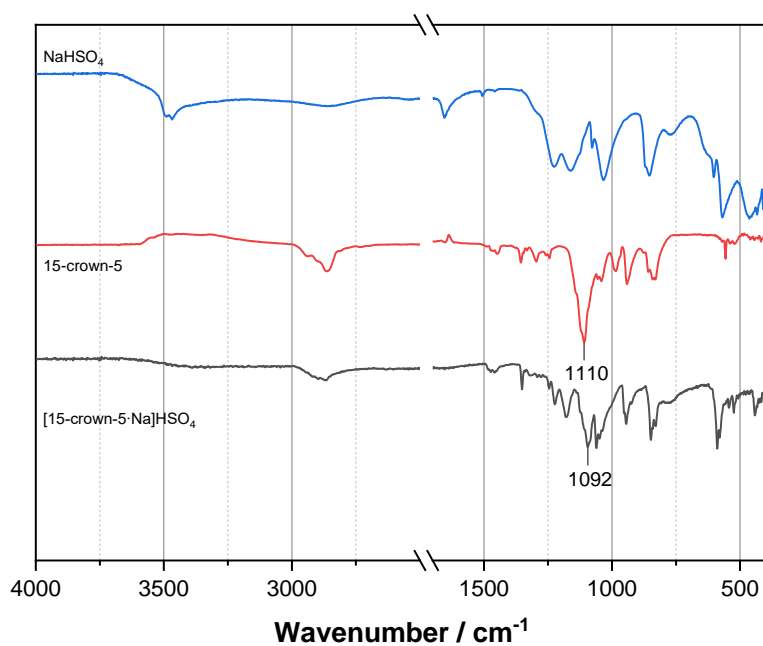


Figure SI4. Comparison between ATR-FTIR spectra recorded on polycrystalline $[\text{15-crown-5-Na}]\text{HSO}_4$ (**1**) and on the starting materials.

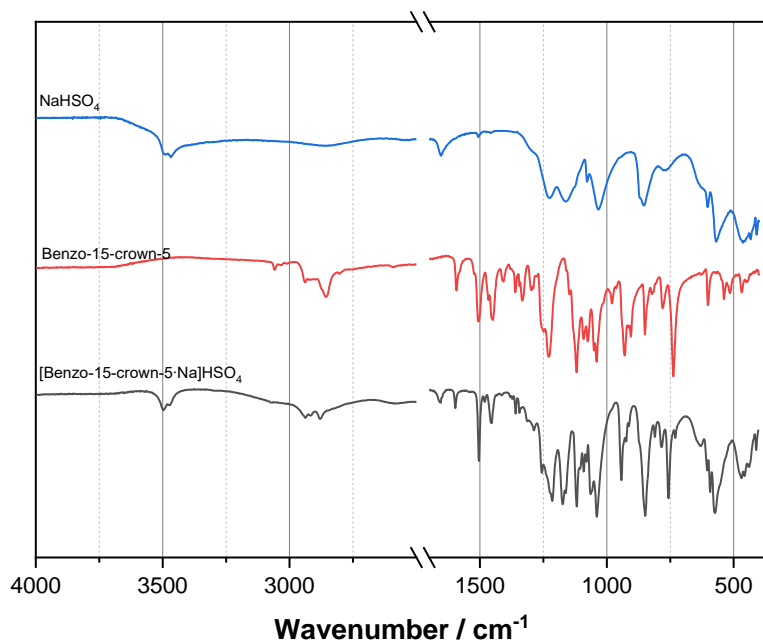


Figure SI5. Comparison between ATR-FTIR spectra recorded on polycrystalline $[\text{benzo-15-crown-5-Na}]\text{HSO}_4$ (**2**) and on the starting materials.

