

# Supporting Information

## X-ray induced modification of the photophysical properties of MAPbBr<sub>3</sub> single crystals

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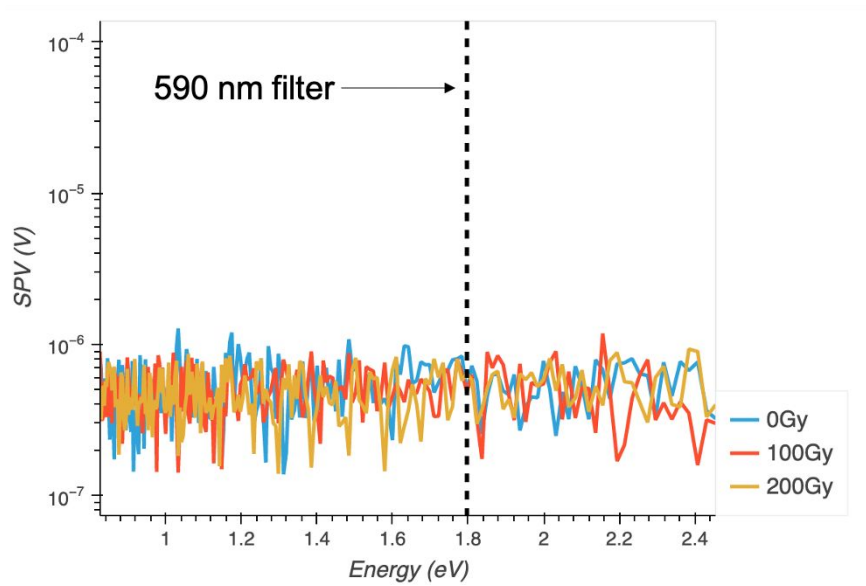
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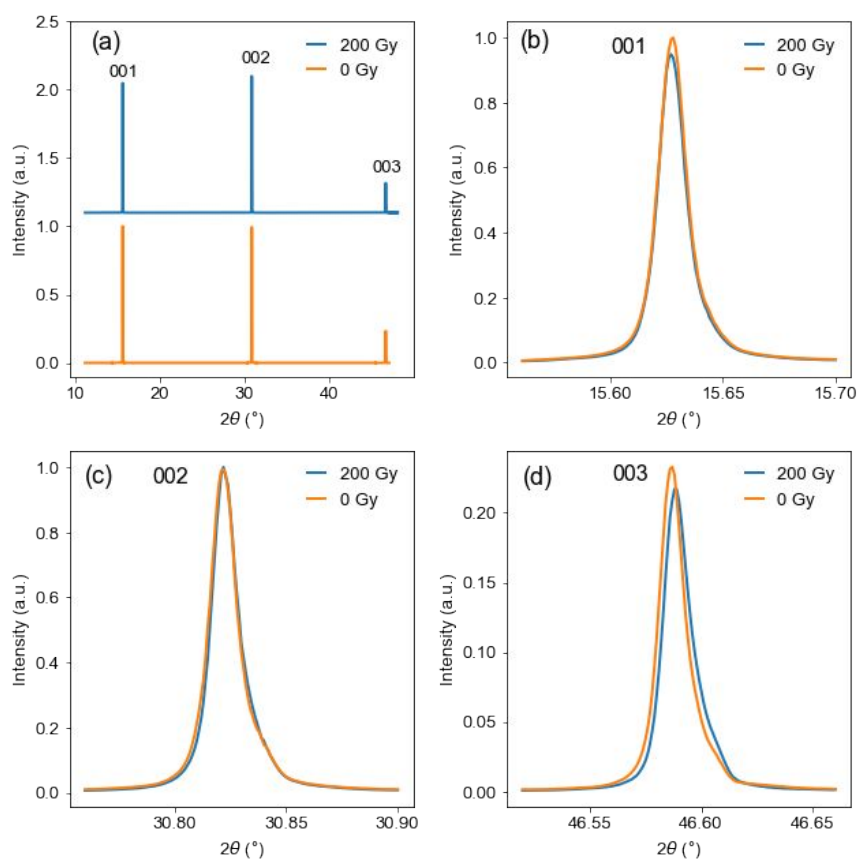
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## SPS spectra in the below-gap region



**Figure S1:** SPS spectra in the below-gap region acquired on a pristine sample (light blue curve) and after 100 Gy (red curve), and 200 Gy (yellow curve) irradiation. The vertical dashed line indicates the cutoff wavelength of the 590 nm long-pass filter. Within the sensitivity of the setup, no deep states were detected after irradiation.

