

Supporting Information

Nano-crystalline and amorphous calcium carbonate from waste seashells by ball milling mechanochemistry processes

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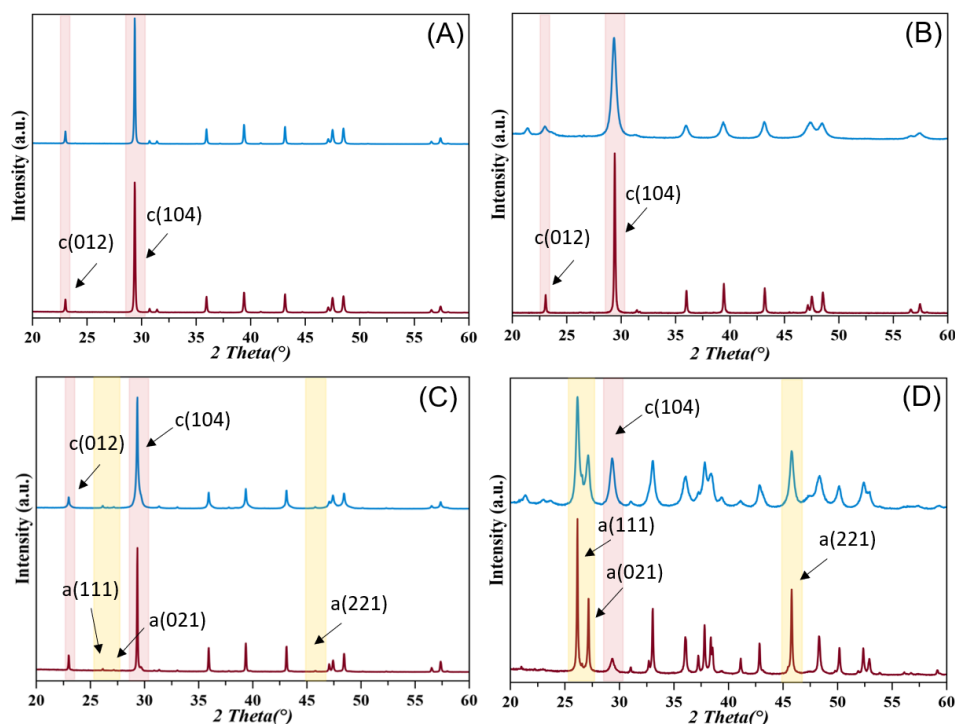


Figure S11. X-ray powder diffraction patterns from starting materials (red line) and dry milled materials (blue line) of geogenic CaCO_3 (A), oyster shells (B), scallop shells (C), and clam shells (D). The intensity is in linear scale. The diffraction patterns were indexed according to PDF 00-005-0586 and PDF 00-005-0453 for calcite and aragonite, respectively.

