Organocatalytic Asymmetric Electrophilic Amination of Allylic Boronates

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GENERAL

All the NMR spectra were recorded on Inova 300 MHz, Gemini 400 MHz or 600 MHz Varian or Bruker spectrometers for ¹H, 101 MHz and 151 MHz for ¹³C, 192 MHz or 193 MHz for ¹¹B. The chemical shifts (δ) for ¹H, ¹³C are given in ppm relative to internal standard TMS (0.0 ppm) or residual signals of CHCl₃ (7.26 ppm). The following abbreviations are used to indicate the multiplicity: s, singlet; d, doublet; t, triplet; q, quartet; m, multiplet; bs, broad signal. Purification of reaction products was carried out by flash chromatography (FC) on silica gel (230-400 mesh). Organic solutions were concentrated under reduced pressure on a Büchi rotary evaporator. High Resolution Mass Spectra (HMRS) were obtained from the Mass Facility unit on a Waters Xevo Q Tof spectrometer. Optical rotations were measured on a Perkin Elmer 241 Polarimeter provided with a sodium lamp and are reported as follows: [α]^{rt}_D (*c* in g/100 mL, solvent). Unless otherwise noted all reactions were set up in the air and using undistilled solvent, without any precautions to exclude moisture.

Materials

Commercial grade reagents and solvents were used without further purification; otherwise, where necessary, they were purified as recommended.¹ (*E*)-2-(but-2-en-1-yl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane **1aa**, azodicarboxylates **2a**, **2b** and **2c**, catalysts **A**, **B**, **D**, **E**, **F**, **G** and *ent*-**G** were purchased from suppliers. Boronic esters **1ba-1ea**, **1cb-1cw** and **N-1c** were prepared according to the literature procedure with some modifications.² Catalysts **C**, **H**, **I**, **L**, **M**, **N** were prepared according to the literature procedures.^{3, 4, 5, 6}

Determination of Enantiomeric/Diasteromeric Purity

The enantiomeric ratio (e.r.) was determined by HPLC analysis on chiral stationary phase on an Agilent 1100- and 1200-series instrumentation using mixtures of i-PrOH/hexane as the eluent or by GC analysis on an Agilent 6890 equipped with a Rt-BetaDex-sm (30 m length x 0.32 mm ID x 0.25 μ m film thickness) with hydrogen as carrier gas (2.0 mL, spilt ratio 50:1) and using the following conditions: 120 °C, 2 min then 2 °C/min up to 230 °C, 5 min; injector : 230 °C, FID detector: 230 °C. HPLC and GC traces of enantioenriched products were compared to racemic ones obtained from the reaction using catalyst *rac*-G. Only for compounds **3u** and **3u'** the diasteromeric ratio was measured by ¹H NMR analysis at 100°C.

Calculations

Each structure has undergone a process of conformational research and subsequent refinement. The conformational search has been performed using CREST (version 2.12)⁷ with the iMTD-GC algorithm using as an implicit solvent the wet octanol with the ALPB model.⁸ To reduce the number of conformers, a statistical clustering of the conformers has been performed using the PCA analysis on the dihedral angles of the molecules. These ensembles have been further refined using ORCA (version 5.0.4),⁹ ordered, and eventually sorted out with two different single-point energy calculations with different theories: B97-3c/def2-SVP,¹⁰ and B3LYP¹¹-D4/def2-SVP. Both analyses have been carried out in gas phase. The best conformer resulting from this protocol has been optimized to the r²scan-3c d4/def2-mTZVPP¹² level; in this case, to better simulate the reaction condition the relative dielectric constant (ϵ_r) of the implicit CPCM solvation model has been set to

12 (which is the molar average of the two solvents $\varepsilon_{r, ACOH} = 6.15$, $\varepsilon_{r, iPrOH} = 19.92$). To choose the functional that best fits the experimental data, a preliminary analysis has been conducted. To reduce the complexity and the various degrees of freedom of the system, 2a was modified to a methyl ester, instead of the tert-butyl group. So, the two Zimmerman-Traxler¹³ like TS's are designed, and the reaction have been simulated following the two diasteromeric faces of the crotyl. Seven different functionals were tested, keeping the basis set constant to def2-SVP: B3LYP-D4, MN15, PBE, r²scan-3c D4, REVPBE98-D4, TPSSH-D4, and ωB97x-D4. All functionals indicated the same stereoselectivity of the reaction and substantially the enantiomeric ratio was consistent, it was chosen to operate with the fastest functional to converge: r²scan-3c D4. Considering the few atoms in the system, the basis set of choice was enlarged to the modified triple- ζ def2-mTZVPP. Furthermore, a single-point energy correction with the **def2-QZVP** basis-set was performed to better simulate the iodine atoms. The default basis set for the effective core potential (ECP) of the iodine atom is used – def2-ECP. Following up, to analyze and to rationalize the energy gaps, the Non-Covalent interaction (NCI) index was calculated with NCIPlot (version 4.0)¹⁴ using FINE integration grid. For visualizing the geometries, frequencies and surfaces ChimeraX software have been used.15

OPTIMIZATION OF THE ASYMMETRIC AMINATION

Table S1: Catalysts screening



Catalyst	^{t-} BuOH (equiv.)	Time (h)	Conversion (%)	e.r.	
/	/	48	17	/	
Α	/	40	17	50:50	
Α	1.0	48	21	50:50	
В	/	48	17	50:50	
В	1.0	48	23	50:50	
С	1.0	48	22	50:50	
D	1.0	24	19	53.5:46.5	
E	1.0	24	27	52:48	
MeO-E	1.0	24	13	53:47	
F	1.0	24	30	50:50	
G ^a	1.0	48	60	80:20	
H۵	1.0	48	60	77:23	
l ^b	1.0	48	40	80:20	
La	1.0	48	53	77:23	
Ma	1.0	48	26	50:50	
N ^a	1.0	48	68	80:20	

 a 0.5 M in toluene instead of 0.1 M. b 0.5 M in PhCF₃ instead of toluene.

Table S2: Screening of temperature and equivalents of reaction partners



Table S3: Screening of alcohols

Me Me Me, YO		G (15 mol%)	Boc N Boc	
Me O-B Me +	Boc N Boc	toluene 0.5 M, 25°C, alcohols , 48h	Me	
1aa: 0.1 mmol	2a: 0.1 mmol		3a	
Additive	Equivalents	Conversion (%)	e.r.	
/	/	60	69.5:31.5	
MeOH	1.0	58	70:30	
EtOH	1.0	55	70:30	
ⁱ PrOH	1.0	55	81:19	
^t BuOH	0.5	55	73:27	
^t BuOH	1.0	60	80:20	
^t BuOH	1.5	55	77:23	
^t BuOH	2.0	30	50:50	
1-adamantanol	1.0	44	77:23	
methanesulfonamide	1.0	54	68:32	
(tetrahydropyranyl)methanol	1.0	53	61.5:39.5	
(S)-2-butanol	1.0	57	81:19	
Ethylene glycole	1.0	34	74:26	
2,3-butandiol	1.0	36	74:26	

Table S4: Screening of additives in the presence of ^tBuOH



^{*a*} No ^{*t*}BuOH added. ^{*b*} AcOH employed as cosolvent (0.1 M in AcOH).

Table S5: Screening of co-solvents



Co-solvent	Conversion (%)	e.r.
DMSO	46	83:17
DMF	46	85:15
THF	47	85:15
Pyridine	48	79:21
Dioxane	51	75:25
ACN	52	86:14
Diethyl ether	61	85:15
Ethyl acetate	65	85:15
DCM	70	85:15
CHCl ₃	68	88:12
МТВЕ	66	85:15
MeOH ^a	71	83:17
EtOH ^a	70	86:14
ⁱ PrOH ^a	75	88:12
^t BuOH	71	83:17

^{*a*} Reaction performed without ^{*t*}BuOH.

Table S6: Optimization of conditions with (*E*)-2-(but-2-en-1-yl)-4,4-dimethyl-1,3,2-dioxaborinane **1ca**



Catalyst	R	X (mol%)	AcOH (mL)	ⁱ PrOH (mL)	[M] ₀	time (h)	Conversion (%)	e.r.
G	Н	15	0.2	0.2	0.25	16	80	96:4
G	Η	10	0.2	0.2	0.25	16	70	95:5
G	Η	10	0.2	0.2	0.25	24	82	95:5
G	Н	10	0.2	0.2	0.25	48	89	94:6
G	Н	5	0.2	0.2	0.25	16	60	94:6
N	I	15	0.2	0.2	0.25	24	98	97:3
N	Ι	15	0.1	0.1	0.5	19	94	96:4
N	Ι	10	0.1	0.1	0.5	24	93	96:4
N	I	5	0.1	0.1	0.5	20	82	96:4

DETERMINATION OF THE ABSOLUTE CONFIGURATION OF 3A

The absolute configuration of **3a** obtained employing (*S*,*S*)-**N** was determined through the optical rotatory power of **11a** and compared with that reported in literature.¹⁶ **11a** is accessible via 2 steps starting from **3a**:



Allylic hydrazide **3a** (143 mg, 0.5 mmol, 1 eq.) was added to a capped vial and dissolved with 1 mL of acetic anhydride and 0.5 mL of pyridine. Afterward DMAP (30 mg, 0.33 mmol, 0.6 eq.) was added and the vial was placed in an oil bath at 50°C. The reaction mixture was vigorously stirred for 24 hours. Then, the vessel was cooled to room temperature and quenched with distilled water. The layers were separated and the aqueous phase was extracted three times with Et₂O. The organic phases were collected, washed two times with sat NaHCO₃, twice with sat NH₄Cl and two times with brine. The organic layer was separated, dried with MgSO₄ and concentrated at reduced pressure. The residue was purified by flash chromatography (from 1:6 to 1:4 Et₂O/Hex) to afford **10a** (140 mg, 0.425 mmol, 85% yield).

Di-tert-butyl (R)-1-acetyl-2-(but-3-en-2-yl)hydrazine-1,2-dicarboxylate (10a)

 $\begin{array}{c} & & & \\ &$

Boc ¹³C NMR (151 MHz, CDCl₃) δ 171.29, 171.14, 171.02, 153.18, 153.01, 152.99, 152.62, 152.48, 138.54, 138.31, 137.78, 137.53, 116.03, 115.97, 115.44, 115.41, 83.90, 83.87, 83.84, 81.70, 80.93, 80.91, 58.55, 58.23, 56.56, 56.38, 28.32, 28.26, 28.17, 28.14, 27.92, 27.88, 25.58, 25.53, 25.49, 25.42, 17.92, 17.60, 17.23, 17.04.

HRMS (ESI⁺): m/z for C₁₆H₂₈N₂O₅Na [M+Na]⁺ calcd. 351.1890, found 351.1896.

Second step:



Compound **10a** (82 mg, 0.25 mmol, 1 eq.) was added to an oven dried two necks round bottom flask under nitrogen flow containing 3 mL of degassed THF and 0.45 mL of degassed HPMA. The flask was placed in an oil bath at 65°C and 15 mL of Sml₂ 0.1 M in THF (1.5 mmol, 15 eq.uiv.) were added dropwise while the solution was under stirring. After two hours the flask was removed from the bath and the reaction was quenched with sat NaHCO₃ and extracted three times with Et₂O. Eventually the organic phases were collected, dried with MgSO₄ and concentrated at reduced pressure. The residue was purified by flash chromatography (from 1:5 to 1:3 Et₂O/Hex) to afford **11a** (20 mg, 0.12 mmol, 40% yield).

tert-butyl (R)-but-3-en-2-ylcarbamate (11a)



¹**H NMR** (400 MHz, CDCl₃) δ 5.82 (ddd, J = 17.3, 10.4, 5.1 Hz, 1H), 5.19 – 5.02 (m, 2H), 4.57-4.10 (m, 2H), 1.46 (s, 9H), 1.22 (d, J = 6.8 Hz, 3H).

 $[\alpha]_{589}^{20}$ = +4.6 (*c* = 1.0, in CHCl₃)



PREPARATIONS OF STARTING ALLYLIC ALCOHOLS

Allylic boronic esters were prepared starting from either commercially available (Figure S1) or synthetic allylic alcohols, accessible through three diverse strategies.



Figure S1: Alcohols commercially available

- Strategy A: Addition of vinyImagnesium bromide to aldehydes.
- Strategy B: Reduction of unsatured aldehydes.
- **Strategy C**: Wittig reaction followed by reduction.

Strategy A



7 mmol of the aldehyde were added to a flame dried two necks round bottom flask under nitrogen flow and dissolved with 7 mL of either THF or Et₂O. The flask was cooled to 0°C and the 7 mL of vinyl magnesium bromide (1.0 M in THF) was added dropwise. The mixture was allowed to warm to room temperature and the reaction continued for 2-3 hours. After the consumption of the starting aldehyde checked by TLC, a saturated solution of NH₄Cl was added and was extracted three times with Et₂O. The organic phases were collected and washed two times with brine, then dried under MgSO₄ and concentrated at reduced pressure. The crudes were considered to be used for the next step without further purifications. The spectra of the alcohols reported in figure S2 matched with those reported in literature 17,18,19,20,21.



Figure S2: Alcohols from vinylation of aldehydes (Strategy A)

All the aldehyde used for this strategy are commercially available; only the aldehyde used for **e** was synthesized starting from the corresponding alcohol:



Following a reported procedure²², 20 mmol of a pent-4-en-1-ol were added to a solution of 20 mL of DCM, TEMPO (1.4 mmol) and of PhI(OAc)₂ (21 mmol) and the mixture was stirred for one hour at room temperature. The solvent was removed and the product was distilled under reduced pressure and collected in a separate flask. Although this fraction was contaminated with 50% of acetic acid, the product was not purified further. The ¹H NMR agrees with those reported in literature. Before the addition of vinyImagnesium bromide the flask was cooled to -78°C and 1 eq. of NaH was added to remove the acetic acid. The mixture was stirred for 30 min in an ice-water bath, then the vinyI magnesium bromide was introduced dropwise to the reaction mixture. The work up is the same of Strategy A.

Strategy B



5 mmol of the allylic aldehyde (1 eq.) was introduced into a round bottom flask and dissolved with either 5 mL of ethanol or methanol. 5 mmol of NaBH₄ were added in small portions and then the reaction was stirred for 2-3 hours. After the consumption of the starting material monitored by TLC, the reaction was quenched with a saturated solution of NH₄Cl and extracted three times with Et₂O. The organic phases were collected and washed two times with brine, then dried under MgSO₄ and concentrated at reduced pressure. The crudes were sufficiently clean to be used without further

purifications. Alcohols t and s (Figure S4) have been prepared with this procedure and the NMR data were consistent with those reported in literature.^{23,24}



Figure S3: Alcohols from vinylation of aldehydes (Strategy A)

Strategy C



First step:



To an ice-cooled flask containing a solution of 3.0 g of 4-phenylbutan-2-one (20 mmol) and 4.0 g of methyl 2-(dimethoxyphosphoryl)acetate (1.1 eq., 22 mmol) in 30 mL of dry THF, 0.96 g of NaH (60% dispersion in mineral oil, 1.2 eq., 24 mmol) were added stepwise. Then the flask was removed from the bath and the reaction was allowed to stir overnight. The mixture was quenched with a saturated solution of NH₄Cl, the layers were separated and the aqueous phase was washed three times with Et₂O. The organic phases were collected, washed two times with brine, dried under MgSO₄ and concentrated at reduced pressure. Flash chromatography of the crude (1:5 Et₂O /hexane) afforded 3.17 grams of methyl 3-methyl-5-phenylpent-2-enoate in 78% yield. NMR data were consistent with those reported in literature.25

Second step:



Following a reported procedure,²⁶ methyl 3-methyl-5-phenylpent-2-enoate (10.4 mmol, 2.27 g) was dissolved in 30 mL of dry DCM and transferred into a flame dried two necks round bottom flask under nitrogen flow. A solution of [1.0 M] DIBAL-H 1.0 M in hexane (2.7 eq., 28 mmol, 28 mL) was added dropwise at -78 °C. The reaction was allowed to stir at the same temperature for two hours, when the total conversion of the starting material was observed by TLC. The reaction was quenched with 50 mL of saturated solution of NH₄Cl, and the mixture was allowed to reach room temperature. Afterward, the layers were separated and the aqueous phase was extracted three times with DCM. The organic phases were collected, washed two times with brine, dried under MgSO₄ and concentrated at reduced pressure. The residue was purified with flash chromatography (1:2 diethyl ether/hexane) to give compound 3-methyl-5-phenylpent-2-en-1-ol (**v**) in 67% yield (1.2 g, 7 mmol). NMR data were consistent with those reported in literature.²⁶

GENERAL PROCEDURE FOR THE SYNTHESIS OF ALLYLIC BORONIC ESTERS



Following a reported procedure² with some modification, to an oven-dried two necks round bottom flask provided of a magnetic stir bar and under nitrogen flow, 1 equivalent of allylic alcohol, dry DMSO and distilled MeOH (1:1 mixture, 0.25 M), 0.036 eq. of di- μ -chlorobis[2-[(dimethylamino)methyl]phenyl-C,N]dipalladium(II), 0.054 equivalents of anhydrous TsOH and 3 equivalents of B₂(OH)₄ were added. The reaction was stirred at 50°C overnight, then the flask was cooled to room temperature and 2.6 equivalents of 3-methylbutane-1,3-diol were added. After the consumption of the boronic acid monitored with TLC, the entire solution was transferred into a packed column and a flash chromatography was performed to isolate the product. High vacuum is not recommended because some of the boronic esters proved to be volatile. Unless otherwise reported, all allylic boronic esters have been obtained with (E)-configuration at the double bond.

(E)-2-(but-2-en-1-yl)-4,4-dimethyl-1,3,2-dioxaborinane (1ca)

The titled compound was obtained from alcohol **a** in 70% yield (353 mg) after column chromatography (1:4.5 Et_2O/Hex).

¹H NMR (600 MHz, CDCl₃) δ 5.48 (m, 1H), 5.39 – 5.29 (m, 1H), 4.04 – 3.98 (m, 2H), 1.83 – 1.76 (m, 2H), 1.64 (dq, 3H), 1.53 (d, 2H), 1.30 (s, 6H). ¹³C NMR (151 MHz, CDCl₃) δ 127.56, 124.13, 70.03, 59.06, 38.01, 29.24, 18.08. ¹¹B NMR (192 MHz, CDCl₃) δ 29.48. HRMS (ESI⁺): m/z for $C_9H_{18}BO_2$ [M+H]⁺ calcd. 169.1400, found 169.1403.

(E)-2-(hex-2-en-1-yl)-4,4-dimethyl-1,3,2-dioxaborinane (1cb)



The titled compound was obtained from alcohol **b** in 41% yield (117 mg) after column chromatography (1:4.5 Et_2O/Hex).

¹H NMR (400 MHz, CDCl₃) δ 5.50 – 5.38 (m, 1H), 5.38 – 5.25 (m, 1H), 4.03 – 3.95 (m, 2H), 2.00 – 1.89 (m, 2H), 1.80 – 1.73 (m, 2H), 1.51 (d, 1H), 1.47 – 1.29 (m, 2H), 1.28 (s, 6H), 0.87 (t, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 129.64, 126.53, 69.98, 59.03, 38.03, 34.88, 29.23, 22.83, 13.62. ¹¹B NMR (192 MHz, CDCl₃) δ 29.56. HRMS (ESI⁺): m/z for C₁₁H₂₂BO₂ [M+H]⁺ calcd. 197.1713, found 197.1716.

(E)-4,4-dimethyl-2-(5-phenylpent-2-en-1-yl)-1,3,2-dioxaborinane (1cc)



The titled compound was obtained from alcohol **c** in 65% yield (251 mg) after after flash chromatography (1:5 Et₂O/Hex).

¹H NMR (400 MHz, CDCl₃) δ 7.30 – 7.23 (m, 2H), 7.22 – 7.13 (m, 3H), 5.51 (m, 1H), 5.43 – 5.32 (m, 1H), 4.02 – 3.96 (m, 2H), 2.70 – 2.62 (m, 2H), 2.36 – 2.23 (m, 2H), 1.81– 1.74 (m, 2H), 1.56 – 1.49 (d, 2H), 1.28 (s, 6H). ¹³C NMR (101 MHz, CDCl₃) δ 142.45, 128.89, 128.46, 128.15, 127.16, 125.55, 70.05, 59.06, 38.03, 36.26, 34.61, 29.25. ¹¹B NMR (192 MHz, CDCl₃) δ 28.19. HRMS (ESI⁺): m/z for C₁₆H₂₄BO₂ [M+H]⁺ calcd 259.1869, found 259.1872.

