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Electronic structure of epitaxial perovskite films in the two-dimensional limit: Role of the surface termination F

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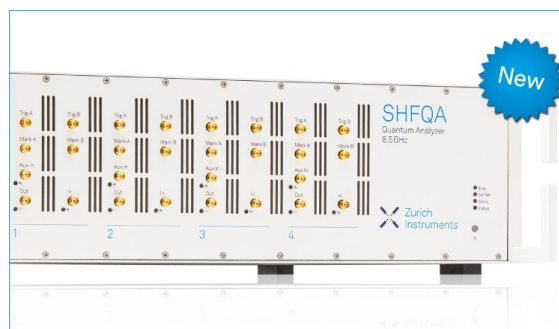
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ABSTRACT

An often-overlooked property of transition metal oxide thin films is their microscopic surface structure and its effect on the electronic properties in the ultrathin limit. Contrary to the expected conservation of the perovskite stacking order in the (001) direction, heteroepitaxially grown SrIrO₃ films on TiO₂-terminated SrTiO₃ are found to exhibit a terminating SrO surface layer. The proposed mechanism for the self-organized conversion involves the adsorption of excess oxygen ions at the apical sites of the IrO₂-terminated surface and the subsequent decomposition of the IrO₆ octahedra into gaseous molecular IrO₃ and the remaining SrO-terminated surface. Whereas the *ab initio* calculated electronic structure of SrO-terminated SrIrO₃ in the monolayer limit exhibits a striking similarity to bulk Sr₂IrO₄, the broken octahedral symmetry at the IrO₂-terminated surface would mix the otherwise crystal field split e_g and t_{2g} states, resulting in distinctly different low-energy electronic states.

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In the last decade, iridates have moved into the focus of solid state research since the interplay between the comparably strong Coulomb and spin-orbit interactions in their $5d$ orbitals is expected to result in hitherto-unforeseen properties.¹ The Ruddlesden-Popper (RP) family of strontium iridates Sr_{n+1}Ir_nO_{3n+1} has attracted special attention due to the observation of an exotic spin-orbit entangled Mott insulating ground state in its quasi-two-dimensional end member Sr₂IrO₄ ($n = 1$)^{2,3} and the phenomenological similarities to the high- T_C cuprate superconductors upon doping.^{4,5} In contrast to its layered counterparts ($n = 1, 2$), the three-dimensional $n = \infty$ RP phase SrIrO₃ is thermodynamically metastable at ambient conditions and can only be synthesized in polycrystalline form upon the application of high pressure and temperature.^{6,7} In an alternative approach, the perovskite phase can be stabilized in a thin-film geometry by effectively substituting the external pressure with epitaxial strain to an underlying substrate. The resulting SrIrO₃ films adopt a distorted perovskite structure,^{8,9} which is predicted to support topologically non-trivial zero-energy surface states protected by chiral and mirror-reflection symmetries.^{10,11} Upon transition into the two-dimensional

limit, SrIrO₃ thin films undergo a metal-insulator transition and adopt an electronic structure that closely resembles that of bulk Sr₂IrO₄.^{9,12} A deeper understanding of the microscopic aspects of this behavior requires in-depth knowledge of the crystallographic structure of not only the bulk but also the surface of SrIrO₃ films. In this Letter, we provide evidence for the conversion from an IrO₂ to a SrO surface termination during the growth of SrIrO₃ on TiO₂-terminated SrTiO₃ as a consequence of the thermodynamic instability of the topmost IrO₂ layer. The pronounced effects on the electronic structure of SrIrO₃ in the monolayer limit are investigated by *ab initio* band structure calculations.