THERMOGRAVIMETRIC ANALYSES

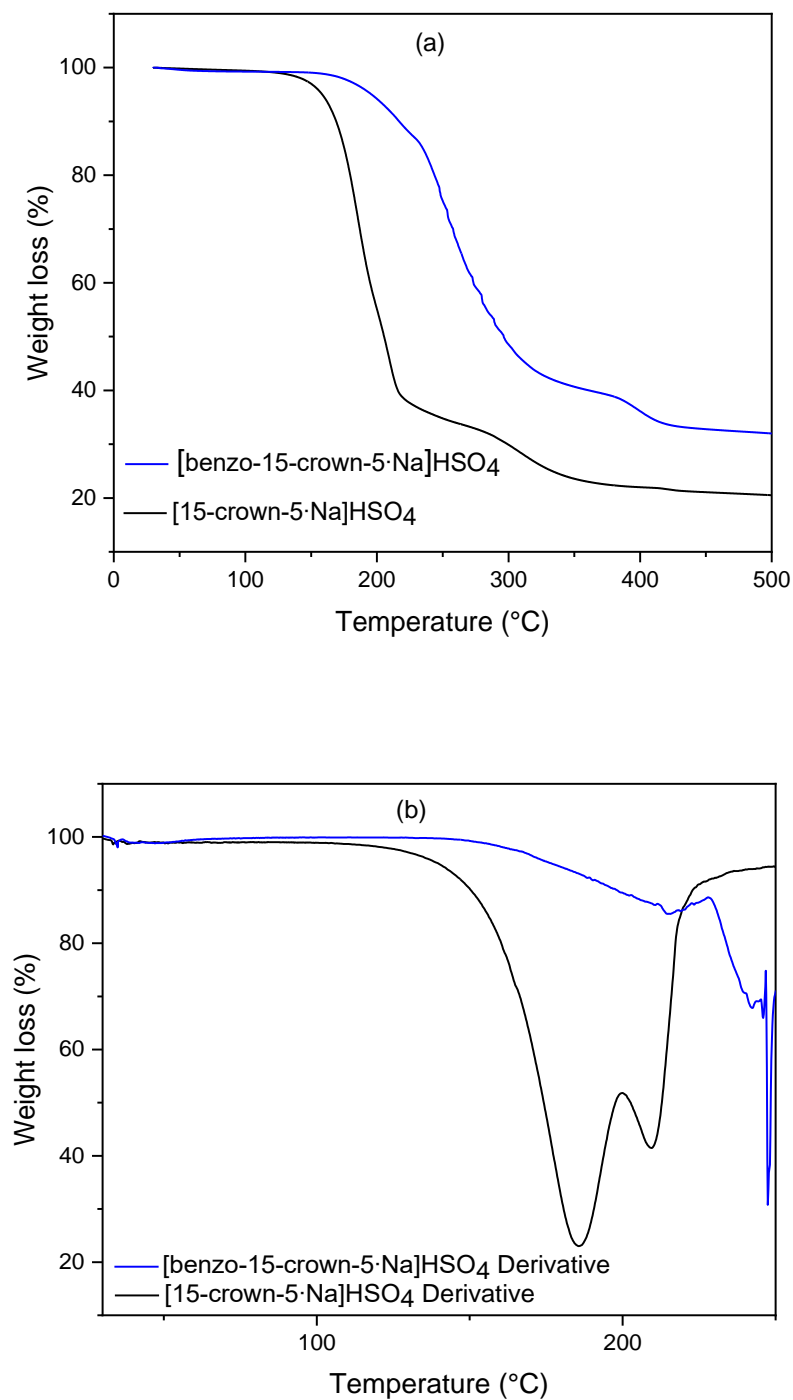


Figure S16. (a) Thermograms recorded for the supramolecular complexes [15-crown-5-Na]HSO₄ (black line), and [benzo-15-crown-5-Na]HSO₄ (blue line). (b) The corresponding derivatives results that highlight thermal stability.