(E)-4,4-dimethyl-2-(5-methylhex-2-en-1-yl)-1,3,2-dioxaborinane (1cd)



The titled compound was obtained from alcohol d in 47% yield (148 mg) after after flash chromatography (1:9 Et₂O/Hex).

¹H NMR (600 MHz, CDCl₃) δ 5.48 – 5.39 (m, 1H), 5.32 (dtt, J = 15.4, 7.1, 1.4 Hz, 1H), 4.03 – 3.98 (m, 2H), 1.86 (tq, J = 6.9, 1.1 Hz, 2H), 1.82 – 1.76

(m, 2H), 1.62 - 1.51 (d, 2H), 1.29 (s, 6H), 0.97 - 0.86 (m, 7H). ¹³C NMR (101 MHz, CDCl₃) δ 128.46, 127.53, 69.99, 59.04, 42.19, 38.04, 29.25, 28.55, 22.24. ¹¹B NMR (192 MHz, CDCl₃) δ 28.25. HRMS (ESI⁺): m/z for C₁₂H₂₄BO₂ [M+H]⁺ calcd. 211.1869, found 211.1872.

(E)-2-(hepta-2,6-dien-1-yl)-4,4-dimethyl-1,3,2-dioxaborinane (1ce)



The titled compound was obtained from alcohol **e** in 30% yield (94 mg) after flash chromatography (1:8 Et₂O/Hex).

¹H NMR (400 MHz, CDCl₃) δ 5.89 – 5.77 (m, 1H), 5.49 (dtt, *J* = 16.3, 7.2, 1.2 Hz, 1H), 5.40 – 5.29 (m, 1H), 5.05 – 4.89 (m, 2H), 4.03 – 3.97 (m, 2H), 2.09 (dtdd, *J* = 5.7, 4.7, 2.2, 1.2 Hz, 4H), 1.81 – 1.75 (m, 2H), 1.56 – 1.49 (m, 2H), 1.29 (s, 6H). ¹³C NMR (151 MHz, CDCl₃) δ 138.81, 128.91, 127.01, 114.23, 70.05, 59.07, 38.05, 34.01, 32.23, 29.27. ¹¹B NMR (192 MHz, CDCl₃) δ 29.53. HRMS (ESI⁺): m/z for C₁₂H₂₂BO₂ [M+H]⁺ calcd. 209.1713, found 209.1715.

(E)-2-(4-ethylhex-2-en-1-yl)-4,4-dimethyl-1,3,2-dioxaborinane (1cf)



The titled compound was obtained from alcohol **f** in 74% yield (248.8 mg) after after flash chromatography (1:6 Et₂O/Hex).

¹H NMR (600 MHz, CDCl₃) δ 5.38 (dt, J = 15.1, 7.5 Hz, 1H), 5.01 (dd, J = 15.3, 8.7 Hz, 1H), 3.99 (t, J = 5.7 Hz, 2H), 1.77 (m, 2H), 1.68 (dq, J = 9.4, 4.3 Hz, 1H), 1.53 (d, J = 7.5 Hz, 2H), 1.41 – 1.29 (m, 2H), 1.28 (s, 6H), 1.23 – 1.13 (m, 2H), 0.83 (t, J = 7.5 Hz, 6H). ¹³C NMR (151 MHz, CDCl₃) δ 133.75, 126.45, 69.95, 59.02, 46.44, 38.07, 29.25, 27.90, 11.75.¹¹B NMR (193 MHz, CDCl₃) δ 29.43. HRMS (ESI⁺): m/z for C₁₃H₂₆BO₂ [M+H]⁺ calcd. 225.2026, found 225.2028.

(E)-2-(3-cyclohexylallyl)-4,4-dimethyl-1,3,2-dioxaborinane (1cg)



The titled compound was obtained from alcohol **g** in 56% (198 mg) after flash chromatography (1:9 Et_2O/Hex).

¹H NMR (400 MHz, CDCl₃) δ 5.41 (dtd, J = 15.4, 7.2, 1.1 Hz, 1H), 5.27 (ddt, J = 15.4, 6.7, 1.4 Hz, 1H), 4.02 – 3.95 (m, 2H), 1.92-1.85 (m,1H), 1.77 (dd, J = 6.2, 5.3 Hz, 2H), 1.73 – 1.64 (m, 4H), 1.64 – 1.57 (m, 1H), 1.50 (d, J = 7.1 Hz, 2H), 1.28 (s, 6H), 1.27 – 0.95 (m, 6H). ¹³C NMR (101 MHz, CDCl₃) δ 135.99, 123.79, 69.97, 59.03, 40.83, 38.03, 33.38, 29.25, 26.28, 26.18. ¹¹B NMR (192 MHz, CDCl₃) δ 28.32. HRMS (ESI⁺): m/z for C₁₄H₂₆BO₂ [M+H]⁺ calcd. 237.2026, found 237.2029.

(S,E)-2-(5,9-dimethyldeca-2,8-dien-1-yl)-4,4-dimethyl-1,3,2-dioxaborinane (1ch)



The titled compound was obtained from alcohol **h** in 74% yield (308.8 mg) after flash chromatography (1:6 Et_2O/Hex).

¹**H NMR** (600 MHz, CDCl₃) δ 5.44 (dtt, J = 14.8, 7.3, 1.3 Hz, 1H), 5.30 (dtt, J = 15.5, 7.2, 1.5 Hz, 1H), 5.10 (tp, J = 7.1, 1.4 Hz, 1H), 4.02 – 3.98 (m, 2H), 2.03 – 1.92 (m, 3H), 1.85 – 1.79 (m, 1H), 1.79 – 1.76 (m, 2H),

1.68 (s, 3H), 1.60 (s, 3H), 1.54 (d, J = 7.4 Hz, 2H), 1.47 – 1.41 (m, 1H), 1.29 (s, 6H), 1.11 (dddd, J = 13.5, 9.7, 7.7, 5.9 Hz, 1H), 0.91 – 0.84 (m, 5H). ¹³C NMR (151 MHz, CDCl₃) δ 130.83, 128.13, 127.68, 125.10, 69.98, 59.02, 40.17, 38.02, 36.58, 32.88, 29.22, 25.69, 25.60, 19.34, 17.59. ¹¹B NMR (192 MHz, CDCl₃) δ 28.17. HRMS (ESI⁺): m/z for C₁₇H₃₂BO₂ [M+H]⁺ calcd. 279.2495, found 279.2498.

2-cinnamyl-4,4-dimethyl-1,3,2-dioxaborinane (1ci)



Me, Me The titled compound was obtained from alcohol I in 50% yield (173 mg) after flash chromatography (1:9 AcOEt/Hex).

¹H NMR (400 MHz, CDCl₃) δ 7.37 – 7.30 (m, 2H), 7.34 – 7.22 (m, 2H), 7.19 – 7.10 (m, 1H), 6.31 (s, 1H), 6.37 – 6.24 (m, 1H), 4.05 – 3.97 (m, 2H), 1.83 – 1.70 (m, 4H), 1.30 (s, 5H). ¹³C NMR (101 MHz, CDCl₃) δ 138.51, 129.18, 128.34, 128.25, 126.23, 125.71, 70.28, 59.14, 38.04, 29.27. ¹¹B NMR (192 MHz, CDCl₃) δ 29.34. HRMS (ESI⁺): m/z for C₁₄H₂₀BO₂ [M+H]⁺ calcd. 231.1556, found 231.1559.

(E)-4,4-dimethyl-2-(3-(o-tolyl)allyl)-1,3,2-dioxaborinane (1cj)



The titled compound was obtained from alcohol j in 75% yield (174 mg) after flash chromatography (1:9 Et₂O/Hex).

¹**H NMR** (400 MHz, CDCl₃) δ 7.43 (d, J = 7.3 Hz, 1H), 7.19 – 7.04 (m, 3H), 6.51 (dt, J = 15.7, 1.7 Hz, 1H), 6.20 (dt, J = 15.5, 7.6 Hz, 1H), 4.06 – 3.99 (m, 2H),

2.33 (s, 3H), 1.85 – 1.76 (m, 4H), 1.32 (s, 6H). ¹³**C NMR** (151 MHz, CDCl₃) δ 137.56, 134.60, 130.05, 129.42, 126.95, 126.25, 125.88, 125.28, 70.26, 59.15, 38.07, 29.29, 19.90. ¹¹B NMR (193 MHz, CDCl₃) δ 29.40. **HRMS (ESI⁺)**: m/z for C₁₅H₂₁BO₂K [M+K]⁺ calcd. 283.1266, found 283.1272.

(E)-4,4-dimethyl-2-(3-(m-tolyl)allyl)-1,3,2-dioxaborinane (1ck)



The titled compound was obtained from alcohol **k** in 60% yield (217.7 mg) after flash chromatography (1:6 Et_2O/Hex).

¹H NMR (600 MHz, CDCl₃) δ 7.18 – 7.10 (m, 3H), 6.96 (d, J = 7.5 Hz, 1H), 6.31 – 6.24 (m, 2H), 4.03 – 3.98 (m, 2H), 2.32 (s, 3H), 1.81 – 1.73 (m, 4H), 1.30 (s, 6H). ¹³C NMR (151 MHz, CDCl₃) δ 138.46, 137.80, 129.23, 128.25, 128.02, 127.04, 126.43, 122.89, 70.27, 59.14, 38.05, 29.28, 21.42. ¹¹B NMR (192 MHz, CDCl₃) δ 29.27. HRMS (ESI⁺): m/z for C₁₅H₂₂BO₂ [M+H]⁺ calcd. 245.1713, found 245.1716.

(E)-2-(3-(4-isopropylphenyl)allyl)-4,4-dimethyl-1,3,2-dioxaborinane (1cl)



The titled compound was obtained from alcohol I in 71% yield (290 mg) after flash chromatography (1:9 Et_2O/Hex).

¹H NMR (600 MHz, CDCl₃) δ 7.25 (d, J = 8.1 Hz, 2H), 7.12 (d, J = 7.9 Hz, 2H), 6.31 – 6.20 (m, 2H), 3.99 (t, J = 5.7 Hz, 2H), 2.86 (p, J = 6.9 Hz, 1H), 1.80 – 1.70 (m, 4H), 1.29 (d, J = 3.5 Hz, 6H), 1.23 (d, J = 6.7 Hz, 6H). ¹³C NMR (151 MHz, CDCl₃) δ 146.94, 136.19, 128.96, 127.29, 126.39, 125.65, 70.24, 59.12, 38.04, 33.77, 29.27, 24.00. ¹¹B NMR (193 MHz, CDCl₃) δ 29.42. HRMS (ESI⁺): m/z for C₁₇H₂₅BO₂K [M+K]⁺ calcd. 311.1579, found 311.1585.

(E)-2-(3-(3-methoxyphenyl)allyl)-4,4-dimethyl-1,3,2-dioxaborinane (1cm)



Me, Me The titled compound was obtained from alcohol **m** in 30% yield (117 mg) after flash chromatography (1:6 Et_2O/Hex).

¹H NMR (400 MHz, CDCl₃) δ 7.19 (t, J = 7.9 Hz, 1H), 6.94 (dt, J = 7.7, 1.3 Hz, 1H), 6.89 (dd, J = 2.6, 1.6 Hz, 1H), 6.72 (ddd, J = 8.1, 2.6, 0.9 Hz, 1H), 6.39 – 6.24 (m, 2H), 4.06 – 3.98 (m, 2H), 3.81 (s, 3H), 1.79 (dt, J = 11.6, 5.8 Hz, 4H), 1.31 (s, 6H). ¹³C NMR (101 MHz, CDCl₃) δ 159.68, 139.98, 129.25, 129.03, 128.64, 118.45, 111.89, 111.01, 70.28, 59.12, 55.14, 38.01, 29.24. ¹¹B NMR (193 MHz, CDCl₃) δ 29.42. HRMS (ESI⁺): m/z for C₁₅H₂₁BO₃K [M+K]⁺ calcd. 299.1215, found 299.1221.

(E)-2-(3-(4-methoxyphenyl)allyl)-4,4-dimethyl-1,3,2-dioxaborinane (1cn)



 $\overset{\text{Me}}{\qquad} \begin{array}{l} \text{The titled compound was obtained from alcohol } \textbf{n} \text{ in 50\% yield (195} \\ \text{mg) after flash chromatography (1:6 Et_2O/Hex).} \end{array}$

¹H NMR (400 MHz, CDCl₃) δ 7.30 – 7.22 (m, 2H), 6.87 – 6.77 (m, 2H), 6.30 – 6.21 (m, 1H), 6.16 (m, 1H), 4.05 – 3.93 (m, 2H), 3.79 (s, 3H), 1.82-1.77 (m, 2H), 1.75– 1.70 (d, 2H), 1.30 (s, 6H). ¹³C NMR (101 MHz, CDCl₃) δ 158.23, 131.43, 128.52, 126.73, 125.96, 113.78, 70.23, 59.12, 55.26, 38.03, 29.26. ¹¹B NMR (192 MHz, CDCl₃) δ 28.12. HRMS (ESI⁺): m/z for C₁₅H₂₂BO₃ [M+H]⁺ calcd. 261.1662, found 261.1665.

(E)-4,4-dimethyl-2-(3-(naphthalen-1-yl)allyl)-1,3,2-dioxaborinane (1co)



The titled compound was obtained from alcohol **o** in 64% yield (269 mg) after flash chromatography (1:5 Et_2O/Hex).

¹**H NMR** (400 MHz, CDCl₃) δ 8.27 – 8.21 (m, 1H), 7.91 – 7.84 (m, 1H), 7.76 (dt, *J* = 8.3, 1.1 Hz, 1H), 7.62 (dt, *J* = 7.2, 1.1 Hz, 1H), 7.57 – 7.43 (m, 3H),

7.11 (dt, J = 15.5, 1.8 Hz, 1H), 6.40 (dt, J = 15.5, 7.7 Hz, 1H), 4.10 – 4.04 (m, 2H), 1.96 (dd, J = 7.7, 1.6 Hz, 2H), 1.87 – 1.82 (m, 2H), 1.37 (s, 6H). ¹³**C NMR** (101 MHz, CDCl₃) δ 136.32, 133.68, 131.53, 131.14, 128.37, 126.74, 126.23, 125.70, 125.57, 125.48, 124.22, 123.30, 70.34, 59.18, 38.06, 29.31. ¹¹**B NMR** (192 MHz, CDCl₃) δ 29.42. **HRMS (ESI+)**: m/z for C₁₈H₂₂BO₂ [M+H]⁺ calcd. 281.1713, found 281.1716.

(E)-2-(3-([1,1'-biphenyl]-4-yl)allyl)-4,4-dimethyl-1,3,2-dioxaborinane (1cp)



The titled compound was obtained from alcohol \mathbf{p} in 23% yield (106 mg) after flash chromatography (1:8 Et₂O/Hex).

¹H NMR (400 MHz, CDCl₃) δ 7.60 – 7.54 (m, 2H), 7.57 – 7.44 (m, 2H), 7.44 – 7.34 (m, 4H), 7.34 – 7.25 (m, 1H), 6.38 – 6.28 (m, 2H), 4.04 – 3.97 (m, 2H), 1.82 – 1.74 (m, 4H), 1.29 (s, 6H). ¹³C NMR (151 MHz, CDCl₃) δ 141.04, 138.98, 137.63, 128.73, 128.71, 128.56, 127.08, 127.00, 126.86, 126.11, 70.33, 59.18, 38.07, 29.29. ¹¹B NMR (193 MHz, CDCl₃) δ 29.52. HRMS (ESI⁺): m/z for C₂₀H₂₃BO₂K [M+K]⁺ calcd. 345.1423, found 345.1428.

(E)-2-(3-(4-chlorophenyl)allyl)-4,4-dimethyl-1,3,2-dioxaborinane (1cq)



Me The titled compound was obtained from alcohol **q** in 55% yield (218.3 mg) after flash chromatography (1:8 Et_2O/Hex).

¹H NMR (600 MHz, CDCl₃) δ 7.28 – 7.19 (m, 4H), 6.32 – 6.22 (m, 2H), 4.01 (m, 2H), 1.81 – 1.77 (m, 2H), 1.75 (d, 2H), 1.30 (s, 6H). ¹³C NMR (151 MHz, CDCl₃) δ 137.01, 131.69, 129.10, 128.44, 128.02, 126.91, 70.35, 59.18, 38.04, 29.27. ¹¹B NMR (192 MHz, CDCl₃) δ 29.32. HRMS (ESI⁺): m/z for C₁₄H₁₉BClO₂ [M+H]⁺ calcd. 265.1167, found 265.1169.

(E)-4,4-dimethyl-2-(3-(4-(trifluoromethyl)phenyl)allyl)-1,3,2-dioxaborinane (1cr)



The titled compound was obtained from alcohol **r** in 43% yield (192 mg) after flash chromatography (1:6 Et_2O/Hex).

¹H NMR (600 MHz, CDCl₃) δ 7.51 (d, *J* = 8.1 Hz, 2H), 7.41 (d, *J* = 7.9 Hz, 2H), 6.44 (dt, *J* = 15.5, 7.7 Hz, 1H), 6.37 – 6.30 (m, 1H), 4.02 (dd, *J* = 7.3, 4.3 Hz, 2H), 1.83 – 1.77 (m, 4H), 1.32 (s, 6H). ¹³C NMR (151 MHz, CDCl₃) δ 141.98, 131.41, 128.15, 128.03, 127.94, 125.77, 125.35, 125.32, 125.29, 125.27, 70.43, 59.21, 38.04, 29.26. ¹⁹F NMR (376 MHz, CDCl₃) δ -62.34. ¹¹B NMR (193 MHz, CDCl₃) δ 29.31. HRMS (ESI⁺): m/z for C₁₅H₁₉BF₃O₂ [M+H]⁺ calcd. 299.1430, found 299.1433.

(E)-4,4-dimethyl-2-(2-methylbut-2-en-1-yl)-1,3,2-dioxaborinane (1cs)



The titled compound was obtained from alcohol s in 50% yield (136 mg) after flash chromatography (1:5 Et₂O/Hex).

¹H NMR (400 MHz, CDCl₃) 5.15 - 5.11 (m, 1H), 4.03 - 4.00 (m, 2H), 1.79 - 1.75 (m, 2H), 1.63 (s, 3H), 1.59 - 1.53 (m, 5H), 1.32 (s, 6H). ¹³C NMR (151 MHz, CDCl₃) δ 134.04, 117.27, 70.03, 59.05, 38.06, 29.28, 17.55, 13.63. ¹¹B NMR (193 MHz, CDCl₃) δ 29.70. HRMS (ESI⁺): m/z for C₁₆H₂₄BO₂ [M+H]⁺ calcd. 259.1869, found 259.1872.

(E)-4,4-dimethyl-2-(2-methyl-3-phenylallyl)-1,3,2-dioxaborinane (1ct)



The titled compound was obtained from alcohol **t** in 56% (205 mg) after flash chromatography (1:8 Et_2O/Hex). Mixture of 1:9 Z/E isomers.

¹H NMR (*E* isomer) (600 MHz, CDCl₃) δ 7.32 – 7.28 (m, 2H), 7.26 – 7.23 (m, 2H), 7.18 – 7.13 (m, 1H), 6.23 (1H), 4.05 – 4.01 (m, 2H), 1.93 – 1.88 (m, 3H), 1.83 – 1.79 (m, 2H), 1.76 (s, 2H), 1.32 (s, 6H). ¹³C NMR (151 MHz, CDCl₃) δ 139.28, 139.13, 138.04, 137.97, 128.81, 128.61, 127.90, 127.89, 125.44, 125.33, 123.91, 123.46, 70.28, 70.21, 59.18, 59.11, 38.07, 29.29, 29.26, 19.74. ¹¹B NMR (193 MHz, CDCl₃) δ 29.27. HRMS (ESI⁺): m/z for C₁₅H₂₂BO₂ [M+H]⁺ calcd. 245.1713, found 245.1716.

(S)-4,4-dimethyl-2-((4-(prop-1-en-2-yl)cyclohex-1-en-1-yl)methyl)-1,3,2-dioxaborinane (1cu)



The titled compound was obtained from alcohol **u** in 87% yield (648 mg) after flash chromatography (1:8 Et_2O/Hex).

¹H NMR (600 MHz, CDCl₃) δ 5.35 (ddd, J = 5.4, 2.6, 1.4 Hz, 1H), 4.70 (dh, J = 3.4, 1.4 Hz, 2H), 4.03 – 3.99 (m, 2H), 2.16 – 2.04 (m, 3H), 2.04 – 1.88 (m, 2H), 1.80 – 1.76 (m, 3H), 1.74 (t, J = 1.1 Hz, 3H), 1.55 – 1.45 (m, 3H), 1.30 (s, 6H). ¹³C NMR (151 MHz, CDCl₃) δ 150.63, 135.73, 119.39, 108.18, 70.06, 59.05, 41.15, 38.06, 31.06, 30.89, 29.30, 28.16, 20.83. ¹¹B NMR (193 MHz, CDCl₃) δ 29.39. HRMS (ESI⁺): m/z for

C₁₅H₂₆BO₂ [M+H]⁺ calcd. 249.2026, found 249.2029.