Pulsed laser deposition (PLD) is a commonly employed technique for the heteroepitaxy of ternary oxides. Two of the major challenges for the synthesis of high-quality specimens are the stoichiometric transfer of the different atomic species from the target to the substrate and the control of the oxidative state of the transition metal cation. The former is not trivial and critically depends on the respective atomic masses and details of the ablation process (e.g., laser fluency, repetition rate, scanned ablation area,...),^{13–17} whereas the

latter primarily depends on the type and pressure of the provided background gas (e.g., O₂, O₃, Ar, NO₂,...).^{18–20} These challenges may become strongly intertwined if the transition metal forms a volatile gaseous compound upon over-oxidation. This is the case for PLD growth of SrRuO₃ and SrIrO₃, where excessive temperature and/or oxygen partial pressure may result in the formation of volatile RuO₃/RuO₄²¹ and IrO₃,^{22,23} respectively. Using a stoichiometric SrRuO₃ (SrIrO₃) target, the substantial loss of ruthenium (iridium) may, therefore, lead to the growth of the RP phases Sr₃Ru₂O₇ or Sr₂RuO₄²⁴ (Sr₃Ir₂O₇ or Sr₂IrO₄²⁵). Indeed, it has been shown that the growth phase diagram can be mapped onto the phase diagram expected from thermodynamic equilibrium despite the inherent non-equilibrium nature of pulsed laser deposition.²⁵

Here, we investigate the microstructure of strontium iridate thin films with varying degrees of iridium deficiency and the types of structural lattice faults therein. For this purpose, strontium iridate films are heteroepitaxially grown on (001)-oriented TiO₂-terminated SrTiO₃ substrates^{26,27} by pulsed laser deposition from a polycrystalline target composed of stoichiometric 6H-SrIrO₃ in its thermodynamically stable hexagonal polymorphic form.^{6,7} The films were grown at variable substrate temperatures and structurally investigated by a combination of out-of-plane x-ray diffraction (XRD) using a Cu K_α-source and scanning transmission electron microscopy in high-angle annular dark field mode (HAADF-STEM). Density functional theory calculations were performed by using the VASP *ab initio* simulation package²⁸ within the projector-augmented-plane wave (PAW) method,^{29,30} using the generalized gradient approximation (GGA), as parametrized by the PBE GGA functional.³¹ Spin-orbit coupling was self-consistently included,³² and the Coulomb repulsion *U* and exchange interaction *J* of Ir *d* orbitals were treated within the rotationally invariant density functional theory (DFT) + *U* scheme of Liechtenstein.³³

The strontium iridate RP compounds Sr_{n+1}Ir_nO_{3n+1} = ([SrIrO₃]_n,SrO) essentially consist of *n* SrIrO₃ perovskite layers that are intercalated by SrO layers and laterally shifted against each other

such that no Ir–O–Ir bonds persist between neighboring [SrIrO₃]_n blocks. Figures 1(a)–1(c) show HAADF-STEM images of a strontium iridate film grown at a substrate temperature of 820 °C from a SrIrO₃ target. In agreement with the out-of-plane XRD pattern shown in Fig. 1(d), the lattice structure predominantly exhibits the *n* = 2 RP phase Sr₃Ir₂O₇. As highlighted in the inset of Fig. 1(a), the neighboring bilayers are laterally shifted with respect to each other, such that the adjacent SrO layers adopt a rock salt structure. However, the similar thermodynamic stability of the neighboring RP-phases results in an abundance of isolated mono- and trilayers embedded in the bilayer matrix as shown by black arrows in Fig. 1(b). Furthermore, regions are found with high disorder due to SrIrO₃-like blocks [Fig. 1(c)].

Growth at a lower substrate temperature of 750 °C results in SrIr_{1–δ}O₃ films, where *δ* denotes a slight iridium off-stoichiometry. As shown in Fig. 1(e), the crystal compensates for the iridium deficiency by arranging into distinct blocks of the perovskite phase, which are separated by highly ordered stacking faults. As highlighted in Fig. 1(f), these stacking faults correspond to the characteristic SrO double layers found in the RP structure and result in a similar spatial separation of 6.5 Å between adjacent IrO₂ layers as found in Sr₂IrO₄ and Sr₃Ir₂O₇.^{34,35} While the out-of-plane XRD pattern in Fig. 1(g) corresponds to the cubic perovskite lattice despite the presence of stacking faults, the inherent disorder leads to some incoherence in the diffraction reflected in the absence of pronounced Laue oscillations in Fig. 1(h).