DIFFERENTIAL SCANNING CALORIMETRY TRACES

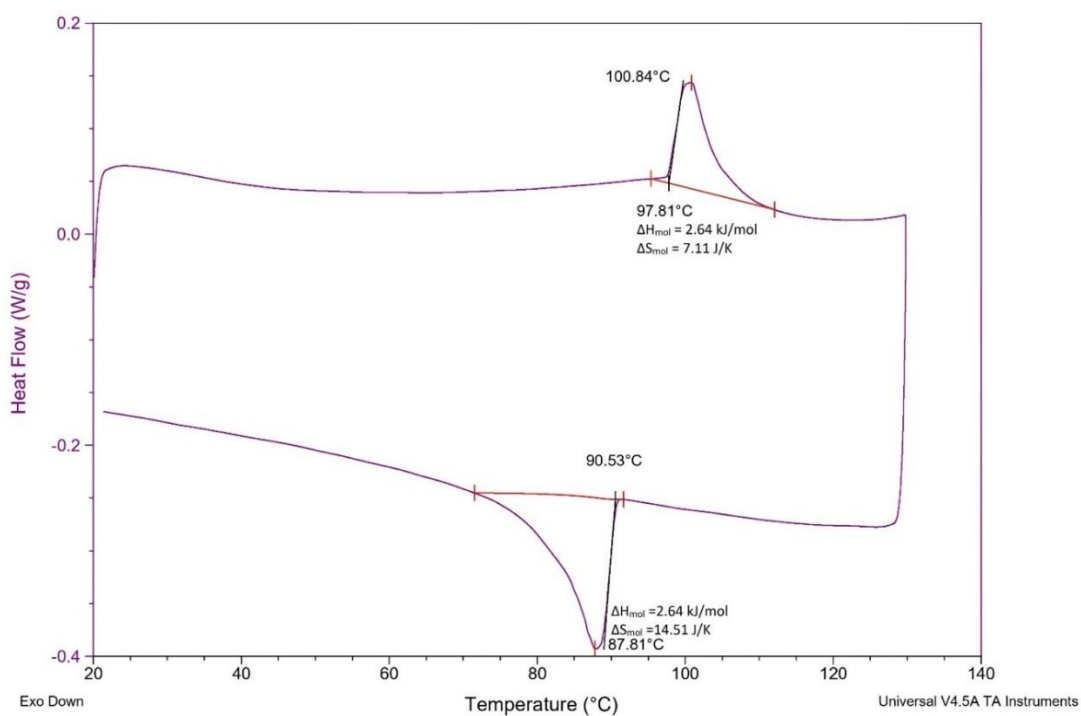


Figure S17. DSC trace recorded on polycrystalline [15-crown-5·Na]HSO₄ (1).