(E)-4,4-dimethyl-2-(3-methyl-5-phenylpent-2-en-1-yl)-1,3,2-dioxaborinane (1cv)



Me, Me The titled compound was obtained from alcohol **v** in 45% (184 mg) after flash chromatography (1:6 Et_2O/Hex). Mixture of 1:1.92 Z/E isomers.

¹H NMR (*E* isomer) 600 MHz, CDCl₃) δ 7.29 – 7.10 (m, 5H), 5.33 – 5.21 (m, 1H), 3.99 (t, *J* = 5.7 Hz, 2H), 2.68 (m, 2H), 2.29 (q, *J* = 9.0 Hz, 2H), 1.78-1.73 (m, 2H), 1.72 (s, 1H) 1.63 (s, 2H), 1.52-1.44 (m, 2H), 1.28 (d, *J* = 1.9 Hz, 6H). ¹³C NMR (151 MHz, CDCl₃) δ 142.83, 133.66, 133.61, 128.42, 128.21, 128.17, 125.60, 125.52, 121.25, 120.57, 70.05, 70.02, 59.08, 59.06, 41.79, 38.05, 34.96, 34.36, 34.01, 29.29, 22.67, 16.10. ¹¹B NMR (193 MHz, CDCl₃) δ 29.71. HRMS (ESI⁺): m/z for C₁₆H₂₆BO₂Na [M+Na]⁺ calcd. 295.1845, found 295.1848.

(E)-2-(3,7-dimethylocta-2,6-dien-1-yl)-4,4-dimethyl-1,3,2-dioxaborinane (1cw)



The titled compound was obtained from alcohol \mathbf{w} in 98% yield (735 mg) after flash chromatography (1:10 Et₂O/Hex).

¹H NMR (600 MHz, CDCl₃) δ 5.26 (ddq, *J* = 7.6, 6.0, 1.3 Hz, 1H), 5.12 (tp, *J* = 7.0, 1.4 Hz, 1H), 4.03 – 3.97 (m, 2H), 2.10 – 1.96 (m, 4H), 1.80 – 1.74 (m, 2H), 1.63 – 1.57 (m, 6H), 1.50 (d, *J* = 7.6 Hz, 2H), 1.29 (s, 6H). ¹³C NMR (151

MHz, $CDCl_3$) δ 134.03, 131.03, 124.66, 120.07, 69.97, 59.05, 39.88, 38.07, 29.29, 26.92, 25.72, 17.69, 15.91. ¹¹B NMR (193 MHz, $CDCl_3$) δ 29.54. HRMS (ESI⁺): m/z for $C_{15}H_{28}BO_2$ [M+H]⁺ calcd. 251.2182, found 251.2185.

(45,55)-2-((E)-but-2-en-1-yl)-4,5-bis(2-iodophenyl)-1,3,2-dioxaborolane (N-1c)



Following the general procedure, but using **N** instead of 3-methylbutane-1,3-diol.

¹**H NMR** (600 MHz, CDCl₃) δ 7.73 (dd, *J* = 8.0, 1.1 Hz, 2H), 7.40 – 7.32 (m, 4H), 6.95 (ddd, *J* = 7.9, 6.7, 2.3 Hz, 2H), 5.59 – 5.42 (m, 2H), 5.38 (s, 2H), 1.87 (dt, *J* = 7.0, 1.4 Hz, 2H), 1.64 – 1.59 (m, 3H).

¹³C NMR (151 MHz, CDCl₃) δ 141.10, 138.71, 129.10, 127.82, 126.59, 125.13, 124.19, 96.93, 87.88, 17.09. ¹¹B NMR (193 MHz, CDCl₃) δ 34.55. HRMS (ESI⁺): m/z for for $C_{18}H_{17}BO_{2}l_2Na$ [M+Na]⁺ calcd. 552.9303, found 552.9306.

PREPARATIONS OF CHIRAL HYDROBENZOINS

All chiral hydrobenzoins except **N** were prepared via Sharpless asymmetric dihydroxylation. The starting *E*-olefines were synthetized through two different strategies depicted in the Scheme S1.



Scheme S1: strategy to access the final chiral hydrobenzoins

Strategy A



In a round bottom flask containing 20 mmol of the alcohol in 40 mL THF were added 20 mmol of NaBH₄. The reaction was monitored by TLC until the total consumption of the starting aldehyde. The reaction mixture was quenched with water and extracted 3 times with Et_2O . The organic phases were collected, dried under MgSO₄ and concentrated at reduced pressure. No further purification was needed.

Naphthalen-1-ylmethanol



Quantitative yield, white solid. ¹H NMR (300 MHz, CDCl₃, ppm): δ 8.09-8.02 (m, 1H); 7.96-7.88 (m, 1H); 7.88-7.80 (dd, 1H); 7.61-7.50 (m, 2H); 7.50-7.40 (m, 2H); 5 (s, 2H); 3.55 (bs, OH). Data consistent with the literature.²⁷

(2-methoxyphenyl)methanol



Quantitative yield, white solid. ¹H NMR (300 MHz, CDCl₃, ppm): δ 7.33-7.27 (m, 2H); 6.99-6.94 (m, 1H); 6,92-6,89 (m, 1H); 4.7 (s, 2H); 3,89 (s, 3H); 2.3 (bs, OH). Data consistent with the literature.²⁸



In a round bottom flask 20 mmol of the previous alcohol were introduced with 20 mL of HCl 37% and the stirring was turned on. The reaction proceeded as heterogeneous mixture overnight, then the solution was transferred into a separating funnel and extracted three times with DCM. The organic phases were collected, dried under MgSO₄ and concentrated at reduced pressure. No further purification was needed.

1-(chloromethyl)naphthalene



Quantitative yield, pale yellow oil. ¹H NMR (300 MHz, CDCl₃, ppm): δ 8.28-8.21 (m, 1H); 8-7.89 (m, 2H); 7.73-7.54 (m, 3H); 7.47-7.41 (m, 1H); 5.07 (s, 1H). Data consistent with the literature.²⁹

1-(chloromethyl)-2-methoxybenzene



Quantitative yield, Pale yellow oil. ¹H NMR (300 MHz, CDCl₃, ppm): δ 7.43-7.31 (m, 2H); 7.01-6.89 (m, 2H); 4.69 (s, 2H); 3.9 (s, 3H). Data consistent with the literature.³⁰



20 mmol of the chloride were dissolved in 40 mL of toluene and 1 equivalent of PPh_3 was added to the flask. The reaction was stirred and refluxed overnight. Afterward the flask was cooled to room temperature and the precipitate was filtrated and washed with cold ethyl acetate. No further purification was needed.

(naphthalen-1-ylmethyl)triphenylphosphonium



Quantitative yield, white solid.¹H NMR (300 MHz, CDCl₃, ppm): δ 7.67-7.50 (m, 11H); 7.48- 7.35 (m, 7H); 7.27 (d, 1H); 7.20-7.09 (m, 2H); 6.97-6.88 (m, 1H); 5.68 (d, 2H).

(2-methoxybenzyl)triphenylphosphonium



Quantitative yield, white solid. **H NMR** (300 MHz, CDCl₃, ppm): δ 7.80 (m, 3H,); 7.66 (m, 6H); 7.61 (m, 6H); 7.30 (m, 1H); 7.25 (m, 1H); 6.80 (m, 1H), 6.62 (d, 1H), 5.05 (d, 2H), 3.20 (s, 3H). Data consistent with the literature.³¹



To a suspension of 20 mmol of the phosphonium salt in 100 mL ethanol, 2 equivalents of sodium were added in small portions. The reaction was stirred for one hour, then 1 eq. of the corresponding aldehyde was added to the mixture. After 24 hours the reaction was quenched by the addition of water and extracted three times with toluene. The organic phases were collected, dried under MgSO₄ and concentrated at reduced pressure. The residue was purified by flash chromatography followed by recrystallization in hot methanol. If the alkene was obtained as a mixture of E/Z it was converted into the *E* isomer in the presence of traces of iodine in refluxing toluene.

(E)-1,2-di(naphthalen-1-yl)ethene



Pale yellow solid. 30% yield, isolated after flashchromatography (9:1 Hex/Et₂O) as a mixture of 1:1 Z/E. Converted into the *trans* isomer with traces of iodine in refluxing toluene. ¹H NMR (300 MHz, CDCl₃, ppm): δ 8.30-8.24 (m, 2H); 7.95-7.84 (m, 8H); 7.59-7.50 (m, 6H). Data consistent with the literature.³²

(E)-1,2-bis(2-methoxyphenyl)ethene



White solid. 40% yield, isolated after flashchromatography 8:2 Hex/Et2O (Rf= 0,77). Resa 40%. Only *trans* isomer present. ¹H NMR (300 MHz, CDCl₃, ppm): δ 7.66 (dd, 2H); 7.48 (s, 2H); 7.27-7.20 (m, 2H); 6.68 (td 2H); 6.90 (dd, 2H); 3,89 (s, 6H). Data consistent with the literature.³³

Strategy B



To a suspension of 1.176 g of zinc (18 mmol, 3 eq.) in 60 mL of freshly distilled THF, 9 mL of a solution of TiCl₄ (1.0 M in DCM, 1.5 eq.) were added dropwise. The reaction was stirred and refluxed for one hour, then 0.73 mL of the aldehyde (6 mmol, 1 eq.) were added. The reaction proceeded at reflux temperature for three hours, then the flask was allowed to cool to room temperature and 60 mL of cold HCl 1N were added. The biphasic mixture was filtrated under celite, then the water layer was separated and extracted three times with DCM. The organic phases were collected, dried under MgSO₄ and concentrated at reduced pressure. Recrystallization with acetone allowed the isolation of the pure alkene in 20% yield.

(E)-1,2-bis(4-methoxyphenyl)ethene



White solid, 20% of pure *E* isomer after recrystallization in acetone. **¹H NMR** (300 MHz, CDCl₃, ppm): 7.47-7.41 (m, 4H); 6.94 (s, 2H); 6.93-6.87 (m, 4H); 3.84 (s, 6H).³⁴

Asymmetric dihydroxylation



To a heterogeneous mixture of water/terbutanol (1:1, 0.1 M) the AD-mix-alpha (1.4 g per mmol of alkene) was added. When the solid was fully dissolved, the flask was cooled to 0°C and the alkene from the previous step (1 eq.) was added to the mixture. The reaction proceeded for 5-6 days under vigorously stirring at the same temperature. Then, a saturated solution of Na₂SO₃ (2.0 g per mmol of alkene) was added to the flask and the reaction was further stirred at room temperature for 1h. The mixture was filtrated, washed with EtOAc and the organic layers were separated from the aqueous one, which was extracted three times with DCM. Afterward, the organic phases were collected, dry over Na₂SO₄ and concentrated at reduced pressure. The final product was obtained after the flash chromatography.

(15,25)-1,2-di(naphthalen-1-yl)ethane-1,2-diol (H)



White solid, 60% yield after flash chromatography 8:2 Hex:AcOEt.

¹H NMR (300 MHz, CDCl₃, ppm): 7.88 (d, 2H); 7.78-7.68 (m, 6H); 7.43-7.25 (m, 6H); 5.80 (s, 2H); 2,99 (bs, 2H). Data consistent with the literature.³

The e.r. was determined by **HPLC** analysis on Lux-amylose 2 column: hexane/*i*-PrOH 80/20 flow rate 1 mL/min, 25 °C, λ = 210 nm: t₁= 13.1 min, t₂= 22.8 min; e.r.= 56:44. The enantiomers were separated by preparative HPLC on Lux-amylose 2 column, hexane/*i*-PrOH 70:30 flow rate 5 mL/min, 25°C.



(1R,2R)-1,2-bis(2-methoxyphenyl)ethane-1,2-diol (I)

White solid. 90% yield after flash chromatography 7:3 Hex/Et₂O. ¹H NMR (300 MHz, CDCl₃, ppm): δ 7.21-7.16 (m, 4H); 6.88-6.82 (m, 2H);

6.77-6.74 (m, 2H); 5.07-5.02 (m, 2H); 3.66 (s, 6H); 3.5-3.44 (m, 2H). Data consistent with the literature.³

The e.r. was determined by **HPLC** analysis on Lux-amylose 2 column: hexane/*i*-PrOH 80/20 flow rate 1 mL/min, 25 °C, λ = 210 nm: t₁= 13.1 min, t₂= 22.8 min; e.r.= 26%. The enantiomers were separated by preparative HPLC on Lux-amylose 2 column, hexane/*i*-PrOH 70:30 flow rate 5 mL/min, 25°C.

(1R,2R)-1,2-bis(4-methoxyphenyl)ethane-1,2-diol (L)



White solid. 50% yield after flash chromatography 6:4 Hex/AcOEt. ¹H NMR (300 MHz, CDCl₃, ppm): 7.07-7.02 (m, 4H); 6.79-6.74 (m, 4H); 4.64 (s, 2H); 3.78 (s, 6H); 2.8 (bs, 2H). Data consistent with the literature.³

HO OH The e.r. was determined by **HPLC** analysis on Lux-amylose 2 column: hexane/*i*-PrOH 80/20 flow rate 1 mL/min, 25 °C, λ = 210 nm, only one peak at 33.4 min; e.r.> 99.05:0.5%.

Synthesis of catalyst N

The compound **N** was readily prepared following a reported procedure:⁶



To an oven dried three necks round bottom flask under nitrogen flow 1 gram of (*S*,*S*)- or (*R*,*R*)-hydrobenzoin (1.0 eq., 4.7 mmol), 18.5 mL of hexane an 18 mL of Et₂O were added. To this suspension, 17.5 mL of n-BuLi in hexane (6.0 eq., 28 mmol, 1.6 M) were added dropwise and the reaction was refluxed overnight. Afterward the flask was allowed to reach room temperature, then it was cooled to -78°C and 8.23 gram of iodine (7.0 eq., 32.69 mmol) dissolved into a minimal volume of Et₂O were added to the mixture. The flask was removed from the cooling bath and the stirring continued for 5 hours at room temperature. The reaction was quenched with 100 mL of saturated Na₂S₂O₃, then the layers were separated and the aqueous phase was extracted four times with 120 mL of AcOEt. The organic phases were collected, dried under MgSO₄ and concentrated at reduced pressure. Flash chromatography (3.5:1 hexane/ethyl acetate) followed by recrystallization with DCM/hexane allowed the isolation of **N** in 30% of yield.

GENERAL PROCEDURE FOR THE ASYMMETRIC SYNTHESIS OF ALLYLIC HYDRAZIDES



To a screw cap vial 0.3 mmol of boronic ester, 0.3 mL of ^{*i*}PrOH, 0.3 mL of AcOH, 0.03 mmol of **N** (14 mg) were added. The solution was stirred for 10 minutes, afterward 0.3 mmol of di-tertbutyl-azodicarboxylate (69 mg) were introduced to the vessel and the reaction was stirred for 24 hours at rt. Then the mixture was diluted with hexane and poured into a packed column to perform the flash chromatography.

Di-tert-butyl (R)-1-(but-3-en-2-yl)hydrazine-1,2-dicarboxylate (3a)

Boc N Boc

White solid. The titled compound was obtained from **1ca** in 92% yield (79 mg on 0.3 mmol) or 90% yield (386 mg on 1.5 mmol) after flash chromatography (1:7 Et_2O/Hex).

¹**H NMR** (400 MHz, CDCl₃) δ 6.08 (s, 1H), 5.85 (m, 1H), 5.15-5.08 (m, 2H), 4.77 (m, 1H), 1.47 (s, 18H), 1.25 (d, *J* = 7.0 Hz, 3H). ¹³**C NMR** (151 MHz, CDCl₃) δ 155.62, 154.75, 139.59, 138.01, 115.26, 87.93, 81.19, 80.84, 28.25, 28.16, 16.52. **HRMS (ESI**⁺): m/z for C₁₄H₂₆N₂O₄Na [M+Na]⁺ calcd. 309.1785, found 309.1791.

The enantiomeric ratio (e.r.) was determined by **GC-FID** analysis on a Rt- β DEXsm column, temperature ramp starting from 120°C, increment of 2°C/min. Two couple of peaks with the same e.r. were observed due to the deprotection of the Boc group promoted by high temperature. First couple of enantiomers t₁= 6.4 min; t₂= 6.5 min; second couple of enantiomers t₃= 9.5 min; t₄= 9.9 min. e.r.= 96:4.

Di-tert-butyl (R)-1-(hex-1-en-3-yl)hydrazine-1,2-dicarboxylate (3b)



White solid. The titled compound was obtained from **1cb** in 93% yield (88 mg) after flash chromatography (1:7 Et_2O/Hex).

^{Me} ¹**H NMR** (400 MHz, CDCl₃) δ 6.23-5.67 (m, 2H), 5.21 – 5.02 (m, 2H), 4.55 (s, 1H), 1.82-1.22 (m, 22H), 0.92 (t, *J* = 7.3 Hz, 3H). ¹³**C NMR** (151 MHz, CDCl₃) δ 155.04, 141.90, 139.75, 137.38, 131.24, 130.17, 129.28, 128.82, 127.79, 124.97, 115.79, 81.12, 80.82, 79.86, 56.72, 38.14, 37.22, 28.25, 28.17, 25.69, 25.35, 19.68, 17.66. **HRMS (ESI**⁺): m/z for C₁₆H₃₀N₂O₄Na [M+Na]⁺ calcd. 337.2098, found 337.2103. The e.r. was determined by HPLC analysis on a Chiralpak AS-H column: hexane/i-PrOH 97.5/2.5 flow rate 1 mL/min, 25 °C, λ = 210 nm: t₁= 5.2 min, t₂= 6.2 min. e.r.= 97:3

Di-tert-butyl (R)-1-(5-phenylpent-1-en-3-yl)hydrazine-1,2-dicarboxylate (3c)



White solid. The titled compound was obtained from **1cc** in 86% yield (97 mg) after flash chromatography (1:8 Et₂O/Hex).

¹H NMR (600 MHz, CDCl₃) δ 7.32 – 7.14 (m, 5H), 6.22 – 5.53 (m, 2H), 5.26-5.05 (m, 2H), 4.82-4.25 (m, 1H), 3.00 - 2.46 (m, 2H), 2.00 (s, 1H), 1.82 (s, 1H), 1.57-

1.39 (m, 18H). ¹³C NMR (151 MHz, CDCl₃) δ 155.52, 155.01, 142.02, 136.55, 135.78, 128.53, 128.31, 125.76, 125.52, 116.69, 81.30, 80.91, 58.99, 33.20, 32.50, 30.33, 29.70, 28.24, 28.18. HRMS (ESI+): m/z for C₂₁H₃₂N₂O₄Na [M+Na]⁺ calcd. 399.2254, found 399.2260.

The e.r. was determined by HPLC analysis on a Chiralpak AS-H column: hexane/i-PrOH 97.5/2.5 flow rate 1 mL/min, 25 °C, λ = 210 nm: t₁= 6.6 min, t₂= 8.5 min; e.r.= 95.5:4.5.

Di-tert-butyl (R)-1-(5-methylhex-1-en-3-yl)hydrazine-1,2-dicarboxylate (3d)



White solid. The titled compound was obtained from 1cd in 75% (74 mg) after flash $Boc_N^N_Boc$ chromatography (1:8 Et₂O/Hex).

¹H NMR (600 MHz, CDCl₃) δ 6.14-5.63 (m, 2H), 5.24-5.04- (m, 2H), 4.80-4.36 (m, Me⁻ 1H), 1.78 – 1.52 (m, 2H), 1.47 (s, 18H), 1.38 – 1.22 (m, 1H), 0.92 (dd, J = 14.4, 6.5 Hz, 6H). ¹³C NMR (151 MHz, CDCl₃) δ 171.16, 155.18, 154.73, 141.05, 140.09, 139.95, 139.02, 138.75, 137.83, 137.22, 137.12, 135.08, 134.75, 134.22, 133.36, 132.69, 130.84, 130.18, 130.10, 129.75, 129.64, 129.50, 128.96, 128.81, 128.76, 128.71, 128.66, 128.59, 128.44, 128.20, 128.10, 128.02, 127.88, 127.71, 127.66, 124.01, 119.07, 118.68, 118.33, 115.60, 81.84, 81.12, 79.83, 74.80, 74.68, 60.40, 58.56, 56.74, 29.70, 28.19, 28.10, 21.05, 14.20. **HRMS (ESI+)**: m/z for C₁₇H₃₃N₂O₄ [M+H]⁺ calcd. 329.2435, found 329.2440.

