

## Supporting Information for

# Synthesis, molecular structure and fluxional behavior of the elusive $[\text{HRu}_4(\text{CO})_{12}]^{3-}$ carbonyl anion

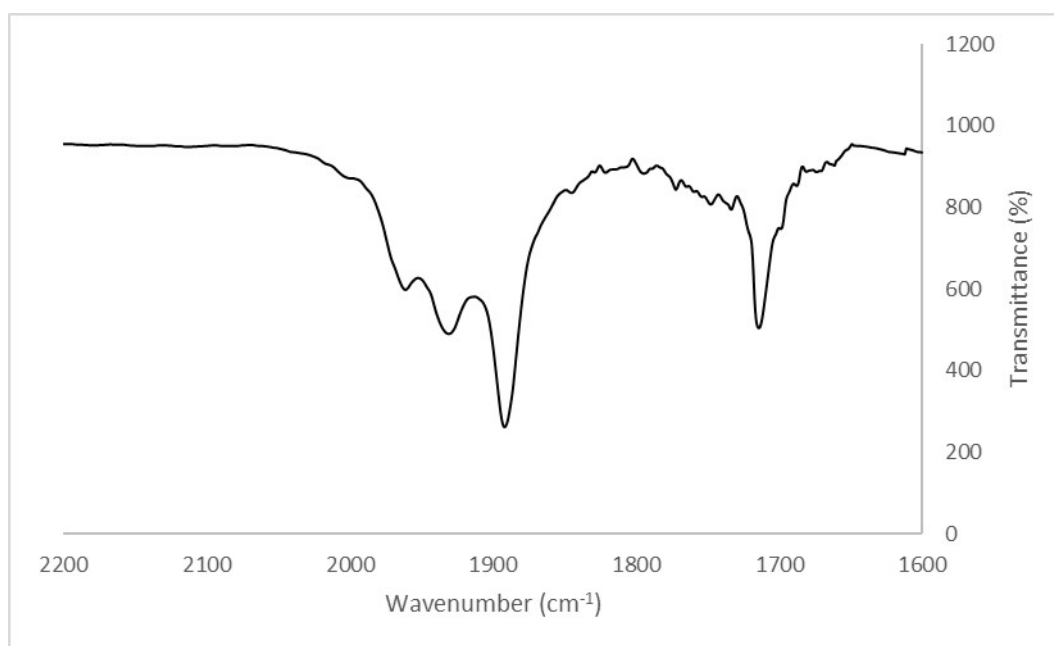
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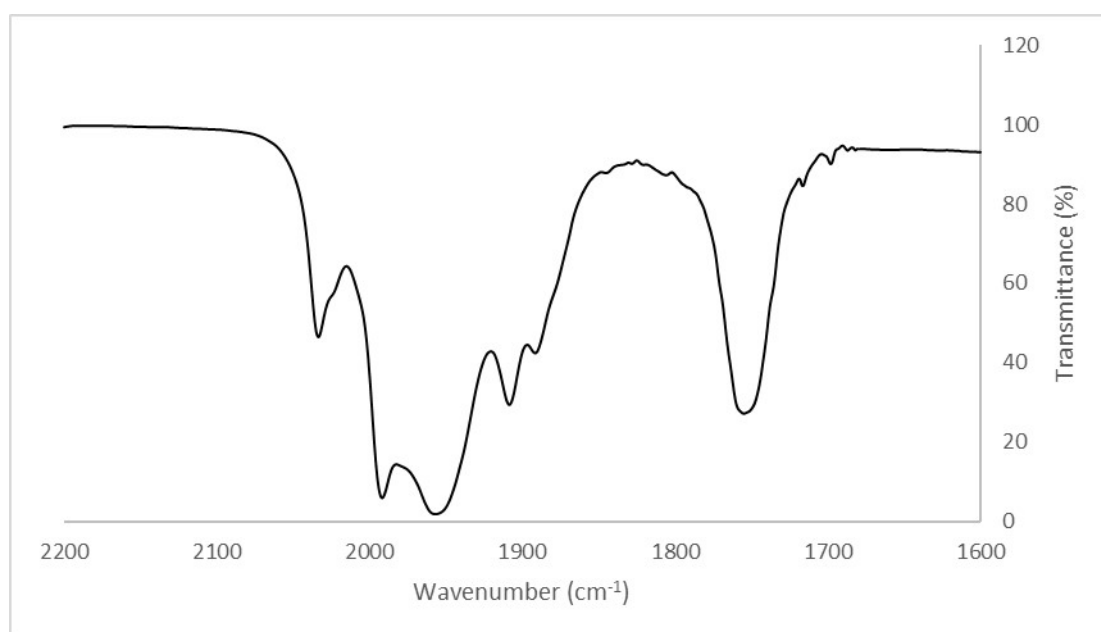
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**Table S1.** Structurally characterized hydridocarbonylates of Fe, Ru and Os.