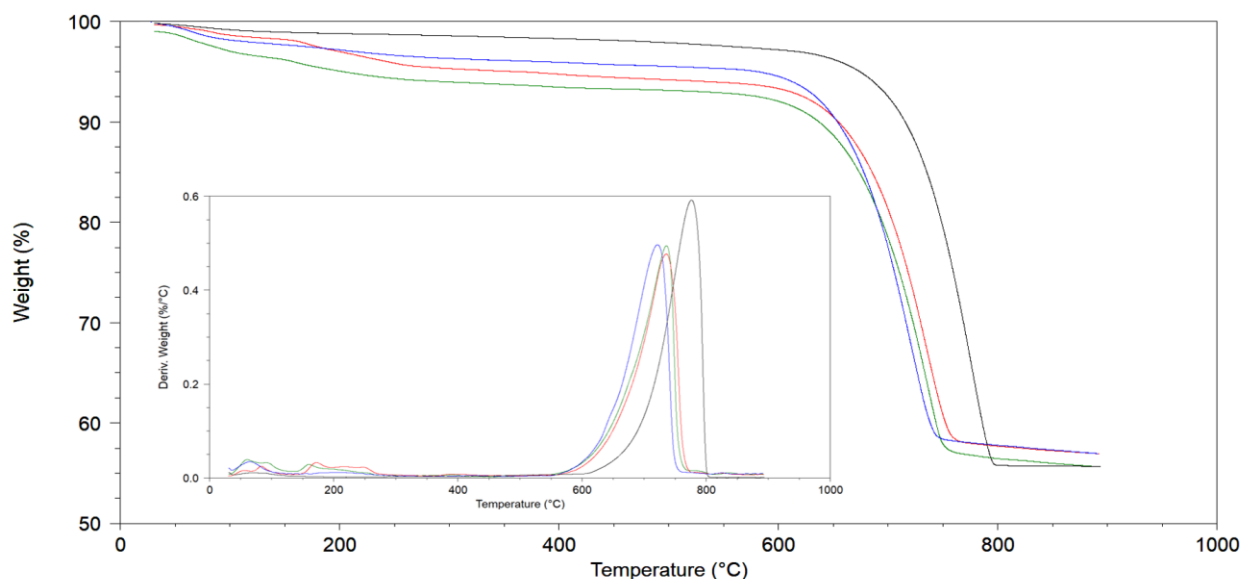


Figure S12. Thermogravimetric analysis (TGA) profiles of geogenic CaCO_3 (black), oyster (blue), scallop (red), and clam (green). The temperature range considered to estimate the content of intraskeletal organic matrix was between 300 °C and 500 °C.

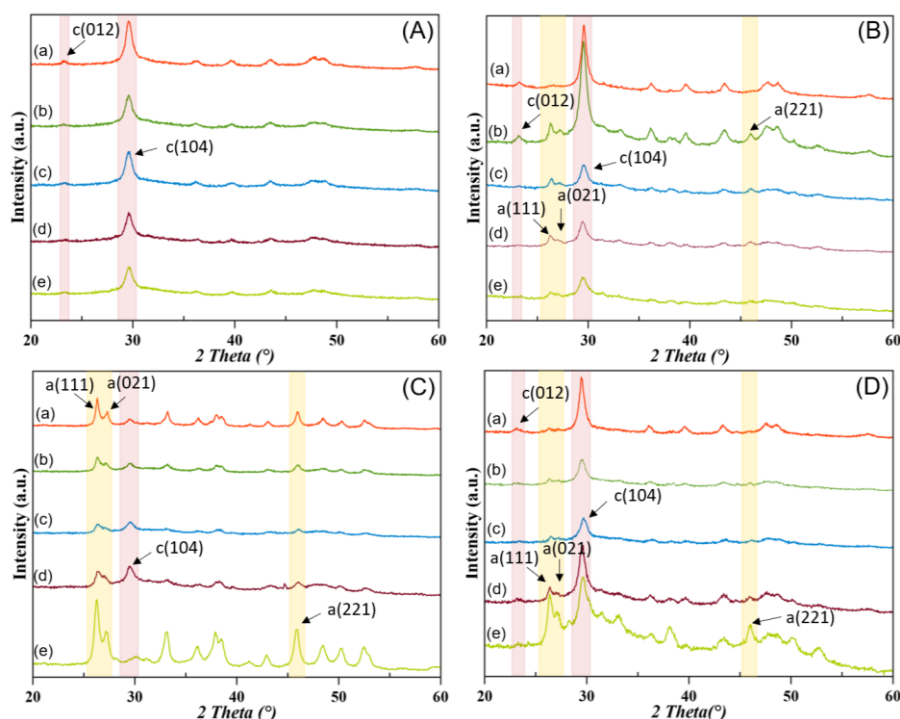


Figure S13. Powder X-ray diffraction patterns of geogenic CaCO₃ (A), oyster (B), scallop (C), and clam (D) shells wet milled for 1 hour (a), 3 hours (b), 6 hours (c), 12 hours (d), and 24 hours (e). The intensity is in linear scale. The diffraction patterns were indexed according to PDF 00-005-0586 for calcite and PDF 00-005-0453 for aragonite.