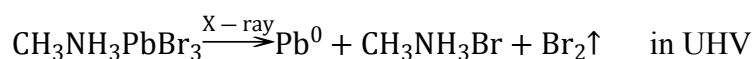
## High resolution X-ray diffraction (HR-XRD) on MAPbBr<sub>3</sub> single crystal before and after X-ray irradiation



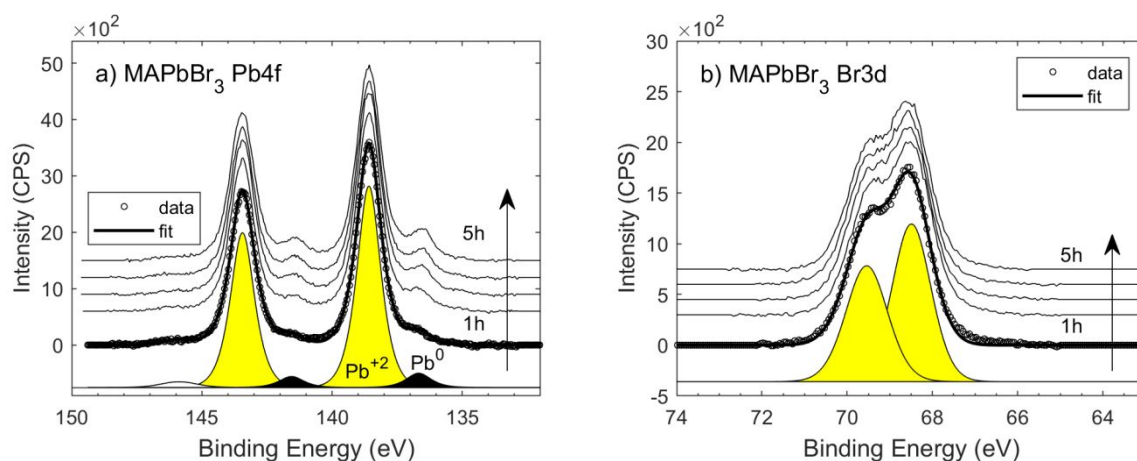
**Figure S2:** (a) High resolution X-ray diffraction spectrum of a MAPbBr<sub>3</sub> single crystal before (orange curve) and after (blue curve) 200 Gy irradiation by W target X-ray tube. Detail of 001 (b), 002 (c) and 003 (d) diffraction peaks.

## Description and results of the XPS calibration experiment

In this experiment we acquired 5 consecutive XPS spectra (duration 1h) on a pristine sample to quantify the effects of the XPS source on the crystals. We observed the appearance of a metallic lead ( $\text{Pb}^0$ ) component in the spectrum as well as a decrease in in the Br/Pb ratio, in agreement with previous studies.<sup>1,2</sup> The N/Pb was unchanged during the experiment. These results fit well with the degradation mechanism proposed by Wang et al.<sup>1</sup> under X-rays and in UHV:



After the first XPS scan the  $\text{Pb}^0/\text{Pb}$  ratio is approximately 7%. Any metallic  $\text{Pb}^0$  detected below this limit could be due to the XPS source itself rather than to other factors