Upon further reduction in the substrate temperature, stoichiometric SrIrO₃ thin films are obtained. Figure 2(a) shows the intensity of the specular reflection high-energy electron diffraction (RHEED) signal during the growth of two equally thick SrIrO₃ films on TiO₂-terminated SrTiO₃. While the persistent intensity oscillations signal layer-by-layer film growth—where one oscillation is typically identified with the deposition of one monolayer—the observation of coherent RHEED patterns before and after deposition of SrIrO₃ (inset) evidences the high crystalline long-range order of the film surface. While the slightly different substrate temperatures (700–720 °C) and

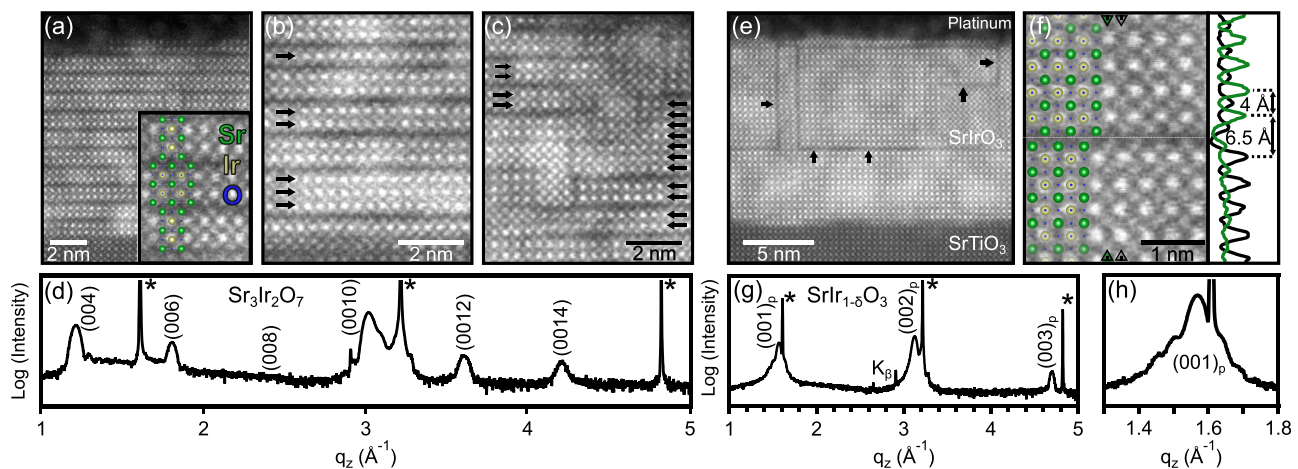


FIG. 1. (a) HAADF-STEM image of a Sr₃Ir₂O₇ film grown from a SrIrO₃ target. (b) RP-like stacking faults showing isolated mono- or trilayers embedded in a Sr₃Ir₂O₇ (i.e., bilayer) matrix. (c) Region with high disorder due to SrIrO₃-like blocks. (d) Out-of-plane XRD exhibits the Sr₃Ir₂O₇ pattern despite the abundance of lattice faults. Asterisks denote the diffraction from the SrTiO₃ substrate. (e) HAADF-STEM image of a slightly iridium-deficient SrIr_{1–δ}O₃ film. Highly ordered stacking faults (arrows) divide the lattice into blocks of SrIrO₃. (f) The stacking faults consist of laterally shifted SrO double layers as found in the RP lattice structure. (g) Out-of-plane XRD exhibits a SrIrO₃ pattern with no other crystal phases present. (h) The present disorder reduces the coherence as reflected in the absence of Laue intensity oscillations in the diffraction signal from the film.