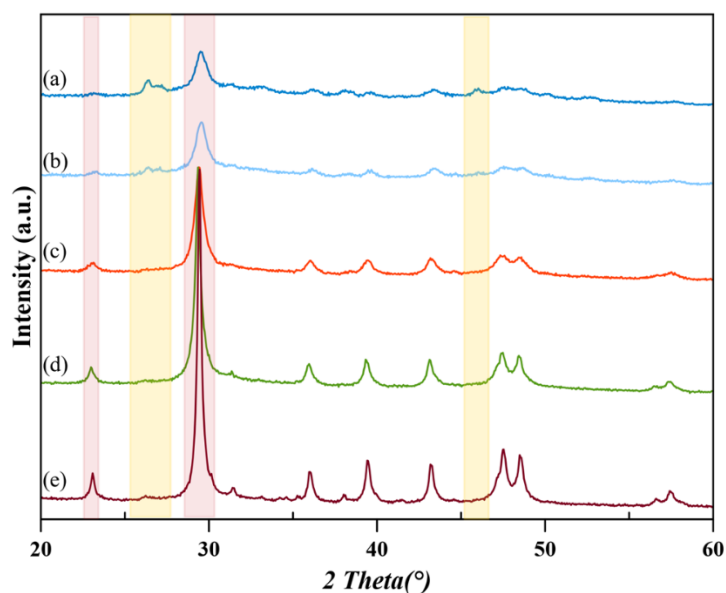


Figure S14. Powder X-ray diffraction patterns of oyster wet milled for 6 hours with different solvents: a) Cyclohexane; b) Heptane; c) Isopropanol; d) Ethanol; e) Butane. The intensity is in linear scale. The diffraction patterns were indexed according to PDF 00-005-0586 for calcite and PDF 00-005-0453 for aragonite.

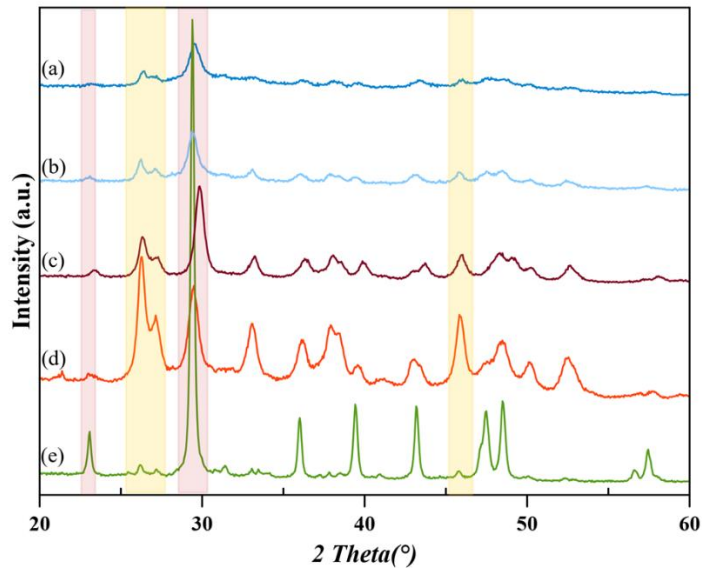


Figure SI5. Powder X-ray diffraction patterns of oyster wet milled for 6 hours with different additives: a) Na_2CO_3 ; b) $\text{Ca}(\text{OH})_2$; c) MgCO_3 ; d) Li_2CO_3 ; e) K_2CO_3 . The intensity is in linear scale. The diffraction patterns were indexed according to PDF 00-005-0586 for calcite and PDF 00-005-0453 for aragonite.

