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Tracking molecular hydrogen migration along a subduction shear zone

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1 Tracking molecular hydrogen migration along a subduction
2 shear zone

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Abstract

This manuscript presents integrated field, petrographic, microstructural and thermodynamic modelling results documenting the high-pressure channelization of H₂ and the rheological impact of its reactivity with carbonate (dolomite and calcite)-rich rocks along a serpentinite-hosted shear zone from Alpine Corsica (France). Microstructures within the carbonates attest to the occurrence of a deformation continuum, evolving from initial brittle fracturing to strain localization by viscous deformation. Raman spectra of fluid inclusions within the carbonates reveals the circulation of H₂ and CH₄ at all stages of the microstructural evolution, the latter interpreted to be the result of H₂-carbonate interactions. Thermodynamic models suggest that carbonate phase stability at pressure-temperature conditions representative of a subduction setting is modified by the presence of H₂, with dolomite being progressively replaced by calcite+graphite+magnetite with increasing H₂ in the system. An initial phase of overpressure created by H₂-rich fluids led to the brecciation of dolomite, creating a fine-grained aggregate, which facilitated a switch to a semi-brittle mode of deformation and created high-permeability pathways for subsequent phases of H₂ infiltration. Subsequent phases of infiltration of H₂ were accompanied by transformation of dolomite to calcite, the degree of transformation dependent upon the efficiency of H₂ percolation. Calcite, being rheologically weaker than dolomite at these temperatures, underwent viscous flow in domains of extensive dolomite reduction, whereas adjacent dolomite-rich domains contain minimal imprints of extensive plastic deformation. Our results demonstrate extensive fossilized H₂-carbonate reactivity and show that the infiltration of H₂-rich fluids strongly affects the rheology of carbonates by inducing reactivity and phase transitions.

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1. INTRODUCTION

Molecular hydrogen (H_2) has been touted to have substantial potential as green fuel, since the only byproduct of H_2 combustion is water (Osselin et al., 2022). Traditional methods of industrial H_2 fuel production by the electrolysis of H_2O are energy intensive processes and not environmentally sustainable in the long term (Osselin et al., 2022). Additionally, underground storage of H_2 in geologic reservoirs, while necessary to store requisite amounts of H_2 to alleviate forecasted global demand, is also challenging and requires an in-depth investigation into the reactivity between H_2 and rock-forming minerals (Braid et al., 2024). The recent discovery of voluminous naturally occurring resources of H_2 in the Earth's subsurface has necessitated a need to evaluate natural H_2 as a possible source of clean fuel (Prinzhofer et al., 2018; Osselin et al., 2022; Truche et al., 2024). Moreover, the natural geologic systems that host these resources also provide information on how H_2 interacts with rocks and minerals (Reeves and Seewald, 2024), and these paleo H_2 -rock interactions can serve as proxies for underground hydrogen storage (UHS; Malki et al., 2024; Braid et al., 2024). In addition to its potential significance to the energy transition, H_2 plays an important role in planetary evolution and serves as source of energy for chemosynthetic life (McGlynn et al., 2020; Lappan et al., 2023; Gill-Olivas et al., 2023; Peverelli et al., 2024). Molecular hydrogen is also an important species in the deep Earth, and its concentration in the Earth's core is coupled with the oxidation of the residual mantle, thereby exerting a substantial control on the evolution of the redox state of the Earth's interior (Hirschmann et al., 2012).

The two most common sources of H_2 in the Earth's subsurface involve either the reduction of H_2O and concomitant oxidation of ferrous iron to ferric iron in minerals during water-rock interactions, such as serpentinization (Charlou et al., 2002; Sleep et al., 2004; Klein et

61 al., 2013), or the radiolytic dissociation of H₂O (Lollar et al., 2014). Although fluid-rock
62 reactions that lead to the formation of H₂ at different crustal levels have been studied in some
63 detail, the mechanisms by which H₂ migrates through the crust, as well as their effects on rock
64 deformation, are relatively poorly understood. A recent study by Prinzhofer et al. (2024)
65 documented high fluxes of H₂ and He along reactivated faults in the municipality of Maricà in
66 Brazil. Similarly, high H₂ fluxes have been documented along a network of shallow fractures in
67 the Bulqizë chromite mine in Albania (Truche et al., 2024). While these recent discoveries have
68 been successful in linking extensive surficial fluxes of reduced fluids, such as H₂ and abiotic
69 CH₄, to structural discontinuities, such as faults and fracture networks acting as preferential flow
70 conduits, the structural, microstructural, mineralogical, and geochemical markers of H₂
71 percolation from potentially deeper source areas within the crust are still largely unexplored.
72 Nevertheless, natural examples of H₂ circulation and channelization in orogenic belts have been
73 recently documented and require more detailed investigations to provide insights into the
74 geochemical and microstructural markers of natural H₂-rock interactions (Galvez et al., 2013;
75 Vitale Brovarone et al., 2017; Prinzhofer et al., 2018; Giuntoli et al., 2020, 2024; Boutier et al.,
76 2024; Truche et al., 2024).

77 High strain zones, such as faults and shear zones, can act as efficient conduits for fluid
78 migration (Austrheim, 2013) and crustal H₂ may plausibly migrate preferentially along these
79 structures. However, although numerous studies have focused on the role played by aqueous
80 fluids in altering the rheology of the Earth's crust, the rheological changes induced by low fO₂
81 fluids rich in H₂ along deformation zones are relatively poorly constrained. Water-rich fluids
82 generally lead to strain softening by promoting processes such as hydrolytic weakening, grain
83 boundary sliding (GBS), dissolution-precipitation creep and the formation of rheologically weak

84 phases, such as phyllosilicates (Post and Tullis, 1998; Skemer et al., 2006; Fuisseis et al., 2009,
85 2023; Getsinger et al., 2013; Menegon et al., 2015; Nicolas et al., 2016; Précigout et al., 2017;
86 Dobe et al., 2021). Only a few studies, on the other hand, have documented the structural and
87 microstructural imprints of natural H₂-rock interactions (Giuntoli et al., 2020, 2024). Most of
88 what we know on the effect of H₂ on material properties is thus derived from metallurgical and
89 material science literature, pertaining to processes such as stress corrosion cracking during H₂
90 injection and H₂ embrittlement (Barnett and Kerrich, 1980; Scully et al., 2012; Robertson et al.,
91 2015; López Freixes et al., 2022). This is largely due to the fact that H₂ interactions with many
92 crustal minerals may not produce marked reaction products and the identification of its paleo-
93 circulation in crustal rocks is still challenging.

94 Carbon-bearing minerals, such as carbonates and graphite, are common in many crustal
95 settings and may be extremely reactive with H₂ (Malvoisin et al., 2012; Galvez et al., 2013;
96 Lazar et al., 2014; Vitale Brovarone et al., 2017; Giuntoli et al., 2020; Peng et al., 2021; Peña-
97 Alvarez et al., 2021; Gelencsér et al., 2023; Huang et al., 2024; Braid et al., 2024; Harada and
98 Tsujimori, 2024; Zhang et al., 2024), and may attest to the circulation of reduced fluids enriched
99 in H₂ (Peverelli et al., 2024; Boutier et al., 2024; Giuntoli et al., 2024). These minerals and their
100 microstructural evolution may therefore be used to track H₂ infiltration and to assess the effects
101 of its percolation at crustal conditions.

102 We studied the microstructural and petrological evolution of carbonate rocks associated
103 with a serpentinite-hosted shear zone from Alpine Corsica (France) to track H₂ infiltration. A
104 combination of optical microscopy, cathodoluminescence (CL) observations and Scanning
105 Electron Microscopy (SEM) in conjunction with Electron Dispersive Spectroscopy (EDS)
106 reveals the presence of multiple carbonate generations in this suite of rocks, which are

107 chemically and microstructurally heterogeneous, and Raman spectra of fluid inclusions within
108 the different carbonate generations testify to the abundant circulation of H₂ through these
109 lithologies. Electron Backscatter Diffraction (EBSD) analyses from different microstructural
110 domains were performed to gain insights into the strain accommodation mechanisms within the
111 different carbonate generations and understand the role played by H₂ in developing the observed
112 microstructural heterogeneity; thermodynamic models of H₂-carbonate interactions show that H₂
113 influx within carbonates at pressure-temperature conditions relevant for subduction zones
114 induced phase transitions between carbonate phases with drastically different rheological
115 properties. By integrating our results from the aforementioned analyses and models, we aim to
116 provide new insights into rheological changes induced by H₂ circulation and reactivity, with an
117 emphasis on the nature of deformation induced by phase transitions caused by H₂-rock
118 interactions and into the definition of characteristic markers of H₂ circulation and reactivity in
119 the crust.

120 **1.1 Geologic setting**

121 The island of Corsica can be subdivided into two major tectonometamorphic domains
122 (Fig. 1A), namely Hercynian Corsica to the west and southwest, and Alpine Corsica to the
123 northeast (Malavieille et al., 1998; Vitale Brovarone et al., 2011, 2013; Molli and Malavieille,
124 2011; Vitale Brovarone and Herwartz, 2013; Di Rosa et al., 2020a; Rossetti et al., 2023).
125 Hercynian Corsica is dominantly granitic and was mostly unaffected by the Alpine orogeny,
126 except for a marginal transition zone, the so-called parautochthonous units, that underwent
127 Alpine metamorphism and deformation at low-grade blueschist to greenschist facies conditions
128 (see below; Molli et al., 2017; Di Rosa et al., 2020c, 2020b).

129 Alpine Corsica represents an exhumed orogenic wedge consisting of ocean- and
130 continent-derived units (Vitale Brovarone et al. 2013 and references therein). It comprises three
131 main domains: the continental margin units or the parautochthonous units, the oceanic units of
132 the Schistes Lustrés and the weakly metamorphosed turbidites and ophiolites of the Nappes
133 Supérieures (Molli, 2008; Molli and Malavieille, 2011; Malusà et al., 2015). The
134 parautochthonous units represent fragments of the Hercynian Corsica basement and their
135 sedimentary cover sequence overprinted during the Alpine orogeny up to blueschist-facies
136 conditions at about 46.7 ± 0.6 Ma to 34.4 ± 0.8 Ma (Brunet et al., 2000; Martin et al., 2011).

137 The Schistes Lustrés are composed of metamorphosed oceanic rocks (variably
138 serpentinized peridotites, metagabbros, metabasalts) and metasedimentary cover along with
139 slivers of continental basement collectively referred to as the “internal continental unit” (Cohen
140 et al., 1981). The tectonic juxtaposition of oceanic and continental basement rocks is interpreted
141 to have originated during rifting and extension of the Piemonte-Ligurian basin in pre-Alpine time
142 (Vitale Brovarone et al., 2011; Beltrando et al., 2014). The Schistes Lustrés can be subdivided
143 into three main tectonometamorphic units from west to east: a lawsonite-blueschist facies
144 metasediment-rich zone, a lawsonite-eclogite facies meta-ophiolite-rich zone, and a lawsonite
145 blueschist-facies zone (Vitale Brovarone et al., 2013), which are characterized by meta-
146 ophiolites, continental slivers and associated metasediments that are distinguished on the basis of
147 consistent tectonostratigraphic and metamorphic features and which show a metamorphic
148 overprint from low-grade blueschist to lawsonite eclogite-facies conditions (Lahondere and
149 Lahondere, 1988; Ravna et al., 2010; Vitale Brovarone et al., 2014; Beltrando et al., 2014). Peak
150 metamorphic conditions are constrained at about 35 Ma in the lawsonite eclogite unit by U-Pb
151 zircon and Lu-Hf garnet geochronology and around 37 Ma in the blueschist-facies unit by Lu-Hf

152 geochronology (Martin et al., 2011; Vitale Brovarone and Herwartz, 2013). The Nappes
153 Supérieures consist of a subgreenschist facies association of oceanic rocks and rocks of
154 continental affinity that experienced both brittle and ductile deformation. The present-day
155 structural position of the Nappes Supérieures atop of the Schistes Lustrés and the
156 Parautochthonous Hercynian units is the result of normal faulting related to the overthickened
157 antiformal stacking of Alpine Corsica and the opening of the Tyrrhenian basin (Jolivet et al.,
158 1998; Malavieille et al., 1998; Molli and Malavieille, 2011).

159 This study focuses on the hitherto undescribed Negru Shear Zone (NSZ hereafter) located
160 inside the Schistes Lustrés complex (Fig. 1B). The NSZ is located along the west coast of Cap
161 Corse, a few hundred meters north of the Negru beach, and deforms the lawsonite-eclogite
162 tectonometamorphic unit after Vitale Brovarone et al. (2013). In this domain, the lawsonite-
163 eclogite unit is primarily composed of ultramafic (variably serpentinized peridotites and
164 serpentinites), and mafic (metabasalts and metagabbros) lithologies, with subordinate
165 metasedimentary and continental basement slices. Peak metamorphic temperatures for this area
166 were estimated in metasedimentary rocks by means of Raman spectroscopy on carbonaceous
167 material at about 430°-470 °C (Vitale Brovarone et al., 2013) and pressure-temperature estimates
168 from mafic rocks, belonging to the same tectonometamorphic unit to the south, point to peak
169 metamorphic conditions in the range of 470-520 °C and 2-2.2 GPa (Ravna et al., 2010; Vitale
170 Brovarone et al., 2011).

171 **1.2 Field relations**

172 The NSZ is located inside a large, ~4x4 km body of serpentinite and is well exposed
173 along a roadcut (Figs. 2A&B). The shear zone is characterized by a high-strain domain, up to ~5
174 m thick, affecting metasomatized metasediments and metagabbros separating two comparatively

175 lower-strain blocks of serpentinite (Figs. 2A&B). The high-strain domain of the NSZ mainly
176 formed within and at the expense of rock types that are currently sandwiched between the two
177 serpentinite blocks. Large lenses of these rocks, up to 5 m thick, are preserved inside the high-
178 strain domain and consist of variably metasomatized metagabbroic rocks –possibly
179 autochthonous inside the serpentinitized mantle section– and allochthonous rocks of
180 metasedimentary origin. High-strain bands wrap around these lenses and generally exhibit higher
181 strain in light-colored, metagabbro-derived domains. These intensely sheared bands are marked
182 by a generally shallow ($\sim 12^\circ$) SW-dipping foliation and a down-dip stretching lineation. The
183 lower-strain domains in the adjacent serpentinite blocks are characterized by a generally
184 massive, yet locally foliated structure. Besides a first meter of intensely sheared serpentinite in
185 contact with the high-strain domain, the structure of the serpentinite blocks is more complex and
186 characterized by internally variable foliated domains wrapped in intensely sheared domains. The
187 internal foliation dips shallowly ($\sim 6^\circ$ - 14°) to the SW away from the high strain zones and
188 becomes progressively parallelized to the high strain zones, in their vicinity. The lower
189 (footwall) serpentinite block is not well exposed, and the following description refers to the
190 upper serpentinite block (hanging wall). The NSZ is cut at a low angle by a later brittle normal
191 fault dipping c. 10° to the NW and associated with c. 10° -plunging slickenlines. The fault has a
192 visible strike-length of 100 m (Figs. 2A&B). In the northwestern part of the roadcut, the fault is
193 parallel to the base of the upper serpentinite block. Indurated gouges and light-colored carbonate-
194 silica veins decorate the fault core and the immediate damage zone. These late-stage brittle
195 structures will not be further described in this manuscript.