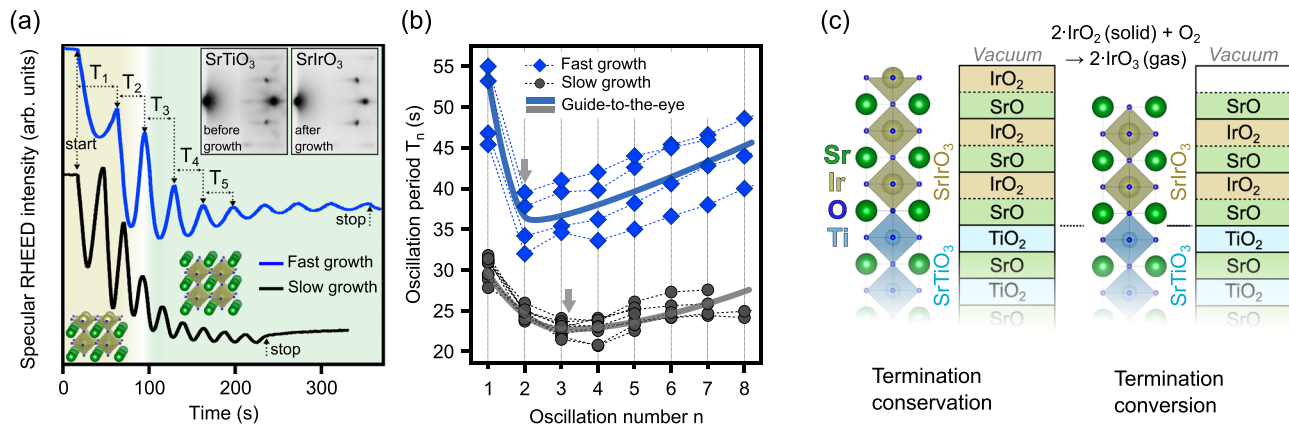


FIG. 2. (a) Specular RHEED intensity oscillations during the growth of SrIrO₃ in the temperature range of 700–720 °C at oxygen partial pressures of 0.05–0.08 mbar and different laser fluencies (slow growth: 1.7 J/cm²; fast growth: 2.5 J/cm²). Inset: RHEED patterns before and after deposition of SrIrO₃. (b) RHEED intensity oscillation period T_n as a function of oscillation number n . After one (three) prolonged oscillation period, a minimum is observed, followed by a gradual increase in T_n . (c) Preservation of the perovskite stacking and stoichiometric deposition would lead to an IrO₂ surface termination. Due to the volatility of IrO₃, a termination conversion to SrO-terminated SrIrO₃ occurs.

oxygen partial pressures (0.05–0.08 mbar) will influence the plume dynamics and hence the oxidative state of the iridium ions, the most pronounced effect is likely to be caused by the use of different laser fluencies (1.7 J/cm² and 2.5 J/cm²), which results in significantly different growth rates.

Interestingly, the RHEED intensity oscillation period T_n of the n th oscillation does not remain constant throughout the deposition process, as shown in Fig. 2(b) for a selection of films grown at similar conditions. After a significantly prolonged first period, a minimal T_n is observed for the second ($n = 2$, blue) or third ($n \approx 3$, black) periods, followed by a gradual increase in T_n . The latter effect, i.e., a reduction in the deposition rate, is well-known and caused by an increasing reflectance of the target surface due to changes in the target surface morphology.^{36–38} The prolonged first oscillation period bears a striking resemblance to the initial growth of SrRuO₃ on TiO₂-terminated SrTiO₃, where it is due to a deviation from the expected preservation of the perovskite stacking order AO-BO₂ in the (001) direction, which would result in a RuO₂-terminated surface.³⁹ Specifically, the topmost RuO₂ layer is unstable and forms volatile compounds until the terminating layer is completely converted into SrO. As evidenced by a doubling of the first oscillation period, for SrRuO₃, the conversion from A- to B-site termination is completed within the first unit cell at 700 °C and distributed over several oscillation periods at lower temperatures. We also note that the specular reflection of the RHEED patterns before and after growth [see the inset in Fig. 2(a)] exhibits a small intensity loss with respect to the (1, 0) and (−1, 0) reflections, which is reminiscent of the growth of SrRuO₃ and has been interpreted as a sign for termination conversion.⁴⁰