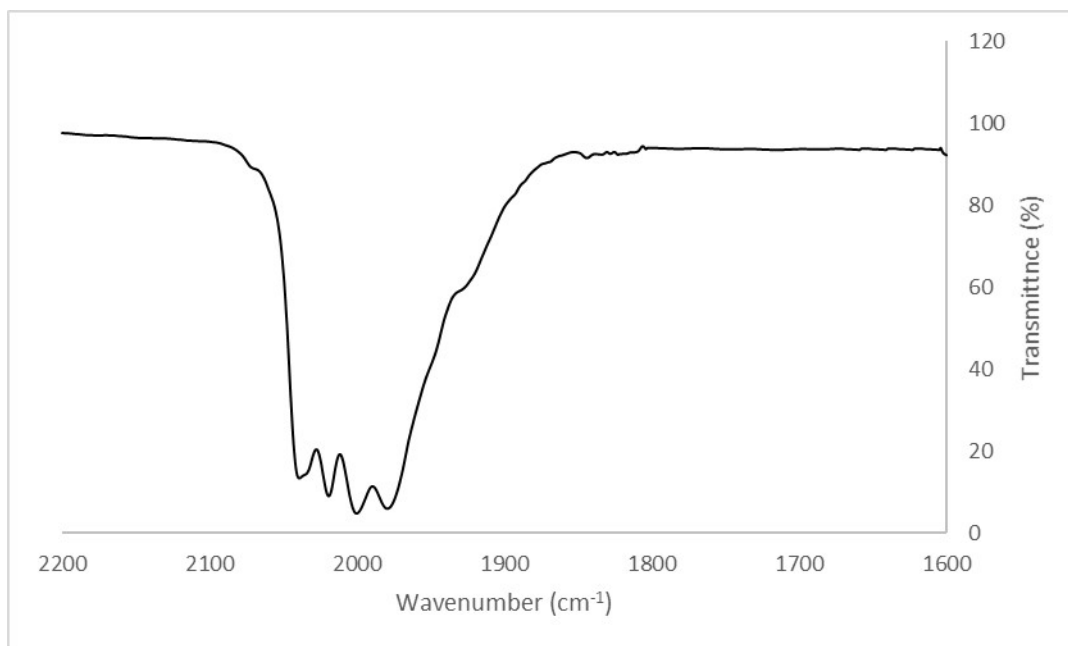
Nuclearity	Fe	Ref	Ru	Ref	Os	Ref
1	[HFe(CO) <sub>4</sub> ] <sup>-</sup>	1				
2	[HFe <sub>2</sub> (CO) <sub>8</sub> ] <sup>-</sup>	2				
3	[HFe <sub>3</sub> (CO) <sub>11</sub> ] <sup>-</sup>	3	[HRu <sub>3</sub> (CO) <sub>11</sub> ] <sup>-</sup>	7	[HOs <sub>3</sub> (CO) <sub>11</sub> ] <sup>-</sup>	19
					H <sub>2</sub> Os <sub>3</sub> (CO) <sub>11</sub>	20
					H <sub>2</sub> Os <sub>3</sub> (CO) <sub>10</sub>	21
4	[HFe <sub>4</sub> (CO) <sub>13</sub> ] <sup>-</sup>	4	[HRu <sub>4</sub> (CO) <sub>13</sub> ] <sup>-</sup>	8	[HOs <sub>4</sub> (CO) <sub>13</sub> ] <sup>-</sup>	22
			H <sub>2</sub> Ru <sub>4</sub> (CO) <sub>13</sub>	9	H <sub>2</sub> Os <sub>4</sub> (CO) <sub>13</sub>	19
	[HFe <sub>4</sub> (CO) <sub>12</sub> ] <sup>3-</sup>	5	[HRu <sub>4</sub> (CO) <sub>12</sub> ] <sup>3-</sup>	This work		
	[H <sub>2</sub> Fe <sub>4</sub> (CO) <sub>12</sub> ] <sup>2-</sup>	6	[H <sub>2</sub> Ru <sub>4</sub> (CO) <sub>12</sub> ] <sup>2-</sup>	10	[H <sub>2</sub> Os <sub>4</sub> (CO) <sub>12</sub> ] <sup>2-</sup>	23
			[H <sub>3</sub> Ru <sub>4</sub> (CO) <sub>12</sub> ] <sup>-</sup>	11	[H <sub>3</sub> Os <sub>4</sub> (CO) <sub>12</sub> ] <sup>-</sup>	11
			H <sub>4</sub> Ru <sub>4</sub> (CO) <sub>12</sub>	12	H <sub>4</sub> Os <sub>4</sub> (CO) <sub>12</sub>	24
5					H <sub>2</sub> Os <sub>5</sub> (CO) <sub>16</sub>	25
					[HOs <sub>5</sub> (CO) <sub>15</sub> ] <sup>-</sup>	26
	[HFe <sub>5</sub> (CO) <sub>14</sub> ] <sup>3-</sup>	6				
6					H <sub>2</sub> Os <sub>6</sub> (CO) <sub>19</sub>	27
			[HRu <sub>6</sub> (CO) <sub>18</sub> ] <sup>-</sup>	13	[HOs <sub>6</sub> (CO) <sub>18</sub> ] <sup>-</sup>	28
			H <sub>2</sub> Ru <sub>6</sub> (CO) <sub>18</sub>	14	H <sub>2</sub> Os <sub>6</sub> (CO) <sub>18</sub>	28
			H <sub>2</sub> Ru <sub>6</sub> (CO) <sub>17</sub>	15		
7					H <sub>2</sub> Os <sub>7</sub> (CO) <sub>22</sub>	29
					H <sub>2</sub> Os <sub>7</sub> (CO) <sub>21</sub>	29
			[HRu <sub>7</sub> (CO) <sub>20</sub> ] <sup>-</sup>	16		
					H <sub>2</sub> Os <sub>7</sub> (CO) <sub>20</sub>	30
8					[HOs <sub>8</sub> (CO) <sub>22</sub> ] <sup>-</sup>	31
			[H <sub>2</sub> Ru <sub>8</sub> (CO) <sub>21</sub> ] <sup>2-</sup>	17		
9					[HOs <sub>9</sub> (CO) <sub>24</sub> ] <sup>-</sup>	32
10			[H <sub>2</sub> Ru <sub>10</sub> (CO) <sub>25</sub> ] <sup>2-</sup>	18		
					[H <sub>4</sub> Os <sub>10</sub> (CO) <sub>24</sub> ] <sup>2-</sup>	33
					[H <sub>5</sub> Os <sub>10</sub> (CO) <sub>24</sub> ] <sup>-</sup>	34
11			[HRu <sub>11</sub> (CO) <sub>27</sub> ] <sup>3-</sup>	18		