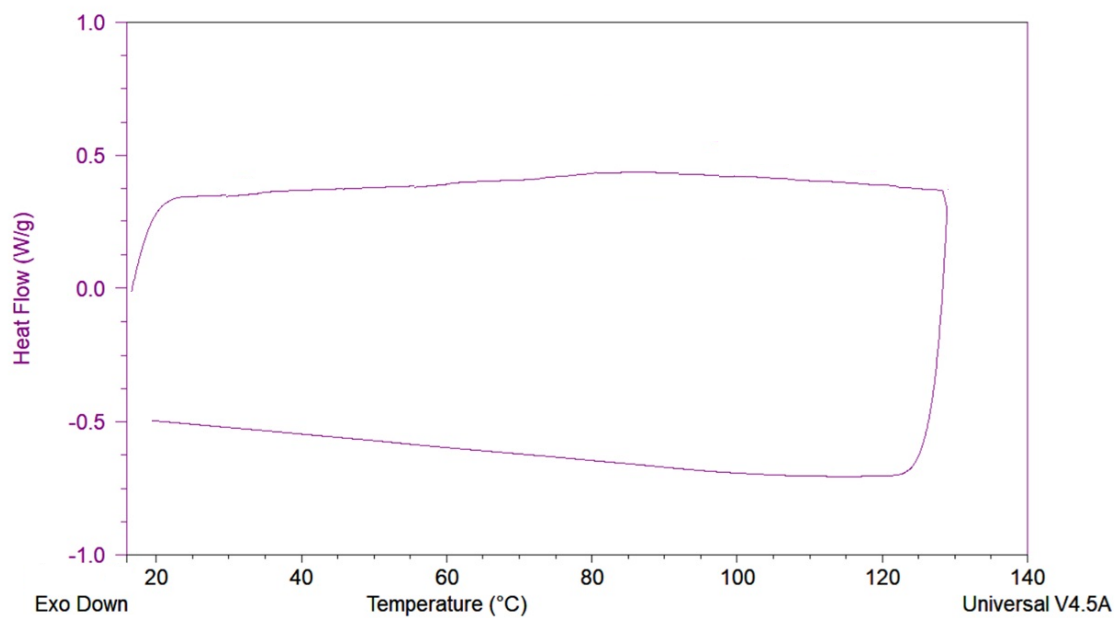


Figure S18. DSC trace measured for polycrystalline [benzo-15-crown-5·Na]HSO₄ (2).

VARIABLE-TEMPERATURE PXRD PATTERNS

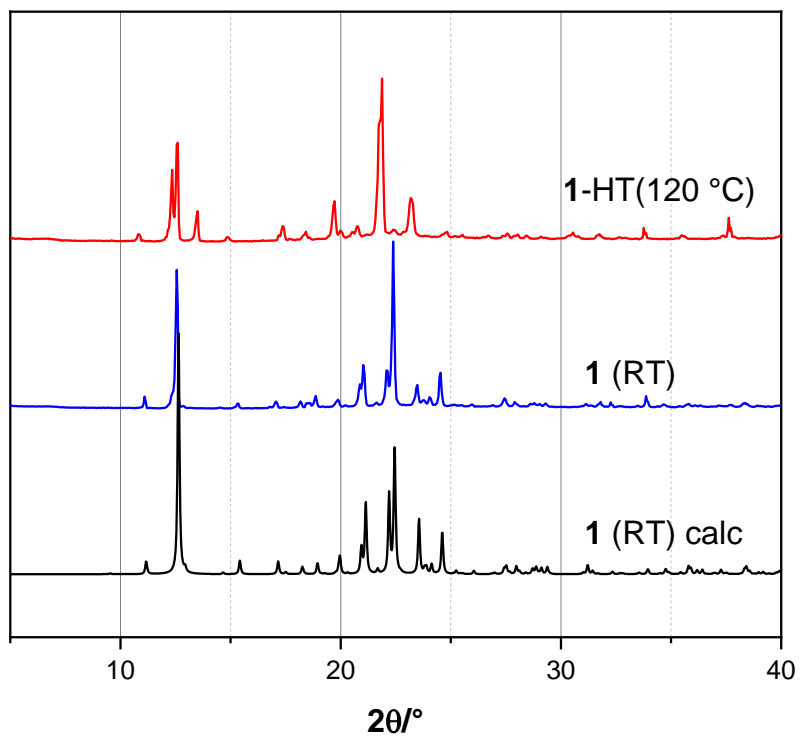


Figure SI9. VT-PXRD patterns recorded at different temperatures for polycrystalline compound **1**.