We, hence, propose that a similar conversion from IrO₂ to SrO termination occurs during the growth of SrIrO₃ by over-oxidation of the terminating IrO₂ layer and subsequent evaporation of IrO₃. As shown in Fig. 2(c), the conservation of the perovskite stacking order on a TiO₂-terminated SrTiO₃ substrate results in a IrO₂-terminated surface with an incomplete IrO₆ octahedron. The tendency of BO₂-terminated perovskite transition metal oxides to adsorb excess oxygen ions at the apical site is well-documented.⁴¹ We argue that the adsorption of an apical oxygen ion leads to the decomposition of the IrO₆

octahedron into gaseous molecular IrO₃ and the remaining SrO-terminated surface. Rather than the doubling of the first oscillation period as in the case of SrRuO₃, a prolongation of the first oscillation is observed experimentally during the growth of SrIrO₃. However, the minimal oscillation period consistently appears after a similar absolute time span (≈ 100 s) and independent of the growth rate, signaling that the IrO₃ formation and evaporation rate is only sensitive to the thermodynamic conditions to a good approximation.

As shown in Ref. 39, the characteristic, consistent signature of the RHEED intensity oscillations is a strong, but arguably indirect indication for a specific termination conversion. For more direct evidence, we investigated the microscopic surface structure of a 4 uc thin SrIrO₃ film, covered with a protective layer of amorphous carbon to facilitate the direct imaging of the surface by HAADF-STEM. The pronounced Laue oscillations and the STEM image of a thick film grown at identical conditions in Figs. 3(a) and 3(b) evidence the coherent growth of stoichiometric SrIrO₃ without stacking faults. Figure 3(c) shows an overview STEM image of the thin SrIrO₃ film, which, at the first glance, appears to be terminated by an IrO₂ layer. Indeed, the iridium columns (atomic number $Z = 77$) are most pronounced due to the Rutherford-like dependence of the scattering cross section in HAADF imaging ($\propto Z^n$ with $n \approx 1.7$),⁴² followed by strontium ($Z = 38$) and titanium ($Z = 22$), while carbon and oxygen atoms are not detected. However, a signal is discerned above the topmost IrO₂ layer [cf. Fig. 3(d)], as highlighted by the averaged line profiles of the IrO₂/TiO₂ and SrO layers perpendicular to the surface (along $[001]_p$) in Fig. 3(e). The strong scattering from the heavy iridium atoms results in a broadened iridium signal, which overshadows the other atomic species. Nonetheless, the topmost atomic layer apparently stems from a SrO layer rather than an IrO₂ layer, as is suggested from the comparison of the SrO and IrO₂ line profiles with the last shoulder toward the surface seen for the former. For clarity, two horizontal line profiles of the sub-surface (gray) and surface (green) SrO layer along the $[100]_p$ lattice direction are shown in Fig. 3(f). Despite the weak intensity of the surface layer, both signals clearly exhibit an in-phase periodicity that substantiates the conjecture of a SrO termination. Whether the weak signal strength of the terminating layer is exclusively caused by the