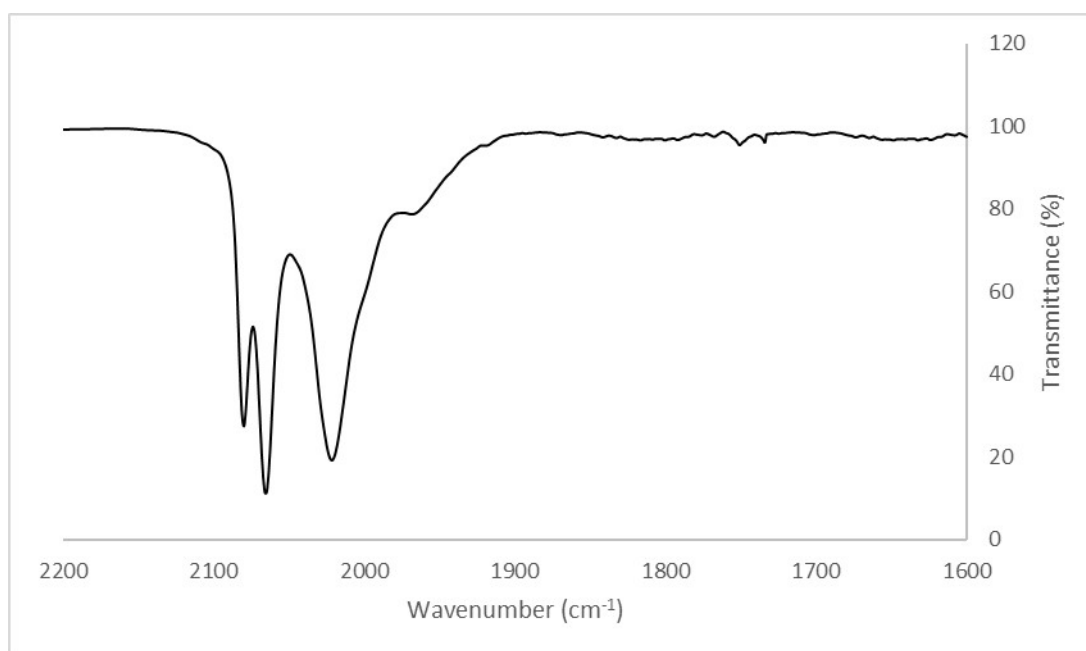
**Figure S1.** IR spectrum in the  $\nu_{\text{CO}}$  region of  $[\text{NEt}_4]_3[\text{HRu}_4(\text{CO})_{12}]$  in  $\text{CH}_3\text{CN}$ .