196 The first few meters of serpentinites above the high-strain domains are crosscut by a
197 network of carbonate veins and anastomosing carbonate-rich deformation bands (Figs. 2C&D)

198 that preserve a continuum of deformation imprints from brittle to ductile. These carbonate veins
199 were not observed farther into the upper serpentinite block to the north and thus appear to be
200 genetically related to shearing within the high-strain domain.

201 On the weathered surfaces the vein network is seen to comprise an earlier set of brown to
202 dark gray colored carbonate veins, and a later set of white carbonate veins (Figs. 3A&B). The
203 brown to dark-gray carbonate veins are present as either part of a chaotic network of small veins
204 (<1 cm in thickness; Figs. 3A&B) or larger (up to 15 cm in thickness) veins preserved as pods
205 along the deformation bands dipping to the northwest at a medium to high angle (Figs. 3C&D).
206 Locally, these stretched carbonate veins are preserved at the contact zone between the high-strain
207 domain of the NSZ and the adjacent serpentinite and are parallelized to it. On fresh surfaces,
208 these veins range in color from milky white to grey or even dark gray. As described in Section 3,
209 the dark color of the deformation bands is related to the reaction-mediated formation of graphite
210 and the degree of transformation of carbonate to graphite varies from moderate (Figs. 3C&D) to
211 almost complete (Figs. 3E&F). Along these bands, zones in which carbonate transformation is
212 complete are marked by a complete absence of carbonate and the presence of graphite in
213 association with diopside. These deformation bands have been reworked multiple times during
214 the tectonometamorphic evolution of the NSZ, and some of the late-stage veins crosscut the
215 bands (Figs. 3C&D), whereas others are parallelized-to and transposed into it with a sinistral
216 (viewed top down) sense of movement (Figs. 3E&F).

217 The larger brown to dark gray carbonate veins are coarse-grained (grains up to 1-2 cm in
218 diameter) whereas the ones that are smaller have grains ranging from <1 mm to 0.5 cm in
219 diameter. All identified carbonate veins/pods contain at least some dark, graphite-bearing
220 domains. These carbonate pods appear brecciated, with graphite occupying the spaces between

221 breccia fragments, and also display evidence of ductile deformation with grain size reduction
222 (Figs. 3C, D, G&H). Smaller veins are generally darker and more reworked by ductile
223 deformation compared to larger carbonate pods. The samples selected for this study were
224 collected from these carbonate veins and pods and their microstructural and petrographic features
225 are presented in Section 3.

226

227 **2. METHODS**

228 **2.1 Petrographic investigations and MicroRaman analysis**

229 Petrographic and microstructural analyses were conducted using a Zeiss correlative
230 microscopy platform composed of an AxioZoom V16 macroscope, an AxioScope7 polarized
231 microscope, and a Zeiss EVO 15 Scanning Electron Microscope (SEM) at the Deep Carbon Lab
232 (<https://deepcarbonlab.org>), Department of Biological, Geological and Environmental Sciences,
233 University of Bologna, Italy. Detailed methodology for SEM-EDS-EBSD is presented in section
234 2.2.

235 Cathodoluminescence (CL) analyses were conducted using a Cathodyne optical CL setup
236 manufactured by NewTec Scientific, operated at 100 mTorr vacuum, an accelerating voltage of
237 12 kV and 100 mA current. CL intensity maps have been presented in Figs. 4C&E, in which
238 brighter shades of red correspond to increasing luminescence (black: least luminescence, orange:
239 maximum luminescence).

240 Raman spectra were acquired using a WITec Alpha 300-R confocal micro-Raman
241 spectrometer with a 532 nm laser source at the Department of Biological, Geological and
242 Environmental Sciences, University of Bologna, Italy. The Raman spectra were acquired with a
243 100x objective using the following parameters: 600 and 1200 g/mm grating; an on-sample laser

244 power between 25-40 mW and acquisition times ranging between 1-10 minutes. The Raman shift
245 was calibrated using an Ar-Hg lamp. The acquired spectra of serpentine were compared with the
246 data reported in (Schwartz et al., 2013) and Compagnoni et al. (2021) for serpentine minerals,
247 and with the ruff.com database for carbonate minerals. The Raman spectra of fluid species were
248 compared with those reported in Frezzotti et al. (2012). We performed quantitative Raman
249 spectroscopy analysis to estimate the composition of the fluid species hosted in the FIs using the
250 methods outlined in Burke (2001).

251 **2.2 SEM-EDS-EBSD**

252 EDS maps of regions of interest (ROI) within the samples studied were obtained using
253 the Oxford Instruments Ultim max 100 silicon drift detector fitted to the Zeiss EVO 15 SEM at
254 the Deep Carbon Lab, in high-vacuum mode. Samples were polished on a Saphir Vibromat with
255 Eposal F (0.06 μm) for 4 hours followed by further polishing with Eposal M (0.01 μm) colloidal
256 silica solutions. EDS data were obtained and subsequently processed using the TruMap
257 background removal post-processing routine in the Aztec software suite (version 5.1),
258 manufactured by Oxford Instruments. EBSD data were acquired from ROIs within the studied
259 samples using an Oxford Instruments C-Nano EBSD detector fitted to the Zeiss EVO 15 SEM at
260 the Deep Carbon Lab, operating in low-vacuum mode with a chamber pressure of 40 Pa. Data
261 acquisition was carried out using the Aztec software suite (version 5.1) manufactured by Oxford
262 Instruments, using an accelerating voltage of 25 kV, step sizes of 1 and 1.5 μm depending on the
263 area of the ROI, a beam current of 8-10 nA, a sample tilt of 70° and working distances between
264 16-22 mm. In domains containing both calcite and dolomite, the TruPhase acquisition routine
265 was utilized to accurately distinguish the two phases using EDS spectra at each analyzed point
266 (Helpa et al., 2014), in addition to acquired diffraction patterns, owing to the substantial

267 crystallographic similarity between the 2 phases (calcite trigonal-high, Laue group -3m; dolomite
268 trigonal low, Laue group -3). The magnesian calcite match unit was used to index dolomite in
269 domains without calcite (trigonal-high, Laue group -3m). Raw indexing rates varied from 80-
270 94%. The acquired EBSD data were processed using Aztec Crystal (version 3.1). A cleanup
271 routine was used involving removal of wild spikes followed by 3 iterations of zero solution and
272 removal by assigning unindexed pixels the average orientation of 3 nearest neighbors. Voids
273 detected based on band contrast were excluded from the cleanup routine. Dolomite is prone to
274 misindexing (Pearce et al., 2013) about the axis/angle pair that defines its growth twins (11-20
275 and 180°). Pseudosymmetry arising from this misindexing was removed by rotating misindexed
276 portions up to 10 pixels wide into the orientation of the host grain. Post pseudosymmetry
277 removal, the data was denoised using a Kuwahara filter constituted by octants of radii of 3
278 pixels. The processed and denoised data obtained from Aztec Crystal was used as the input file
279 for Mtex 5.10.2 (Bachmann et al., 2010), which was used to plot interphase misorientations
280 between dolomite and calcite based on the workflow presented in Morales (2022) and Zertani et
281 al. (2024). Determination of the presence of a crystallographic preferred orientation (CPO) was
282 done by plotting upper hemisphere equal area pole figures using a halfwidth of 10° using one
283 pixel per grain to ensure that larger grains are not overrepresented. Poles to (0001), and (2-1-10)
284 were plotted to determine the distributions of c and a axes, respectively while poles to (10-14),
285 (01-12) and (01-18) were plotted to understand the prevalence of twinning in the microstructure.
286 Grain reference orientation deviation (GROD) maps, that plot the pixelwise orientation deviation
287 from the mean orientation of each grain, have been presented in Figs. 9-11 to visualize the
288 internal substructures of dolomite and calcite grains from different textural domains. Low-angle
289 (2°-10°) and high-angle (>10°) boundaries along with f-twins (misoriented 78° about 2-201) and

290 growth twins (misoriented 180° about 11-20) in dolomite and e-twins for calcite (misoriented 78°
291 about 20-21) were superposed onto the GROD maps. Disorientation (minimum misorientation)
292 angle distributions between neighbor pairs of grains (grains that share a common boundary;
293 shown in blue) and random pairs of grains (randomly chosen data points not related to grains
294 across a common boundary; shown in yellow) for dolomite and calcite have also been shown in
295 Figs. 9-11. The theoretical distribution of misorientations for a completely random fabric
296 (Mackenzie plot) is shown with a grey line in these figures. The M-index (Skemer et al., 2005), a
297 measure of the fabric strength (values between 0 and 1; 0 corresponds to a completely random
298 fabric and 1 corresponds to a fabric with a uniform orientation), has been calculated for each
299 textural domain and the value has been incorporated within the disorientation plots.

300 **2.3 Thermodynamic modelling**

301 To model the mineralogical evolution of carbonates under the infiltration of H_2 -rich
302 fluid, pseudosections were computed with the Perple_X software packages (Connolly, 2005)
303 Two pseudosections were computed at 1 GPa and 2 GPa with temperature ranging from 300 °C
304 to 800 °C as y-axis and moles of H_2 added to the initial composition as the x-axis (see Section
305 3.4 for details on the considered pressure-temperature conditions). The initial composition was
306 set to that of a slightly ferroan dolomite with a Mg# of 5%, $Mg\# = Mg/(Mg+Fe)$. The x-axis
307 represents the evolution from 0 moles of H_2 added to the Fe-dolomite up to 5 moles of H_2 added.
308 The Holland and Powell (2011) version 633 thermodynamic database was used. The
309 pseudosections were computed in the CaO-MgO-FeO-C-H₂-O₂ system of components. No
310 excess O₂ was added to the initial dolomite composition. The solid solutions used were:
311 oCcM(EF) for Fe-Mg-Ca-carbonates (Franzolin et al., 2011), B for Fe-Mg-brucite (ideal
312 molecular mixing), Fper(H) for Fe-Mg-periclase (Holland et al., 2013), MF for Fe-Mg-magnetite

313 (Sack and Ghiorso, 1991) and COH-Fluid for H₂-CH₄-H₂O-CO-CO₂ fluid. Mineral abbreviations
314 were adopted from Holland and Powell (2011).

315

316 **3. RESULTS**

317 **3.1 Petrography**

318 Four samples from different parts of the NSZ (COR18-7a, COR18-7b, 3COR19-20Zb,
319 and 3COR19-20Zc, see Figs. 3C&G for thin-section locations), representing carbonate veins and
320 pods and the foliated serpentinite within which they are hosted, were chosen for this study. The
321 veins and pods contain variable proportions of dolomite, calcite, and serpentine as major mineral
322 phases, as well as minor chlorite, magnetite, sulfides, and apatite. COR18-7b is a sample of the
323 host serpentinite (Fig. 4A), COR18-7a (Figs. 4B&C) of a brecciated carbonate hosted along one
324 of the carbonate-bearing deformation bands developed along the NSZ and 3COR19-20Zb (Fig.
325 4D&E) and 3COR19-20Zc (Fig. 4F) are samples from the interface between carbonate veins and
326 serpentinite.

327 The host serpentinite (sample COR18-7b; Fig. 4A) contains a foliation defined by
328 elongated antigorite and minor chlorite and is cut by later concordant and discordant veins of
329 chrysotile that will not be described further. Several kink bands deform the serpentine grains at
330 right angles to the foliation (Fig. 5A). The host serpentinite underwent a phase of brittle
331 deformation post-dating the development of its pervasive regional foliation, as evidenced by the
332 presence of graphite-filled fractures cutting across the foliation. This phase of deformation
333 involved the formation of fractures at high angle to the serpentinite foliation and the subsequent
334 widening of these fractures and their infilling by carbonate (Fig. 4A).

335 A carbonate pod hosted within one of the deformation bands within the NSZ (COR18-7a;
336 Figs. 4B&C) consists of, in order of abundance, dolomite, calcite, serpentine (mostly antigorite),
337 and graphite, with minor magnetite as an accessory phase (Fig. 4B). Dolomite and calcite were
338 distinguished based on CL and EDS maps (Figs. 4C&E and Fig. 7). In the CL maps, dolomite
339 appears darker, and calcite appears brighter (Figs. 4C&E). The microstructures of dolomite vary
340 from domains consisting of brecciated coarse-grained dolomite to high-strain domains composed
341 of aggregates of fine-grained dolomite formed by fracturing of the coarser grains, with
342 transitional domains between the two endmembers (Figs. 4B&C). In the brecciated domains of
343 the sample, individual grains are up to 1 mm wide, with an average width of 500 microns. They
344 contain a high density of twins, show no evidence of substantial recrystallization, and exhibit
345 nearly perfect extinction patterns in the untwinned domains (Fig 4B, 5B). The brecciated
346 fragments have angular shapes, with straight to curvilinear boundaries (Fig. 5B). Curved
347 segments of the boundaries between the brecciated fragments have a serrated appearance,
348 resembling stylolites, and attest to the occurrence of dissolution-precipitation at the margins of
349 the breccia fragments.

350 The deformed interface between a carbonate vein and the host serpentinite within the
351 NSZ is presented in Figs. 3G&H and exhibits a layered structure defined by a stack of a
352 carbonate-dominated layer (sample 3COR19-20Z), a layer of carbonate + serpentinite clasts and
353 a layer of sheared diopside in contact with the serpentinite. Graphite is present in all layers. The
354 diopside + graphite layer is interpreted to be indicative of complete reaction between carbonate
355 and serpentine in the presence of H₂ (Vitale Brovarone et al., 2017). We focused on slab
356 3COR19-20Z because it records an incipient (i.e., not complete) stage of fluid-carbonate
357 interaction. This sample shows a continuum of microstructures from brittle (3COR19-20Zb) to

358 ductile (3COR19-20Zc; Figs. 3G&H, 4D-F). Within the carbonate-dominated layer, a domain of
359 brittle microstructures similar to those observed in COR18-7a and characterized by angular
360 fragments of twinned dolomite grains lined by graphite, transitions into domains consisting of
361 aggregates of dolomite characterized by substantial grain size reduction (Figs. 4D-F). Within the
362 fine-grained domains, larger relict grains of dolomite are surrounded by small subgrains,
363 resembling core-mantle textures (Fig. 5C). The size of the dolomite grains decreases from 1 mm
364 in the brecciated domains to 100-200 microns in the fine-grained domains. There is also a
365 variation in grain shapes of the fine-grained dolomite compared to the brecciated dolomite. The
366 fine-grained dolomite grains transition to more rounded shapes from elongate grains and are
367 observed to show a depletion in their Fe contents from core to rim (Fig. 6B). Boundaries between
368 the finer grains are rarely straight, whereas the brecciated dolomite grains are mostly equant with
369 well-defined grain boundaries that occasionally deviate from linearity owing to the development
370 of stylolites at the interfaces between grains. While some of the larger fragments contain
371 abundant twin boundaries, the smaller grains are completely devoid of them (Figs. 5E&6C).