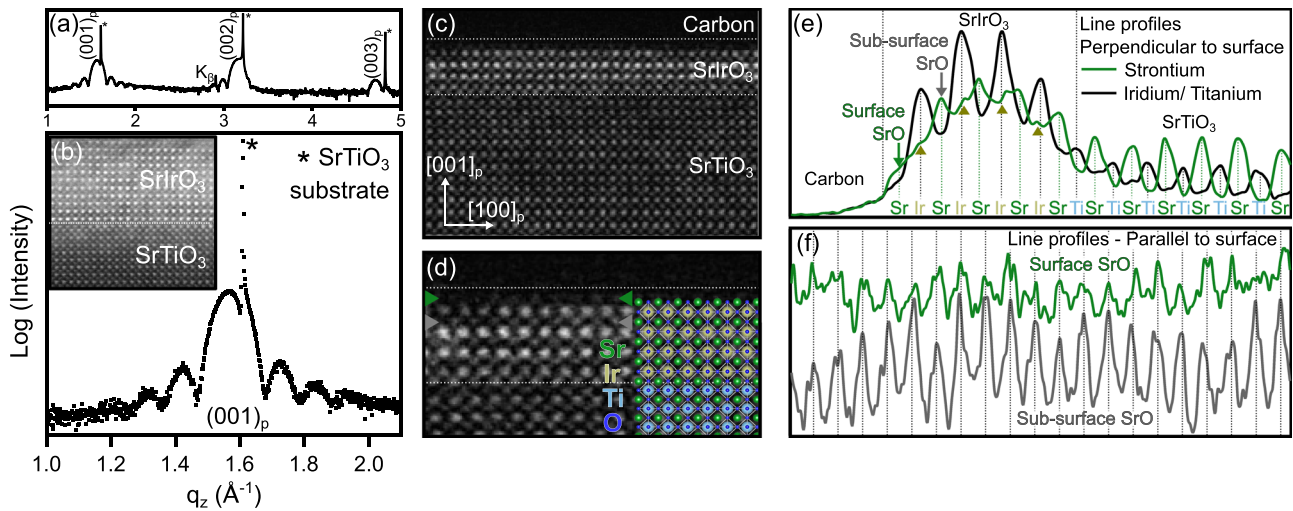


FIG. 3. (a) Out-of-plane XRD of a thick stoichiometric SrIrO₃ film. (b) Pronounced Laue fringes of the (001) diffraction peak signal a coherent film structure. Inset: HAADF-STEM image. (c) HAADF-STEM image of a thin SrIrO₃ film covered with amorphous carbon to facilitate the investigation of the surface. (d) Close-up of the film superimposed with a structural model. A faint terminating SrO layer can be discerned (green arrows). (e) Line profiles through the IrO₂/TiO₂ (black) and SrO (green) layers along the [001]_p direction. In contrast to IrO₂, the SrO profile exhibits a shoulder at the film surface. (f) Line profiles of the surface- and sub-surface SrO layer along the [100]_p direction. Both line profiles exhibit a clearly seen in-phase periodicity despite the small intensity of the terminating SrO layer.

broken translational symmetry at the surface or also due to an only partial, inhomogeneous SrO coverage of the sample remains an open question. Further investigations using atomic-resolution scanning probe techniques (e.g., scanning transmission or atomic force microscopy) may shed light on this in the future.

In the two-dimensional limit, the surface structure becomes critical for the electronic band structure because of the divergent surface-to-bulk ratio. Figure 4 compares the calculated DFT + *U* band

structure of two supercells comprising four layers of SrTiO₃ and SrO- and IrO₂-terminated monolayers of SrIrO₃. For a realistic computation of the actual surface properties, a vacuum gap of 28.1 Å between adjacent supercells was introduced. Laterally, the supercells comprise two iridium sites to incorporate the collective octahedral rotations that enlarge the surface-projected unit cell of thin SrIrO₃ films by $\sqrt{2} \times \sqrt{2}$.⁹ Note that the band structure is calculated for the enlarged unit cell (highlighted in red) and not backfolded into the pseudo-cubic