The e.r. was determined by HPLC analysis on a Chiralpak AS-H column: hexane/i-PrOH 98/2 flow rate 0.5 mL/min, 25 °C, λ = 210 nm: t₁= 9.50 min, t₂= 12.3 min; e.r. = 94:6.

Di-tert-butyl (R)-1-(hepta-1,6-dien-3-yl)hydrazine-1,2-dicarboxylate (3e)



White solid. The titled compound was obtained from 1ce in 83% yield (81 mg) after flash chromatography (1:8 Et₂O/Hex).

¹H NMR (600 MHz, CDCl₃) δ 6.14 – 5.68 (m, 3H), 5.21 – 5.08 (m, 2H), 5.07 – 4.93 (m, 2H), 4.71-4.29 (m, 1H), 2.25 – 1.93 (m, 2H), 1.88-1.70 (m, 1H), 1.66-1.54 (m, 1H), 1.46 (s, 18H). ¹³C NMR (151 MHz, CDCl₃) δ 155.53, 154.91, 138.13, 136.52, 125.52, 116.59, 114.93, 81.23, 80.89, 58.72, 30.48, 30.33, 30.29, 29.70, 28.25, 28.17. **HRMS (ESI**⁺): m/z for C₁₇H₃₁N₂O₄ [M+H]⁺ calcd. 327.2278, found 327.2284.

The e.r. was determined by **HPLC** analysis on a Chiralpak AS-H column: hexane/*i*-PrOH 95/5 flow rate 0.5 mL/min, 23 °C, λ = 210 nm: t₁= 8.4 min, t₂= 9.7 min; e.r.= 96:4.

Di-tert-butyl (S)-1-(4-ethylhex-1-en-3-yl)hydrazine-1,2-dicarboxylate (3f)

Boc White solid. The titled compound was obtained from **1cf** in 75% (77 mg) after flash HN_{N}^{POC} chromatography (1:7 Et₂O/Hex).

¹H NMR (600 MHz, CDCl₃) δ 6.13 – 5.67 (m, 2H), 5.27-4.96 (m, 2H), 4.50-4.16 (m, 1H), 1.62 – 1.13 (m, 22H), 0.91 – 0.77 (m, 6H). ¹³C NMR (151 MHz, CDCl₃) δ 174.40, 155.28, 139.74, 135.50, 133.76, 130.10, 128.83, 127.50, 126.45, 117.60, 88.83, 81.15, 80.85, 69.96, 59.84,

59.03, 46.44, 40.73, 38.08, 30.33, 29.25, 29.02, 28.27, 28.16, 27.90, 20.89, 20.43, 11.83, 11.75, 10.24, 9.96. **HRMS (ESI⁺)**: m/z for C₁₈H₃₄N₂O₄Na [M+Na]⁺ calcd. 365.2411, found 365.2416.

The e.r. was determined by **HPLC** analysis on a Chiralpak AS-H column: hexane/*i*-PrOH 95/5 flow rate 0.5 mL/min, 25 °C, λ = 210 nm: t₁= 7.4 min, t₂= 8.4 min; e.r.= 95:5.

Di-tert-butyl (S)-1-(1-cyclohexylallyl)hydrazine-1,2-dicarboxylate (3g)



White solid. The titled compound was obtained from 1cg in 75% (80 mg) after flash chromatography (1:7 Et₂O/Hex).

¹**H NMR** (400 MHz, CDCl₃) δ 6.15 – 5.61 (m, 2H), 5.24-5.08 (m, 2H), 4.33-3.91 (m, 1H), 1.79 - 1.61 (m, 5H), 1.50-1.42 (m, 19H), 1.30 - 1.10 (m, 3H), 0.92 (dd, *J* = 47.5, 1.45) (m, 2H), 1.50-1.42 (m, 2H), 1.30 - 1.10 (m, 2H), 0.92 (dd, *J* = 47.5) (m, 2H) (m

12.0 Hz, 2H). ¹³C NMR (151 MHz, CDCl₃) δ 174.70, 155.26, 139.76, 135.18, 117.85, 88.84, 81.05, 65.86, 59.83, 38.39, 33.31, 30.20, 30.05, 29.02, 28.47, 28.28, 28.16, 27.75, 26.42, 25.98, 25.36, 20.54, 15.27. HRMS (ESI⁺): m/z for C₁₉H₃₄N₂O₄Na [M+Na]⁺ calcd. 377.2411, found 377.2416.

The e.r. was determined by **HPLC** analysis on a Chiralcel OD-H column: hexane/*i*-PrOH 97/3 flow rate 0.5 mL/min, 25 °C, λ = 210 nm: t₁= 7.7 min, t₂= 8.4 min; e.r.= 94:6.

Di-tert-butyl 1-((3R,5S)-5,9-dimethyldeca-1,8-dien-3-yl)hydrazine-1,2-dicarboxylate (3h)



White solid. The titled compound was obtained from **3ch** in 70% yield (83 mg) after flash chromatography (1:9 Et₂O/Hex).

¹**H NMR** (600 MHz, CDCl₃) δ 6.07-5.68 (m, 2H), 5.23-5.02 (m, 3H), 4.85-4.46 (m, 1H), 1.98 (dp, J = 41.4, 7.7 Hz, 2H), 1.75-1.65 (m, 4H), 1.58 (s, 3H), 1.46 (s, 18H), 1.37 – 1.12 (m, 4H), 0.92 (d, J = 6.4 Hz, 3H). ¹³**C NMR** (151 MHz, CDCl₃) δ 154.99, 139.63, 136.80, 129.92, 129.87, 128.42, 116.30, 81.14, 80.83, 79.83, 59.03, 33.35,

29.71, 28.72, 28.26, 28.17, 25.69, 25.35, 19.35, 17.67, 13.89. **HRMS (ESI⁺)**: m/z for C₂₂H₄₀N₂O₄Na [M+Na]⁺ calcd. 419.2880, found 419.2886.

The d.r. was determined by **HPLC** analysis on a Chiralpak AD-H column: hexane/*i*-PrOH 95/5 flow rate 0.5 mL/min, 25 °C, λ = 210 nm: t₁= 13.7 min, t₂= 15.4 min; e.r. > 99.5:0.5; d.r.= 65:1.

Di-tert-butyl (S)-1-(1-phenylallyl)hydrazine-1,2-dicarboxylate (3i)



Boc N Boc N Boc after flash chromatography (1:7 Et₂O/Hex).

¹H NMR (600 MHz, CDCl₃) δ 7.35 – 7.09 (m, 5H), 6.22 – 5.50 (m, 3H), 5.30-4.98 (m, 2H), 1.57 – 0.89 (m, 18H).
 ¹³C NMR (151 MHz, CDCl₃) δ 156.25, 155.22, 142.01, 141.92, 139.70, 139.55, 139.05, 135.16, 130.14, 129.99, 129.81, 128.81, 128.36, 127.89, 127.82, 126.41, 117.83, 115.44, 99.14, 97.87, 88.90, 87.97, 81.05, 80.83, 67.52, 65.36, 63.96, 58.78, 43.50, 40.30, 38.40, 33.38, 32.79, 30.19, 30.06, 29.70, 28.28, 28.16, 26.42, 26.16, 26.09, 26.01, 25.99, 25.96, 25.78. HRMS (ESI⁺): m/z for C₁₉H₂₈N₂O₄Na [M+Na]⁺ calcd. 371.1941, found 371.1947.

The e.r. was determined by **HPLC** analysis on a Chiralcel OD-H column: hexane/*i*-PrOH 95/5 flow rate 1 mL/min, 25 °C, λ = 210 nm: t₁= 4.2 min, t₂= 4.7 min; e.r.= 95:5.

Di-tert-butyl (S)-1-(1-(o-tolyl)allyl)hydrazine-1,2-dicarboxylate (3j)



Colourless oil. The titled compound was obtained from **1cj** in 69% yield (75 mg) after flash chromatography (1:7 Et_2O/Hex).

¹**H NMR** (600 MHz, CDCl₃) δ 7.34-7.06 (m, 4H), 6.31 – 5.72 (m, 3H), 5.33 – 5.03 (m,

Me² 2H), 2.48 – 2.25 (m, 3H), 1.67 – 0.92 (m, 18H). ¹³**C NMR** (151 MHz, CDCl₃) δ 155.24, 154.79, 142.27, 139.76, 139.55, 139.50, 138.32, 137.89, 136.00, 135.50, 131.13, 130.68, 130.48, 130.30, 130.09, 129.73, 129.14, 128.84, 127.82, 127.54, 127.40, 126.78, 126.55, 126.43, 126.28, 126.18, 125.97, 125.88, 125.77, 125.58, 118.11, 115.41, 115.10, 97.98, 88.78, 81.41, 80.75, 79.92, 75.05, 71.98, 60.22, 52.73, 35.94, 29.71, 28.17, 28.07, 27.48, 24.64, 20.01, 19.65, 19.40, 19.11. **HRMS (ESI+)**: m/z for C₂₀H₃₀N₂O₄Na [M+Na]⁺ calcd. 381.2098, found 385.2103.

The e.r. was determined by **HPLC** analysis on a Lux cellulose-1 column: hexane/*i*-PrOH 97/3 flow rate 0.5 mL/min, 25 °C, λ = 210 nm: t₁= 10.7 min, t₂= 12.3 min; e.r.= 95:5.

Di-tert-butyl (S)-1-(1-(m-tolyl)allyl)hydrazine-1,2-dicarboxylate (3k)



Pale yellow solid. The titled compound was obtained from **1ck** in 55% yield (60 mg) after flash chromatography (1:6 Et_2O/Hex).

¹H NMR (600 MHz, CDCl₃) δ 7.26 – 7.04 (m, 4H), 6.28 – 6.01 (m, 2H), 5.96-5.65 (m, 1H), 5.34 – 5.13 (m, 2H), 2.35 (s, 3H), 1.65-1.09 (m, 18H). ¹³C NMR (151 MHz, CDCl₃) δ 154.87, 135.17, 128.27, 127.14, 125.00, 123.59, 123.39, 118.08, 114.98, 81.56,

80.88, 75.04, 28.23, 28.13, 21.46, 21.37. **HRMS (ESI**⁺): m/z for $C_{20}H_{30}N_2O_4Na$ [M+Na]⁺ calcd. 381.2098, found 385.2103.

The e.r. was determined by **HPLC** analysis on a Chiralcel OD-H column: hexane/*i*-PrOH 97/3 flow rate 0.5 mL/min, 25 °C, λ = 210 nm: t₁= 9.0 min, t₂= 9.9 min; e.r.= 94.5:5.5.

Di-tert-butyl (S)-1-(1-(4-isopropylphenyl)allyl)hydrazine-1,2-dicarboxylate (3l)



Pale yellow solid. The titled compound was obtained from **1cl** in 60% yield (69 mg) after flash chromatography (1:8 Et_2O/Hex).

¹H NMR (600 MHz, CDCl₃) δ 7.38 – 6.98 (m, 4H), 6.57-6.02 (m, 2H), 5.96-5.6 (m, 1H), 5.50-5.08 (m, 2H), 2.90 (hept, J = 6.9 Hz, 1H), 1.65 – 0.97 (m, 24H). ¹³C NMR (151 MHz, CDCl₃) δ 155.01, 154.41, 154.13, 147.43, 146.42, 142.61,

140.92, 140.69, 139.29, 138.86, 138.28, 136.63, 135.44, 134.51, 130.07, 129.83, 129.78, 129.64, 129.59, 129.27, 129.03, 128.40, 128.14, 127.87, 127.72, 127.66, 127.49, 127.29, 126.08, 126.02, 126.00, 125.99, 125.91, 125.85, 125.71, 123.51, 116.23, 116.14, 115.82, 98.58, 87.38, 80.40, 79.65, 79.61, 74.83, 74.32, 33.44, 27.59, 23.47, 10.35, 7.72. **HRMS (ESI+)**: m/z for $C_{22}H_{34}N_2O_4Na$ [M+Na]⁺ calcd. 413.2411, found 413.2416.

The e.r. was determined by **HPLC** analysis on a Chiralcel OD-H column: hexane/*i*-PrOH 97/3 flow rate 0.5 mL/min, 20 °C, λ = 230 nm: t₁= 9.0 min, t₂= 9.6 min; e.r.= 96:4.

Di-tert-butyl (S)-1-(1-(3-methoxyphenyl)allyl)hydrazine-1,2-dicarboxylate (3m)



ÓMe

White solid. The titled compound was obtained from alcohol **1cm** in 47% yield (53 mg) after flash chromatography (1:7 Et_2O/Hex).

¹H NMR (600 MHz, CDCl₃) δ 7.17 (t, J = 7.9 Hz, 1H), 6.91 – 6.69 (m, 3H), 6.09 – 5.51 (m, 3H), 5.20 (m, 2H), 3.74 – 3.69 (m, 3H), 1.54 – 0.96 (m, 18H).
 ¹³C NMR (151 MHz, CDCl₃) δ 158.82, 158.78, 158.61, 154.14, 153.83, 143.30, 140.84, 139.89, 139.13,

138.58, 133.88, 128.87, 128.56, 128.51, 128.33, 127.38, 119.20, 118.08, 117.58, 117.24, 114.15, 112.67, 112.28, 111.81, 110.68, 98.09, 80.61, 79.89, 78.82, 74.23, 54.21, 29.30, 28.68, 27.19, 27.11, 26.92. **HRMS (ESI+)**: m/z for C₂₀H₃₀N₂O₅Na [M+Na]⁺ calcd. 401.2047, found 401.2052.

The e.r. was determined by **HPLC** analysis on a Chiralcel OD-H column: hexane/*i*-PrOH 95/5 flow rate 0.5 mL/min, 25 °C, λ = 254 nm: t₁= 9.3 min, t₂= 10.7 min; e.r.= 96:4.

Di-tert-butyl (S)-1-(1-(4-methoxyphenyl)allyl)hydrazine-1,2-dicarboxylate (3n)



Pale yellow solid. The titled compound was obtained from **1cn** in 40% yield (45 mg) after flash chromatography (1:7 Et_2O/Hex).

¹H NMR (600 MHz, CDCl₃) δ 7.34-7.20 (m, 2H), 6.90 – 6.83 (m, 2H), 6.27 – 5.92 (m, 2H), 5.76 (m, 1H), 5.24 (m, 2H), 3.80 (s, 3H), 1.58 – 1.03 (m, 18H). ¹³C NMR

 $(151 \text{ MHz}, \text{CDCl}_3) \ \delta \ 159.02, \ 154.86, \ 141.82, \ 139.63, \ 135.48, \ 130.95, \ 130.10, \ 129.92, \ 129.87, \ 129.38, \ 128.42, \ 127.96, \ 127.67, \ 127.61, \ 126.29, \ 120.68, \ 117.67, \ 114.05, \ 114.02, \ 113.97, \ 113.71, \ 113.69, \ 81.53, \ 80.86, \ 79.82, \ 63.93, \ 55.28, \ 31.93, \ 30.33, \ 29.70, \ 28.24, \ 28.13, \ 22.70, \ 14.13. \ \textbf{HRMS} \ \textbf{(ESI^+)}: \ m/z \ for \ C_{20}H_{30}N_2O_5Na \ [M+Na]^+ \ calcd. \ 401.2047, \ found \ 401.2052.$

The e.r. was determined by **HPLC** analysis on a Chiralcel OD-H column: hexane/*i*-PrOH 98/2 flow rate 1 mL/min, 25 °C, λ = 228 nm: t₁= 11.3 min, t₂= 12.1 min; e.r.= 94:6.

Di-tert-butyl (S)-1-(1-(naphthalen-1-yl)allyl)hydrazine-1,2-dicarboxylate (30)



Colourless oil. The titled compound was obtained from **1co** in 58% yield (69 mg) after flash chromatography (1:7 Et_2O/Hex). 69 mg, 58% yield.

¹H NMR (25°C) (400 MHz, C₂D₂Cl₄) δ 8.44-7.97 (m, 1H), 7.94 – 7.82 (m, 2H), 7.63 – 7.41 (m, 4H), 6.92-6.11 (m, 2H), 5.55-5.09 (m, 2H), 1.79 – 1.09 (m, 16H), 0.53 (bs, 2H). ¹H NMR (105°C) (400 MHz, C₂D₂Cl₄) δ 8.26 (d, J = 8.4 Hz, 1H), 7.90 (d, J = 8.0

Hz, 1H), 7.84 (d, J = 8.1 Hz, 1H), 7.62 – 7.44 (m, 4H), 6.63 (d, J = 5.9 Hz, 1H), 6.43-6.30 (m, J = 8.7 Hz, 1H), 5.87 (s, 1H), 5.42 – 5.28 (m, 2H), 1.52 (s, 10H), 1.21 (bs, 8H). ¹³**C** NMR (151 MHz, C₂D₂Cl₄) δ 135.59, 133.76, 128.83, 126.75, 125.92, 125.38, 124.36, 81.81, 74.31, 74.23, 74.05, 73.87, 28.37, 27.89, 27.05, 20.48. HRMS (ESI⁺): m/z for C₂₃H₃₀N₂O₄Na [M+Na]⁺ calcd. 421.2098, found 421.2104.

The e.r. was determined by **HPLC** analysis on a Chiralpak AS-H column: hexane/*i*-PrOH 95/5 flow rate 0.5 mL/min, 25 °C, λ = 210 nm: t₁= 10.4 min, t₂= 12 min; e.r.= 94.5:5.5.

Di-tert-butyl (S)-1-(1-([1,1'-biphenyl]-4-yl)allyl)hydrazine-1,2-dicarboxylate (3p)



Colourless oil. The titled compound was obtained from **1cp** in 57% yield (72 mg) after flash chromatography (from 1:8 Et_2O/Hex to 1:5 Et_2O/Hex).

¹H NMR (600 MHz, CDCl₃) δ 7.76 – 7.30 (m, 9H), 6.45 – 6.03 (m, 2H), 6.06 – 5.68 (m, 1H), 5.44 – 5.15 (m, 2H), 1.72 – 1.00 (m, 18H). ¹³C NMR (151 MHz, CDCl₃) δ 155.21, 154.92, 154.59, 141.70, 140.84, 140.71, 140.44, 140.25,

139.76, 138.29, 137.75, 135.83, 135.07, 130.71, 129.91, 128.88, 128.79, 128.73, 128.51, 128.40, 128.15, 127.42, 127.34, 127.32, 127.25, 127.23, 127.16, 127.13, 127.10, 127.04, 127.01, 126.98, 126.94, 126.88, 126.79, 126.68, 118.59, 118.32, 118.09, 115.22, 81.74, 81.01, 75.12, 58.71, 57.22, 31.61, 29.72, 28.45, 28.25, 28.21, 28.15, 22.67, 20.74, 14.15. **HRMS (ESI⁺)**: m/z for $C_{25}H_{32}N_2O_4Na$ [M+Na]⁺ calcd. 447.2254, found 447.2260.

The e.r. was determined by **HPLC** analysis on a Lux 5u cellulose-1 column: hexane/*i*-PrOH 97/3 flow rate 0.5 mL/min, 23 °C, λ = 210 nm: t₁= 16.9 min, t₂= 19.1 min; e.r.= 96:4.

Di-tert-butyl (S)-1-(1-(4-chlorophenyl)allyl)hydrazine-1,2-dicarboxylate (3q)



White solid. The titled compound was obtained from 1cq in 60% yield (69 mg) after flash chromatography (1:7 Et₂O/Hex). 69 mg, 60% yield.

¹H NMR (400 MHz, CDCl₃) δ 7.35-7.17 (m, 4H), 6.34-5.59 (m, 3H), 5.35-5.10 (m, 2H), 1.54-0.99 (m, 18H). ¹³C NMR (151 MHz, CDCl₃) δ 171.16, 155.18, 154.73,

141.05, 140.09, 139.95, 139.02, 138.75, 137.83, 137.22, 137.12, 135.08, 134.75, 134.22, 133.36, 132.69, 130.84, 130.18, 130.10, 129.75, 129.64, 129.50, 128.96, 128.81, 128.76, 128.71, 128.66,

128.59, 128.44, 128.20, 128.10, 128.02, 127.88, 127.71, 127.66, 124.01, 119.07, 118.68, 118.33, 115.60, 81.84, 81.12, 79.83, 74.80, 74.68, 60.40, 58.56, 56.74, 29.70, 28.19, 28.10, 21.05, 14.20. **HRMS (ESI+)**: m/z for C₁₉H₂₇N₂O₄ClNa [M+Na]⁺ calcd. 405.1552, found 405.1557.