372 Within the domains dominated by brecciated dolomite (3COR19-20Za and b and
373 COR18-7a), calcite occurs in close association with the dolomite (Figs. 5F, 6 A&C, 7E-H). In
374 the brecciated domains, calcite proportions are low, with incipient calcite formation observed
375 along the boundaries of the dolomite grains, especially at junctions among dolomite fragments
376 (Fig. 5F, 6A). Away from the brecciated domains in COR18-7a and 3COR19-20Zb, calcite
377 occurs in wider domains up to tens of microns thick, closely associated with graphite and
378 magnetite. In these domains, the calcite grains are euhedral to subhedral and rounded (Figs. 5G,
379 6D, 7 I-L). The proportion of graphite and magnetite correlates well with the amount of calcite
380 present, with relatively higher graphite proportions in the wider calcite-dominated domains

381 compared to the domains of incipient calcite formation associated with the brecciated dolomite
382 grains. Nonetheless, within the brecciated domains, zones with relatively higher calcite
383 proportions also contain abundant graphite along fractures and twin boundaries within dolomite
384 (Fig. 5F, 6C).

385 The domain of fine-grained dolomite is characterized by drastic grain-size reduction and
386 passes laterally in a domain composed of diopside and graphite along with pervasive discordant
387 calcite veins (3COR19-20Zc, Fig. 5H). The diopside grains are fine-grained (average grain
388 diameter of 50 microns), with individual grains not exceeding 100 microns in diameter, and are
389 rounded and anhedral. They are localized into an extensively sheared domain dissected by
390 conjugate sets of fractures. The entire unit is cross-cut by late-stage carbonate veins (Fig. 5H).

391

392 **3.2 Fluid inclusion petrography and micro-Raman Spectroscopy**

393 The dolomite grains within the brecciated domains have cloudy cores, owing to the
394 occurrence of several fluid inclusions. These fluid inclusions are mostly biphasic containing
395 liquid and vapor and are characterized by variable shapes, from irregular to negative crystal.
396 They are mostly localized along trails that cross-cut twin boundaries. The size of the inclusions is
397 variable, ranging from 5 microns in length to less than a micron. Some inclusions are notably
398 monophasic, consisting of a singular vapor phase (Fig. 8A). Raman spectra obtained from
399 monophasic fluid inclusions within the brecciated dolomite grains (Fig. 8B) confirm the presence
400 of CH₄, characterized by a Raman shift ranging between 2913 cm⁻¹ and 2916 cm⁻¹, as the
401 dominant vapor phase with subordinate amounts of H₂, displaying a Raman shift at 4153 cm⁻¹.
402 Fluid inclusions in the relict dolomite grains from the fine-grained high-strain domains (Fig. 8C)
403 are also characterized by their abundance and negative crystal shapes, as in the brecciated

404 dolomite domains but the proportion of monophasic vapor-only inclusions is substantially higher
405 and biphasic liquid + vapor inclusions are rare. The Raman spectra of fluid inclusions within the
406 relict dolomite grains from the high-strain domain confirm the presence of H₂ as the dominant
407 fluid component featuring Q1 vibrational bands at 4127, 4144, 4156 and 4162 cm⁻¹ and the S0
408 (1) rotational band at 585 cm⁻¹ (Fig. 8D). Fluid inclusions within calcite are much less abundant
409 compared to the brecciated and relict dolomite grains. When present, they are localized adjacent
410 to grain boundaries and occur within transgranular trails (Fig. 8E). The fluid inclusions within
411 calcite are commonly monophasic with negative crystal shapes and biphasic inclusions are very
412 rare, similarly to the fluid inclusions in the relict dolomite grains from the high strain domains.
413 The inclusions in calcite also show a greater variability in size, ranging from up to 15 microns in
414 length to less than a micron (Fig. 8E). Calcite-hosted fluid inclusions (Fig. 8F) consist of a vapor
415 phase dominated by CH₄ (2912 cm⁻¹) with subordinate amounts of H₂ (4149 cm⁻¹).

416 Fluid inclusions were also studied in a sample from a large diopside vein hosted in
417 serpentinite from the hanging wall of the NSZ to ascertain whether the H₂ hosted in the fluid
418 inclusions within carbonate of the NSZ was externally or internally derived, since this diopside
419 vein is located at the edge of the NSZ, more than 200 m from the carbonate-rich deformation
420 bands within the NSZ. The vein is concordant to the regional foliation of the serpentinite and is
421 composed of large (> 1 cm) diopside crystals elongated parallel to the main lineation in the
422 antigorite serpentinite. This feature indicates that diopside formed or at least recrystallized
423 during the metamorphic evolution of the unit. Diopside contains large, biphasic liquid + vapor
424 fluid inclusions. These inclusions are rich in CH₄ and H₂ (Fig. 8G&H). We undertook
425 quantitative estimates of the proportion of CH₄ and H₂ in the vapor phase in 5 carbonate-hosted
426 fluid inclusions and one diopside-hosted fluid inclusion from within the NSZ. Four out of five

427 carbonate hosted fluid inclusions contain a higher molar proportion of CH₄ compared to H₂; in
428 contrast, the diopside-hosted fluid inclusion contains higher proportion of H₂ relative to CH₄
429 (Table 1).

430 Raman spectra were also used for the identification of antigorite as the dominant
431 serpentine species (Supplementary figure S1B) and to confirm the presence of graphite
432 (Supplementary figure S1D).

433 **3.3 EBSD analyses**

434 EBSD data were acquired from the three texturally distinct domains that were recognized
435 in these samples. These domains consist of (1) high-strain domains composed exclusively of
436 fine-grained dolomite (domain 1; Figs. 5C&9A), (2) brecciated dolomite with incipient
437 development of calcite at the margins of breccia fragments (domain 2; Figs. 5E, 6C&10A), and
438 (3) domains of advanced calcite formation (domain 3; Figs. 5G&11A).

439 In the high-strain domains (domain 1; Fig. 9), dolomite appears to preserve core-mantle
440 textures. The GROD map (Fig. 9A) from this domain shows that larger relict grains of dolomite
441 have much higher GROD values compared to the finer grains that surround it (up to 15° in the
442 relict grain). The proportion of low-angle boundaries is also higher in the relict grains compared
443 to the grains in the matrix (Fig. 9A). These observations appear to suggest that the dolomite
444 grains in this domain underwent subgrain rotation (SGR) recrystallization, wherein original
445 subgrains within larger grains of dolomite progressively rotated into new orientations (Lloyd et
446 al., 1997; Bestmann and Prior, 2003; Passchier and Trouw, 2005). However, both the cumulative
447 misorientation (misorientation at a point relative to the misorientation at the first point; shown in
448 red) and the point-to-point misorientation (misorientation at each point of the profile; shown in
449 blue) curves (Fig. 9B) along the white line through the center of the relict grain in Fig. 9A show

450 sharp peaks and troughs, typical of healed microcracks (Schenk and Urai, 2005; Anders et al.,
451 2014; Hentschel et al., 2019) and do not correlate with a gradual increase in misorientation as
452 would be expected for SGR recrystallization (Bestmann and Prior, 2003; Passchier and Trouw,
453 2005; Lopez-Sanchez et al., 2021). Therefore, the finer grainsize fraction of dolomite probably
454 originated due to fracturing of coarser dolomite grains, and not by the formation and rotation of
455 subgrains in coarse-grained dolomite into new orientations.

456 In the domains of incipient calcite formation (domain 2; Fig. 10A), the larger relict grains
457 of dolomite are dominated by twin boundaries, while the smaller dolomite grains within the
458 matrix are free of twin boundaries. Both f-twins and growth twins are observed within dolomite.
459 The relict, twinned grains of dolomite show higher GROD values ($\sim 7^\circ$) compared to the smaller
460 dolomite grains surrounding it. In this case, the proportions of low-angle boundaries in both the
461 relict, coarser grains of dolomite as well as the finer-grained dolomite within the matrix are low,
462 reinforcing a fracture-based origin for the finer grainsize fraction of dolomite instead of SGR
463 recrystallization. In the domains of advanced calcite formation (domain 3), recrystallized
464 dolomite grains are absent, but larger relict grains of dolomite with growth and f-twins persist
465 (Fig. 11A). In this domain, the distribution of GROD values is broadly consistent with that
466 observed in the domain of incipient calcite formation. A higher GROD value is observed within
467 the relict and twinned dolomite grain ($\sim 6^\circ$), whereas GROD values in the adjacent calcite-rich
468 matrix is generally low (between 0° - 6°) except for intermediate values in some larger grains.

469 Within the high-strain domain (domain 1), dolomite shows a substantial spread in both neighbor
470 pair and random pair misorientations (Fig. 9C). The lack of a discernible CPO is reflected in the
471 pole figure plots (Fig. 9D) in which the $\langle c \rangle$ axes of dolomite are concentrated into a broad girdle
472 of orientations sub-parallel to the main shear plane of the NSZ and there are no CPOs observed

473 for the $\langle a \rangle$ axes [2-1-10], rhomb planes (10-14) or f planes (01-12). The absence of a strong
474 CPO is also reflected in the very low M-index value for the fabric in this domain (0.082; Fig.
475 9C).

476 In the domains of incipient calcite formation (domain 2; Fig. 10), random pair
477 misorientation distributions of dolomite show a wide spread in values, while neighbor pair
478 misorientation distributions show strong peaks at misorientations corresponding to f-twin (78°)
479 and growth twin (180°) boundaries (Fig. 10B). The spread in neighbor pair misorientation
480 distributions is manifested as an intermediate M-index value of 0.202 for the dolomite fabric.
481 The twin-dominated CPO of dolomite is reflected in the pole figures as well (Fig. 10D). The
482 basal planes of the dolomite grains in this domain are strongly aligned sub-parallel to the
483 foliation, as evidenced by the strong concentration of poles to (0001) at almost right angles to the
484 mesoscopic foliation and near-alignment of the $\langle a \rangle$ axis along the foliation. The CPOs of the f-
485 planes (01-12) of the dolomite grains are characteristic of f-twinning, with 2 sets of orientation
486 clusters in close proximity and one set of overlapping orientations. The misorientation
487 distribution of calcite, on the other hand, is substantially different compared to that of dolomite,
488 with strong random pair misorientation peaks between 20° - 100° (Fig. 10C) while the neighbor
489 pair misorientation distribution shows a wider spread in orientations. The strong peaks of random
490 pair misorientations are reflected in a strong M-index value of 0.764 for the calcite fabric in this
491 domain. The $\langle c \rangle$ axis distributions of calcite (Fig. 10D) show a strong preferred orientation at a
492 high angle to the mesoscopic fabric. There is one set of overlapping $\langle c \rangle$ axis orientations
493 between calcite and dolomite (highlighted with a red circle in Fig. 10D) suggesting an
494 inheritance of orientations from dolomite to calcite. No CPOs are observed for the $\langle a \rangle$ axes,
495 rhomb or e-planes of calcite grains in this domain. Interphase misorientations between calcite

496 and dolomite were plotted to investigate the crystallographic relationships between calcite and
497 dolomite in the domains of incipient and advanced calcite formation (Figs. 10E&11E). In the
498 domains of incipient calcite formation, 37.59% of the boundaries between calcite and dolomite
499 are low-angle boundaries (Fig. 10E), suggesting growth of calcite on dolomite by fluid-mediated
500 interface-coupled dissolution-precipitation rather than deposition from a Ca²⁺-rich fluid (Putnis
501 et al., 2005; Ruiz-Agudo et al., 2014; Spruzeniece et al., 2017; Moore et al., 2024).

502 In domains where calcite formation is more advanced (domain 3; Fig. 11), calcite grains
503 form layers up to 1 mm wide and are texturally different compared to the domains of incipient
504 calcite development. There is a contrast in fabric strength (Figs. 11B&C) as well, with dolomite
505 characterized by a strong fabric (M-index 0.9) compared to a weak fabric for calcite (M-index
506 0.11), although the high M-index of dolomite in this domain is strongly influenced by the
507 presence of a single, large, twinned grain of dolomite. There is no observable CPO for the calcite
508 grains in this domain (Fig. 11D) and the mean grain size of calcite is quite small (25.47 μm). The
509 relatively small grain size in conjunction with the absence of a strong CPO leads us to conclude
510 that calcite grains in this domain deformed primarily by diffusion creep accommodated by grain
511 boundary sliding. The substantially lower proportion of low angle boundaries between calcite
512 and dolomite in this domain (6.21%, Fig. 11E) can be attributed to the onset of diffusion creep in
513 calcite, wherein initial low-angle boundaries were subsequently modified by grain rotation to
514 high angle configurations.

515 To summarize the results of our EBSD-based textural analyses, strain accommodation
516 within the carbonates of the NSZ was initially restricted to twinning in coarse grained dolomite,
517 followed by fracturing and comminution of the twinned dolomite grains. Fracturing of the
518 twinned dolomite resulted in the formation of fine-grained aggregates of dolomite (domain 1)

519 that were subsequently reorganized into high-strain zones, as well as the development of a
520 fracture network that created high permeability pathways for subsequent ingress of H₂. Although
521 the relatively higher proportion of low-angle boundaries in the relict dolomite grains from the
522 high strain domain may suggest an origin of the fine-grained dolomite aggregate by SGR
523 recrystallization, we do not observe a similar distribution of low angle boundaries in coarser
524 dolomite grains from domain 2 and 3 that contain abundant growth and f-twins. Additionally,
525 misorientation profiles constructed across relict dolomite grains in domain 1 (Fig. 9B) show
526 sharp peaks in misorientation which are more commonly indicative of healed microcracks rather
527 than the pervasive operation of SGR recrystallization. This further suggests a fracture-based
528 origin for the fine-grained dolomite aggregates observed in domain 1.

529 While Raman spectra of fluid inclusions (Section 3.2) confirm the presence of H₂ in all
530 the microstructural domains described previously, the products of carbonate reduction are
531 variable. Within the high strain domains (domain 1), calcite is noticeably absent with dolomite
532 being associated with graphite and magnetite, whereas in the other microstructural domains,
533 calcite is present as a product of interface-coupled dissolution of dolomite (domain 2), or as fine-
534 grained aggregates in high-strain zones around dolomite (domain 3), in association with graphite
535 and magnetite. Therefore, graphite and magnetite are interpreted to have formed at multiple
536 stages across the dolomite-calcite transformation, and thermodynamic models presented in the
537 next section have been developed to explain these observations.

538

539 **3.4 Thermodynamic modelling**

540 The microstructurally constrained chronology of carbonate transformation in the studied
541 samples indicates a progressive transition from Fe-bearing dolomite to Fe-poor dolomite and

542 finally to Ca-carbonate (calcite). Graphite and magnetite are found in microstructural association
543 with Fe-poor dolomite in the high-strain domains and with calcite in domains where carbonate
544 transformation has proceeded to completion.

545 In order to constrain the conditions of dolomite transformation and the role of reducing
546 fluids like H₂ and CH₄ in facilitating this transformation, we calculated thermodynamic models
547 of H₂-dolomite interactions at 1 and 2 GPa, and in the 300 °-800 °C temperature range (Figs.
548 12&13). The two considered pressures represent the peak metamorphic pressure of the
549 Lawsonite eclogite-facies unit of Alpine Corsica (about 2 GPa; Vitale Brovarone et al., 2013),
550 and a retrograde pressure condition (1 GPa), consistent with the minimum temperature
551 conditions of the antigorite stability field (380-400 °C, Schwartz et al., 2013) according to the
552 tentative retrograde path suggested by Vitale Brovarone et al. (2011).