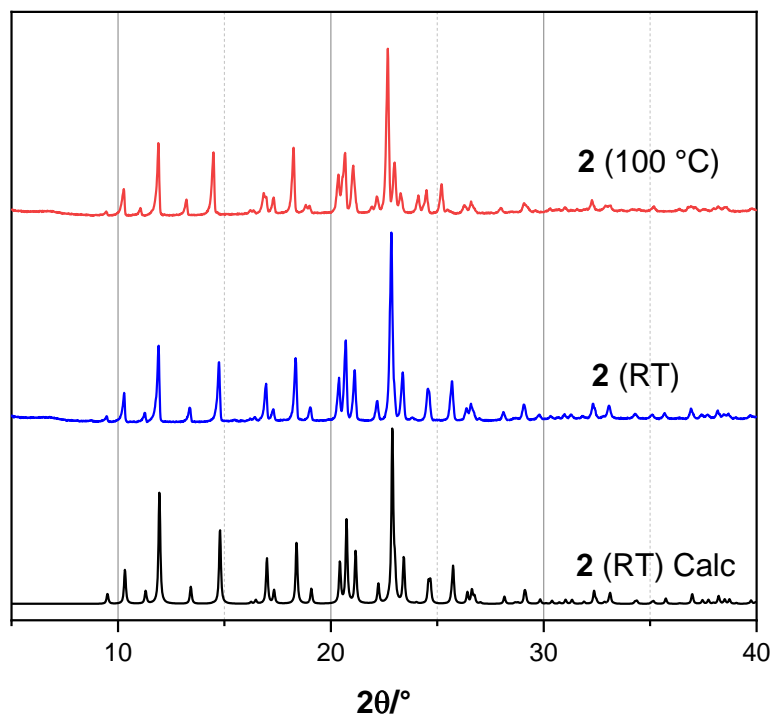


Figure SI10. VT-PXRD patterns recorded at different temperatures for polycrystalline compound **2**.

SOLID-STATE NMR SPECTROSCOPY

Table SI2. List of set temperatures (in °C) employed for the $^1\text{H T}_1$ SSNMR study, with the corresponding sample T values (both in °C and in K), according to the temperature calibration procedure with $\text{Pb}(\text{NO}_3)_2$ (see Experimental Section).

Set T (°C)	Sample T (°C)	Sample T (K)
20.0	31.3	304.4
25.0	35.7	308.9
30.0	40.2	313.4
40.0	49.1	322.3
50.0	58.1	331.3
60.0	67.0	340.2
65.0	71.5	344.7
70.0	76.0	349.2
75.0	80.4	353.6
80.0	84.9	358.1
85.0	89.4	362.6
90.0	93.8	367.0
95.0	98.3	371.5
100.0	102.8	376.0
105.0	107.2	380.4
110.0	111.7	384.9
115.0	116.2	389.4
120.0	120.7	393.9

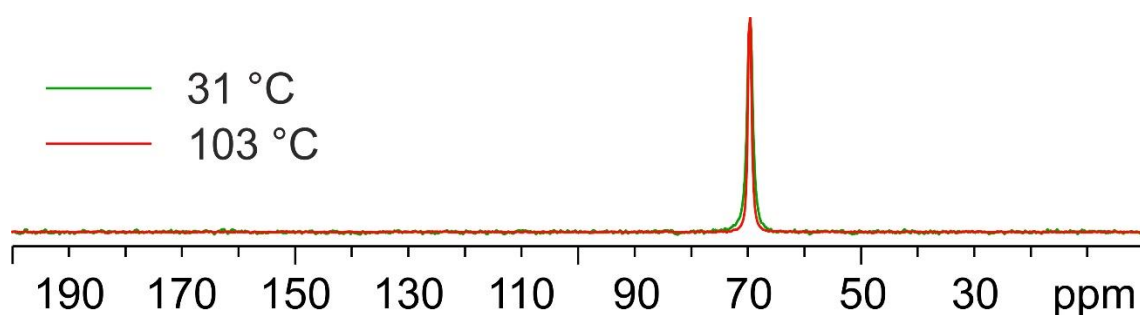


Figure SI11. ^{13}C (150.91 MHz) CPMAS SSNMR spectra of **1**, acquired at the indicated temperatures at a spinning speed of 12 kHz.