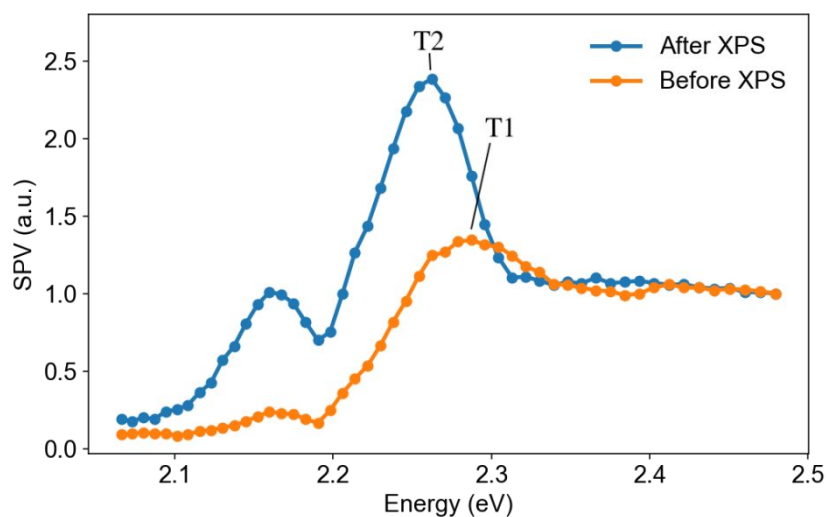


**Figure S3.** Time evolution during XPS calibration experiment to quantify the damage of the XPS beam on the crystals. Pb 4f and  $\text{Pb}^0$  (a), and Br 3d doublet (b) are reported. The fitting curves are displayed in yellow for  $\text{Pb}^{2+}$  and Br 3d, and in black for  $\text{Pb}^0$ . An offset was added to the curves for clarity purpose.

Time	$\text{Pb}^{2+} + \text{Pb}^0$	Br	N	$\text{Pb}^0/\text{Pb}$	O/Pb
1h	1	$2.8 \pm 0.1$	$1.2 \pm 0.1$	$0.067 \pm 0.009$	$0.04 \pm 0.02$
2h	1	$2.7 \pm 0.1$	$1.2 \pm 0.1$	$0.102 \pm 0.009$	$0.04 \pm 0.02$
3h	1	$2.6 \pm 0.1$	$1.1 \pm 0.1$	$0.127 \pm 0.009$	$0.035 \pm 0.02$
4h	1	$2.6 \pm 0.1$	$1.2 \pm 0.1$	$0.127 \pm 0.009$	$0.03 \pm 0.02$
5h	1	$2.6 \pm 0.1$	$1.2 \pm 0.1$	$0.140 \pm 0.009$	$0.05 \pm 0.02$

**Table S1.** Elemental ratios in the MAPbBr<sub>3</sub> single crystal during XPS calibration experiment.

All values are normalized to total Pb = Pb<sup>0</sup> + Pb<sup>+2</sup> (column 2).



**Figure S4.** SPV spectra of the crystal used for the calibration experiment (orange line) and after (blue line) 4 repeated XPS scans of 1 hour. This shows that quenching of the T1 peak and the appearance of the T2 peak is observed also after irradiation in UHV conditions.

**Elemental ratios as measured by XPS on samples irradiated with W target X-ray source in air**

Dose (Gy)	Pb	Br	N	Pb <sup>0</sup> /Pb (%)	O/Pb (%)
0	1.0	2.7 ±0.1	1.20 ±0.05	6.2	6.0
60	1.0	2.6 ±0.1	1.10 ±0.05	5.8	14.0
120	1.0	2.5 ±0.1	1.00 ±0.05	6.8	29.0
0 (7 days)	1.0	2.4 ±0.1	0.81 ±0.05	6.8	41.0
60 (7 days)	1.0	2.2 ±0.1	0.48 ±0.05	4.7	42.0
120 (7 days)	1.0	2.2 ±0.1	0.52 ±0.05	4.7	51.0

**Table S2.** Columns 2-4: relative concentration of Pb, Br and N for samples probed by XPS just after irradiation and after one week. The Pb signal includes both the Pb<sup>0</sup> and the Pb<sup>2+</sup> contributions. Columns 5-6: percentual contribution of Pb<sup>0</sup> to the total Pb signal and percentual ratio of oxygen amount with respect to Pb, respectively.

**References**

- (1) Wang, C.; Ecker, B. R.; Wei, H.; Huang, J.; Gao, Y. Environmental Surface Stability of the MAPbBr<sub>3</sub> Single Crystal. *J. Phys. Chem. C* **2018**, *122* (6), 3513–3522. <https://doi.org/10.1021/acs.jpcc.7b12740>.
- (2) Sadoughi, G.; Starr, D. E.; Handick, E.; Stranks, S. D.; Gorgoi, M.; Wilks, R. G.; Bär, M.; Snaith, H. J. Observation and Mediation of the Presence of Metallic Lead in Organic-Inorganic Perovskite Films. *ACS Appl. Mater. Interfaces* **2015**, *7* (24), 13440–13444. <https://doi.org/10.1021/acsami.5b02237>.