553 At 1 GPa and temperatures above ~460 °C, the addition of H₂ to dolomite results in the
554 formation of an assemblage consisting of dolomite + brucite + graphite. This might reflect the
555 observed chemical zonation in dolomite from the high strain domains in sample COR18-7a (Fig.
556 6B). Dolomite grains in this domain show progressive Fe depletion from core to rim but are not
557 associated with calcite. Increasing H₂ in the system is followed by the addition of calcite to this
558 assemblage with further addition of H₂ leading to the formation of magnetite (Fig. 12A). The
559 formation of magnetite is a consequence of the destabilization of the ferroan component of
560 dolomite after approximately adding 1.5 moles of H₂ to the system (Fig. 12B&C). At
561 temperatures below ~460 °C, the progressive addition of H₂ results in the formation of dolomite
562 + calcite + brucite + graphite, and subsequently magnetite. In this case, the thermodynamic
563 model predictions reflect the preserved assemblages in all the microstructural domains except the

564 dolomite-rich high-strain zones (domain 1), with calcite, graphite, and magnetite in intimate
565 microstructural association with dolomite (See Figs. 6A, C&D).

566 At 2 GPa and between 400-600 °C, dolomite + aragonite is converted to an assemblage
567 of dolomite + brucite + aragonite + graphite with magnetite being introduced into the assemblage
568 after the addition of ~1.5 moles of H₂ (Figs. 13A-D). The overall evolution is similar to that
569 computed for 1 GPa, with the only major difference being the existence of the Ca-carbonate-free
570 assemblage (dolomite + brucite + graphite) at higher temperature (above 600 °C). The formation
571 of magnetite in both cases requires the addition of ~1.5 moles of H₂ to the system. The absence
572 of brucite in the samples from the NSZ may be explained by the high solubility of Mg at the
573 pressure-temperature (PT) conditions encountered by the rocks of the NSZ (see Vitale Brovarone
574 et al., 2017 for a similar interpretation) or by the presence of silica in the natural samples, which
575 may have favored the formation of silicates such as serpentine or diopside.

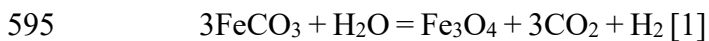
576 **4. DISCUSSION**

577 Our study investigates the link between H₂ infiltration and the microstructural evolution
578 of carbonates in a subduction zone setting, as testified by field, petrological, and microstructural
579 data in conjunction with thermodynamic modelling results. Depending upon the amount and the
580 PT conditions of H₂ infiltration, we see substantial differences in microstructure and overall
581 rheology within the rocks of the NSZ. These differences and their controlling parameters are
582 discussed in the following sections.

583 **4.1 Origin and conditions of formation of molecular hydrogen, methane, and** 584 **graphite**

585 Despite the substantial microstructural heterogeneity observed at the scale of our study,
586 the ubiquitous circulation of H₂ in all stages of the microstructural evolution of these rocks is

587 evidenced by the presence of H₂ in fluid inclusions in dolomite from both the brecciated and
588 fine-grained domains, as well as in calcite from the domains of advanced calcite formation
589 (Section 3.2). The origin of the H₂ in these rocks can be attributed to three processes: (i) aqueous
590 alteration of Fe²⁺ to Fe³⁺ during serpentinization contemporaneous with subduction [see Vitale
591 Brovarone et al. (2020) for equivalents in Alpine Corsica], (ii) the release of H₂ from preexisting
592 serpentinites [see Malvoisin et al. (2012) for equivalents in Alpine Corsica], or (iii) aqueous
593 alteration of the siderite component of carbonate minerals (Milesi et al., 2015; Tao et al., 2018),
594 following the generic reaction:



596 The first two processes would imply the infiltration of external, i.e., serpentinization- or
597 serpentinite-derived, H₂-rich fluids, whereas the third process would imply internally derived
598 fluids inside the carbonate pods. In our study area, we have documented the presence of H₂ in
599 fluid inclusions hosted within diopside veins in the serpentinite adjacent to the NSZ (see Table
600 1), in addition to the H₂-bearing inclusions in carbonates within the NSZ. In the diopside veins
601 adjacent to the NSZ, the molar fraction of H₂ relative to CH₄ is higher compared to carbonate-
602 hosted fluid inclusions within the NSZ, except for fluid inclusions within the brecciated
603 dolomite, which are gaseous and contain only H₂. The presence of H₂ outside the NSZ
604 carbonates and inside ultramafic rocks in their vicinity favors an external origin for the H₂ that is
605 responsible for the reduction of the studied carbonates. This feature suggests that at least a
606 fraction of the CH₄ observed in the carbonate pods formed through H₂-carbonate interactions
607 inside the pods. The presence of CH₄ in the diopside veins may be interpreted as a result of the
608 syn-serpentinization conversion of carbon-bearing species present in aqueous metamorphic
609 fluids to CH₄ during serpentinization (Vitale Brovarone et al., 2020; Boutier et al., 2021). The

610 presence of H₂ gaseous inclusions in the brecciated dolomite domains may testify to the incipient
611 infiltration of H₂ in response to fluid overpressure.

612 A subsequent phase of comparatively minor, in-situ H₂ production by aqueous alteration
613 of the siderite component of dolomite and self-reduction of dolomite to CH₄ is also suggested by
614 the presence of magnetite particles in reacted portions of the carbonate pods and the
615 microstructural relationships with calcite and graphite (Figs. 6 A-D). Our thermodynamic
616 modelling results indicate that the observed mineralogical evolution in the samples from the NSZ
617 can be explained by protracted infiltration of H₂-rich fluids at conditions either close to peak
618 metamorphic conditions (470-520 °C and 2-2.2 GPa, Vitale Brovarone et al., 2013), as well as
619 PT conditions experienced during retrograde cooling across ~460 °C and 1GPa (threshold
620 temperature for different reaction pathways as derived from thermodynamic modeling). The
621 presence of Ca-carbonate-free reacted domains suggests that, at least locally, the process started
622 at near peak metamorphic conditions (2 GPa), whereas the common presence of calcite in most
623 of the samples indicates that the process protracted (continuously or episodically) during
624 exhumation. The presence of antigorite in all the studied microstructures indicates that H₂
625 infiltration occurred at temperatures > 380-400 °C (Schwartz et al., 2013). It is possible to
626 estimate a maximum amount of H₂ infiltration using Figures 12 & 13. At 2 GPa and 500 °C and
627 at 1 GPa and 400 °C, 2 moles of H₂ are necessary to fully react 2 moles of dolomites to form
628 graphite + Ca-carbonate + magnetite + brucite + fluid. At these pressure and temperature
629 conditions, the molar volume of dolomite is almost similar (32.34 and 32.65 cm³/mol,
630 respectively), therefore the volume and mass of H₂ infiltrated are very similar. This implies that
631 full reaction of 1 cm³ of dolomite requires the infiltration of 0.06 g of H₂ at the considered
632 conditions, and 60 kg H₂ are required to completely reduce a cubic meter of dolomite.

633 The presence of graphite in the rocks from the NSZ can be attributed to the saturation of
634 carbon in the fluid after reduction of carbonate minerals during H₂ infiltration. Firstly, this is
635 suggested by the microstructural association of graphite and carbonate, with graphite forming
636 along preferential fluid pathways, such as grain and phase boundaries and fractures in the
637 carbonate phases, especially dolomite (Figs. 5B, E, F&G and Figs. 6A-D). This is also suggested
638 by the thermodynamic modelling results presented in Section 3.4. This mechanism of graphite
639 precipitation is equivalent to the one described by Malvoisin et al. (2012) and Galvez et al.
640 (2013) along a blueschist-facies serpentinite-metasediment contact of Alpine Corsica, or to the
641 H₂-mediated reduction of carbonated serpentinites in the Lanzo massif, Western Italian Alps
642 (Vitale Brovarone et al., 2017; Giuntoli et al., 2020). The partial conversion of CH₄ to graphite
643 following carbon saturation in the fluid may also be responsible for at least a fraction of the CH₄
644 found in fluid inclusions, following the generic reaction:



646 **4.2 Chronology of deformation and metamorphic events associated with H₂**
647 **infiltration and carbonate transformation along the Negru Shear Zone**

648 Based on our observations, we suggest, following the initial phase of dolomite
649 precipitation within the pods and veins of the NSZ, the resulting post-precipitation sequence of
650 chemical-mechanical events recorded by the carbonates within the studied samples:

651 1) An initial phase of deformation of dolomite above ~300 °C is suggested by the
652 abundance of twin boundaries in the brecciated fragments of dolomite. The development of f-
653 twins is the principal mode of strain accommodation in dolomite at temperatures ranging from
654 300°-600 °C (Barber and Wenk 1979; Barber et al. 1981) and the substantial proportion of f-
655 twins in the brecciated fragments of dolomite constrain the first phase of deformation to

656 temperatures in the aforementioned range. A temperature > 380-400 °C is also suggested by the
657 presence of antigorite (Schwartz et al., 2013). The preservation of antigorite without any
658 evidence of dehydration, in association with the carbonates, additionally constrains temperature
659 of deformation to less than 600 °C (Perrillat et al. 2005).

660 2) Subsequently, the twinned dolomite underwent a phase of brecciation forming a
661 fine-grained dolomite aggregate, which initiates a switch from a brittle to a semi-brittle
662 deformation regime (Gratier et al., 1999; Trepmann and Stöckhert, 2003; Füsseis and Handy,
663 2008; Menegon et al., 2013; Reber and Pec, 2018). Although the high-strain domains appear to
664 preserve possible textural evidence of SGR recrystallization (see Section 3.3, Fig. 9) in the form
665 of low-angle boundaries in relict dolomite grains, the low-angle boundaries within these relicts
666 do not define polygonal subgrains, are associated with sharp changes in misorientation (Fig. 9B)
667 and may well represent undulatory extinction/lattice distortions and/or healed microcracks and it
668 is plausible that the smaller grains within the high-strain domains originated as fragments
669 derived from fracturing of the coarser grains of dolomite. As highlighted previously, the
670 persistence of antigorite and the presence of twins in dolomite coupled with previous
671 temperature constraints (430-470 °C) obtained using Raman spectroscopy of carbonaceous
672 material on metasedimentary lithologies from the same lawsonite-eclogite unit (Vitale Brovarone
673 et al. 2013) constrain deformation temperatures to a range not suitable for the operation of
674 dynamic recrystallization of dolomite (Higgs and Handin 1959; Barber et al. 1994; Newman and
675 Mitra 1994; Delle Piane et al. 2009; Holyoke et al. 2013, 2014; Berger et al. 2016; Li et al.
676 2021). In detail, dolomite deforms by climb-assisted deformation above 600 °C (Barber et al.,
677 1981), whereas twinning is the dominant mode of strain accommodation between 300-600 °C
678 (Newman and Mitra, 1994). Fine-grained high-strain domains can originate in two ways: 1) they

679 may form as a result of fluid-induced dynamic recrystallization (Newman and Mitra, 1994;
680 Rutter and Brodie, 1995; Pennacchioni and Mancktelow, 2007; Fousseis and Handy, 2008;
681 Goncalves et al., 2016) or, alternatively, 2) an initial phase of entirely brittle deformation can
682 lead to the formation of a finer-grained matrix in which strain can subsequently localize (Imon et
683 al. 2004; Brander et al. 2012; Menegon et al. 2013; Moore et al. 2024). Based on temperature
684 constraints obtained from the thermodynamic modelling presented in Section 3.4 coupled with
685 our textural observations presented in Section 3.3, we attribute the grain size reduction of
686 dolomite to fluid-induced fracturing, which led to the subsequent development of the high strain
687 domains within the fine-grained dolomite layers. The relict dolomite grains in the high strain
688 zones contain H₂-bearing fluid inclusions which points to the initial brecciation of the dolomite
689 being induced by fluids containing H₂, although the absence of calcite suggests that proportions
690 of H₂ within the fluid must have been low, at least initially (less than 2 moles of added H₂; see
691 Fig. 12). Once a pronounced grain size contrast was established between the coarse-grained
692 brecciated dolomite and finer-grained dolomite within the high strain domains, a strength
693 contrast was established such that subsequent deformation localized within the weaker fine-
694 grained shear zones, whereas the stronger angular breccia fragments deformed locally by twin
695 gliding and minor crystal plasticity. These observations correlate well with previous studies on
696 natural examples of fine-grained dolomite shear zones (Newman and Mitra, 1994), in which the
697 effect of grain size variations in the localization of strain in dolomite aggregates was established.
698 Additionally, continuous viscous deformation of the fine-grained dolomite grains around the
699 rigid lenses of the coarse-grained brecciated dolomite leads to the build-up of stresses, eventually
700 triggering brittle failure of these lenses (Fagereng and Diener 2014). The newly formed

701 transgranular fractures in the twinned dolomite fragments increased the secondary porosity,
702 leading to later and renewed infiltration of H₂-rich fluids.

703 3) Based on our thermodynamic modelling of Section 3.4, the latter stages of H₂
704 infiltration led to the reduction of dolomite to calcite + graphite + magnetite, with the proportion
705 of calcite directly linked to the amount of H₂ infiltration. A distinction is made between the
706 different phases of H₂ infiltration based on the presence of calcite + graphite + magnetite. This
707 assemblage is noticeably absent in the high-strain domains but is ubiquitous around the margins
708 of brecciated dolomite and along the tips of transgranular fractures within twinned dolomite
709 fragments. Our modelling results indicate that the assemblage of calcite + graphite + magnetite
710 can be derived from dolomite in the presence of H₂-rich fluids at temperatures between 300-600
711 °C at 2GPa (near peak metamorphic conditions) or close to 460 °C at 1GPa (retrograde path). At
712 these temperatures, deformation of dolomite is limited to twin gliding and minor crystal
713 plasticity accommodated by low angle boundaries without substantial dislocation motion (Higgs
714 and Handin 1959; Barber and Wenk 1979; Barber et al. 1981, 1994; Newman and Mitra 1994;
715 Delle Piane et al. 2009; Holyoke et al. 2013, 2014; Berger et al. 2016; Li et al. 2021). On the
716 other hand, calcite is rheologically weaker compared to dolomite at those temperatures (Viola et
717 al. 2006; Delle Piane et al. 2009; Kushnir et al. 2015), and based on our microstructural
718 observations from the domains of advanced calcite formation, we can conclude that infiltration
719 of H₂-rich fluids at temperatures ~460 °C and 1GPa led to the formation of a layer of fine-
720 grained calcite into which strain was localized by a combination of grain boundary sliding and
721 diffusion creep (see Section 3.3 and Fig. 14 for a summary).

722 Based on our results, we can identify two alternative mechanisms by which H₂ infiltration
723 can influence the deformation of carbonate-rich lithologies. The first mechanism involves fluid

724 overpressure at temperatures above 460 °C and P of ~1GPa, which leads to an initial phase of
725 brecciation of dolomite and formation of a fine-grained aggregate wherein subsequent strain is
726 localized, forming shear zones. The second mechanism involves deformation at conditions at
727 which calcite stability is enhanced in the presence of H₂ (see Section 3.4), leading to a phase
728 transition from dolomite to calcite and resulting in substantial rheological weakening due to
729 viscous flow of calcite at these pressure and temperature conditions. Both brittle and ductile
730 phases of deformation within the NSZ can be attributed to the infiltration of H₂-rich fluids, but at
731 different stages and PT conditions. This also explains the variability in calcite occurrence in
732 different domains as the proportion of calcite formed from dolomite is dependent on the
733 efficiency of hydrogen ingress.

734 Thus, the rocks collected in close association with the NSZ constitute a unique instance
735 of deformation induced by infiltration of H₂-rich fluids and provides insights into the
736 mechanisms by which H₂ can circulate along a subduction interface shear zone.