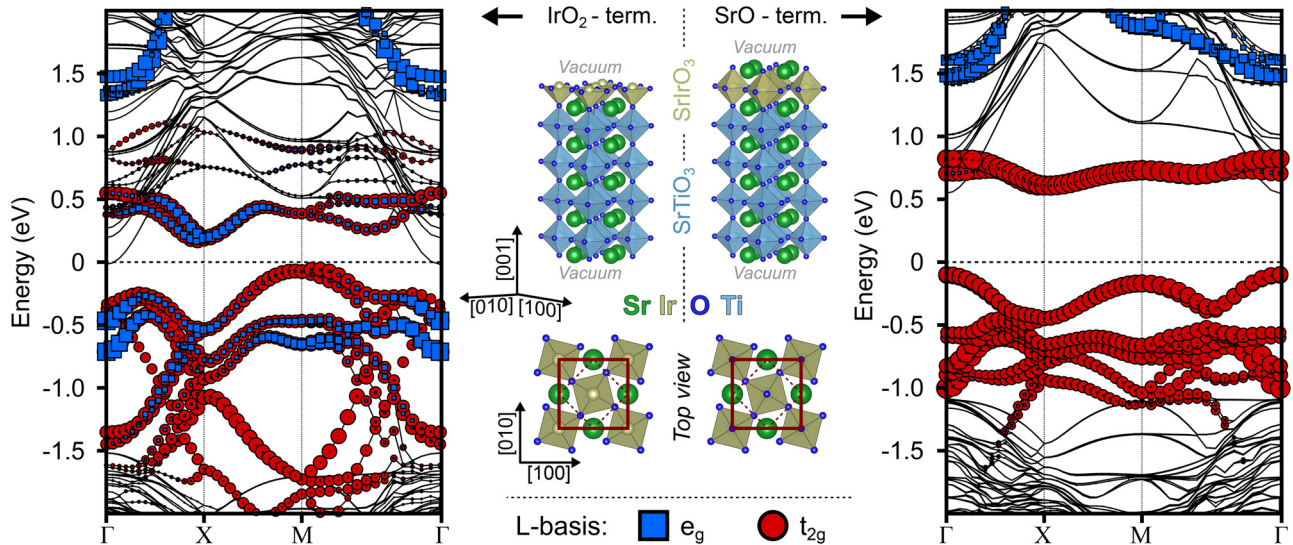


FIG. 4. DFT+*U* band structure calculations of two supercells comprising four unit cells of SrTiO₃ and a monolayer of SrIrO₃ with different terminations. The blue and red marker sizes indicate the *e_g* and *t_{2g}* orbital characteristics, respectively. Left: IrO₂-termination. Since the local *O_h* symmetry of the crystal field is broken, the *e_g* and *t_{2g}* states are mixed. Right: SrO-termination. The low-energy bands around the Fermi level are described by pure *t_{2g}* states as a consequence of the preserved *O_h* symmetry of the IrO₆ octahedra. The band structure is highly reminiscent of the bulk band structure of Sr₂IrO₄.

Brillouin zone (dashed red line), as in Ref. 9. Here, the sizes of the blue squares and red circles denote the Ir $5d$ e_g and t_{2g} orbital characteristics, respectively, as obtained from a projection onto the basis of the orbital angular momentum L .

The valence and conduction bands of the IrO₂-terminated monolayer in Fig. 4 exhibit a strong mixing of e_g and t_{2g} characteristics, whereas the SrO-termination results in a distinct separation thereof. Strictly speaking, the distinction between a t_{2g} and e_g manifold only holds in octahedral crystal fields with the point group symmetry O_h , which essentially originates from octahedra of negative point charges, i.e., the nearest-neighbor oxygen ligands in perovskite oxides. Hence, the observed orbital mixing occurs as a consequence of the broken local octahedral symmetry due to the absence of an apical oxygen at the IrO₂-terminated surface. In particular, the striking similarity between the electronic structure of SrIrO₃ in the monolayer limit and bulk Sr₂IrO₄^{43,44} exclusively holds for SrO-terminated SrIrO₃. Hence, the observed self-organized conversion to a SrO termination is imperative for the close analogy between bulk Sr₂IrO₄ and monolayer SrIrO₃ since the quasi-two-dimensional network of fully closed IrO₆ octahedra constitutes their characteristic common structural feature.

In summary, the effects of iridium loss via the formation of volatile IrO₃ on the microstructure of strontium iridate films are investigated. While excessively oxidizing conditions may result in significant off-stoichiometry, even for nominally stoichiometric SrIrO₃, the terminating surface IrO₂ layer remains prone to over-oxidation. The consequential conversion from B- to A-site termination is evidenced by direct surface imaging using scanning transmission electron microscopy and found to strongly influence the electronic structure of SrIrO₃ thin films. The local octahedral symmetry in SrO-terminated SrIrO₃ monolayers restores the crystal field splitting of e_g and t_{2g} states and results in an electronic structure similar to bulk Sr₂IrO₄. With regard to SrIrO₃ as a potential topologically non-trivial metal, these findings are relevant for further theoretical predictions of symmetry-protected zero-energy surface states.

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