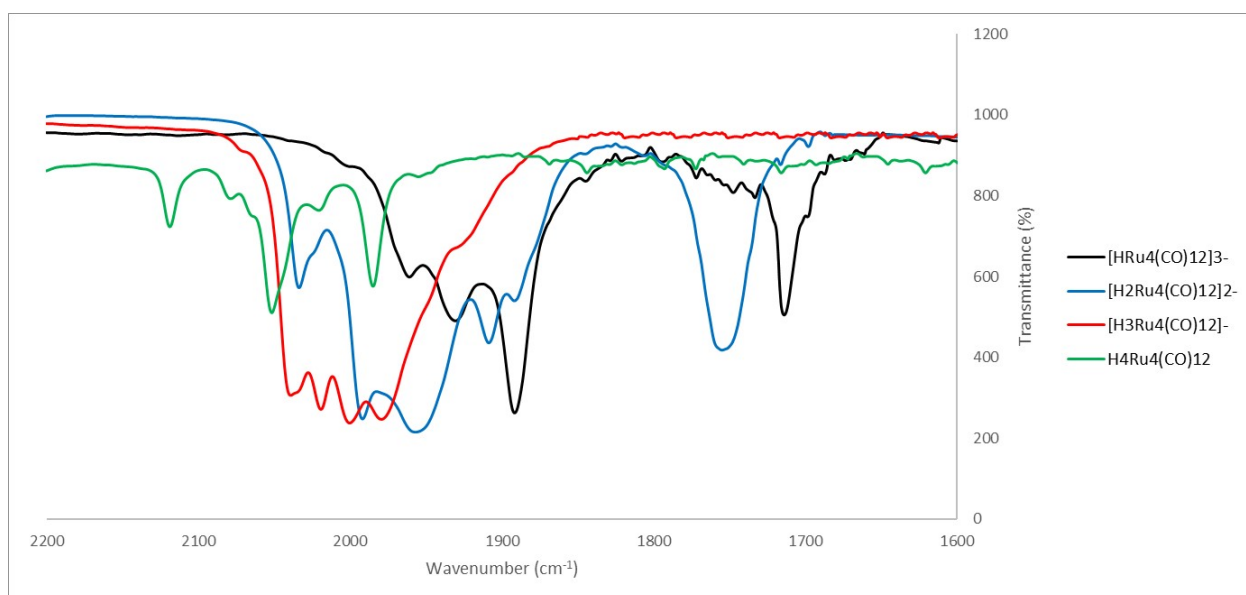
**Figure S2.** IR spectrum in the  $\nu_{\text{CO}}$  region of  $[\text{NEt}_4]_2[\text{H}_2\text{Ru}_4(\text{CO})_{12}]$  in  $\text{CH}_3\text{CN}$ .



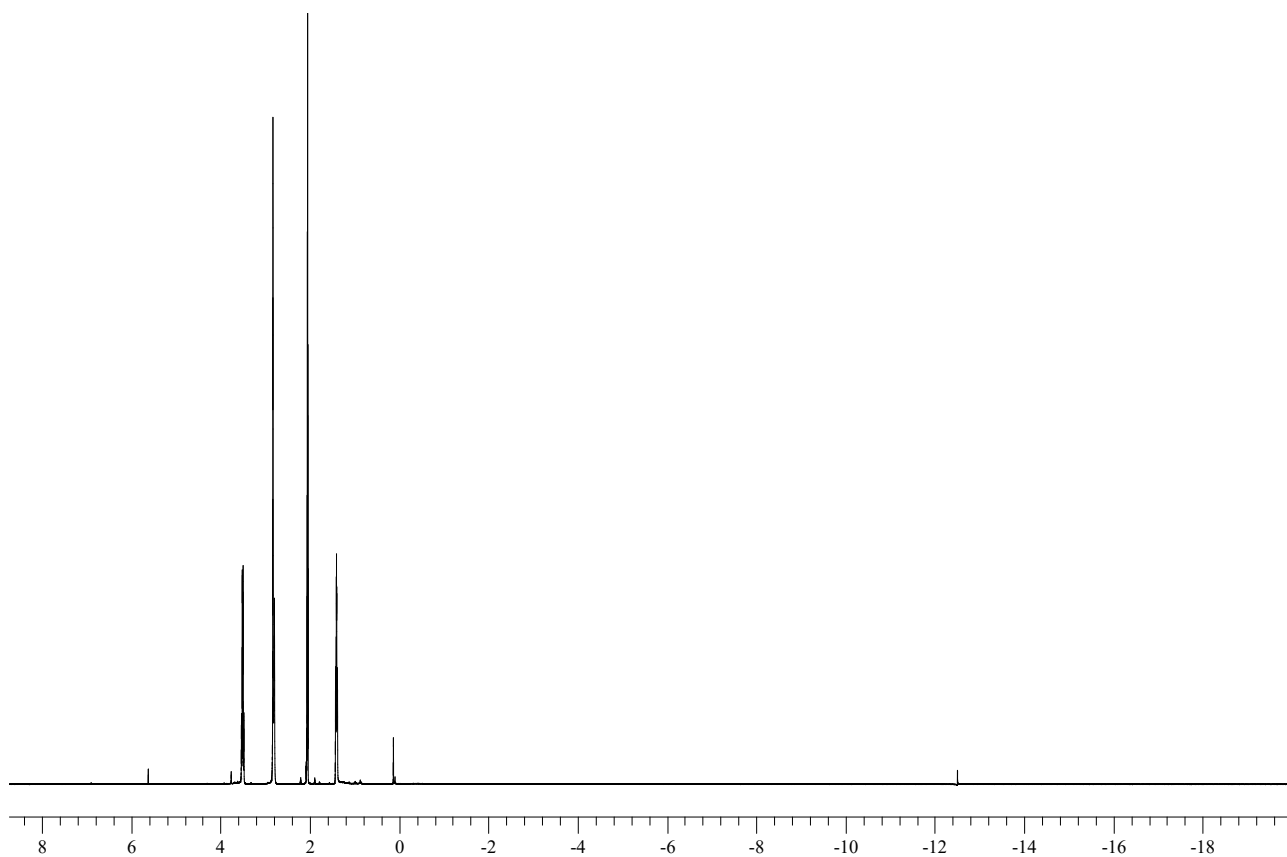
**Figure S3.** IR spectrum in the  $\nu_{\text{CO}}$  region of  $[\text{NEt}_4][\text{H}_3\text{Ru}_4(\text{CO})_{12}]$  in  $\text{CH}_3\text{CN}$ .