737 **4.3 Mechanical effects of H₂ infiltration**

738 The effect of H₂ infiltration into crustal rocks is relatively unknown barring a couple of
739 recent studies (Giuntoli et al., 2020, 2024) whereas the effect of infiltrating H₂ as a facilitator for
740 embrittlement processes such as stress corrosion cracking of metals has been known for some
741 time (Su et al., 1992; Scully et al., 2012; Robertson et al., 2015; Braid et al., 2024). While brittle
742 creep of geological materials linked to stress corrosion induced by H₂O is well known (Anderson
743 and Grew, 1977; Michalske and Freiman, 1982; Atkinson, 1984; Brantut et al., 2013), there
744 currently exists a knowledge gap between laboratory results of H₂ embrittlement and corrosion
745 cracking and their potential manifestation in natural environments such as natural H₂-rock
746 reactions or underground hydrogen storage (UHS) in geologic reservoirs. Although this study

747 could not provide evidence for direct analogues of the typical microstructures of H₂
748 embrittlement and corrosion cracking as observed in material science experiments, our results
749 provide an opportunity to assess the potential contribution of these processes in a geologic
750 environment. The brecciated dolomite grains within the NSZ probably underwent multiple
751 phases of brittle deformation, as evidenced by multiple generations of transgranular and
752 intragranular fractures of varying orientations (see Section 3.1). While some of the fracture
753 generations might be related to stress buildups along rigid lenses of carbonate and others to fluid
754 overpressure, the morphologies of some intragranular fractures appear well-correlated with those
755 formed due to H₂-induced stress corrosion cracking (Woodtli and Kieselbach, 2000). Within the
756 domains of incipient calcite formation, we observe calcite nucleation at the boundaries between
757 fragments, where they are intersected by transgranular fracture tips (Figs. 5F, 6A). A key
758 criterion for the identification of stress corrosion cracking is the localization of reactions at the
759 leading edge of fracture tips, instead of alteration along the entire length of the fracture owing to
760 enhanced access of reactants (Anderson and Grew, 1977; Barnett and Kerrich, 1980; Brantut et
761 al., 2013), which correlates well with our observations of incipient calcite nucleation at the
762 margins of the dolomite breccia fragments. The net effect of the stress corrosion cracking of
763 dolomite and subsequent formation of calcite is a positive volume change of 16% owing to the
764 volume change resulting from the conversion of one molecule of dolomite to two molecules of
765 calcite (Michałowski and Asuero 2012). This volume increase accompanying the transformation
766 of dolomite to calcite may lead to the development of a positive feedback loop as progress of the
767 reaction leads to the formation of newer fractures that enhance permeability (Yoshida et al.,
768 2020; Renard, 2021).

769 5. CONCLUSIONS

770 The rocks of the NSZ constitute a unique natural laboratory to study deep H₂ migration in
771 the crust and the chemical-mechanical effects of H₂ ingress within crustal carbonates. The
772 carbonates of the NSZ preserve a continuum of microstructures attesting to deformation
773 occurring in both the brittle and ductile regimes, with a pronounced role of H₂-rich fluids as a
774 common influence across all stages. An initial phase of brittle deformation of dolomite occurred
775 due to overpressure developed by the accumulation of H₂-rich fluids formed by serpentinization
776 occurring in the vicinity of the NSZ. The brittle fracturing of the dolomite grains had two
777 important implications:

778 1) The formation of a finer-grained matrix that weakened the rock and localized
779 subsequent strain.

780 2) Formation of high permeability pathways that facilitated the infiltration of H₂
781 which led to the formation of calcite+graphite+magnetite. In addition to brittle deformation due
782 to fluid overpressure, at least one phase of intra and intergranular fractures in dolomite have
783 morphologies typical of those produced by H₂-induced stress corrosion cracking. These
784 mechanisms are well established in material sciences, but still not documented in natural rocks.

785 Transformation of dolomite to calcite during exhumation appears to be directly correlated
786 with the amount of infiltrating H₂ and varies even on the scale of a single thin-section, with
787 different identifiable textural domains ranging from domains of incipient calcite formation to
788 domains of advanced calcite formation. Our thermodynamic results indicate that full reaction of
789 1 m³ dolomite required the infiltration of ~60 kg of H₂, and these calculations provide a first-
790 order estimate of paleo-H₂ fluxes in subducted lithosphere. A strong rheological contrast is
791 established in the domains of advanced calcite formation owing to the pronounced difference in
792 strength between calcite and dolomite. The coarse dolomite grains in this domain preserve

793 limited imprints of plastic deformation, whereas the calcite grains in this domain preserve
794 evidence of having undergone substantial viscous flow and strain localization due to diffusion
795 creep. Our results show that reducing fluids such as H₂ can lead to pronounced strain localization
796 without any substantial variations in the prevailing stress regime, by inducing phase
797 transformations in carbonates. Lastly, our results suggest substantial reactivity between
798 carbonates and H₂ at PT conditions relevant for subduction zone settings, and experiments
799 should be performed to understand if such reactivity persists at lower PT conditions, before
800 large-scale underground hydrogen storage (UHS) applications are pursued in carbonate
801 reservoirs.

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1214 FIGURE CAPTIONS

1215 Figure 1. Geological setting of the study area. A) Map showing the distribution of peak Alpine
1216 metamorphic conditions in Alpine Corsica modified after Vitale Brovarone et al. (2013). The
1217 study area is highlighted with a red star. B) Lithological map of the study area and its
1218 surroundings showing the broad distribution of rock types, modified after Rossi et al. (1994) and
1219 Lahondère (1983).

1220 Figure 2. Structural setting of the Negru Shear Zone. A) Panoramic view of the Negru Shear
1221 Zone. B) Sketch of the first-order structural arrangement of the Negru Shear Zone. C) Photo of
1222 the studied segment of the Negru Shear Zone (GPS coordinates: 42.76328°N, 9.338714°E) with
1223 major structural elements highlighted. D) Sketch of the outcrop shown in C, illustrating the
1224 structural relationships between different generations of veins, graphitized deformation bands
1225 and the main foliation within the serpentinite.

1226 Figure 3. Examples of carbonate distribution and transformations on the upper serpentinite block
1227 of the Negru Shear Zone (see Fig. 2 C-D for location of panels A-B, C-D and E-F). In the
1228 legend, carbonate refers to mixed dolomite and calcite-bearing domains and for
1229 carbonate+graphite bearing domains, the shading in the legend is indicative of the relative
1230 abundance of graphite (darker shades reflect higher abundance). A) Extensive, chaotic network
1231 of dolomite-rich veins (yellow/brown) transected by a later stage calcite vein (white). B)
1232 Interpretative sketch of the structural relationships of A. C) Brecciated and partially reacted
1233 (calcite + graphite) dolomite vein cross-cutting the host serpentinite. The locations of samples
1234 COR18-7a and 7b are shown black boxes. D) Sketch of the structural relationships of C. Note the
1235 calcite vein cross-cutting both the serpentinite and the deformation band. E) Strongly graphitized
1236 deformation band within serpentinite. Truncated calcite veins are drawn into parallelism with the
1237 band, suggesting their involvement in the activity of the deformation band. F) Sketch of the
1238 structural elements of E. G) Slab of layered domains preserving different degrees of reaction,
1239 from partially reacted, brecciated dolomite to a fully homogenized layer composed of dolomite +
1240 graphite. An intermediate layer of carbonate+graphite+serpentinite clasts is also observed. The
1241 locations of sample 3COR19-20Zb and 20Zc are shown in black boxes. H) Sketch of the

1242 structural elements of G. Note the pods of the brecciated dolomite in the lowermost carbonate-
1243 rich horizon.

1244 Figure 4. Thin-section photomicrographs and cathodoluminescence intensity maps. Yellow
1245 boxes correspond to the textural domains shown in greater detail in Fig. 5 and green boxes
1246 correspond to domains from which the BSE images in Fig. 6 and the EBSD maps in Figs. 9-11
1247 were acquired. Mineral abbreviations from Whitney and Evans (2010). A) Cross-polarized
1248 photomicrograph of the host serpentinite (COR18-7b). B) Cross-polarized photomicrograph of a
1249 brecciated dolomite pod (COR18-7a) within a graphitized deformation band from the Negru
1250 Shear Zone. C) Cathodoluminescence intensity map of COR18-7a. Brighter domains correspond
1251 to calcite-rich domains, darker domains correspond to dolomite-rich domains. The serpentinite
1252 clasts are black. D-F) Cross-polarized photomicrograph (D&F) and cathodoluminescence
1253 intensity map (E) of interface between serpentinite and carbonate vein (3COR19-20Zb&c; See
1254 Fig. 3G for sample locations). The thin sections are flipped horizontally with respect to their
1255 schematic positions shown in Fig. 3G.

1256 Figure 5. Microphotographs of the representative microstructural domains highlighted with
1257 yellow and green boxes in Fig. 4. A) Foliated serpentine from the host serpentinite (COR18-7b).
1258 A kink band is observable with the trace of the axial plane normal to the foliation. B) Brecciated
1259 fragments of twinned dolomite (COR18-7a) containing twins of variable widths. The fragments
1260 are lined by graphite, magnetite and calcite. C) Relict twinned dolomite grain surrounded by a
1261 recrystallized matrix of finer-grained dolomite (3COR19-20Zb). This domain corresponds to
1262 domain 1 (dolomite-rich high-strain zone) of the EBSD-based textural analyses presented in
1263 Section 3.3 below. D) Recrystallized carbonate in association with porphyroclasts of the host
1264 serpentinite (3COR19-20Zb). E) Relict, twinned dolomite grains surrounded by a matrix of fine-

1265 grained dolomite with an intervening fringe of calcite, graphite and magnetite (3COR19-20Zb).
1266 This domain corresponds to domain 2 (zone of incipient calcite formation) of the EBSD-based
1267 textural analyses presented in Section 3.3. EDS maps of this domain are shown in Figs. 7E-H. F)
1268 Calcite, graphite and magnetite at junctions between brecciated fragments of twinned dolomite
1269 (COR18-7a). G) Transition between relict, twinned dolomite and highly recrystallized calcite in
1270 domains of advanced calcite formation (COR18-7a). This domain corresponds to domain 3 (zone
1271 of advanced calcite formation) of the EBSD-based textural analyses presented in Section 3.3.
1272 EDS maps of this domain have been presented in Figs. 7I-L. H) Sheared diopside layer crosscut
1273 by late-stage carbonate veins (3COR19-20Zc).

1274 Figure 6. BSE images of representative textural domains. A) Incipient growth of calcite at the
1275 intersection between brecciated dolomite fragments (see Figs. 4B&C and 5F for precise location
1276 of the domain). B) Fine-grained dolomite (see Figs. 4B&C for domain correlation). Individual
1277 grains of dolomite become progressively darker from core to rim, corresponding to a decrease in
1278 Fe/Mg ratio due to Fe loss. EDS maps from this domain have been presented in Figs. 7A-D. C)
1279 Fringe of incipient calcite growing onto twinned dolomite, in association with graphite and
1280 magnetite, transitioning laterally into a domain of recrystallized dolomite (see Figs. 4D&E and
1281 5E for domain correlation). D) Domain of advanced calcite formation within which fragments of
1282 dolomite are observed to be completely engulfed within zones of calcite growth (see Figs. 4D&E
1283 for domain correlation).

1284 Figure 7. EDS maps of distributions of Ca, Mg and Fe from representative textural domains. The
1285 brightness in each map is directly correlated with the abundance of the element. A-D) High-
1286 strain, fine-grained, dolomite-rich domain from COR18-7a. The location of this domain is shown
1287 in Figs. 4B&C (green box corresponding to Fig. 6B) and detailed microstructural relationships

1288 have been described in Fig. 6B. Note the increase in Mg at the margins of the dolomite
1289 fragments corresponding to a decrease in the Fe content. E-H) domain of incipient calcite
1290 formation from 3COR19-20Zb. The location of this domain has been highlighted in Figs. 4D&E
1291 (green box corresponding to Figs. 5E and 6C), and the microstructural relationships have been
1292 described in Figs. 5E&6C. I-L) Domain of advanced calcite formation from COR18-7a. The
1293 location of this domain is shown in Figs. 4B&C (green box corresponding to Fig. 5G), and
1294 detailed microstructural relationships have been described in Fig. 5G. Inclusions of unreacted
1295 dolomite are engulfed within the calcite-rich layer (J&K).

1296 Figure 8. Representative Raman spectra of species of interest. A) photomicrograph of fluid
1297 inclusion trails and clusters within brecciated dolomite. B) Raman spectra of the fluid inclusion
1298 highlighted with an arrow in A. Enhanced view of characteristic peaks for H₂ in the inset. C)
1299 photomicrograph of fluid inclusions within relict twinned and brecciated fragment of dolomite
1300 inside a high-strain domain. D) Raman spectra of the fluid inclusion highlighted with an arrow in
1301 C. Enhanced view of characteristic peaks for H₂ in the inset. E) photomicrograph of fluid
1302 inclusion trail within calcite from the domains of advanced calcite formation. F) Raman spectra
1303 of the fluid inclusion highlighted with an arrow in E. Enhanced view of characteristic peaks for
1304 H₂ in the inset. G) Plane-polarized photomicrograph of a diopside grain with fluid inclusions. H)
1305 Raman spectra obtained from a fluid inclusion highlighted with a black arrow in (G), with
1306 representative Raman peaks for diopside. CH₄ and H₂ highlighted. Note the presence of CH₄ in
1307 B, F and H and its absence in D. Mineral abbreviations from Whitney and Evans (2010).

1308 Figure 9. EBSD maps and pole figures from high strain domains (domain 1) consisting of fine-
1309 grained dolomite and no calcite. (A) GROD map. The white line transecting the relict dolomite
1310 grain in the center corresponds to the misorientation profile shown in B. (B) GROD map with

1311 misorientation profile in the inset. Cumulative misorientations (misorientation at a point relative
1312 to the misorientation at the first point) and point-to-point misorientations (misorientation at each
1313 point) are shown as red and blue curves, respectively. (C) Neighbor pair (blue) and random pair
1314 (yellow) misorientation angle distribution. (D) Pole figures of dolomite.

1315 Figure 10. EBSD maps and pole figures from domains of incipient calcite formation (domain 2)
1316 consisting of relict and fine-grained dolomite along with calcite at the interface between the relict
1317 and fine-grained dolomite domains. (A) GROD map (B&C) Neighbor pair (blue) and random
1318 pair (yellow) misorientation angle distributions of dolomite and calcite respectively. (D) Pole
1319 figures of dolomite (top) and calcite (bottom). (E) Map of interphase misorientations between
1320 calcite and dolomite.

1321 Figure 11. EBSD maps and pole figures from domains of advanced calcite formation (domain 3)
1322 consisting of relict, twinned grains of dolomite adjacent to domain consisting of fine-grained
1323 calcite engulfing isolated grains of dolomite. (A) GROD map (B&C) Neighbor pair (blue) and
1324 random pair (yellow) misorientation angle distributions of dolomite and calcite respectively. (D)
1325 Pole figures of dolomite (top) and calcite (bottom). (E) Map of interphase misorientations
1326 between calcite and dolomite.

1327 Figure 12. Thermodynamic modeling at 1 GPa. A) pseudosection. B) dolomite proportions
1328 (mol%) in dolomite. C) ankerite proportions (mol%) in dolomite. D) relative amount of Ca in
1329 carbonates.