The e.r. was determined by **HPLC** analysis on a Chiralpak AS-H column: hexane/*i*-PrOH 95/5 flow rate 0.5 mL/min, 25 °C, λ = 210 nm: t₁=11.7 min, t₂= 14 min; e.r.= 96:4.

Di-tert-butyl (S)-1-(1-(4-(trifluoromethyl)phenyl)allyl)hydrazine-1,2-dicarboxylate (3r)



White solid. The titled compound was obtained from **1cr** in 34% yield (43 mg) after flash chromatography (1:7 Et_2O/Hex).

¹H NMR (600 MHz, CDCl₃) δ 7.64 – 7.41 (m, 4H), 6.23 – 5.62 (m, 3H), 5.35 (d, J $CF_3 = 10.3$ Hz, 1H), 5.23 (d, J = 17.4 Hz, 1H), 1.56 – 1.00 (m, 18H). ¹³C NMR (151

MHz, CDCl₃) δ 154.70, 139.66, 134.40, 131.29, 129.83, 128.39, 126.83, 126.59, 125.23, 123.24, 82.01, 81.25, 79.80, 65.86, 63.35, 28.17, 28.08, 15.28. ¹⁹F NMR (376 MHz, CDCl₃) δ -62.59. HRMS (ESI⁺): m/z for C₂₀H₂₈N₂O₄F₃ [M+H]⁺ calcd. 417.1996, found 417.2001.

The e.r. was determined by **HPLC** analysis on a Lux cellulose-1 column: hexane/*i*-PrOH 97/3 flow rate 0.5 mL/min, 25 °C, λ = 210 nm: t₁= 10.0 min, t₂= 10.7 min; e.r.= 97:3.

Di-tert-butyl (R)-1-(3-methylbut-3-en-2-yl)hydrazine-1,2-dicarboxylate (3s)

Boc White solid. The titled compound was obtained from **1cs** in 86% yield (78 mg) after HN_{N}^{Boc} flash chromatography (1:7 Et₂O/Hex).

Me ¹H NMR (600 MHz, CDCl₃) δ 6.15-5.52 (m, 1H), 4.98-4.33 (m, 3H), 1.92-1.62 (m, 3H), Me 1.57-1.42 (m, 18H), 1.28 (d, J = 6.9 Hz, 3H). ¹³C NMR (151 MHz, CDCl₃) δ 155.04, 144.65,

111.79, 81.14, 80.75, 28.25, 28.17, 20.76, 15.37. **HRMS (ESI+)**: m/z for C₁₅H₂₈N₂O₄Na [M+Na]⁺ calcd. 323.1941, found 323.1947.

The e.r. was determined by **HPLC** analysis on a Chiralcel OD-H column: hexane/*i*-PrOH 97/3 flow rate 0.5 mL/min, 25 °C, λ = 210 nm: t₁= 8 min, t₂= 9.2 min; e.r.> 99.5:0.5.

Di-tert-butyl (S)-1-(2-methyl-1-phenylallyl)hydrazine-1,2-dicarboxylate (3t)



¹H NMR (600 MHz, CDCl₃) δ 7.37 – 7.22 (m, 5H), 6.24-5.95 (m, 1H), 5.82-5.52 (m, 1H), 5.11-4.66 (m, 2H), 1.87 – 1.01 (m, 21H). ¹³C NMR (151 MHz, CDCl₃) δ 155.19,

154.72, 154.37, 143.02, 141.92, 139.74, 139.58, 137.60, 136.19, 130.67, 130.17, 130.08, 129.94, 129.86, 129.54, 129.05, 128.88, 128.83, 128.68, 128.38, 128.33, 128.19, 128.17, 128.13, 128.02, 127.80, 127.53, 126.59, 126.45, 126.41, 124.96, 111.14, 99.12, 97.84, 87.94, 81.51, 81.34, 80.79, 80.56, 79.88, 68.93, 67.15, 65.85, 62.26, 30.33, 29.70, 28.37, 28.16, 28.04, 27.94, 27.57, 21.79,

21.08, 18.23, 15.27, 14.12. HRMS (ESI⁺): m/z for $C_{20}H_{30}N_2O_4Na$ [M+Na]⁺ calcd. 385.2098, found 385.2103.

The e.r. was determined by **HPLC** analysis on a Chiralpak AS-H column: hexane/*i*-PrOH 97/3 flow rate 0.5 mL/min, 25 °C, λ = 220 nm: t₁= 17 min, t₂= 20.2 min; e.r.= 99.5:0.5.

Di-tert-butyl 1-((1*R*,5*S*)-2-methylene-5-(prop-1-en-2-yl)cyclohexyl)hydrazine-1,2-dicarboxylate (3u)

Boc N Boc

White solid. The titled compound was obtained from **1cu** in 55% yield (61 mg) after flash chromatography (1:9 Et_2O/Hex).

^{Me} ¹H NMR (600 MHz, CDCl₃) δ 6.33-5.85 (m, 1H), 5.00 – 4.48 (m, 5H), 2.58 – 1.33 (m, 28H). ¹³C NMR (151 MHz, CDCl₃) δ 155.60, 155.33, 154.91, 150.61, 148.81, 147.43, 146.83, 139.75, 135.73, 130.11, 128.85, 127.54, 125.52, 122.39, 119.38, 114.07, 110.16, 109.77, 108.99, 108.78, 108.66, 108.19, 88.91, 81.35, 81.20, 80.86, 67.23, 65.85, 60.96, 59.05, 56.97, 41.16, 41.01, 39.33, 38.06, 36.79, 34.32, 33.50, 32.81, 32.21, 31.72, 31.43, 31.06, 30.89, 30.41, 30.33, 29.70, 29.30, 29.05, 28.35, 28.23, 28.18, 27.48, 26.12, 21.59, 21.19, 21.00, 20.92, 20.83, 20.80. HRMS (ESI⁺): m/z for C₂₀H₃₅N₂O₄ [M+H]⁺ calcd. 367.2591, found 367.2597.

The d.r was determined by ¹H NMR analysis at 100°C. dr> 20:1.

Di-tert-butyl 1-((1*S*,5*S*)-2-methylene-5-(prop-1-en-2-yl)cyclohexyl)hydrazine-1,2-dicarboxylate (3u')



Pale yellow oil. The titled compound was obtained from **1cu** with *ent*-**N** in 44% yield (49 mg) after flash chromatography (1:9 Et₂O/Hex).

¹H NMR (400 MHz, CDCl₃) δ 6.32-5.98 (m, 1H), 4.88-4.63 (m, 4H), 4.54 – 4.29 (m, 1H), 2.50-2.40 (m, 1H), 2.33 – 2.02 (m, 4H), 1.87-1.75 (m, 1H), 1.58-1.38 (m, 18H), 1.36 – 1.15 (m, 2H). ¹³C NMR (151 MHz, CDCl₃) δ 154.79, 154.15, 153.89, 147.80,

138.73, 129.09, 127.83, 126.52, 118.37, 108.09, 107.96, 107.42, 107.16, 104.12, 103.75, 87.89, 80.41, 80.21, 79.79, 64.84, 58.03, 57.84, 43.37, 43.16, 40.13, 38.31, 37.04, 34.01, 33.60, 33.29, 30.89, 30.64, 30.03, 29.99, 28.28, 28.01, 27.17, 19.90, 14.25. **HRMS (ESI⁺)**: m/z for $C_{20}H_{35}N_2O_4$ [M+H]⁺ calcd. 367.2591, found 367.2597.

The d.r. was determined by ¹H NMR analysis at 100°C. dr: 89:11.

Di-tert-butyl (R)-1-(3-methyl-5-phenylpent-1-en-3-yl)hydrazine-1,2-dicarboxylate (3v)



Pale yellow solid. The titled compound was obtained from 1cv in 50% yield (59 mg) after flash chromatography (1:7 Et₂O/Hex).

Me⁻¹H NMR (400 MHz, CDCl₃) δ 7.35-7.09 (m, 5H), 6.29-5.89 (m, 2H), 5.17 – 5.03 (m, 2H), 2.80 – 2.20 (m, 3H), 2.14-1.60 (m, 2H), 1.53-1.40 (m, 20H). ¹³C NMR (151 MHz, CDCl₃) δ 143.50, 127.35, 124.76, 110.31, 84.60, 80.25, 63.70, 39.62, 29.84, 27.25, 27.24, 27.21, 22.69. HRMS (ESI⁺): m/z for C₂₂H₃₄N₂O₄Na [M+Na]⁺ calcd. 413.2411, found 413.2416.

To determine the *e.r.*, 50 mg of the compound (0.1 mmol) were converted into the N-benzylated analogue in 0.2 mL of dry DMF with 49 mg of Cs_2CO_3 (0.15 mmol) and 19 mg of 4- (bromomethyl)benzonitrile (0.096 mmol, 0.96 eq.), under vigorously stirring at room temperature for 5 hours. The reaction was quenched with water and extracted one time with Et_2O . The organic layer was directly poured into a packed column to perform the purification (1:5 Et_2O /Hex).



Di-tert-butyl (*R*)-1-(4-cyanobenzyl)-2-(3-methyl-5-phenylpent-1-en-3-yl)hydrazine-1,2dicarboxylate (3vb)



¹H NMR (600 MHz, CDCl₃) δ 7.62 – 7.40 (m, 2H), 7.32 – 6.99 (m, 7H), 6.30 – 5.92 (m, 1H), 5.42 – 4.94 (m, 2H), 4.52 (d, *J* = 136.2 Hz, 1H), 2.77 – 2.57 (m, 1H), 2.59 – 2.19 (m, 2H), 1.95 (m, 1H), 1.52 – 1.25 (m, 18H), 1.23 – 1.19 (m, 3H). ¹³C NMR (151 MHz, CDCl₃) δ

144.54, 132.03, 130.56, 128.43, 128.41, 128.35, 128.27, 125.86, 125.78, 118.82, 111.71, 111.34, 82.07, 81.91, 80.89, 65.86, 64.45, 57.29, 55.91, 44.07, 41.33, 41.06, 40.64, 31.34, 30.86, 30.33, 28.38, 28.27, 28.24, 28.12, 28.05, 25.38, 23.71, 15.29. **HRMS (ESI**⁺): m/z for $C_{30}H_{39}N_3O_4Na$ [M+H]⁺ calcd. 528.2833, found 528.2838.

The d.r and e.r. were determined by **HPLC** analysis on a Chiralpak ID-3 column: hexane/*i*-PrOH 98/2 flow rate 0.5 mL/min, 25 °C, λ = 254 nm: t₁= 21 min, t₂= 21.5 min; t₃=26.1 min; t₄=29 min. d.r. (peak 1 and peak 3 one diastereoisomer; peak 2 and peak 4 one diastereoisomer) =1:1. The e.r. of the starting trisubstituted hydrazide is 70:30.
Di-tert-butyl (R)-1-(3,7-dimethylocta-1,6-dien-3-yl)hydrazine-1,2-dicarboxylate (3w)



Colourless oil. The titled compound was obtained from **1cw** in 80% yield (88 mg) after flash chromatography (from 100% Hex to 1:8 Et₂O/Hex).

¹H NMR (600 MHz, CDCl₃) δ 6.23 – 5.80 (m, 2H), 5.25 – 4.90 (m, 3H), 2.11 – 1.87 (m, 3H), 1.86 – 1.69 (m, 1H), 1.70-1.64 (m, 3H), 165-1.53 (m, 4H), 1.53 – 1.40 (m, 18H), 1.40-1.29 (m, 2H).
¹³C NMR (151 MHz, CDCl₃) δ 155.99, 144.93, 142.61, 131.47,

Me^{- N}_{Me} 1.40-1.29 (m, 2H). ¹³**C NMR** (151 MHz, CDCl₃) δ 155.99, 144.93, 142.61, 131.47, 124.28, 111.55, 110.89, 81.02, 80.89, 80.68, 64.46, 38.82, 38.05, 37.74, 28.40, 28.21, 28.19, 28.18, 25.64, 23.72, 23.05, 22.99, 22.76, 17.58, 15.22. **HRMS (ESI**⁺): m/z for C₂₀H₃₆N₂O₄Na [M+Na]⁺ calcd. 391.2567, found 391.2573.

To determine the e.r., 37 mg of the compound (0.1 mmol) were converted into the N-benzylated analogue in 0.2 mL of dry DMF with 49 mg of Cs_2CO_3 (0.15 mmol) and 19 mg of 4-(bromomethyl)benzonitrile, (0.096 mmol, 0.96 eq.) under vigorously stirring at room temperature for 5 hours. The reaction was quenched with water and extracted one time with Et_2O . The organic layer was directly poured into a packed column to perform the purification (1:4 Et_2O /Hex).



Di-tert-butyl (*R*)-1-(4-cyanobenzyl)-2-(3,7-dimethylocta-1,6-dien-3-yl)hydrazine-1,2-dicarboxylate (3wb)



¹H NMR (600 MHz, CDCl₃) δ 7.59 – 7.51 (m, 2H), 7.49 – 7.43 (m, 2H), 6.19 – 5.95 (m, 1H), 5.15 – 4.99 (m, 3H), 4.55 – 4.30 (m, 2H), 2.10 –1.10 (m, 31H). HRMS (ESI⁺): m/z for $C_{28}H_{41}N_3O_4Na$ [M+Na]⁺ calcd. 506.2989, found 506.2995.

The d.r and e.r. were determined by **HPLC** analysis on a Chiralpak ID-3 column: hexane/*i*-PrOH 98/2 flow rate 0.5 mL/min, 25 °C, λ = 210 nm: t₁= 13.5 min, t₂= 14 min; t₃=16 min; t₄=17.3 min. d.r. (peak 1 and peak 3 one diasterisomer; peak 2 and peak 4 one diastereoisomer) =1.15:1, e.r. (major

diastereoisomer)= 94.5:5.5, e.r. (minor diastereoisomer)= 95:5. Enantiomeric ratio of the starting trisubstitued hydrazide= 90%.

Diisopropyl (R)-1-(but-3-en-2-yl)hydrazine-1,2-dicarboxylate (3x)



 $\label{eq:homoscale} {}^{1}\text{H NMR} (400 \text{ MHz, CDCl}_3) \, \delta \, 6.01 \, (\text{m}, 2\text{H}), 5.21 - 5.04 \, (\text{m}, 2\text{H}), 4.92 \, (\text{hept}, J = 6.2 \, \text{Hz}, 2\text{H}), \, 4.78 \, (\text{m}, 1\text{H}), \, 1.22 \, (\text{m}, 15\text{H}). \, \text{HRMS} \, \text{(ESI^+)}: \, \text{m/z} \, \text{for} \\ C_{12}\text{H}_{22}\text{N}_2\text{O}_4\text{Na} \, [\text{M+Na}]^+ \, \text{calcd.} \, 281.1472, \, \text{found} \, 281.1478.$

The e.r. was determined by **HPLC** analysis on a Chiralcel IC column: hexane/*i*-PrOH 97/3 flow rate 1 mL/min, 25 °C, λ = 210 nm: t₁= 31.10 min, t₂= 33,3 min; e.r.=99.5:0.5.

DERIVATIZATIONS REACTIONS

Synthesis of 4a



In a 50 mL two necks round bottom flask under nitrogen flow were added 6.3 mL of freshly distilled THF, 85.8 mg of x (0.3 mmol, 1 eq.) and 1.2 mL of 9-BBN (0.5M in THF, 0.6 mmol, 2 eq.). The solution was stirred overnight at room temperature. When the starting material was fully consumed (monitored by TLC), the reaction was quenched with 2 mL of distilled water and 0.6 mL of 1.0 M NaOH. Then the flask was cooled to 0°C and 0.22 mL of H₂O₂ (30% w/w) were added dropwise. The stirring continued for 24 hours at room temperature; afterward the layers were separated and the aqueous phase was washed three times with Et₂O. The organic phases were collected, washed with brine, dried with MgSO₄ and concentrated at reduced pressure. The residue was purified by flash chromatography (2:1 Et₂O/Hex) to obtain 64 mg of the desired product (0.21 mmol, 70% yield).

Di-tert-butyl (R)-1-(4-hydroxybutan-2-yl)hydrazine-1,2-dicarboxylate (4a)

 $\begin{array}{c} & \mbox{H} & \m$

The e.r. was determined by **HPLC** analysis on a Chiralpak IA-3 column: hexane/*i*-PrOH 90/10 flow rate 0.5 mL/min, 25 °C, λ = 210 nm: t₁= 19.8 min, t₂= 26.2 min; e.r.= 96:4.

Synthesis of 5a



In a 25 mL round bottom containing 286 mg of **3a** (1 mmol, 1 eq.) 6 mL of AcOEt and 6 mL of ACN were added. The resulting solution was cooled to 0°C and 14.5 mg of RuCl₃ x $3H_2O$ (0.05 mmol, 0.05 eq.), 321 mg of NaIO₄ (1.5 mmol, 1.5 eq.) and 2 mL of distilled water were added. The dark mixture was stirred for 10 minutes in the ice bath. Afterward, 15 mL of water were added to the solution and extracted three times with AcOEt. The organic phases were collected, dry under Na₂SO₄ and

concentrated at reduced pressure. The residue was purified with flash chromatography (1.5:1 AcOEt/Hex) to afford 266 mg of **5a** as a mixture of diastereosiomers (0.83 mmol, 83% yield).

Di-tert-butyl 1-((2R)-3,4-dihydroxybutan-2-yl)hydrazine-1,2-dicarboxylate (5a)

 $\begin{array}{ccc} & & & & & & \\ & & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & & & \\ & & & & & \\ & & & & \\ & & & & \\ & & & & & \\ & & & & & \\$

The d.r and e.r. were determined by **HPLC** analysis on a Chiralpak IA column: hexane/*i*-PrOH 90/10 flow rate 0.5 mL/min, 25 °C, λ = 210 nm: t₁ = 11 min, t₂ = 11.6 min; t₃ = 17.2 min; t₄ = 19.4 min. dr = 2.2:1, e.r. (major diastereoisomer) = e.r.= 96:4.

Synthesis of 6a



128 mg of the previous diol **5a** (0.4 mmol, 1 eq.) were added to a screw cap vial containing an heterogeneous mixture of 4.2 mL of Et_2O/H_2O (2:1). Then, 214 mg of $NalO_4$ (0.4 mmol, 1.0 eq.) were added and was vigorously stirred for 5 minutes. The mixture was diluted with Et_2O and extracted with water. The layers were separated, dried under Na_2SO_4 and concentrated at the rotary evaporator without heating. The crude was immediately dissolved in 4 mL of Et_2O , split in two parts to proceed with the next derivatizations.

Synthesis of 8a



Half of the crude was concentrated at reduced pressure, redissolved in 4 mL of iPrOH and subjected to the reduction with 7.55 mg of NaBH₄ (0.2 mmol, 1 eq.). After 2 hours the reaction was quenched with a saturated solution of NH_4CI and extracted three times with AcOEt. The organic phases were collected, washed with brine, dried with MgSO₄ and concentrated at reduced pressure. The crude did not require further purifications.

Di-tert-butyl (R)-1-(1-hydroxypropan-2-yl)hydrazine-1,2-dicarboxylate (8a)

White solid, 48 mg, 83% yield after 2 steps (from 5a).

^{DC} ¹H NMR (600 MHz, CDCl₃) δ 6.36 – 5.86 (m, 1H), 4.66 – 4.16 (m, 2H), 3.65 – 3.21 (m, 2H), 1.48 (d, J = 11.7 Hz, 19H), 1.13 – 0.93 (m, 3H). ¹³C NMR (151 MHz, CDCl₃) δ 158.55, 157.95, 155.71, 154.84, 82.30, 81.98, 81.40, 78.56, 67.75, 67.29, 63.36,

56.01, 53.73, 38.71, 29.70, 28.21, 28.11, 27.95, 26.98, 13.71, 10.61, 9.97, 9.42. **HRMS (ESI+)**: m/z for $C_{13}H_{27}N_2O_5$ [M+H]⁺ calcd. 291.1914, found 291.1920.

Compound *ent-8a* (29 mg, 0.1 mmol), prepared from *ent-3a* following the so far reported procedures of derivatization, was directly converted into the N-benzylated analogue **9** in 0.2 mL of dry DMF with 49 mg of Cs_2CO_3 (0.15 mmol) and 19 mg of 4-(bromomethyl)benzonitrile (0.096 mmol, 0.96 eq.), under vigorously stirring at room temperature for 5 hours. The reaction was quenched with water and extracted one time with Et_2O . The organic layer was directly poured into a packed column to perform the purification (100% Et_2O).



Di-tert-butyl -1-(4-cyanobenzyl)-2-(1-hydroxypropan-2-yl)hydrazine-1,2-dicarboxylate (9)



OH

The product was obtained as a mixture of diastereoisomers **9a** and **9a'**. Viscous oil. 30 mg, 75% yield after two steps (from **5a**).