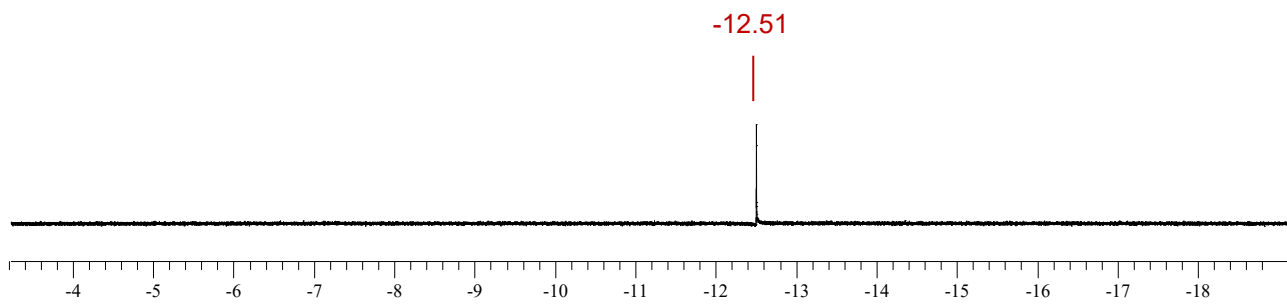
**Figure S4.** IR spectrum in the  $\nu_{\text{CO}}$  region of  $\text{H}_4\text{Ru}_4(\text{CO})_{12}$  in  $\text{CH}_2\text{Cl}_2$ .



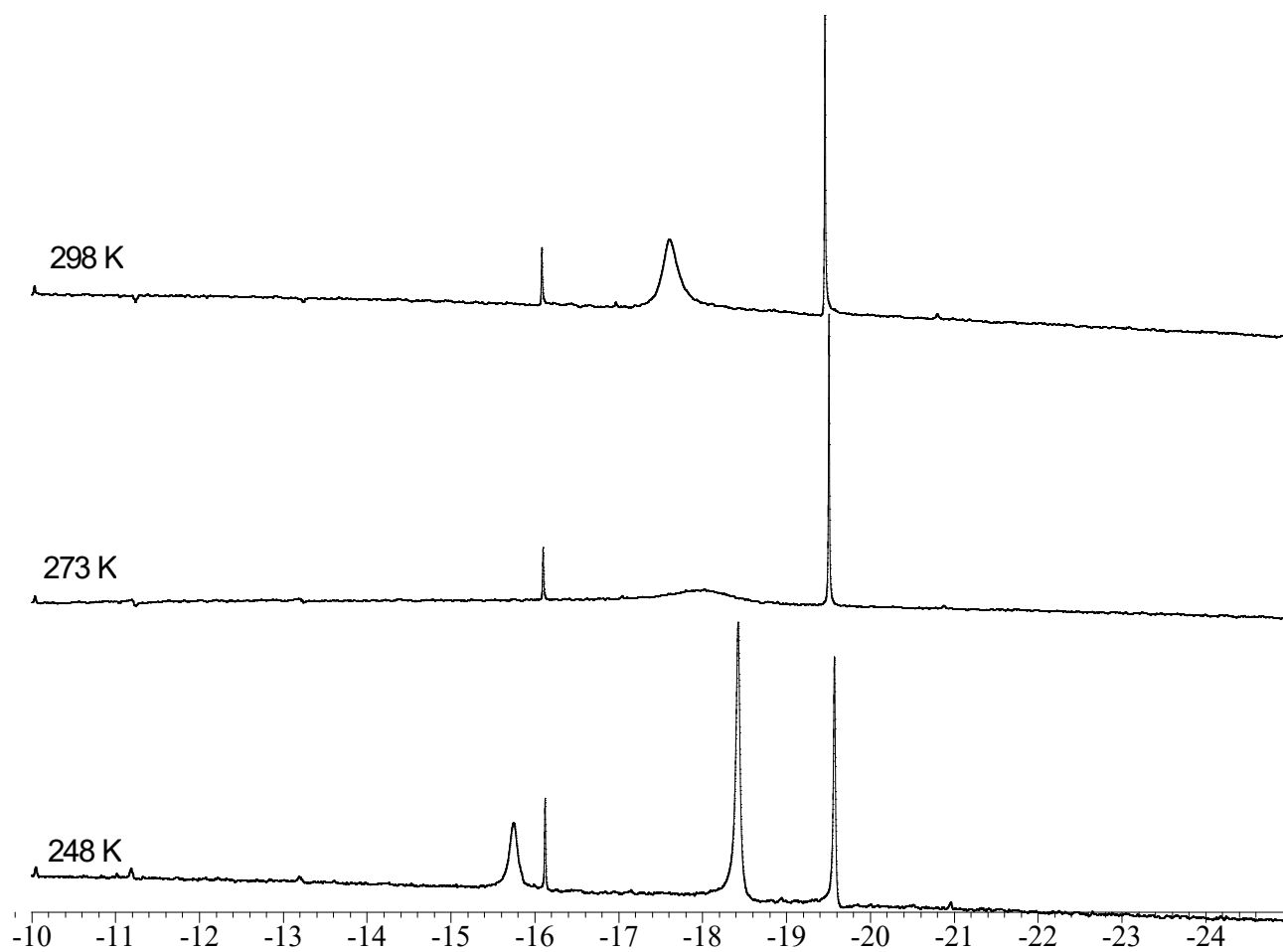
**Figure S5.** IR spectra in the  $\nu_{\text{CO}}$  region of  $[\text{H}_{4-n}\text{Ru}_4(\text{CO})_{12}]^{n-}$  ( $n = 0-3$ ) obtained from the stepwise protonation of  $[\text{HRu}_4(\text{CO})_{12}]^{3-}$  with  $\text{HBF}_4 \cdot \text{Et}_2\text{O}$  in  $\text{CH}_3\text{CN}$ .