1330 Figure 13. Thermodynamic modeling at 2 GPa. A) Pseudosection. B) Dolomite proportions
1331 (mol%) in dolomite. C) Ankerite proportions (mol%) in dolomite. D) Relative amount of Ca in
1332 carbonates.

1333 Figure 14. Schematic diagram depicting the progressive transformation of initial coarse-grained
1334 dolomite in yellow to fine grained Ca-carbonate+graphite in grey with increasing H₂ infiltration
1335 over time.

1336 TABLE CAPTIONS

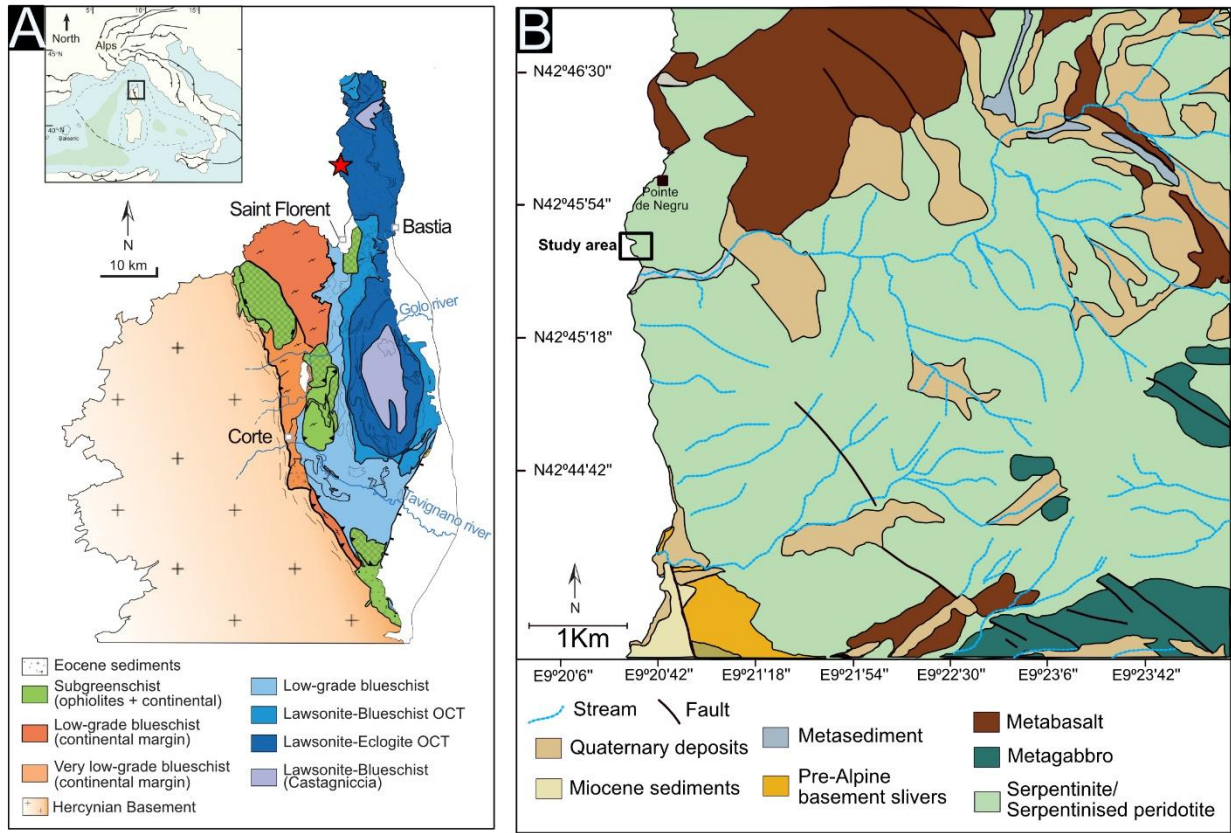
1337 Table 1. Qualitative and quantitative micro-Raman spectroscopy results on six pre-selected fluid
1338 inclusions from the NSZ area. The table includes host mineral, fluid composition based on
1339 Raman analysis, CH₄ (v1) and H₂ (Q1(1)) Raman shifts (expressed in cm⁻¹), CH₄ and H₂ peak
1340 areas, and molar fractions in the gas phase (expressed in mol%). The peak areas were computed
1341 after subtraction of a linear baseline. The CH₄ and H₂ molar fractions were calculated using their
1342 respective peak areas and Raman scattering cross-sections reported by (Burke, 2001) for a green
1343 514 nm laser.

1344 ¹Supplemental Material. [*Materials supplementary to the manuscript*] Please visit
1345 <https://doi.org/10.1130/gsab.s.29189777> to access the supplemental material, and contact
1346 editing@geosociety.org with any questions.

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Fig. 1



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Fig. 2

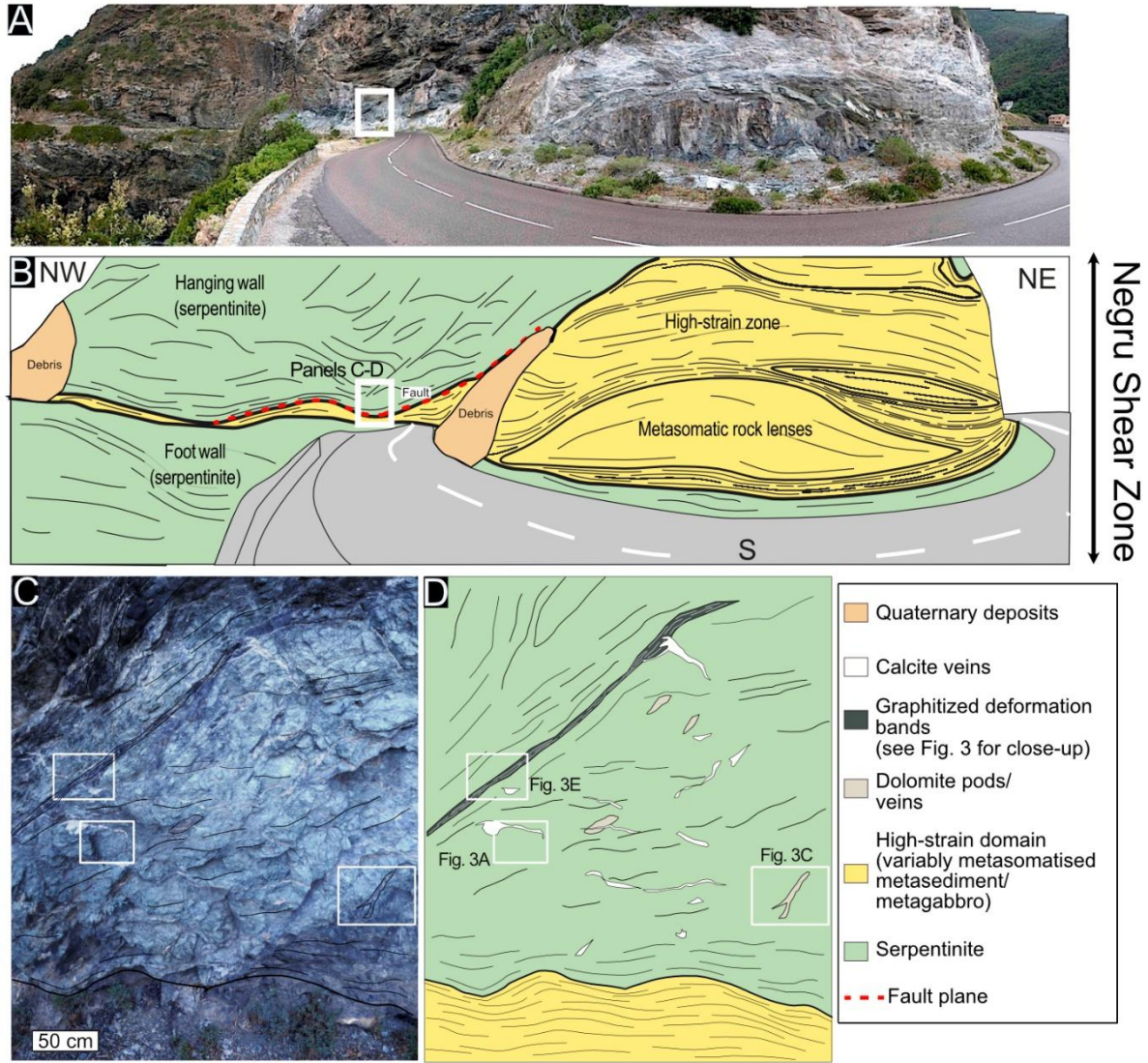
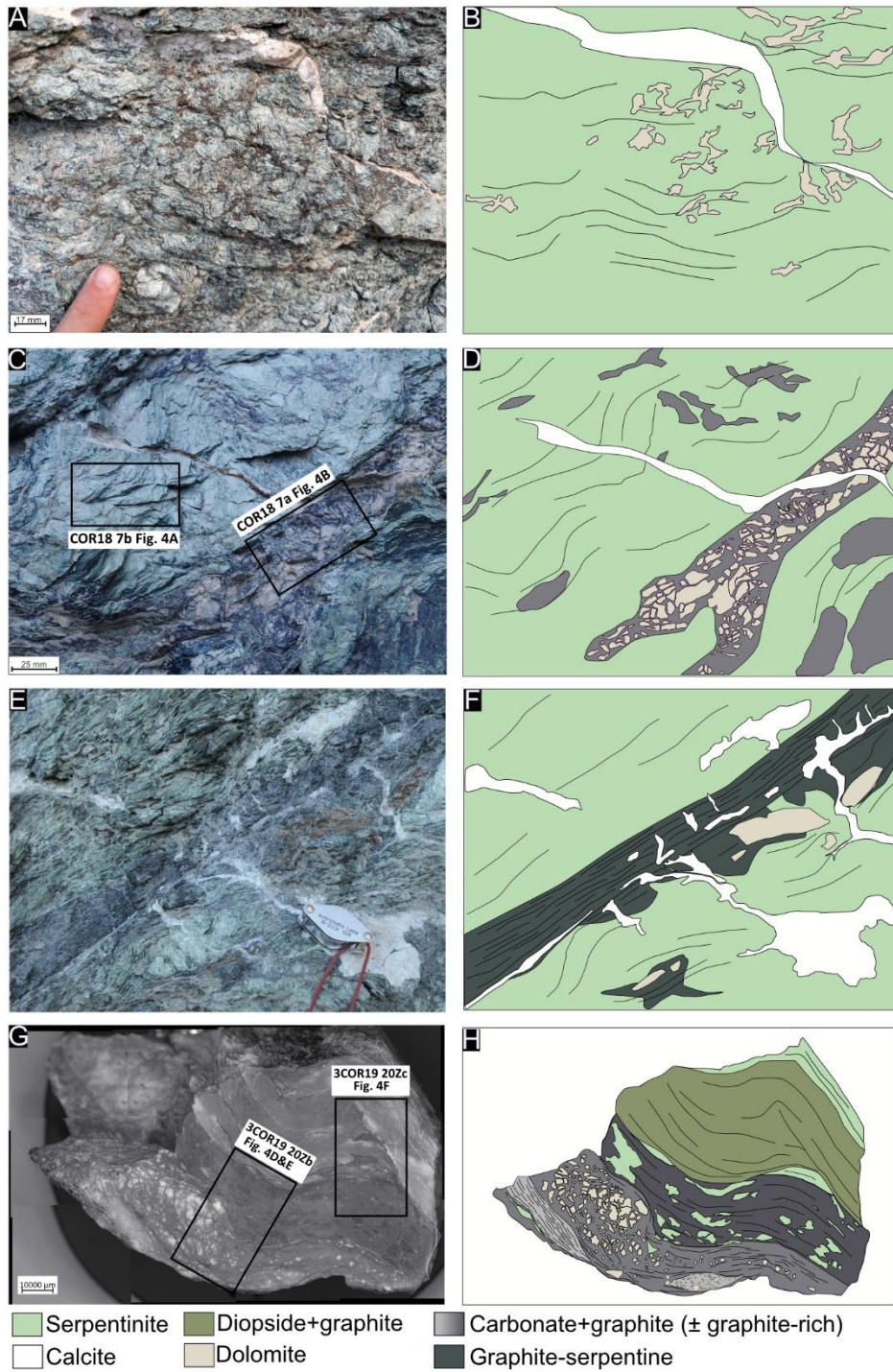
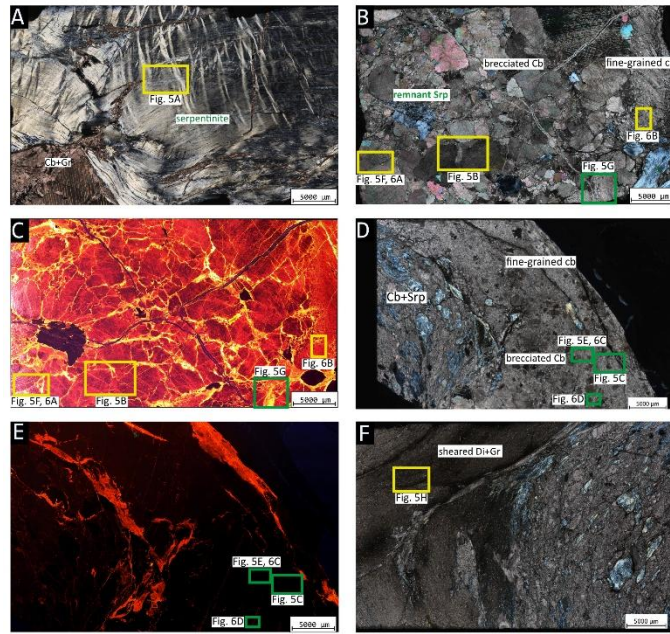


Fig. 3



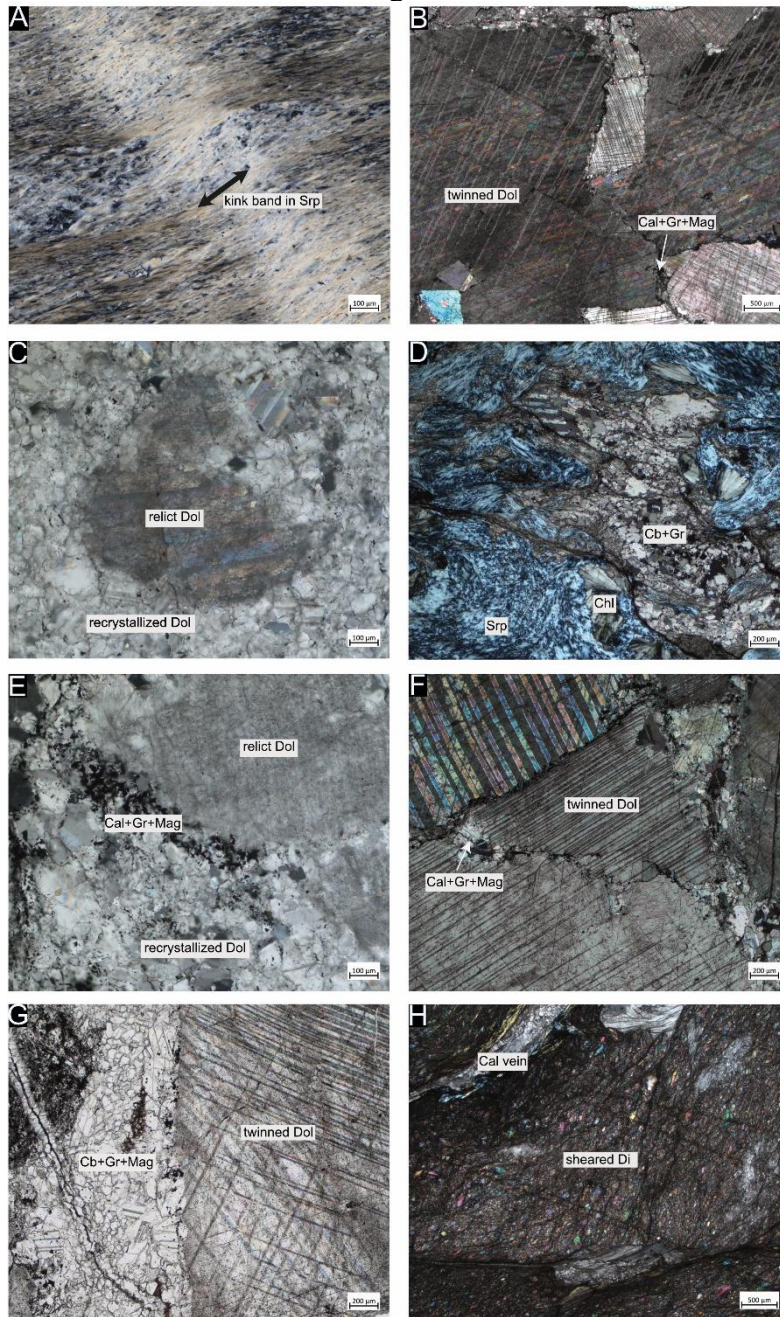
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Fig. 4