¹H NMR (600 MHz, CDCl₃) δ 7.65 – 7.50 (m, 2H), 7.45 – 7.36 (m, 2H), 5.46 – 4.08 (m, 3H), 3.70 –3.39 (m, 2H), 1.53 – 0.76 (m, 21H).
¹³C NMR (151 MHz, CDCl₃) δ

158.36, 157.64, 154.65, 154.41, 146.21, 143.71, 143.09, 132.33, 132.22, 132.18, 129.21, 128.00, 127.60, 127.02, 118.85, 118.73, 118.58, 111.24, 110.96, 83.66, 83.57, 82.32, 81.72, 64.59, 64.25, 63.64, 63.62, 60.40, 58.64, 58.37, 58.30, 56.69, 55.13, 31.59, 30.33, 29.70, 28.26, 28.21, 28.18, 28.11, 27.99, 27.88, 22.66, 21.06, 15.25, 15.02, 14.21, 14.12. **HRMS (ESI+)**: m/z for $C_{21}H_{31}N_3O_5Na$ [M+Na]⁺ calcd. 428.2156, found 428.2162.

The d.r and e.r. were determined by **HPLC** analysis on a Lux cellulose-2 column: hexane/*i*-PrOH 90/10 flow rate 1 mL/min, 25 °C, λ = 254 nm: t₁= 14.1 min, t₂= 14.9 min; t₃=20.6 min; t₄=23.6 min. d.r.=1:12 (peak 1 and peak 2 one diasterisomer; peak 3 and peak 4 one diasteroisomer). The e.r. of **9a** and **9a'** is 90:10. The e.r. of starting trisubstitued hydrazide **ent-8a** is 90:10.

Synthesis of 7a



Half of the crude from 6a was concentrated at reduced pressure and redissolved in 2.5 mL of tBuOH to be subjected to the Pinnick oxidation. To the solution were added 0.5 mL distilled water, 46 mg of KH₂PO₄ (0.34 mmol, 1.7 eq.), 0.1 mL of H₂O₂ 30% w/w (9.6 mmol, 4.8 eq.) and 62 mg of NaOCl₂ (0.69 mmol, 3.45 eq.) and the stirring was continued overnight. Then the reaction was quenched with distilled water and extracted three times with AcOEt. The organic phases were collected, washed with brine, dried under MgSO₄ and concentrated at reduced pressure. The crude did not require further purifications.

N-(tert-butoxycarbonyl)-N-((tert-butoxycarbonyl)amino)-D-alanine (7a)



13.50, 13.10. HRMS (ESI⁺): m/z for C₁₃H₂₄N₂O₆Na [M+Na]⁺ calcd. 327.1527, found 327.1532.

To determine the e.r., 30 mg of the compound 8a (0.1 mmol, 1 eq.) was converted into the corresponding methyl-ester in MeOH (0.1 M) with 25 μL of TMSCI (0.2 mmol, 2 eq.), stirring at 25°C for three days. Afterward, the solvent was removed under reduced pressure to afford the desired product.



Di-tert-butyl (R)-1-(1-methoxy-1-oxopropan-2-yl)hydrazine-1,2-dicarboxylate (15a)



¹**H NMR** (600 MHz, CDCl₃) δ 6.27 (d, J = 165.1 Hz, 1H), 4.83 (d, J = 171.9 Hz, 1H), Boc N^{-N} Boc 3.73 (s, 3H), 1.47 (d, J = 6.1 Hz, 22H). ¹³C NMR (151 MHz, CDCl₃) δ 172.04, 154.09, Me 80.79, 51.26, 28.68, 27.15, 27.11, 13.32. HRMS (ESI⁺): m/z for C₁₄H₂₅N₂O₆Na [M+Na]⁺ calcd. 341.1683, found 341.1689.

Synthesis of 10j



51 mg of **3j** (0.14 mmol, 1 eq.) were added to a capped vial and dissolved with 0.3 mL of acetic anhydride and 0.14 mL of pyridine. Afterward, 8.5 mg of DMAP (0.07 mmol, 0.5 eq.) were added to the solution and the vial was placed in an oil bath at 50°C and the mixture was vigorously stirred for 24 hours. Then, the reaction was cooled to room temperature and quenched with distilled water. The layers were separated and the aqueous phase was extracted three times with Et₂O. The organic phases were collected, washed two times with sat NaHCO₃, twice with sat NH₄Cl and eventually two times with brine. The organic layer was separated, dried with MgSO₄ and concentrated at reduced pressure. The residue was purified by flash chromatography (from 1:5 to 1:3 Et₂O/Hex) to afford 45 mg **10j** (0.11 mmol, 80% yield).

Di-tert-butyl (S)-1-acetyl-2-(1-(o-tolyl)allyl)hydrazine-1,2-dicarboxylate (10n)



¹H NMR (600 MHz, CDCl₃) δ 8.14 – 7.05 (m, 4H), 5.96 – 4.93 (m, 4H), 2.53 – 2.21 (m, 6H), 1.64-1.00 (m, 18H). ¹³C NMR (151 MHz, CDCl₃) δ 134.46, 130.54, 130.18, 126.68, 126.27, 125.82, 117.73, 84.39, 81.83, 30.94, 28.04, 28.01, 27.85, 27.72, 27.61, 25.77, 19.32. HRMS (ESI⁺): m/z for $C_{22}H_{32}N_2O_5Na$ [M+Na]⁺ calcd. 427.2203, found 427.2209.

Synthesis of 11j



45 mg of **10j** (0.11 mmol, 1 eq.) were added to an oven dried two necks round bottom flask under nitrogen flow containing 1.5 mL of degassed THF and 0.2 mL of degassed HPMA. Then the flask was placed in an oil bath at 65°C and 8 mL of Sml₂ (0.1 M in THF) were added dropwise while the solution is under stirring. After two hours the flask was removed from the bath and the reaction was quenched with sat NaHCO₃ and extracted three times with Et₂O. The organic phases were collected, dried with MgSO₄ and concentrated at reduced pressure. The residue was purified by flash chromatography (from 1:5 to 1:3 Et₂O/Hex) to afford 21 mg of **11j** (0.08 mmol, 76% yield).

Tert-butyl (S)-(1-(o-tolyl)allyl)carbamate (11j)



¹H NMR (600 MHz, CDCl₃) δ 7.21 – 7.15 (m, 4H), 6.06 – 5.96 (m, 1H), 5.48 (bs, 1H), 5.22 (d, 1H) 5.18 (d, 1H), 2.38 (s, 3H), 1.43 (bs, 9H). ¹³C NMR (151 MHz, CDCl₃) δ 154.98, 138.96, 137.65, 130.71, 127.51, 126.36, 126.19, 115.24, 115.23, 53.07, 28.38, 19.19. HRMS (ESI⁺): m/z for C₂₂H₃₂N₂O₅Na [M+Na]⁺ calcd. 427.2203, found

427.2209.

COMPUTATIONAL STUDY

Reaction pathways

The electrophilic amination reaction has been studied with four different crotyl boronates (1aa, 1ca, G, and N) and 2a. The results are shown in Table S7 and visualized in Figure S4.

Table S7: Comparison between reaction pathways.



	1aa	100 100		Ν		G	
		ICa	<i>Re</i> face	Si face	<i>Re</i> face	Si face	
Pre	0.0	0.0	0.0	1.0	0.0	0.4	
TS	11.6	12.7	7.7	11.3	9.7	12.1	
Post	-40.5	-40.4	-36.1	-37.4	-37.9	-37.5	

Comparison between reaction pathways. Energies are in kcal/mol. Theory level: r2scan-3c d4/def2-mTZVPP-CPCM[e =12]//r2scan-3c d4/def2-QZVP-CPCM[e=12] calculated at standard condition.



Figure S4. 3D Geometries of the various Transition States (TSs) with their relative energies. Level of theory: r2scan-3c d4/def2-mTZVPP-CPCM[e=12]//r2scan-3c d4/def2-QZVP-CPCM[e=12] calculated at standard condition.

The reactions with achiral boronates **1aa** and **1ca** show a higher ΔG^{\dagger} than those with the chiral boronates **G-1ca** and **N-1ca**, this result is in accordance with the experimental observation that the background racemic reaction is slow compared with the catalytic ones. In perfect accord with the experimental results, DFT calculations predict the formation of the product with (*R*) stereochemical configuration when using boronates **G-1ca** and **N-1ca** with the (*S,S*) configuration of the catalyst. In both cases the enantiomeric ratio (e.r.) of the reaction is slightly overestimated: calculations predict an e.r. of 97.8:2.2 for **G** ($\Delta\Delta G = 2.4$ kcal/mol) and of 99.8:0.2 for **N** ($\Delta\Delta G = 3.6$ kcal/mol) while the experimental values is 96:4. for both **G** and **N**.

Relative position of iodine atoms in TS1-N-Re

In addition to the standard conformational analysis of the TS, an additional analysis of the possible arrangements of the iodine atom was conducted for the TS of **N**, as shown in Figure S5. The structures differ in the relative positions of the iodine atoms and have been simply named according to their position, where "in" refers to a position "inside" the aromatic pocket, and "out" outside of it.



Figure S5. Possible disposition of the iodine atoms and relative denomination. Representation of the same geometries but with different prospective view of the **TS1-N-***Re* geometry: upper line frontal view centred on the boron atom; bottom line perpendicular view emphasising the aromatic pocket (**2a** has been omitted for clarity).

All geometries of the possible conformations were optimized towards the TS. The results are compiled in Table S8.

Table S8. Relative energy of the four possible conformations of the iodine atoms for both the possible faces of **N**.

$\Delta\Delta G^{\dagger}$ [kcal/mol]	<i>Re</i> face	Si face
00	(142)	(166)
oi	(141)	N.C.
io	4.72	4.08
ii	0.00	3.58

Theory level: r2scan-3c d4/def2-mTZVPP-CPCM[e=12] at standard condition. N.C.: not converged.

The results for the "oo" and "oi" geometries are extremely high due to steric clashes between the iodine atoms and the reaction zone; therefore, they were not further refined. The favourite one is that with both the iodine atoms inside the aromatic pocket ("ii") for both the Re and Si approaches, and the further conformational search have been conducted one these geometries.

Analysis of TSs of catalyst N and G.

For the geometries of the catalytic reactions, the NCI surfaces have been plotted and are shown in Figure S6.



Figure S6. NCI analysis of the four catalytic TSs. Dashed blue lines represent a hydrogen bond and semitransparent bonds represent the TS bond. Surface color scheme: blue attractive; green London; and red repulsive interaction. Circle color scheme: orange interactions present in both the diasteromeric TSs; blue for the reaction site; and green interactions present only in one TS.

Main common interactions that can be spotted are: i) the hydrogen bond (HB) between the oxygen of the carbonyl and the β -vinylic proton of the crotyl, ii) the London interaction of the methyl moiety of the BOCs pointing toward the aryl part of the catalyst, and iii) iodine-iodine interaction inside the aryl pocket. Comparing TS1-N-Re and TS2-N-Si, there are three interactions which are present in both (shown in orange in the figure) and three additional interactions that are unique to the preferred TS1-N-Re (shown in green in the figure). These unique interactions include two hydrogen bonds and one London dispersion interaction between the remaining BOC group and the nearby aromatic ring. The hydrogen bond between the carbonyl oxygen and the ortho proton of the aryl group on the catalyst is particularly strong (d=2.30 Å). These interactions can explain the enantioselective preference for this TS and therefore for the formation of the (R) centre, as observed experimentally. The same analysis performed on TS1-G-Re and TS2-G-Si shows that, once again, the number of interactions is greater in the favoured TS, thus favouring the approach on the Re face of the crotyl. Finally, the comparison between the two preferred transition states (TS1-N-Re and TS1-G-Re) reveals that TS1-N-Re has an extra hydrogen bond, which is also the shortest present in the structures. This may explain why the ΔG^{\dagger} of the reaction with the iodine-containing catalyst is lower by 2 kcal/mol. In Table S9 are reported the selected geometrical parameters relevant to compare the TSs in Figure S6.

Table S9.	Selected	geometrical	parameter o	of the 4 as	ymmetrical	reactions
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	γ Ar-Ar	NBC	C-N [Å]	B-N [Å]	Pyramidalization B	Boc-l	Ph [Å]
TS1-N-Re	133°	93°	2.95	1.98	21.9°	2.97	3.17
TS2-N- <i>Si</i>	91°	95°	2.80	1.94	26.0°	> 4	3.12
TS1-G-Re	90°	92°	2.76	1.98	21.4°	3.20	2.68
TS2-G-Si	86°	95°	2.77	1.91	26.7°	> 4	2.99

Ar-Ar is the dihedral angle between the two aryl groups; NBC is the bite angle between the nitrogen, boron and allylic carbon atoms; C-N is the distance between the two reacting atoms; pyramidalization B represents the dihedral angle between O-O-C-B and helps understanding how much the boron atom is outside of the plane described from the two ester oxygens and the allylic carbon; Boc-Ph is the minimum distance between the hydrogens of the methyls of the Bocs and the nearest carbon atom of the phenyl ring.

From these parameters and from the distances highlighted in Figure S6, the favorable **TS1-N-***Re* shows three hydrogen bonds, one of these is a highly stabilizing interaction between the oxygen of the carbonyl moiety and the hydrogen of the C6 of the iodophenyl ring of the catalyst. This interaction produces the highly distorted γ Ar-Ar dihedral angle without disrupting the constructive interaction inside the aromatic pocket, with the aromatic ring oriented towards the methyl group thus reinforcing this London interaction between them. This kind of interaction was possible also in the **TS1-G-***Re*, but the absence of the iodine atoms does not allow the aromatic rings to move towards the *tert*-butyl group and the BOC's carbonyl group; in addition, the carbonyl oxygen is pointing towards the vinylic β -proton, so the so HB between them can't be present.

Orbitals analysis

From the steric data above described, the focus of this study shifted toward the understanding of the orbital's contribution and interactions. To do so, the projection of the molecular orbitals (MOs) of both the **TSs** for catalyst **G** (here labelled Ξ_i) on the two different fragments: borocrotyl (**G-1ca**) ϕ_i and **2a** θ_i . This analysis was performed assuming that ϕ_i and θ_i are orthonormalized.

$$\Xi_i = \sum_j^N a_{ij} \phi_j + \sum_j^N b_{ij} \theta_j$$
 (Eq. S1)

$$a_{ij} = \int_{a} \Xi_i^* \phi_j d\mathbf{r}$$
 (Eq. S2)

$$b_{ij} = \int \Xi_i^* \theta_i d\mathbf{r}$$
 (Eq. S3)

$$\sum_{j}^{N} a_{ij}^{2} + \sum_{j}^{N} b_{ij}^{2} = 1$$
 (Eq. S4)

Where *i* is the number of the orbitals selected to describe the TSs geometry (HOMO-1, HOMO, LUMO, LUMO+1), *j* is related to the orbital's number of the selected fragments, and *a* and *b* represent the coefficients to weight the fragments' orbitals and are required to be normalized for each TS's orbital. The orbitals of the two fragments are represented in Figure S7.



Figure S7. Orbital chosen for the projection analysis. a) Orbitals for the **Ts1-G**-*Re*. b) Orbitals for the **Ts2-G**-*Si*. List of the orbitals selected: Borocrotyl **G-1ca**: HOMO-1: p of C=C; HOMO: p_z of B; LUMO: p* of C=C and s* of C-B; LUMO+1: p* of G. Azodicarboxylate **2a**: HOMO-1: lone pair of N-atoms in **2a**; HOMO: p* of N=N; LUMO p* of C=O; LUMO+1 p* of C=O.

From the results of this analysis, it is evident that the composition of the MOs of the two TSs is substantially different (Figure S8). For the attack on the *Re* face, the two occupied orbitals of the TS (HOMO-1 and HOMO) are predominantly described by the HOMO and HOMO-1 of the **G-1ca** and by the HOMO-1, HOMO, and LUMO of the **2a**. On the other hand, the orbitals involved for the attack on the Si face are more mixed, making a clear separation difficult.



Figure S8. Projection diagram of the TS's MOs on the two different fragments with main characters highlighted.

Moreover, a second orbital analysis was conducted to find a reason that could rationalize the energy difference and thus the enantioselection of the catalyst. To do so, the overlap of the orbitals of the fragments as basis function was employed.

$$S_{ij} = \int \Phi_i^* \theta_j dr$$
(Eq. S5)
$$s_{ij} = \left(\frac{S_{ij}}{\max(S)}\right)^2 \times 100$$
(Eq. S6)

The results (Table S10) show a crucial difference in relative overlap: the maximum overlap is between θ_{HOMO-1} (so the lone pair of the nitrogen atoms) with the ϕ_{HOMO} (so the p_z of the boron atom).

Table S10. Relative overlap coefficient of the two diasteromeric TSs of G resulting from Eq. S6.

TS1-G-Re	HOMO-1	НОМО	LUMO	LUMO+1
HOMO-1	1.1%	26.0%	0.0%	2.7%
НОМО	100.0%	3.9%	7.2%	8.9%
LUMO	0.8%	1.1%	2.4%	5.2%
LUMO+1	4.6%	0.1%	1.0%	0.7%

TS1-G-Si	HOMO-1	НОМО	LUMO	LUMO+1
HOMO-1	0.27%	30.99%	0.78%	0.93%
НОМО	57.71%	1.42%	3.93%	3.21%
LUMO	2.16%	0.31%	1.48%	14.02%
LUMO+1	8.66%	0.59%	0.51%	0.71%

This interaction is maximum in the **TS1-G**-*Re* and only the 58% of that in **TS1-G**-*Si*. Another important interaction is the ϕ_{HOMO-1} (so the p system of the crotyl) with the θ_{HOMO} (so the π^* of **2a**) but the difference of this overlap is not so predominant as the one of before (26% in the **TS1-G**-*Re* against 31% of **TS2-G**-*Si*).

NMRs TRACES OF BORONIC ESTERS




















































































NOE













NOE





Similar ratio of the starting alcohol (1:2.07)







NOE









NMRs TRACES OF ALLYLIC HYDRAZIDES

All the ¹H NMR spectra display very broad signals due to the rotamers of the Boc groups. To further prove this well-known feature, one aliphatic and one aromatic compound were selected to be analysed by NMR at 23°C and at higher temperature. As you can see, at higher temperature the peaks are sharper and more resolved. A similar trend is also observed by ¹³C NMR where the signals rarely correspond to the number of the carbons of the compound, appearing broad and not intense although the high concentration of the sample.







Comparison of NMRs






















¹H NMR (600 MHz, CDCl₃) Вос Boc NH 3i 7.67-5.20 2.02 6.5 4.5 4.0 3.5 f1 (ppm) 3.0 2.5 2.0 1.5 7.5 7.0 6.0 5.5 5.0 1.0

141 141 133 133 105

0.5

0.0






















































DETERMINATION OF STEREOSELECTIVITY

Determination of d.r. of 3u and 3u' through ¹H NMR

3u and **3u'** were poorly separated by HPLC technique, therefore we opted to determine their diasteromeric excess through NMR analysis. Due to a complex combination of rotamers, we recorded the ¹H NMR spectra of **3u**, **3u'** and their mixture at 100°C in $C_2D_2Cl_4$ after a screening of temperature conducted on the mixture (**Figure S9**). The comparison of those spectra revealed the d.r. of each compound (for **3u** d.r.> 20:1; for **3u'** d.r.= 89:11).