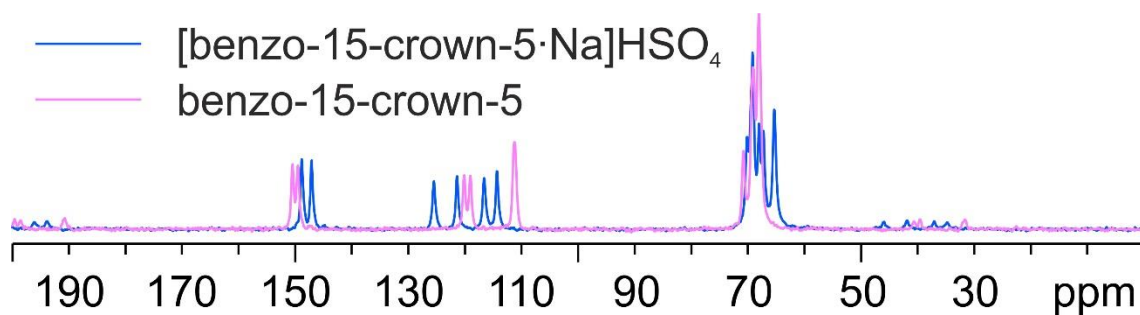


Figure SI12. ^{13}C (150.91 MHz) CPMAS SSNMR spectra of **2** (in blue) and pure benzo-15-crown-5 (in pink), acquired at probe temperature at a spinning speed of 12 kHz.

Table SI3. ^{13}C SSNMR chemical shifts with assignment for [15-crown-5-Na]HSO₄, [benzo-15-crown-5-Na]HSO₄ and benzo-15-crown-5.

[15-crown-5-Na]HSO ₄		[benzo-15-crown-5-Na]HSO ₄		benzo-15-crown-5	
^{13}C chemical shift (ppm)	Assignment	^{13}C chemical shift (ppm)	Assignment	^{13}C chemical shift (ppm)	Assignment
69.6	CH ₂	148.8	Aromatic C _q	150.4	Aromatic C _q
		147.0	Aromatic C _q	149.5	Aromatic C _q
		125.4	Aromatic CH	120.1	Aromatic CH
		121.3	Aromatic CH	119.0	Aromatic CH
		116.5	Aromatic CH	111.2	2 x Aromatic CH
		114.3	Aromatic CH	70.7	CH ₂
		70.0	CH ₂	69.0	3 x CH ₂
		69.4	CH ₂	68.0	4 x CH ₂
		69.1	2 x CH ₂		
		68.0	CH ₂		
		67.2	CH ₂		
		65.3	2 x CH ₂		