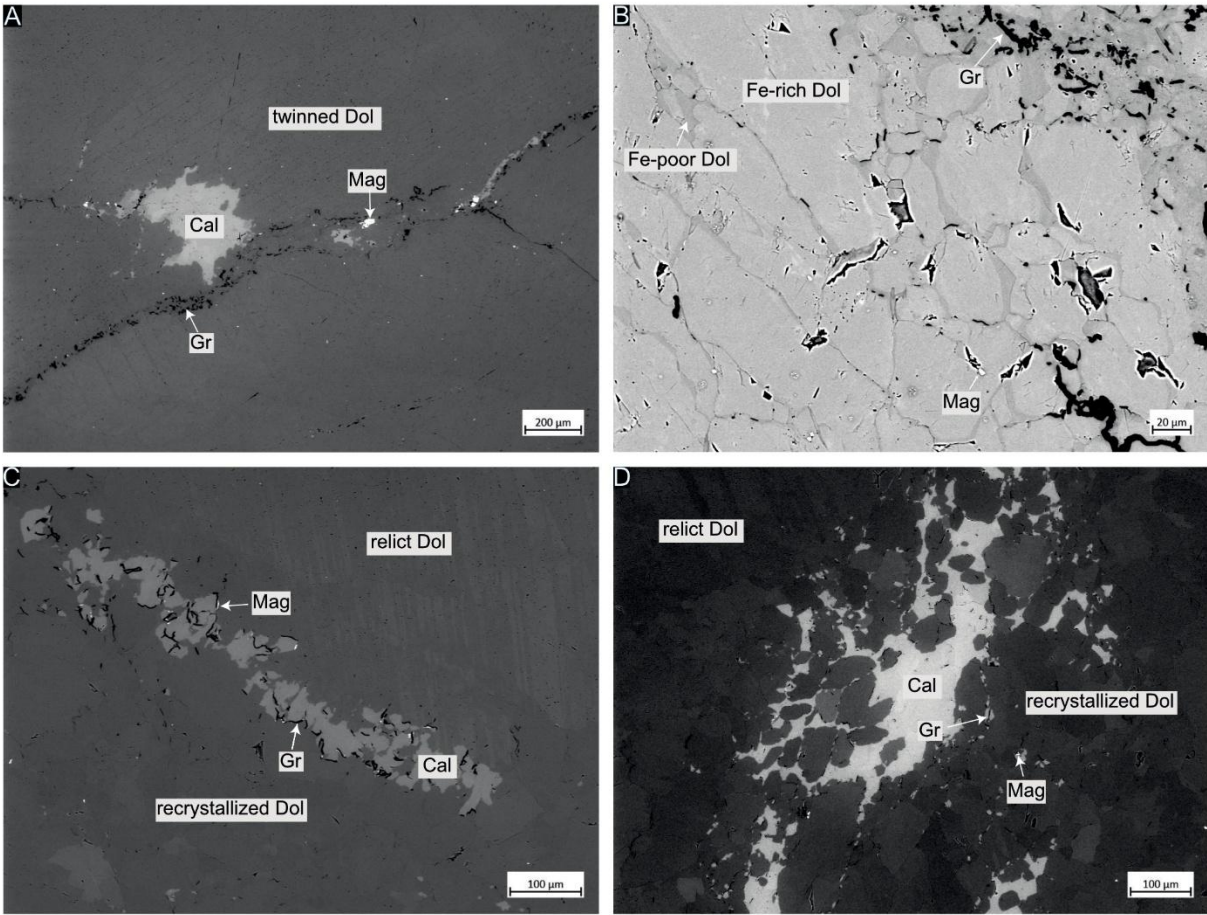
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Fig. 5



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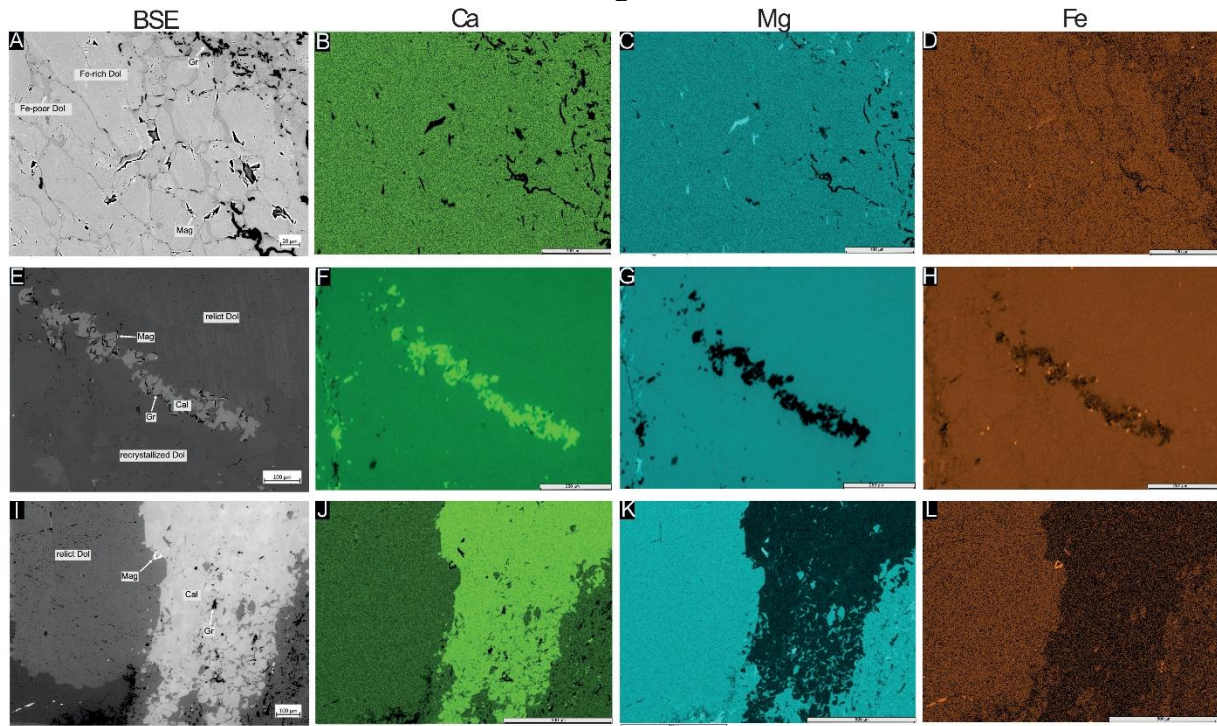
Fig. 6



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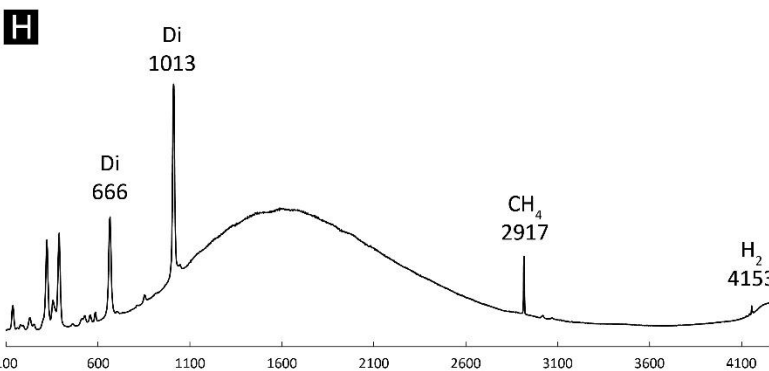
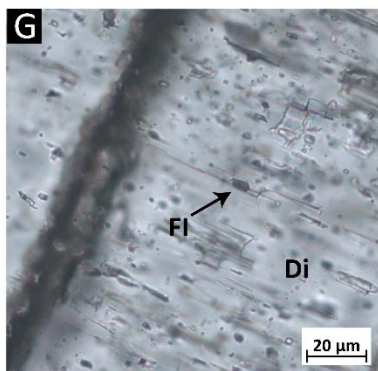
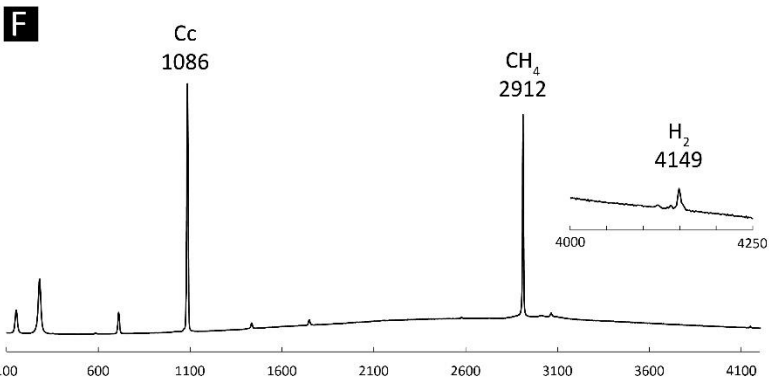
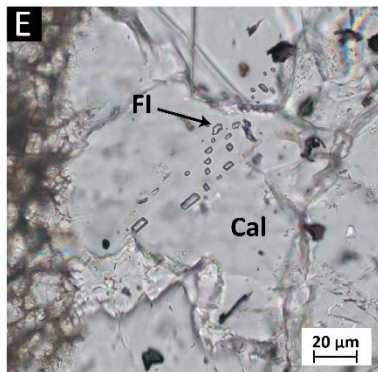
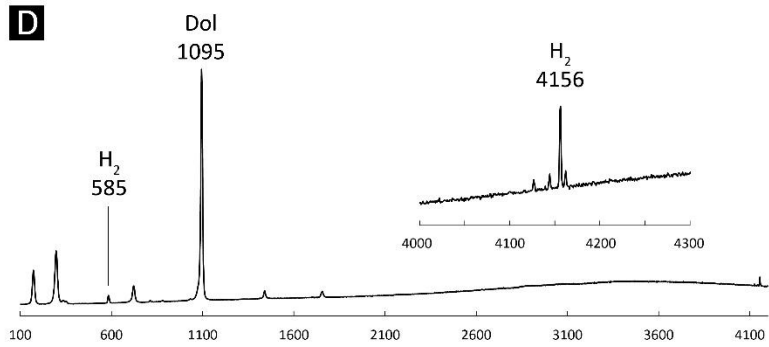
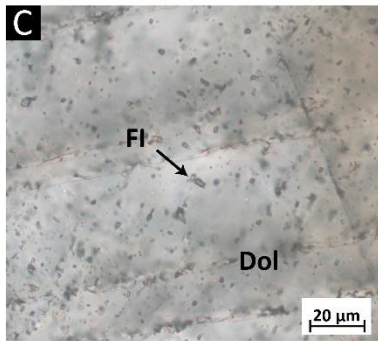
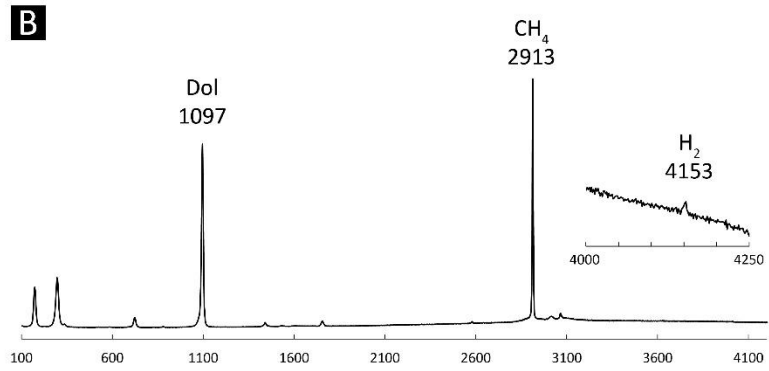
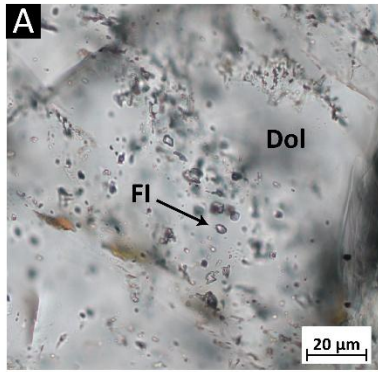
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Fig. 7



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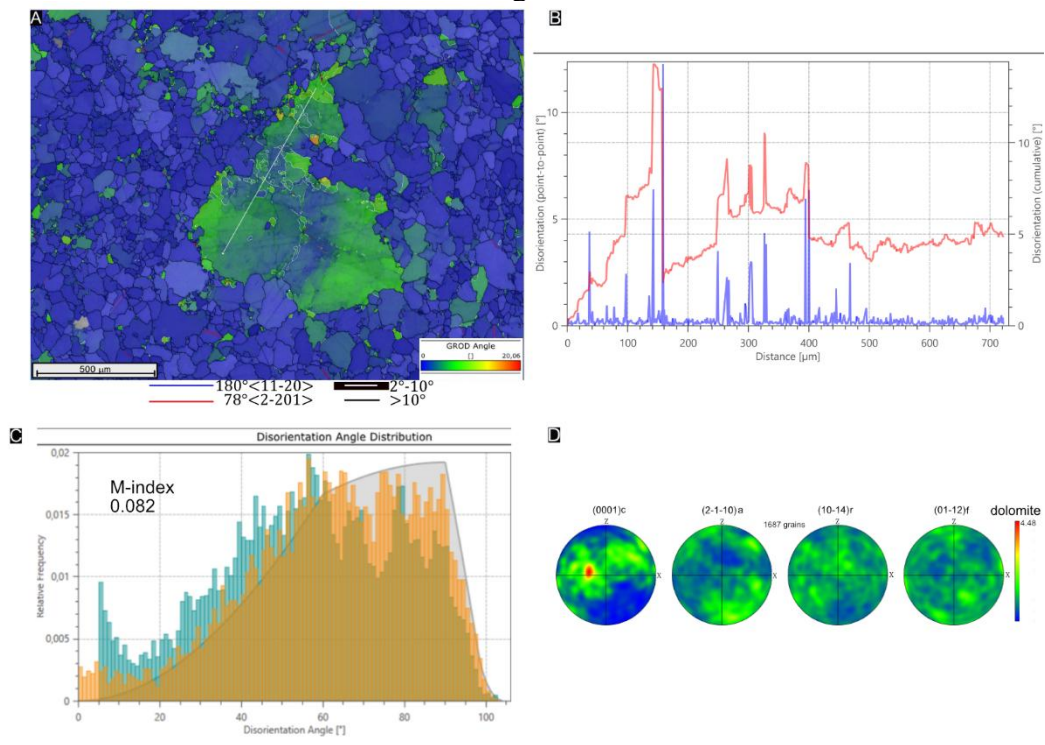
Fig. 8