Figure S9: screening of temperature for the mixture

6.5

6.0

5.5

4.5

5.0

4.0

3.5

3.0 f1 (ppm) 2.5

1.5

2.0

1.0

0.5

0.0





















Determination of e.r. and d.r. by GC-FID and HPLC

	Area Percent Report
Acq. Operator : Giovanni Acq. Instrument : 6890 GC Location : Vial 1 Injection Date : 17/06/2020 17:46:27 Inj : 1 Inj Volume : Manually	Sorted By : Signal Multiplier: : 1.0000 Dilution: : 1.0000 Use Multiplier & Dilution Factor with ISTDs
<pre>Acq. Method : C:\CHEM32\4\METHODS\100FID.M Last changed : 17/06/2020 13:20:58 by Glovanni (modified after loading) Analysis Method : C:\CHEM32\1\DATA\DEF_GC 2024-05-07 16-10-46\SYNLAB.M Last changed : 17/07/2024 15:20:40 by Synlab (modified after loading) Sample Info : 120 °C, 2 min - 2°C/min -> 230, 5 min plù conc</pre>	Signal 1: FID1 A, Peak RetTime Type Width Area Height Area [min] [min] [pA*s] [pA] %
F01 A (\$100#0ATA600001_00006.0)	
Boc N Boc N M rac-3a	Boc e

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Acq. Operator : Giovanni Acq. Instrument : HFLC-1 Location : Vial Injection Date : 16/03/2023 18:20:40 Acq. Method : C:\CHEM32\1\METHODS\DEF_LC.M Last changed : 16/03/2023 18:04:40 by Giovanni (modified after loading) Analysis Method : C:\CHEM32\1\METHODS\DEF_GC.M Sample Info : GC_G28_1.1 mL/min, 97.5:2.5 Hex:iPr, 25°C, AS-	Sorted By : Signal Multiplier: : 1.0000 Dilution: : 1.0000 Use Multiplier & Dilution Factor with ISTDs Signal 1: DADL C, Sig=210,8 Ref=360,100
	Peak RetTime Type Width Area Height Area # [min] [mal0*s] [mAU]



Totals :

2.03780e4 441.26964





N File S:\HPLC\l\DATA\GC\GC_687_1.D Dle Name: GC_687	Area Percent Report
Acq. Operator : Giovanni Acq. Instrument : 1100 Location : Vial 25 Injection Date : 08/01/2024 15:17:17	Sorted By : Signal Multiplier: : 1.0000 Dilution: : 1.0000 Use Multiplier & Dilution Factor with ISTDs
Acq. Method : C:\CHEM32\\\METHODS\NICCLO1100.M Last changed : 08/01/2024 15:07:56 by Giovanni (modified after loading) Analysis Method : C:\CHEM32\\\METHODS\DEF_GC.M Sample Info : GC_687, AS-H, 98:2 n-hexane/iPrOH, 0.5 mL/min, 25°C	Signal 1: DAD1 C, Sig=210,8 Ref=360,100 Peak RetTime Type Width Area Height Area # [min] [min] [mAU*s] [mAU] %
	1 9.651 MM 1.1110 768.12183 11.6229 51.76624 2 12.635 MM 1.2644 734.34106 9.67951 48.23376
	Totals : 1522.46289 21.50243









8.5



11.5

12

totals : 6925.03564 303.06015

10

10.5

11



9.5





0-

5













: Giovanni		
: HPLC-1	Location	: Vial 1
: 13/07/2023 16:32:57		
: C:\CHEM32\1\METHODS\GC.M		
: 13/07/2023 16:32:29 by Giovanni (modified after loading)		
: C:\CHEM32\1\METHODS\DEF GC.M		
: 31/12/2023 17:49:21		
	<pre>: Giovanni : HHC-1 : 13/07/2023 16:32:57 : C:\CHEM32\1\METHODS\GC.M : 13/07/2023 16:32:29 by Giovanni (modified after loading) : C:\CHEM32\1\METHODS\DE_gC.M : 13/1/2/2023 17:49:11</pre>	: Giovanni : HEC-1 Location : 13/07/2023 16:32:57 : C:\CHEM21\1METHODS\GC.M : 13/07/2023 16:32:29 by Giovanni (modified after Loading) : C:\CHEM32\1METHODS\DEF_GC.M : 13/1/2/2023 17:49:21

2 1 1 2 2					
Sorted By		Signal	N STORIGEN		
Multiplier:		:	1.0000		
Dilution:		:	1.0000		
Use Multiplier	r & Dilution	Factor with	ISTDS		
Signal 1: DAD1	1 C, Sig=210,	.8 Ref=360,:	00		
Signal 1: DAD1 Peak RetTime 7	1 C, Sig=210, Type Width	.8 Ref=360,:	00 Height	lrea	
Signal 1: DAD1 Peak RetTime 1 # [min]	1 C, Sig=210, Type Width [min]	.8 Ref=360,: Area [mAU*s]	00 Height [mAU]	Area %	
Signal 1: DAD] Peak RetTime 3 # [min] 	1 C, Sig=210, Type Width [min] 	.8 Ref=360,: Area [mAU*s]	00 Height [mAU]	Area %	
Signal 1: DAD1 Peak RetTime 1 # [min] - 1 13.990 V	1 C, Sig=210, Type Width [min] VV 0.3895	.8 Ref=360,: Area [mAU*s] 4357.65625	00 Height [mAU] 	Area % 85.96217	











	Area Percent Report
Acq. Operator : Giovanni Acq. Instrument : HPLC-1 Location : Vial 1 Injection Date : 02/05/2023 12:17:09 Acq. Method : C:\CHEM32\1\METHODS\DEF_LC.M Last changed : 02/05/2023 12:05:26 by Nunzio (modified after loading)	Sorted By : Signal Multiplier: : 1.0000 Dilution: : 1.0000 Use Multiplier & Dilution Factor with ISTDs Signal 1: DAD1 C, Sig=210,8 Ref=360,100
Analysis Method : C:\CHEM32\1\METHODS\DEF_GC.M Last changed : 31/12/2023 17:49:21 (modified after loading) Sample Info : GC_555, 1.0 mL/min, Hex:Ipr = 95:5, 23 °C, OD-H	Peak RetTime Type Width Area Height Area # [mail] [mail] [maV*] [maV] \$
	Totals : 1.68817e4 1450.16763

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ata File S:\HFLC\1\DATA\GC\GC_732_5.D ample Name: GC_732	Area Percent Report
Acq. Operator : Giovanni Acq. Instrument : 1100 Location : Vial 21	Sorted By : Signal Multiplier: : 1.0000 Dilution: : 1.0000 Use Multiplier 4 Dilution Factor with ISTDs
<pre>Injection Date : 2//10/2023 13:5:55 Acq. Method : c:\CleHM231\MuRTHOS\80_20_0.75ML_60MIN_254NM_COLUMN2.M Last changed : 27/10/2023 13:44:04 by Pietro (modified after loading) Analysis Method : c:\CleHM231\MuRTHOS\DEF_GC.M Sample Info : GC_732, lux Su cellulose-1, 97:3 n-hexane-iproh, 0.50 m</pre>	Signal 1: DADI C, Sig=210,8 Ref=360,100 Peak RetTime Type Width Area Height Area # [min] [min] [mAU*9] [mAU] %
L/min, 25°C	1 10.702 YV 0.7663 1.72310-4 299.02267 48.60895 2 12.458 VV 0.6739 1.82172-4 279.69751 51.39105

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an Onerator	· Giov	anni			Sorte	d By		Signal			
log. Tratrumont	. unto	-1	Togot	tion , Migl 1	Multi	plier:		:	1.0000		
Injection Date	: 17/0)7/2023 10:04:43	loca	.ion . viai i	Use M	ion: ultiplier & I	Dilution	: Factor wit	h ISTDs		
Acq. Method Last changed	: C:\C : 17/C (mod	HEM32\1\METHODS\GC.M)7/2023 10:03:19 by (lified after loading)) ;iovanni		Signa.	l 1: DAD1 C,	Sig=210,	8 Ref=360,	100		
Analysis Method	: C:\C	HEM32 \1\METHODS \DEF	GC.M		Peak	RetTime Type	Width	Area	Height	Area	
Last changed	: 31/1 (mod	2/2023 17:49:21 lified after loading)	-		#	[min]	[min]	[mAU*s]	[mAU]	* 	
Sample Info	: GC_6	593, 0.5 mL/min, Hex:	Ipr = 97:3, 3	25 °C, OD-H	1	9.119 VV 10.179 VV	0.2537	1.51406e4 1.53337e4	921.26459 832.18408	49.68320 50.31680	
					Total			3.04743e4	1753.44867		



	Area Percent Report
ata File S:\HFLC\1\DATA\GC\GC_692.D ample Name: GC_692	Sorted By : Signal Multiplier: : 1.0000 Dilution: : 1.0000 Use Multiplier & Dilution Factor with ISTDs
Acq. Operator : Glovanni Acq. Instrument : HPLC-1 Location : Vial 1 Injection Date : 17/07/2023 11:00:39 Acq. Method : C: VCHEN32\1MMETHODS\GC.M Last changed : 17/07/2023 10:54:44 by Giovanni // Undified affect logitor	Signal 1: DAD1 C, Sig=210,8 Ref=360,100 Peak RetTime Type Width Area Height Area # [min] [mAU*s] [mAU] %
Analysis Method : C:\CHEM3\.\MEHDOS\DF_GC.M Last changed : 31/12/2023 17:49:21 (modified after loading)	1 9.034 MM 0.2349 617.79425 43.83487 5.43073 2 9.916 MM 0.2822 1.07581e4 635.32642 94.56927
Sample Info : GC_692, 0.5 mL/min, Hex:Ipr = 97:3, 25 °C, OD-H	Totals : 1.13759e4 679.16128



	Area Percent Report
File S:\HFRC\2\DATA\GC\GC_528_1.D le Name: GC_828 Acq. Operator : Giovanni Acq. Instrument : 1200 Location : Vial 41 Unaction Data : 10/01/2024 17:42:50	Sorted By : Signal Multiplier: : 1.0000 Dilution: : 1.0000 Use Multiplier & Dilution Factor with ISTDs
Argen Make : 1000/2021 1:1:1:00 Inj Volume : 5.0 µl Acq. Method : C:\CHEM32\2\METHODS\HLW_IC3_ALLYLATION.M Last changed : 10/0/1/2024 17:42:31 by Giovanni (modified after loading) Analysis Method : C:\CHEM32\2\METHODS\DEF_LC.M Sample Info : GC_828, OD-H, HEX/ISOPROPANOL 97/3, 0.5 ml/min, 20°C	Signal 1: DADI D, Sig=230,16 Ref=360,100 Peak RetTime Type Width Area Height Area # [min] [mAU'a] [mAU'] %

















le Name: GC_619_3	Area Percent Report
Acq. Operator : Giovanni Acq. Instrument : HPLC: Injection Date : 06/03/2023 15:01:54 Location : Vial 1 Acq. Method : c:\cHEM9211\METHODS\DEF LC.M	Sorted By : Signal Multiplier: : 1.0000 Dilution: : 1.0000 Use Multiplier 4 Dilution Factor with ISTDs
Last changed : 06/03/2023 14:59:08 by Giovanni (modified after loading)	Signal 1: DAD1 D, Sig=228,16 Ref=360,100
Analysis Method : C:\CHEM32\1\METHODS\DEF_GC.M	Peak RetTime Type Width Area Height Area
Last changed : 31/12/2023 16:59:51 (modified after loading)	# [min] [min] [mAU*s] [mAU] %
Sample Info : GC_619_3, O.5 mL/min, 98:2 hex:ipr, 25°C, OD-H	1 11.338 VV 0.3385 1418.30688 65.01511 6.12952 2 12.142 VB 0.3718 2.17206e4 892.91467 93.87048
	Totals : 2.31389e4 957.92979



le Name: GC_671	_1	n de contra de contra en contra en contra en contra en de la contra en d En de la contra en de En de la contra en de la contra					
and Operator		Giovanni			==:		
Acq. Instrument	÷	HPLC-1	Loca	tion		Vial	1
njection Date	÷	11/07/2023 12:32:43					-
Acq. Method		C:\CHEM32\1\METHODS\GC.M					
ast changed	:	11/07/2023 12:30:46 by Giovanni (modified after loading)					
Analysis Method	:	C:\CHEM32\1\METHODS\DEF GC.M					
Sample Info		GC 671 1. 0.5 mL/min. Hex:Inr =	95:5	5. 25	۰,	C. AS	- H

Sorted By	: Signal			
Multiplier:	:	1.0000		
Dilution:	:	1.0000		
Use Multiplier & Signal 1: DAD1 A	Dilution Factor wit , Sig=254,4 Ref=360,	h ISTDs 100		
Use Multiplier & Signal 1: DAD1 A Peak RetTime Type	Dilution Factor wit , Sig=254,4 Ref=360, = Width Area	h ISTDs 100 Height	Area	
Use Multiplier & Signal 1: DAD1 A, Peak RetTime Type # [min]	Dilution Factor wit , Sig=254,4 Ref=360, e Width Area [min] [m&U*s]	h ISTDs 100 Height [mAU]	Area %	
Use Multiplier & Signal 1: DAD1 A Peak RetTime Type # [min] 	Dilution Factor wit , Sig=254,4 Ref=360, e Width Area [min] [mAU*s]	h ISTDs 100 Height [mAU]	Area %	
Use Multiplier & Signal 1: DAD1 A Peak RetTime Type # [min] 	Dilution Factor wit , Sig=254,4 Ref=360, = Width Area [min] [mAU*3] -	h ISTDs 100 Height [mAU] 9.6586	Area % - 7 50.12307	







		Area Percent
ata File S:\HPLC\1	DATA\GC\GC_731_4.D	Canhad Ba
ampie name: GC_/SI		Multinlier:
		Dilution: :
Acq. Operator	: Giovanni	Use Multiplier & Dilution Factor with
Acq. Instrument	: 1100 Location : Vial 21	
Injection Date	: 27/10/2023 14:49:16	Signal 1, DAD1 & Sig=254 4 Paf=360
Acq. Method	: C:\CHEM32\1\METHODS\80_20_0.75ML_60MIN_254NM_COLUMN2.M	Signal I: DEDI K, SIG-254,4 Rel-500,
Last changed	: 27/10/2023 14:45:40 by Giovanni	Peak RetTime Type Width Area
	(modified after loading)	# [min] [mAU*s]
Analysis Method	: C:\CHEM32\1\METHODS\DEF_GC.M	
Last changed	: 31/12/2023 17:49:21	1 16.700 BV 0.9406 2987.14990
	(modified after loading)	2 19.039 VB 1.0420 3019.16602
Sample Info	: GC_731, lux 5u cellulose-1, 97:3 n-hexane-iproh, 0.50 m	
	T/min 25°C	Totals : 6006.31592













Acq. Operator	: Giovanni	
Acq. Instrument	: 1100	Location : Vial 42
Injection Date	: 11/12/2023 10:58:44	
Acq. Method	: C:\CHEM32\1\METHODS\FRANC	ESCOPIRROLI.M
Last changed	: 11/12/2023 10:52:26 by Gi (modified after loading)	ovanni
Analysis Method	: C:\CHEM32\1\METHODS\DEF G	C. M
Sample Info	: GC 796, lux cellulose 1,	97:3 n-hexane/iPrOH, 0.5 mL/m
	n. 25°C	

mAU 1000









	Area Percent Report
Vata File S:\HFLC\1\DATA\GC\GC_762_3.D ample Name: GC_762	Sorted By : Signal Multiplier: : 1.0000 Dilution: : 1.0000
Acq. Operator : Giovanni Acq. Testrumat: 1100 Location : Vial 22	Use Multiplier & Dilution Pactor with ISTDs
Acq. Mst ummer 1100 Injection Date : 25/10/2023 12:36:23 Acq. Method : C:\CHEM32\L\METHODS\98_2_0.5ML_40MIN_COLUMN_2.M Last changed : 25/10/2023 12:35:25 by Giovanni (modified after Loading)	Signal 1: DAD1 C, Sig=210,0 Ref=360,100 Peak RetTime Type Width Area Height Area # [min] [min] [mAU*s] [mAU] \$
Analysis Method : C:\CHEM32\1\METHODS\DEF_GC.M Sample Info : GC_762, OD-H, 97:3 n-hexane-iproh, 0.5 mL/min, 25°C	1 7.960 MF 0.5172 1167.57690 37.62770 54.30025 2 9.172 FM 0.5340 982.64685 30.66655 45.69975
	J Totals : 2150.22375 68.29425

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para file Stimpleticitation(structure_fsu_c.p) Sample Name: GC_750 Acq. Operator : GLOVANNI Acq. Instrument : 1200 Injection Date : 13/10/2023 12:18:57 Inj Volume : 5.0 pl Acq. Method : C:\CHEM32\2\METHOS\IDFAZIDITETRACMIM Last changed : 13/10/2023 12:18:45 by GloVANNI (modified after loading) Analysis Method : C:\CHEM32\2\METHOS\ELSM Sample Info : GC_750, AS-H, Mex:ipr 97:3, 25*C, 0.5 mL/min









Acq. Operator	: Giovanni	
Acq. Instrument	: 1100	Location : Vial 5
Injection Date	: 18/12/2023 19:43:02	
Acq. Method	: C:\CHEM32\1\METHODS\N	ICOLO1100.M
Last changed	: 18/12/2023 19:38:19 b (modified after loadi	y Giovanni ng)
Analysis Method	: C:\CHEM32\1\METHODS\D	EF GC.M
Last changed	: 02/01/2024 10:44:36 (modified after loadi	ng)
Sample Info	: GC 819, ID-3, 98:2 n-	hexane/iPrOH, 0.5 mL/min, 25°C

			1	Area Percen	t Report		
Sort	nd By			Signal			
Mult:	iplier:			:	1.0000		
Dilu	tion:				1.0000		
Use 1	Multiplier	a Dil	ution	Factor wit	h ISTDs		
Sign	al 1: DAD	L C, Si	.g=210,	,8 Ref=360,	100		
Sign Peak	al 1: DAD RetTime (l C, Si Cype W	.g=210, Midth	,8 Ref=360, Area	100 Height	Area	
Sign Peak #	al 1: DAD RetTime ([min]	l c, si Cype W [.g=210, lidth [min]	8 Ref=360, Årea [mAU*s]	100 Height [mAU]	Area 8	
Sign Peak #	al 1: DAD RetTime ([min]	l C, Si Cype W [.g=210, Midth min]	8 Ref=360, Area [mAU*s]	100 Height [mAU]	Area 8	
Sign Peak # 1	RetTime ([min] - 18.573)	L C, Si Cype W [/V 0	.g=210, idth min] .3936	8 Ref=360, Area [mAU*s] 4.27146e4	100 Height [mAU] 1409.37256	Area 8 1 18.72880	
Sign Peak # 1 2	RetTime ([min] - 18.573 (19.164 (l C, Si Cype W /V O /V O	.g=210, min] .3936 .6813	8 Ref=360, Area [mAU*s] 4.27146e4 7.58204e4	Height [mAU] 1409.37256 1388.01245	Area 8 18.72880 33.24453	
Sign Peak # 1 2 3	RetTime ' [min] - 18.573 ' 19.164 ' 22.609 '	l C, Si Type W /V O /V O /V O	.g=210, min] .3936 .6813 .6172	<pre>A Ref=360, Area [mAU*s] 4.27146e4 7.58204e4 5.09520e4</pre>	Height [mAU] 1409.37256 1388.01245 1119.15625	Area 8 	
Sign # 1 2 3 4	RetTime ([min] 	L C, Si Cype W [/V 0 /V 0 /V 0 /B 0	.g=210, min] .3936 .6813 .6172 .9794	8 Ref=360, Area [mAU*s] -4.27146e4 7.58204e4 5.09520e4 5.85818e4	100 Height [mAU] 1409.37256 1388.01245 1119.15625 789.75067	Area 8 	







Acq. Operator	:	Giovanni				
Acq. Instrument	:	1100	Location	:	Vial	1
Injection Date	:	18/12/2023 17:56:35				
Acq. Method	:	C:\CHEM32\1\METHODS\NICOLO	100.M			
Last changed	:	18/12/2023 17:55:27 by Giov (modified after loading)	anni			
Analysis Method	:	C:\CHEM32\1\METHODS\DEF GC.	. M			
Last changed	:	02/01/2024 10:44:36 (modified after loading)				
Sample Info	:	CP_328_frac2, ID-3, 98:2 n-	hexane/iPrOH,	0.	.5 mL	/mir

		area Percent	Report	
Sorted By		Signal		
Multiplier:		:	1.0000	
Dilution:		: .	1.0000	
Use Multiplier &	Dilution	Factor with	n ISTDs	
Signal 1: DAD1 C	, Sig=210	,8 Ref=360,1	100	
Peak RetTime Typ	e Width	Area	Height	Area
# [min]	[min]	[mAU*s]	[mAU]	8
	-			
1 13.348 VV	0.3635	6149.44629	254.85426	20.45261
2 13.950 VB	0.4968	9092.26758	263.35168	30.24022
3 16.073 BV	0.5371	7298.11670	202.21936	24.27300
4 17.368 VB	0.6111	7526.97510	175.25293	25.03417
Totals :		3.00668e4	895.67824	







Acq. Operator		Arianna
Acq. Instrument	÷	1100 Location : Vial 3
Injection Date	:	11/27/2023 11:23:42 AM
		Inj Volume : 5.0 µl
Acq. Method	0	C:\CHEM32\1\METHODS\FRANCESCOPIRROLI.M
Last changed	:	11/27/2023 11:07:31 AM by Arianna (modified after loading)
Analysis Method	:	C:\CHEM32\1\METHODS\A95 B05 1.50ML 35MIN COL1.M
Last changed	:	7/1/2024 1:45:21 PM by MOT
Sample Info	R.	AG 91 10, IC, 97:3 n-hexane/iPrOH, 1.0 mL/min, 25