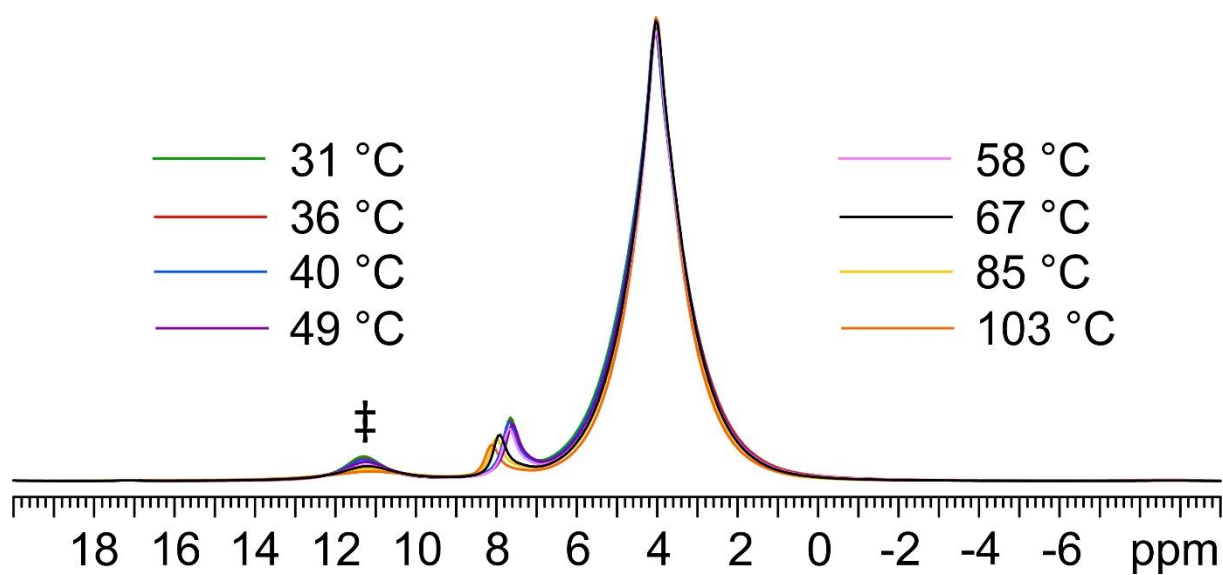


Figure SI13. ^1H (600.13 MHz) MAS SSNMR spectra of **1**, acquired at the indicated temperatures at a spinning speed of 12 kHz. The symbol “‡” indicates an unassignable signal.

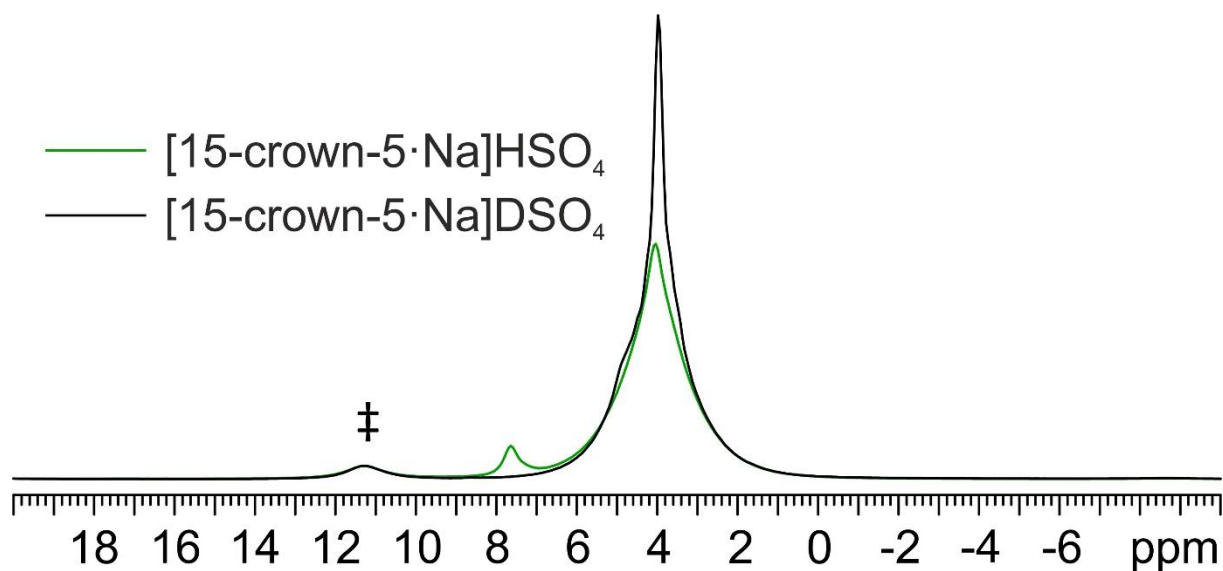


Figure SI14. ^1H (600.13 MHz) MAS SSNMR spectra of **1** (in green) and its NaDSO_4 -based analogous (in black), acquired at probe temperature at a spinning speed of 12 kHz. The symbol “‡” indicates an unassignable signal.