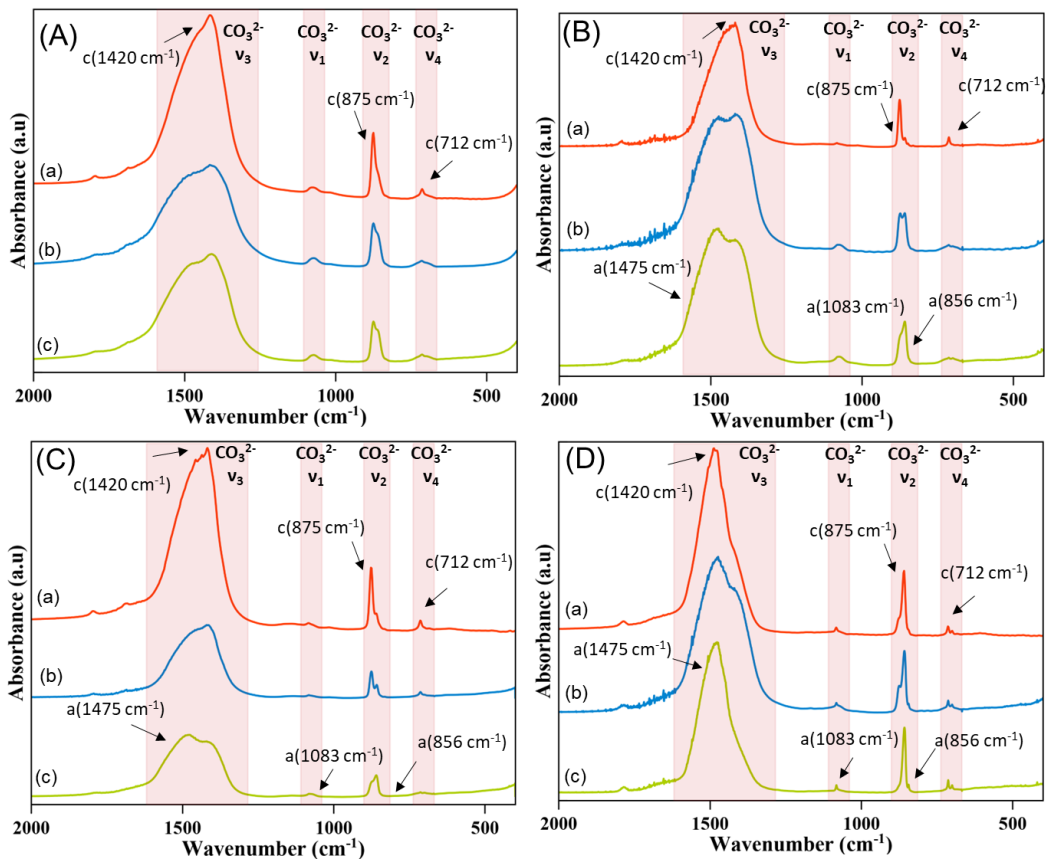


Figure SI6. FTIR spectra of geogenic CaCO_3 (A), oyster (B), scallop (C), and clam (D) shells wet milled for 1 hour (a), 6 hours (b), and 24 hours (c).

Table S11. Percentage of CaCO₃ polymorph, organic matrix content, elemental composition, and crystallite size of geogenic calcium carbonate, oyster shell, scallop shell and clam shell powder after hammer mill grinding. The crystallite size was calculated along the (104) and the (111) zone axis for calcite and aragonite respectively.

Sample	Calcite (wt. %)*	Aragonite (wt. %)*	Organic material content (wt. %) [#]	Mg (at. %)	Sr (at. %)	$d_{(104)}/d_{(111)}$ (nm)
geo CaCO ₃	100	/	0.2	1.24	0.00	1525 ± 3/ -
oyster shell	99.41	0.59	1.1	0.99	0.09	92 ± 3/ -
scallop shell	96.23	3.77	0.7	1.36	0.04	88 ± 3/ 70 ± 16
clam shell	5.78	94.22	0.7	0.28	0.18	43 ± 16/161 ± 8

Table S12. SAED patterns analysis of the sample mechanochemical treated by ball milling for 6 hours. The measured d-spacings are reported for the bCC powders and the geogenic one. The d-spacing of aragonite and calcite are reported for comparison.

Geo-CaCO ₃ (Å)	Oyster (Å)	Scallop (Å)	Clam (Å)	aragonite (Å)	calcite (Å)
3.84	3.80	3.82	3.37	3.39	3.86
3.00	3.01	3.36	2.70	2.70	3.03
2.47	2.75	2.99	2.37	2.37	2.49
2.32	2.40	2.46	1.97	1.98	2.28

Table S13. Percentage of CaCO₃ polymorphs and crystallite size of geogenic calcium carbonate, oyster shell, scallop shell and clam shell powder after different aging times in diverse solvents. The instrumental error is reported.