60-

3x

Peak	RetTime Type	Width	Area	Height	Area
#	[min]	[min]	[mAU*s]	[mAU]	90
1	31.106 BV	0.7064	2874.99854	55.67829	49.5077
2	33.306 VB	1.0111	2932.17407	34.10155	50.4923







Acq. Operator	:	Arianna	
Acq. Instrument	:	1100	Location : Vial 6
Injection Date		27/11/2023 13:34:51	
Acq. Method	:	C:\CHEM32\1\METHODS\FRANCESC	OPIRROLI.M
Last changed	:	27/11/2023 13:33:38 by Arian (modified after loading)	na
Analysis Method	:	C:\CHEM32\1\METHODS\DEF GC.M	
Last changed	:	02/01/2024 10:44:36 (modified after loading)	
Sample Info	:	AG 84 8, IA-3, 90:10 n-hexan	e/iPrOH, 0.5 mL/min, 25

	A	rea Percent	Report		
					====
Sorted By	:	Signal			
Multiplier:			1.0000		
Dilution:			1.0000		
Use Multiplier & 1)ilution	Factor with	ISTDS		
Signal 1: DAD1 C,	Sig=210,	8 Ref=360,1	00		
Signal 1: DAD1 C, Peak RetTime Type	Sig=210, Width	8 Ref=360,1 Area	00 Height	Area	
Signal 1: DAD1 C, Peak RetTime Type # [min]	Sig=210, Width [min]	8 Ref=360,1 Area [mAU*s]	00 Height [mAU]	Area %	
Signal 1: DAD1 C, Peak RetTime Type # [min] 	Sig=210, Width [min]	8 Ref=360,1 Area [mAU*s] 	00 Height [mAU]	Area %	
Signal 1: DAD1 C, Peak RetTime Type # [min] 	Sig=210, Width [min] 0.6510	8 Ref=360,1 Area [mAU*s] 593.16095	00 Height [mAU] 10.83033	Area 8 55.19302	
Signal 1: DAD1 C, Peak RetTime Type # [min] 1 21.223 BB 2 28.128 BB	Sig=210, Width [min] 0.6510 0.7258	8 Ref=360,1 Area [mAU*s] 593.16095 481.54196	00 Height [mAU] 10.83033 7.80531	Area % 55.19302 44.80698	



le Na	: S:\HPLC\1' ame: AG_95_:	1	ATA\AG\AG_95_1.D
Acq.	Operator	:	arianna
Acq.	Instrument	:	1100 Location : Vial 6
Injed	tion Date	:	29/11/2023 15:46:55
Acq.	Method	:	C:\CHEM32\1\METHODS\MICHELE LAB COL2-IA3 ETOH IPROH
Last	changed	:	29/11/2023 15:45:07 by arianna (modified after loading)
Analysis Method :		:	C:\CHEM32\1\METHODS\DEF GC.M
Last	changed	:	02/01/2024 10:44:36 (modified after loading)
Samp	Le Info	:	AG 95 1, 90:10 n-hexane/iPrOH, 0.5 mL/min, 25°C





	Area Percent Report				
	Sorted By : Signal				
	Multiplier: : 1.0000				
	Dilution: : 1.0000				
a File S:\HPLC\1\DATA\AG\CP 312 1.D	Use Multiplier & Dilution Factor with ISTDs				
Acq. Operator : Arianna Acq. Instrument : 1100 Location : Vial 21 Ferencies - 122/1/2022 12:10:20	Signal 1: DAD1 C, Sig=210,8 Ref=360,100 Peak RetTime Type Width Area Height Area				
Injection bate : 23/11/2023 13:10:35	W Landard Landard Land The 1				
Acq. Method : C:\CHEM32\2\METHODS\PROVA.M	# [min] [min] [mAU*s] [mAU] %				
Anjetizali jede : 25/11/2023 / 35/10/39 Acq. Method : c:\cHEN322\METHOS\PROVA.M Last changed : 23/11/2023 13:08:39 by Arianna (modified after Loading)	# [min] [min] [mkU*s] [mkU] % 1 10.971 BV 0.3217 1283.05078 58.08234 17.97810				
Angevenho de : C:\CEN3212\METROS\$PROVA.M Last changed : C:\CEN3212\METROS\$PROVA.M [modified after Loading] Analysis Method : C:\CEN321\METROS\DEF_GC.M	# [ma.1] [ma.1] [ma.1] 8				
Acq. Method : c:\CHBM3212\MethodsNEWOA.M Last changed : 23/11/2023 13:06:39 by Ariannn [modified after loading] Analyzis Method: c:\CHBM21\MethodsNEF_GC.M Last changed : 02/01/2024 10:44:36 [modified after loading]	# [ma.1] [ma.1] [ma.0] 8 1 10.971 BW 0.3217 1283.05078 58.08224 17.97810 2 11.600 WB 0.3966 1680.66030 61.44754 23.66178 3 17.194 BB 0.5855 2076.14844 48.86218 29.05098 4 10.272 DP 6.258 2000.04204 40.4000.02204				
Injetuchi back : 15/17/205/15/15/5 Acq. Method : C:\CEMP212\METHODS\PPOVA.M Last changed : 23/11/2023 13:08:39 by Arianna (modified after loading) Analysis Method : C:\CEMP32\1\METHODS\DEF_GC.M Last changed : 02/01/2024 10:44:15 (modified after loading) Sample Info : CP_312_1, AI, 90:10 n-hexame/iPrOM, 0.5mL/min, 25°C	# [min] [min] [mk0F*] [mk0J] 8				



Area Percent Report

Data File S:\HFLC\1\DATA\AG\GC_780.D Sample Name: GC_780	Sorted By Multiplier: Dilution: Use Multiplier & D	: Dilution	Signal : : Factor with	1.0000 1.0000 1STDs	
Acq. Operator : arianna Acq. Instrument : 1100 Location : Vial 21	Signal 1: DAD1 C, Sig=210,8 Ref=360,100				
Injection Date : 29/11/2023 16:58:38 Acq. Method : C:\CHEM321\1\METHODS\MICHELE_LAB_COL2-IA3_ETOH_IPROH.M Last changed : 29/11/2023 16:57:35 by arianna (modified after loading)	Peak RetTime Type # [min] 	Width [min]	Area [mAU*s] 64.03806	Height [mAU]	Area 8
Analysis Method : C:\CHEM32\1\METHOR\DEF_GC.M Last changed : 02/01/2024 10:44:36 (modified after loading)	2 11.535 VB 3 17.025 BB	0.3508	1873.87915 863.98370	76.76527 21.49986	66.87885 30.83563
Sample Info : GC_780, IA, 90:10 n-hexane/iPrOH, 0.5 mL/min, 25°C	Totals :		2801.90092	101.93259	





MECHANISM INVESTIGATION

To investigate the role of isopropanol and acetic acid and to gain a better insight into the mechanism, different experiments were conducted in CDCl₃ and checked by ¹H NMR and ¹¹B NMR. The NMRs of the main compounds involved are reported below before further discussions.

Compound N-1ca





S178



Compound 3a




Compound 1ca





NMRs EXPERIMENT n1

This experiment has been designed to study the reaction between the catalytically active species **N**-**1ca** and the azodicarboxylate **2a**.

<u>Conditions</u>: 40 mg of N-1ca (0.07 mmol), 0.65 mL of CDCl₃ and 34.5 mg (0.15 mmol) of di-tertbutylazodicarboxylate were added to a screw cap vial provided of a magnetic stir bar. The reaction was stirred at 25°C and occasionally transferred into an NMR tube to be monitored.



30 minutes:



This spectrum recorded after 30 minutes evidence the formation of a new compound which does not correspond to the expected product. To help in visualizing the new signals, those of the starting (**N-1ca** and **2a**) have been marked with coloured dots. We speculated that the new compound corresponds to the adduct **N-3a**, being the reasonable product of the allylation occurring through the Zimmerman-Traxler transition state. Interestingly, the multitude of the peaks may be ascribable to a restricted rotation around the N-N bond. This feature is also observable by ¹¹B NMR spectrum:



4 hours:

After 4 hours the catalytically active specie is almost fully consumed. A rough integration of the peaks corresponding to the signals of the intermediate is shown.



Because of the formation of such adduct **N-3a**, the catalyst **N** as well the product **3a** cannot be detected in the reaction mixture:





24 hours:

After 24 hours **N-1ca** is no longer present and only the clean intermediate **N-3a** with the remaining excess of the **2a** can be detected.





The reaction mixture was split into two parts of equal volume which were diluted with 0.3 mL of $CDCl_3$ each. 24 mg (0.05 mmol, 1.57 eq.) of **N** were added to the first portion (**NMRs experiment**

1.1), while 7 μ L of isopropanol and 7 μ L of acetic acid were added to the second solution (**NMRs** experiment 1.2).

NMRs EXPERIMENT n1.1



The NMRs show the signals of **3a** due to the hydrolysis of the B-N bond of the adduct **N-3a**, but not those belonging to **N**. This means that the B-N is particularly sensitive and in the regular reaction conditions even the free catalyst may provoke its hydrolysis, while the B-O bond appears to be more robust.





The integration of the signals belonging to **3a** is shown below:



Beside the formation of the product, another compound is detected which is believed to be something similar to **13**.







NMRs EXPERIMENT n1.2

The addition of iPrOH and AcOH cause the hydrolysis of the B-N bond and the formation of **3a**. Even in this case **N** seems to be still involved with the boron atom.





The side-product of the hydrolysis involving the catalyst seems to be different from the adduct **13** detected in the **NMRs experiment 1.1**.





In conclusion, these two experiments evidence the easiness toward the hydrolysis of the intermediate **N-3a** detected in experiment 1 and may further provide an explanation on why this specie has never been detected during the regular reaction conditions. The e.r. of the isolated product: 96:4.

NMRs EXPERIMENT n2

This experiment aims to reproduce the reaction between the starting materials **1ca** and **2a** in the presence of **N** but without iPrOH and AcOH.

<u>Conditions</u>: In a screw cap vial provided of a magnetic stir bar 25.5 mg of **1ca** (0.15 mmol), 0.65 mL of CDCl₃ and 7 mg of **N** (0.015 mmol) were added. The reaction was stirred for 10 minutes, then a ¹H NMR of the mixture was recorded.



After 10 minutes 7.8% of the starting amount of **N** is converted into **N-1c**.

Afterward, the solution was transferred into a screw cap vial and 34.5 mg (0.15 mmol) of **2a** were added and the mixture was stirred at room temperature.



NMRs were recorded after 30 minutes, 4 hours and 24 hours. The concentration of the **N-1c** gradually increases over 4 hours, but the ¹H NMR at this time also shows new aromatic signals. Interestingly, after 24 hours a complexity of signals are visible in the aromatic region and there is no clear evidence of the presence either of **N** or **N-1c**. Moreover, these signals do not match with the suggested compounds **13-14** detected in **NMR experiments 1.1** and **1.2**. The percentages refer to the initial amount of **N**.











Since the unknown signals detected by ¹H NMR are close to the signals of **N** and those of **12** and according to ¹¹B NMR where multiple peaks between 19 and 15 ppm are present, we speculated that N and 12 are somehow bonded to the boron atom. It seems that within these conditions there is no efficient way to release the catalyst. Thus, as confirmed by the e.r of the isolated product after 24 hours (83:17), also the background reaction could have a relevant contribution on the overall conversion. The percentage of N-1ca refers to the initial amount of N.



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NMRs EXPERIMENT n2.1

To better study the disappearance of **N** along with **N-1ca** during the reaction we run the following experiment. 25.5 mg of **1ca** (0.15 mmol) were dissolved into a solution of 0.65 mL of $CDCl_3$ containing 7 mg of **N** (0.015 mmol). The mixture was transferred into an NMR tube and an ¹H NMR spectrum was recorded every 20 minutes:





After 13h we observed a well-established equilibrium, so we added 34.5 mg (1.0 eq.) of **2a** and we recorded an ¹H NMR every 20 minutes:

The gradual loss of the catalytically active species **N-1ca** during the reaction coordinate would be accountable for the inefficient regeneration the free catalyst **N**.

NMRs EXPERIMENT n3

This experiment aims to understand the role of the isopropanol.

<u>Conditions</u>: In a screw cap vial provided of a magnetic stir bar 25.5 mg of **1ca** (0.15 mmol), 0.65 mL of CDCl₃,7 mg of **N** (0.015 mmol) and 23 μ L of isopropanol (0.3 mmol, 2 eq.) were added. The reaction was stirred for 10 minutes, then the ¹H NMR of the mixture was recorded.



Then, 34.5 mg (0.15 mmol) of **2a** were added to the reaction mixture and the stirring was continued at rt.



The equilibrium toward the formation of **N-1ca** seems to occur quicker compared to the experiment n2 and **N-1ca** accumulates over time. After 24 hours only few new peaks are observed in the aromatic region and both the catalytically active species **N-1ca** and the catalyst **N** are clearly visible.



The signal of isopropanol partially overlaps with the peaks corresponding to the CH_2 (pink) of the ligand of **1ca**. After 24 hours in that region another signal appears that seems to be similar to the CH_2 of the ligand, while also the multiplet of the isopropanol shifts to 4.30 ppm. Therefore, the 3-methylbutane-1,3-diol **12**, unlike the catalyst, seems to be almost totally bonded with the boron atom together with some molecules of isopropanol.



The same trend is visible in the most shielded area of the spectra. After 24h, each signal of **12** is duplicated as well as those belonging to the isopropanol.





In conclusion, the main effect of the isopropanol seems to be the releasing of **N**, beside favouring the formation of a species which involves **12** with the boron atom. This is also beneficial toward the formation of **N-1ca** which accumulates over time. The e.r of the isolated product is 96:4. The percentage of **N-1ca** refers to the initial amount of **N**.

Time (h)	N-1ca	3a
0.17	14	/
0.67	36	3.8
3	43	8
24	61	56.6



NMRs EXPERIMENT n4

This experiment aims to understand the role of the acetic acid.

<u>Conditions</u>: In a screw cap vial provided of a magnetic stir bar were added 25.5 mg of **1ca** (0.15 mmol), 0.65 mL of $CDCl_3$, 7 mg of **N** (0.015 mmol) and 17 µL of acetic acid (0.3 mmol). The reaction was stirred for 10 minutes, then a ¹H NMR of the mixture was recorded.



Then 34.5 mg (0.15 mmol) of **2a** were added to the reaction mixture and the stirring was continued at 25°C.



Unlike the previous cases, **N-1ca** is already the 85% of the starting **N** within 10 minutes. Nevertheless, this is the first case where its concentration constantly decreases during the reaction coordinate. After 4 hours the aromatic region is already contaminated by a co-product and even the signals of the catalyst slightly shift, like the case of **NMRs experiments n2**, which may indicate that no free catalyst is anymore available in the solution. This strongly affect the formation of **N-1ca** that is only 30% after 24 hours.







Indeed, these spectra share the same co-products of those detected in the **NMRs experiments n2**: there is the same singlet at 5.55 ppm and between 4.15 ppm and 3-37 ppm a kind of similar multiple signals arises over 24 hours.







In contrast with **NMRs experiments n2**, the ¹¹B NMR after 24 hours shows a kind of convergence of different co-products into a predominant one. This might be due to the difference in conversion between the two experiments.



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In conclusion, this experiment clarified the role of the acetic acid in favouring the formation **N-1ca** and strongly evidenced that the isopropanol is strictly necessary to release the catalyst from the coproducts. The conversion is comparable to that of **NMRs experiment n3** and the e.r. of the isolated product is 96:4, thus the background reaction does not affect the conversion. The percentage of **N-1ca** refers to the starting amount of N

Time (h)	N-1ca	3a
0.17	84.7	/
0.67	82.6	6.5
4	65	23
24	30	50



NMRs EXPERIMENT n5

This experiment was design to study the effect of the combination of isopropanol and acetic acid.

<u>Conditions</u>: In a screw cap vial provided of a magnetic stir bar 25.5 mg of **1ca** (0.15 mmol), 7 mg of the catalyst (0.015 mmol), 23 μ L of iPrOH (0.3 mmol) and 17 μ L of acetic acid (0.3 mmol) were added to 0.65 mL of CDCl₃. The reaction was stirred for 10 minutes, then a ¹H NMR of the mixture was recorded.



Then 34.5 mg (0.15 mmol) of **2a** were added to the reaction mixture and the stirring was continued at 25°C.



Within 10 minutes **N-1ca** is already 86% of the starting amount of the catalyst **N**, as in the case of **NMRs experiment n4** (2 eq. of acetic acid); but unlike that case its concentration is significantly lowered only after 24 hours, when 66% of **1ca** has been consumed. Interestingly, even in this case it seems that the free catalyst is no longer present in the solution after 24 hours; nevertheless, the reaction would proceed thanks to **N-1ca** which is still abundant.



The following spectra shows the same co-products observed when 2 eq. of iPrOH and 2 eq. of AcOH have been used independently.










In conclusion, using a mixture of acetic acid and the isopropanol has the beneficial effects exhibited by those experiments in which they were used separately. Indeed, the concentration of **N-1ca** maintains high thanks to acetic acid while isopropanol accelerates the turnover of the catalyst. At the end, the conversion is higher than **NMRs experiments n3** and **n4**. The product was collected with an e.r. of 96:4. The percentage of **N-1ca** refers to the initial amount of **N**.

Time (h)	N-1ca	3a (%)	
0.17	86	/	
0.67	85	6.9	
4	83	24	
24	76	66	



Comparative study for the reaction of boronates 1aa, 1da and 1ca

These experiments were design to compare the reactivity of boronates **1aa**, **1da** and **1ca** in the ligand exchange process with **N** to generate **N-1ca** and their reactivity with **2a**. Indeed, for each boronate the reaction with azodicarboxylate **2a** was studied without catalyst **N** and the results obtained have been compared with the catalytic reaction to clarify the effect of a background reaction.

First experiment

<u>**Conditions</u>**: In a screw cap vial provided of a magnetic stir bar boronate (0.15 mmol), 7 mg of the catalyst (0.015 mmol), 23 μ L of iPrOH (0.3 mmol) and 17 μ L of acetic acid (0.3 mmol) were added to 0.65 mL of CDCl₃. The reaction was stirred for 10 minutes, then a ¹H NMR of the mixture was recorded. Then 34.5 mg (0.15 mmol) of **2a** were added to the reaction mixture and the stirring was continued at 25°C. The conversion of **3a** and the concentration of **N-1ca** was monitored during time.</u>



Time (h)	[N1ca] %		3a (%)			
	From 1da	From 1aa	From 1ca	From 1da	From 1aa	From 1ca
0,17	9	8,8	8,6	0	0	0
0,67	9,2	8,2	8,5	15	9	6,9
4	9	8,2	8,3	37	24	24
24	9	7	7,6	80	64	66



Within 10 minutes the concentration of **N-1ca** for the reaction of boronate **1aa** and **1da** is almost identical to the one of boronate **1ca**. The concentration of the catalytic species maintains constant during the reaction with **2a**. Boronate **1da** has the fastest kinetic profile while **1aa** and **1ca** show a similar trend even if at the beginning of the reaction with azodicarboxylate **1aa** produces **3a** in a larger amount than **1ca**. The e.r. for each reaction has been determined after 24 hours: 77:23 with **1da**, 92.5:7.5 with **1aa** and 96:4 with **1ca**.

Second experiment

<u>**Conditions</u>**: In a screw cap vial provided of a magnetic stir bar boronate (0.15 mmol), 23 μ L of iPrOH (0.3 mmol) and 17 μ L of acetic acid (0.3 mmol) were added to 0.65 mL of CDCl₃ then 34.5 mg (0.15 mmol) of **2a** were added to the reaction mixture and the stirring was continued at 25°C. The conversion of **3a** was monitored by ¹H NMR at the indicated time. The results obtained have been compared with those of catalytic reactions.</u>





The results obtained clearly indicates how the background reaction contribute to convert boronates into allylic hydrazide for **1aa** and **1da** in a greater amount than what is observed for **1ca**. Despite the catalytic reaction is faster than the racemic one, if no background reaction took place the same value of e.r. should be obtained for each of catalytic reactions. This is because the amount of **N-1ca**

is always the same and it has been clearly established that when **N-1ca** reacts with **2a** in a stoichiometric reaction the e.r. is 96:4.

NONLINEAR EFFECT

Procedure: solutions of the catalyst with different e.e. were prepared by mixing solutions of isopropanol containing **N** or **ent-N** in appropriate proportions. To this mixtures 0.1 mL of AcOH and 0.1 mmol of **1ca** were added and the solution was stirred for 10 minutes. Afterward, **2a** was introduced to the vessel and the reaction proceeded under stirring for 24 hours. The crude was diluted with hexane and flash chromatography was performed to isolate the product. GC-FID analysis provided the *e.e.* of the pure **3a**.





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