ELECTROCHEMICAL IMPEDANCE SPECTROSCOPY (NYQUIST PLOTS)

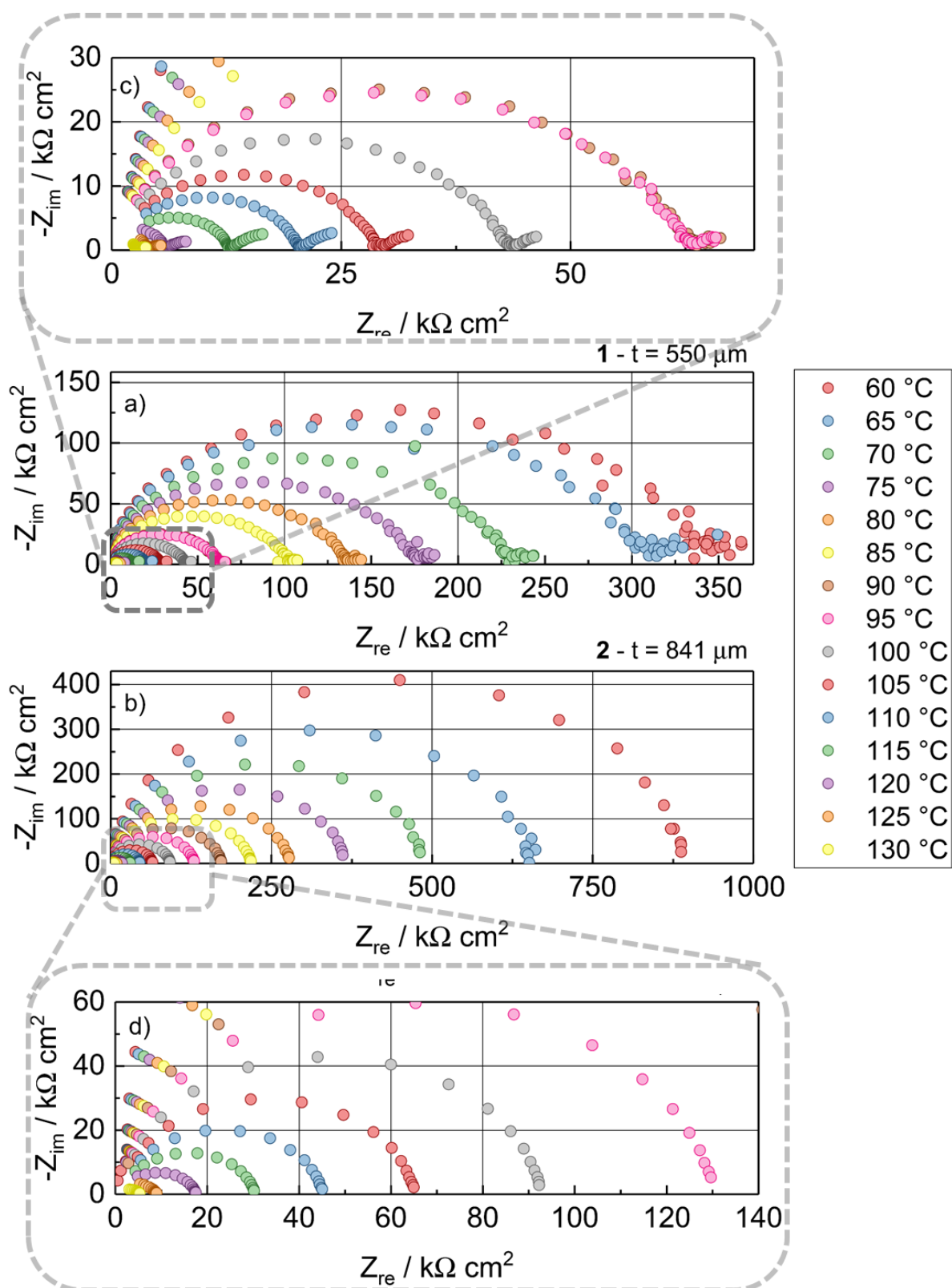


Figure SI15. Nyquist plots at different temperatures of: (a) [15-crown-5-Na]HSO₄ (1) and (b) [benzo-15-crown-5-Na]HSO₄ (2).