environment	Aging time	Calcite (wt. %)	Aragonite (wt. %) [§]	ACC (wt. %) [#]	d₍₁₀₄₎/d₍₁₁₁₎ (nm)
geo CaCO₃					
N ₂	10 days	74 ± 2	/	26 ± 2	12.18 ± 0.86
	20 days	84 ± 2	/	16 ± 2	13.61 ± 0.56
	30 days	85 ± 2	/	15 ± 2	15.44 ± 0.43
EtOH	15 min	90 ± 2	/	10 ± 2	12.14 ± 0.38
	30 min	90 ± 2	/	10 ± 2	12.56 ± 0.49
	120 min	97 ± 2	/	3 ± 2	16.00 ± 3.10
H ₂ O	5 min	100 ± 2	/	0 ± 2	21.60 ± 0.42
	30 min	100 ± 2	/	0 ± 2	23.43 ± 0.65
	120 min	100 ± 2	/	0 ± 2	25.38 ± 0.74
Oyster shell					
N ₂	10 days	43 ± 2	18 ± 2	39 ± 2	8.41 ± 0.49/14.20 ± 1.20
	20 days	53 ± 2	27 ± 2	20 ± 2	13.59 ± 0.46/20.17 ± 0.91
	30 days	68 ± 2	27 ± 2	5 ± 2	17.55 ± 0.46/19.82 ± 0.87
EtOH	15 min	51 ± 2	22 ± 2	27 ± 2	7.99 ± 0.17/14.00 ± 1.20
	30 min	47 ± 2	26 ± 2	27 ± 2	9.69 ± 0.18/15.43 ± 0.93
	120 min	51 ± 2	26 ± 2	23 ± 2	12.87 ± 0.21/18.56 ± 0.84
H ₂ O	5 min	74 ± 2	23 ± 2	4 ± 2	29.38 ± 0.73/19.02 ± 0.60
	30 min	84 ± 2	13 ± 2	3 ± 2	30.05 ± 0.39/19.80 ± 1.20
	120 min	89 ± 2	11 ± 2	0 ± 2	38.24 ± 0.91/23.70 ± 1.40
Scallop shell					
N ₂	10 days	48 ± 2	8 ± 2	44 ± 2	7.79 ± 0.17/14.80 ± 1.70
	20 days	46 ± 2	19 ± 2	35 ± 2	8.89 ± 0.19/13.40 ± 1.10
	30 days	49 ± 2	23 ± 2	28 ± 2	12.36 ± 0.46/20.20 ± 1.00
EtOH	15 min	42 ± 2	14 ± 2	44 ± 2	8.68 ± 0.69/13.10 ± 1.50
	30 min	48 ± 2	19 ± 2	33 ± 2	10.17 ± 0.51/17.30 ± 1.30
	120 min	57 ± 2	23 ± 2	20 ± 2	13.58 ± 0.41/18.80 ± 1.00
H ₂ O	5 min	82 ± 2	14 ± 2	3 ± 2	27.83 ± 0.61/17.84 ± 0.88
	30 min	89 ± 2	9 ± 2	2 ± 2	37.30 ± 1.00/18.70 ± 1.50
	120 min	91 ± 2	8 ± 2	1 ± 2	35.05 ± 0.84/21.30 ± 1.60
Clam shell					
N ₂	10 days	36 ± 2	26 ± 2	38 ± 2	6.73 ± 0.38/12.08 ± 0.59
	20 days	38 ± 2	42 ± 2	20 ± 2	12.39 ± 0.81/16.53 ± 0.55
	30 days	43 ± 2	46 ± 2	11 ± 2	16.06 ± 0.64/16.76 ± 0.42
EtOH	15 min	35 ± 2	29 ± 2	36 ± 2	6.65 ± 0.18/11.33 ± 0.54
	30 min	35 ± 2	36 ± 2	29 ± 2	8.81 ± 0.44/14.31 ± 0.54
	120 min	47 ± 2	50 ± 2	3 ± 2	14.93 ± 0.72/16.08 ± 0.47
H ₂ O	5 min	55 ± 2	43 ± 2	2 ± 2	25.56 ± 0.75/25.90 ± 1.70
	30 min	65 ± 2	33 ± 2	2 ± 2	32.85 ± 1.00/26.90 ± 1.80
	120 min	66 ± 2	33 ± 2	1 ± 2	32.08 ± 0.87/27.40 ± 1.70

[§] Percentage of crystalline phases. [#] Percentage of ACC in the particles.