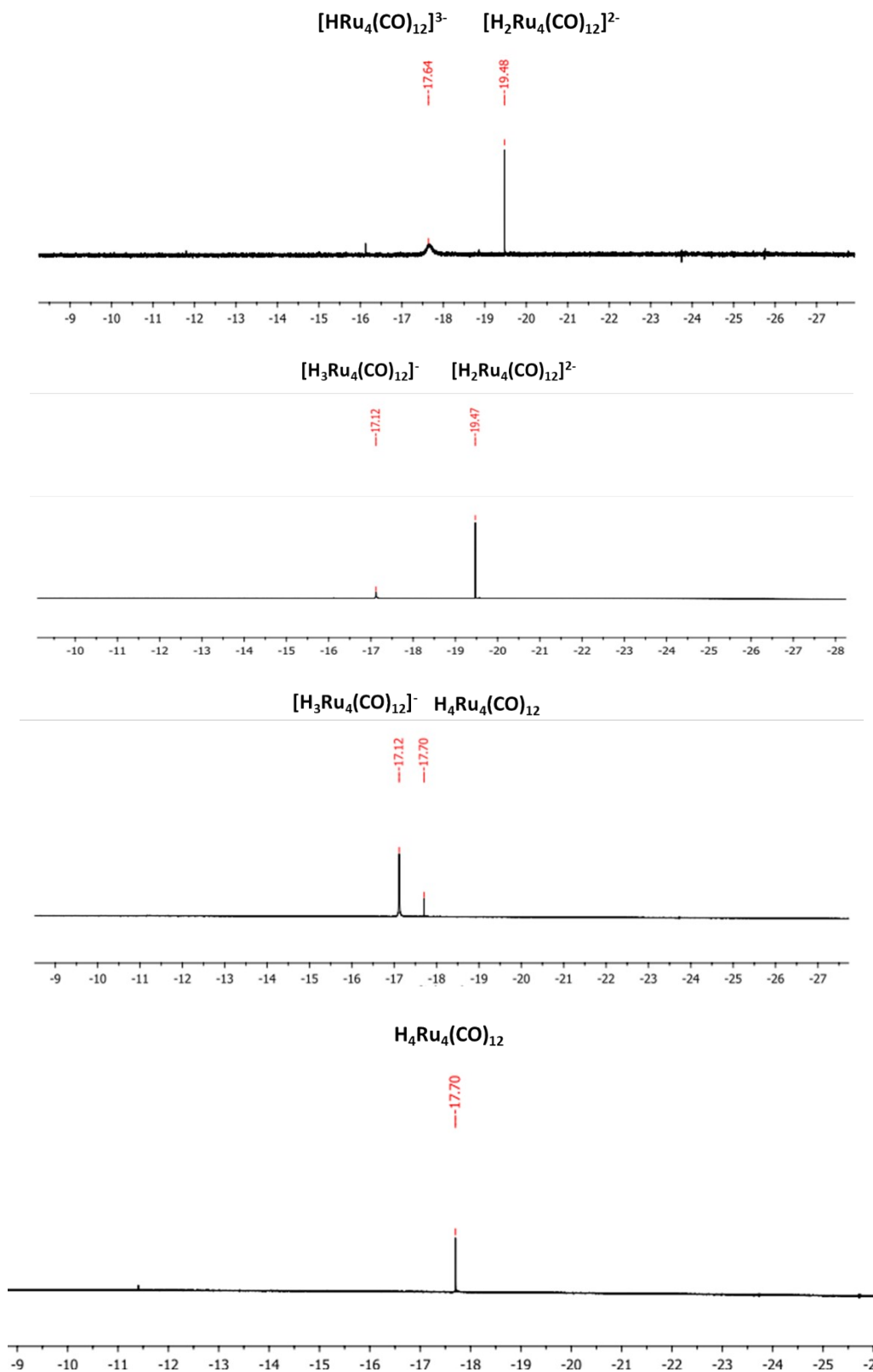
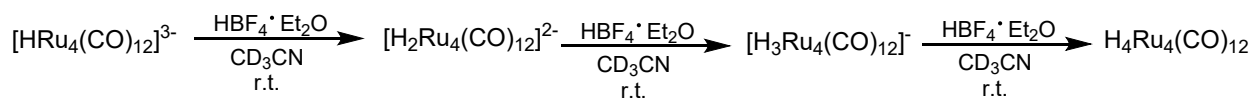
**Figure S6.**  $^1\text{H}$  NMR spectrum of  $[\text{NEt}_4][\text{HRu}_3(\text{CO})_{11}]$  in acetone- $\text{d}_6$  at 298 K.