Raman shift (cm⁻¹)

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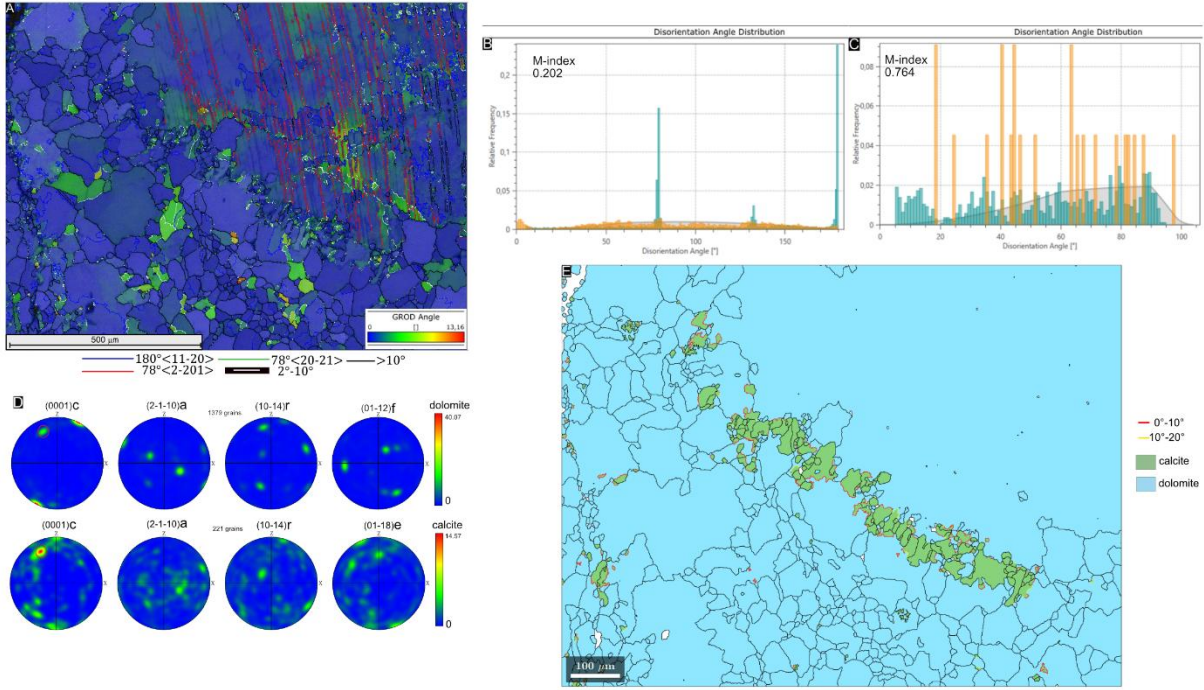
Fig. 9



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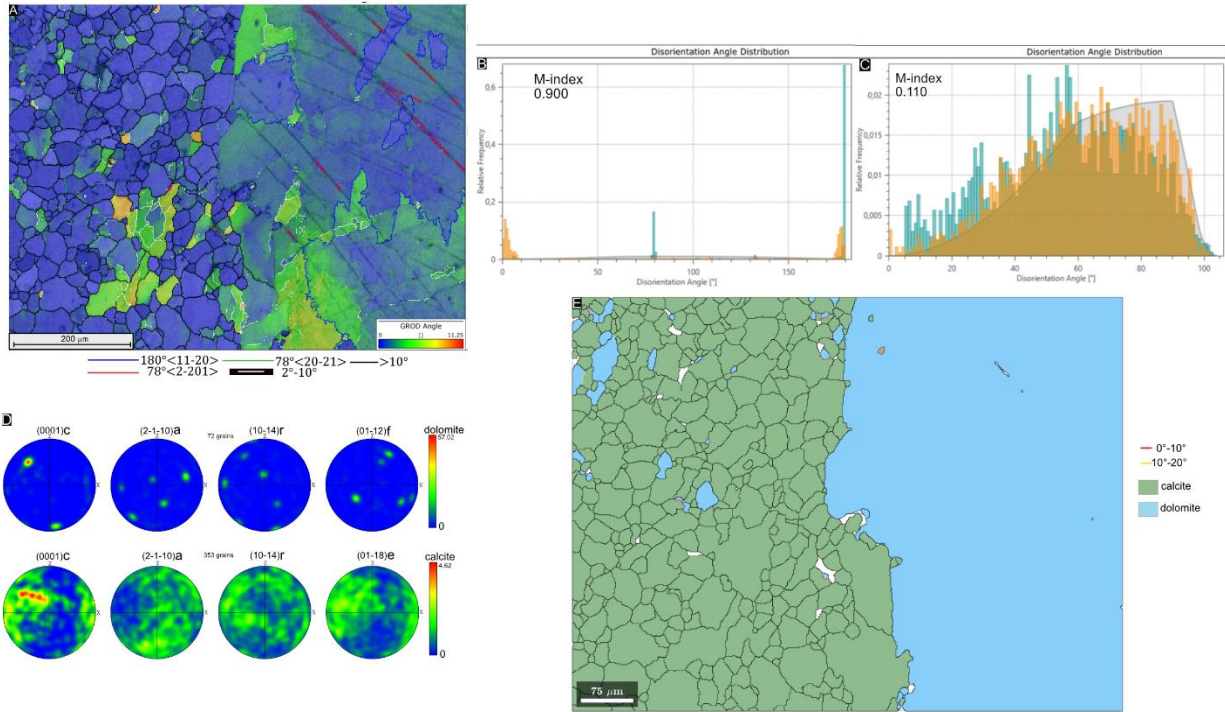
Fig. 10



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Fig. 11



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Fig. 12

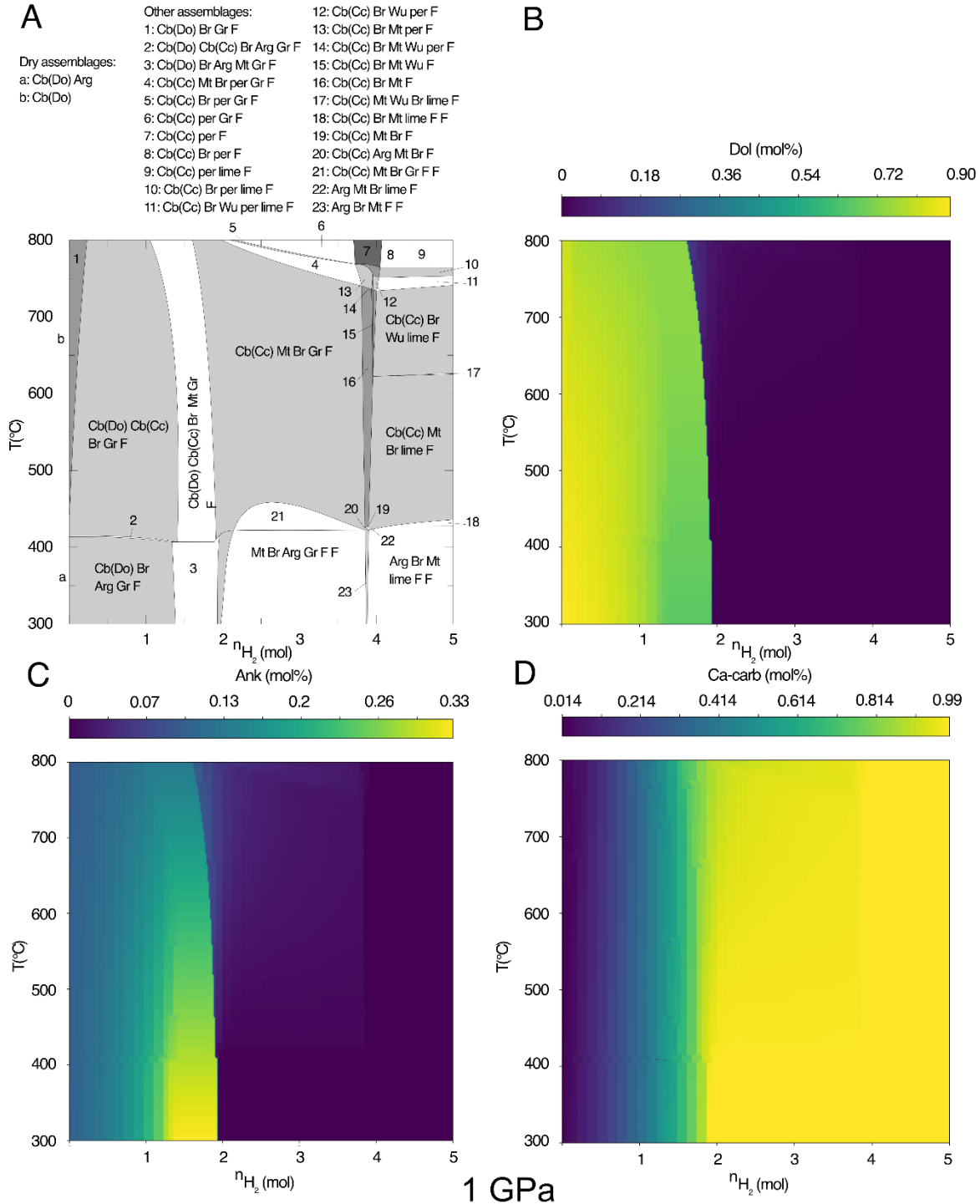
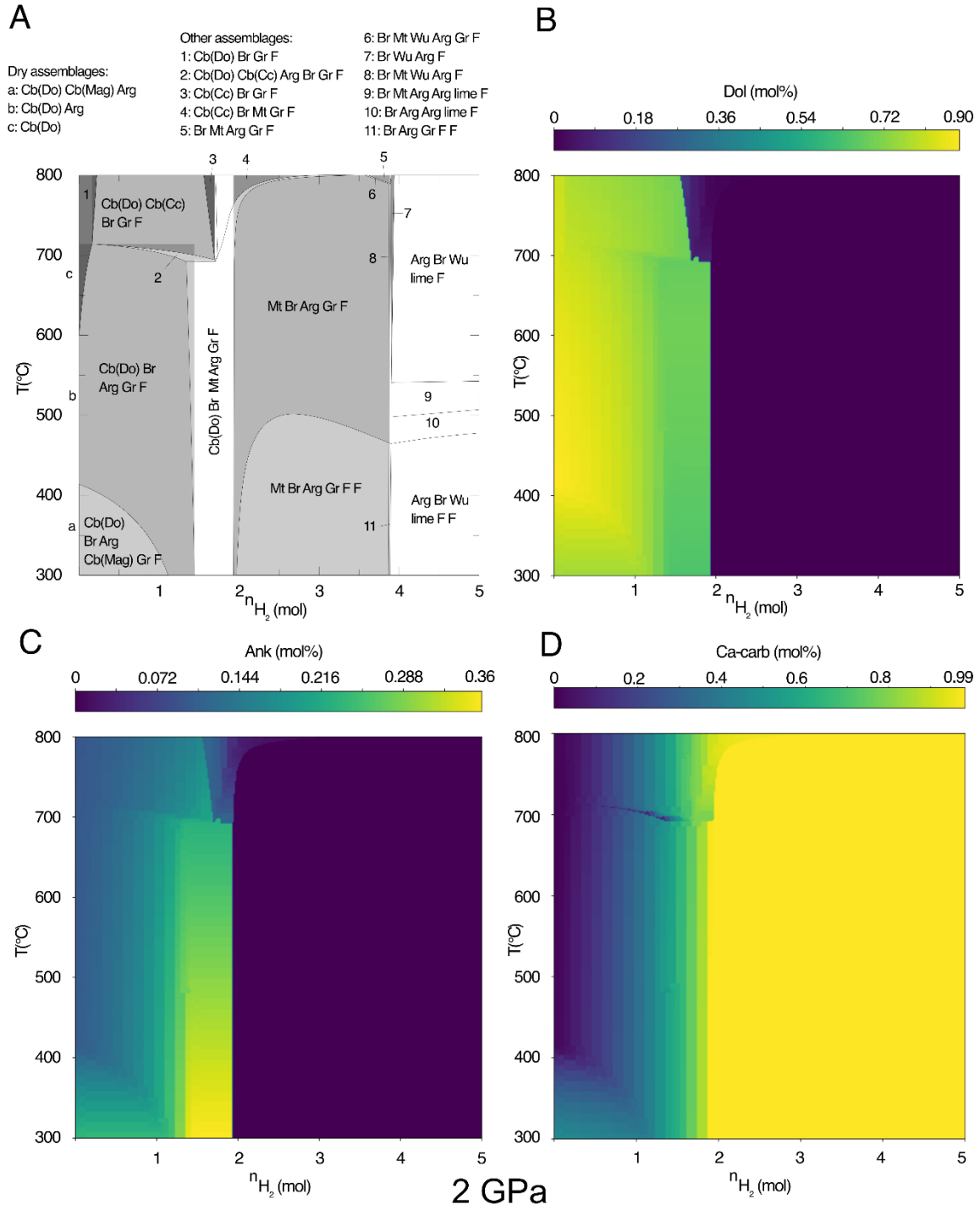
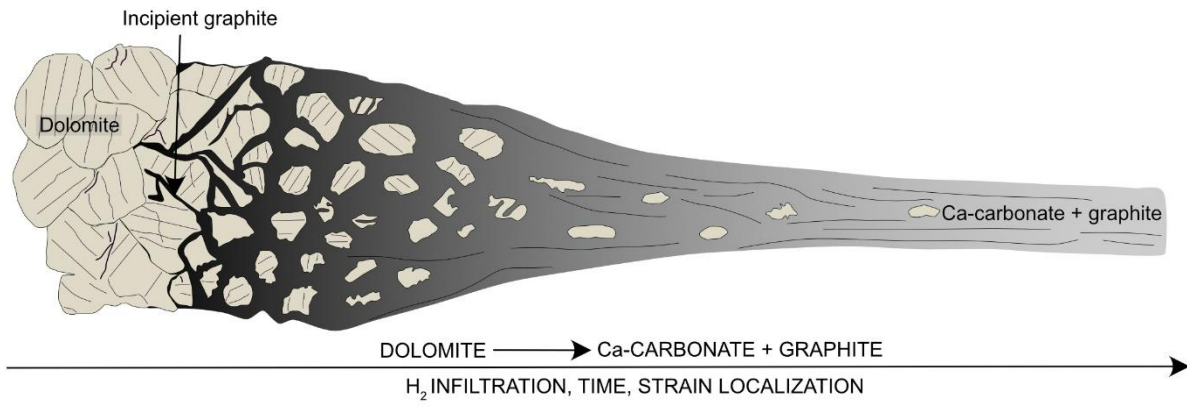


Fig. 13



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Fig. 14



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