**Figure S7.** Hydride region of the  $^1\text{H}$  NMR spectrum of  $[\text{NEt}_4][\text{HRu}_3(\text{CO})_{11}]$  in acetone- $\text{d}_6$  at 298 K.

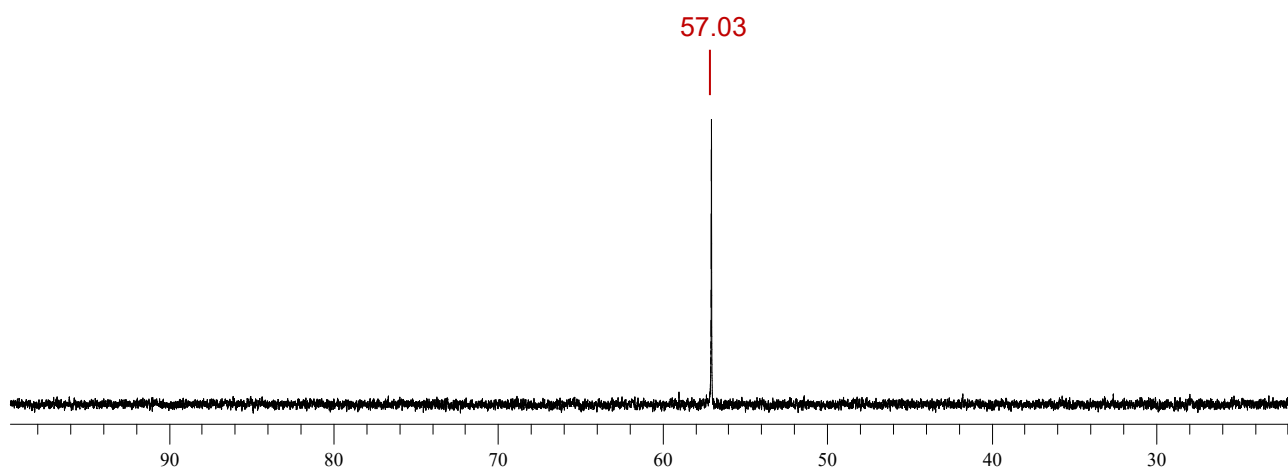


**Figure S8.** Hydride region of the VT <sup>1</sup>H NMR spectra of [NEt<sub>4</sub>]<sub>3</sub>[HRu<sub>4</sub>(CO)<sub>12</sub>] in CD<sub>3</sub>CN.

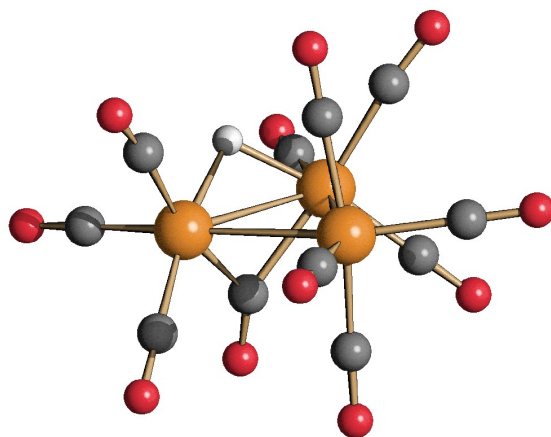


**Figure S9.** Hydride region of the  $^1\text{H}$  NMR spectra of  $[\text{H}_{4-n}\text{Ru}_4(\text{CO})_{12}]^{n-}$  ( $n = 0-3$ ) obtained from the stepwise protonation of  $[\text{HRu}_4(\text{CO})_{12}]^{3-}$  with  $\text{HBF}_4 \cdot \text{Et}_2\text{O}$  in  $\text{CD}_3\text{CN}$  at 298 K.

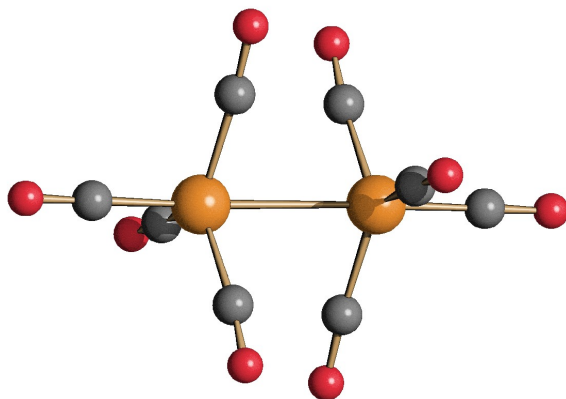




**Figure S10.**  $^{31}\text{P}\{^1\text{H}\}$  NMR spectrum of  $[\text{NEt}_4][\text{Ru}_3(\text{CO})_9(\text{NPPh}_3)]$  in  $\text{CD}_2\text{Cl}_2$  at 298 K.



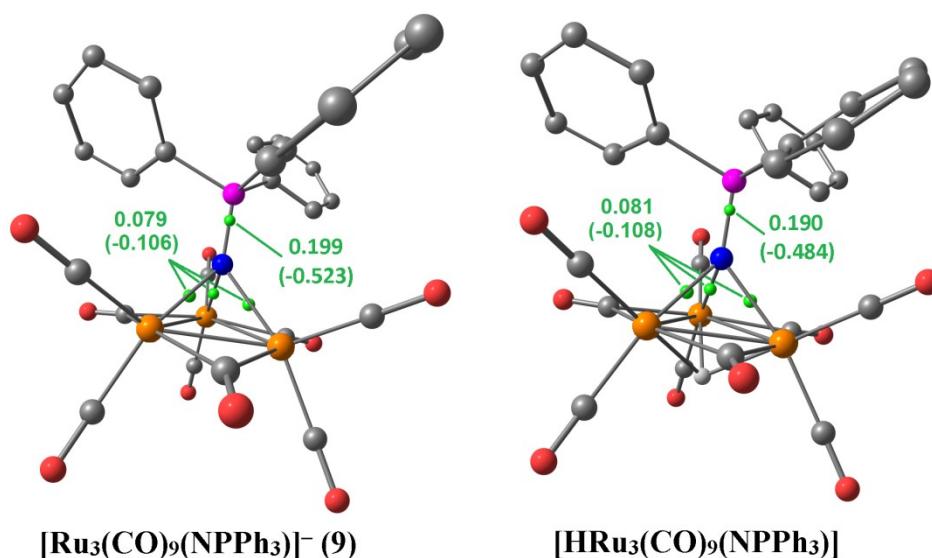
**Figure S11.** Molecular structure of  $[\text{HRu}_3(\text{CO})_{11}]^-$  (orange Ru; red O; grey C; white H).



**Figure S12.** Molecular structure of  $[\text{Ru}_2(\text{CO})_8]^{2-}$  (orange Ru; red O; grey C).

### Computational analysis of $[\text{HRu}_3(\text{CO})_9(\text{NPPH}_3)]$ (**9**)

DFT calculations afforded a stationary point closely comparable to the structure determined by SC-XRD. AIM and Mayer analyses were carried out on the P-N and Ru-N bonds of the compound and on the DFT-optimized  $[\text{HRu}_3(\text{CO})_9(\text{NPPH}_3)]$  hydride with the same CO stereochemistry.<sup>35</sup> Electron density ( $\rho$ ) and potential energy density ( $V$ ) average values at selected b.c.p. are summarized in Figure S13. The computed data suggest that the P-N bond is stronger in **9** with respect to  $[\text{HRu}_3(\text{CO})_9(\text{NPPH}_3)]$ . Such a result is supported by the Mayer bond order analysis, with values for the P-N bonds of 1.525 and 1.403 for **9** and  $[\text{HRu}_3(\text{CO})_9(\text{NPPH}_3)]$ , respectively. The different P-N bond strengths can be explained on the basis of lower N $\rightarrow$ Ru  $\sigma$ -donation in the case of the more electron rich anionic cluster. Such hypothesis is confirmed by the Ru-N b.c.p. data in Figure S13, indicating slightly weaker Ru-N bonds in  $[\text{Ru}_3(\text{CO})_9(\text{NPPH}_3)]^-$ , and by the Mayer analysis, with average Ru-N bond orders of 0.370 for **9** and 0.402 for  $[\text{HRu}_3(\text{CO})_9(\text{NPPH}_3)]$ .



**Fig. S13** DFT-optimized structures of  $[\text{Ru}_3(\text{CO})_9(\text{NPPH}_3)]^-$  (**9**) and  $[\text{HRu}_3(\text{CO})_9(\text{NPPH}_3)]$  (orange Ru; purple P; blue N; red O; grey C; white H) and selected (3,-1) b.c.p. (green) with density values at b.c.p. (potential energy density values in parenthesis). Data in a.u.

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