

## Alma Mater Studiorum Università di Bologna Archivio istituzionale della ricerca

Spectroscopic detection of the gallium methylene (GaCH2 and GaCD2) free radical in the gas phase by laserinduced fluorescence and emission spectroscopy

This is the final peer-reviewed author's accepted manuscript (postprint) of the following publication:

Published Version:

Smith, T.C., Tarroni, R., Clouthier, D.J. (2024). Spectroscopic detection of the gallium methylene (GaCH2 and GaCD2) free radical in the gas phase by laser-induced fluorescence and emission spectroscopy. THE JOURNAL OF CHEMICAL PHYSICS, 160, 1-14 [10.1063/5.0182504].

Availability:

This version is available at: https://hdl.handle.net/11585/956630 since: 2024-02-11

Published:

DOI: http://doi.org/10.1063/5.0182504

Terms of use:

Some rights reserved. The terms and conditions for the reuse of this version of the manuscript are specified in the publishing policy. For all terms of use and more information see the publisher's website.

This item was downloaded from IRIS Università di Bologna (https://cris.unibo.it/). When citing, please refer to the published version.

(Article begins on next page)

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset. PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

Chemical Physics

The Journal of

AIP Publishing

## Spectroscopic Detection of the Gallium Methylene (GaCH2 and GaCD2) Free Radical in the

Gas Phase by Laser-Induced Fluorescence and Emission Spectroscopy

by

Tony C. Smith,<sup>1</sup> Riccardo Tarroni<sup>2</sup> and Dennis J. Clouthier<sup>1a)</sup>

<sup>1</sup>Ideal Vacuum Products, LLC, 5910 Midway Park Blvd. NE, Albuquerque, New Mexico 87109,

USA

<sup>2</sup> Dipartimento di Chimica Industriale "Toso Montanari", Università di Bologna, Viale

Risorgimento 4, 40136 Bologna, Italy

<sup>a)</sup>Author to whom correspondence should be addressed: djc@idealvac.com

## ABSTRACT

GaCH<sub>2</sub>, a free radical thought to play a role in the chemical vapor deposition of gallium-containing thin films and semiconductors, has been spectroscopically detected for the first time. The radical was produced in a pulsed discharge jet using a precursor mixture of trimethylgallium vapor in high pressure argon and studied by laser-induced fluorescence and wavelength resolved emission techniques. Partially rotationally resolved spectra of the hydrogenated and deuterated species were obtained and they exhibit the nuclear statistical weight variations and subband structure expected for a  ${}^{2}A_{2} - {}^{2}B_{1}$  electronic transition. The measured spectroscopic quantities have been compared to our own *ab initio* calculations of the ground and excited state properties. The electronic spectrum of gallium methylene is similar to the corresponding spectrum of the aluminum methylene radical which we reported in 2022.

AIP Publishing

## The Journal of Chemical Physics

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset.

PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

Gallium-containing compound semiconductors, such as GaAs, GaN, GaP, GaSb, InGaAs, InGaN, AlGaInP, InGaP and AlInGa are important in devices as diverse as infrared laser sources, light-emitting diodes, radiation detectors, solar cells and in the fabrication of quantum wells, wires and dots. Trimethylgallium (TMGa) is the preferred source of gallium for metalorganic chemical vapor deposition (MOCVD) of single- or polycrystalline thin films in which the precursor is decomposed at the hot surface of a substrate. Although extensively employed in industry, the detailed mechanisms of the gas- and gas-surface phase chemical reactions that take place in gallium MOCVD are still poorly understood.

As early as 1991, Mountziaris and Jensen<sup>1</sup> developed a kinetic model for the MOCVD deposition of GaAs from trimethylgallium and arsine. In their model, methyl radicals produced in the initial step were postulated to undergo hydrogen-abstraction reactions with TMGa or its decomposition byproducts, producing short-lived, highly reactive gallium methylene (GaCH<sub>2</sub>). These reactive intermediates were proposed to play an important role in the deleterious incorporation of carbon in the growth process.

A thorough literature search shows that GaCH<sub>2</sub> has never been observed spectroscopically, either in the gas phase or in matrices and there are no computational studies available on the free radical. In fact, of the possible group III-methylene species BCH<sub>2</sub>, AlCH<sub>2</sub>, GaCH<sub>2</sub>, InCH<sub>2</sub> and TlCH<sub>2</sub> only aluminum methylene, which we reported very recently,<sup>2,3</sup> is now known. We used laser-induced fluorescence techniques to detect AlCH<sub>2</sub> among the decomposition products of trimethylaluminum in a pulsed electric discharge jet. The  $\tilde{B}^2A_2 - \tilde{X}^2B_1$  band systems of AlCH<sub>2</sub> and AlCD<sub>2</sub>, in the 513–483 nm region, were recorded and the 0-0 bands rotationally resolved which

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset

PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

> AIP Publishing

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset

PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

afforded the molecular structures in both states. The observed "spin-splittings" were found to be predominantly caused by a large aluminum Fermi contact interaction in the excited state.

In the present work, we have succeeded in detecting the LIF spectra of  $GaCH_2$  and  $GaCD_2$ produced by the electric discharge degradation of trimethylgallium at the exit of a supersonic expansion into vacuum. The spectra have been assigned based on their similarity to those of aluminum methylene, on the predictable effects of deuteration and on the correspondence between the spectroscopically observed properties (electronic excitation energies, vibrational frequencies, deuterium isotope effects and rotational band contours) and those we calculate using high-level *ab initio* methods.

## **II. EXPERIMENT**

The pulsed discharge jet technique, described in detail elsewhere,<sup>4,5</sup> was used to produce rotationally cold gallium methylene free radicals in the gas phase. Moderately cold spectra (30-40 K) were obtained with the following conditions: Pyrophoric trimethylgallium (TMGa or TMGa-d<sub>9</sub>) liquid was transferred in vacuum to a Pyrex U-tube held in a fume hood, cooled to -15 C (vapor pressure ~ 29 Torr) and pressurized with 40-50 psi of argon. The gas mixture was carried through stainless steel tubing from the fume hood to a pulsed molecular beam valve (General Valve, series 9) mounted in a vacuum chamber. The precursor gas pulses were expanded into a black Delrin flow channel where a pulsed electrical glow discharge between two stainless steel ring electrodes fragmented the trimethylgallium precursor, producing a variety of products. A 1.0 cm long, 5 mm ID cylindrical Delrin reheat tube<sup>6</sup> added to the end of the discharge flow channel was found to increase production of Ga methylene radicals and to suppress the background glow from excited argon atoms. Much colder (8 -10K) spectra were obtained by diluting 15-30 Torr of trimethylgallium vapor with 500 psi of argon in a 1000 cm<sup>3</sup> stainless steel sample cylinder

> AIP Publishing

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset

PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

(Swagelok 304L-HDF4-1000) and expanding the gas mixture at a pressure of 150-200 psi through the discharge and into vacuum.

Our initial searches for the gallium methylene radicals were conducted using the 2D LIF spectrometer, described in detail in a recent publication<sup>7</sup>, developed in the laboratory at Ideal Vacuum. This instrument uses a broadly tunable, high power optical parametric oscillator (Continuum Horizon OPO, 400–710 nm, linewidth 3–7 cm<sup>-1</sup>, and energy 10–50 mJ/pulse) to produce low resolution LIF spectra and the corresponding emission spectrum at each laser step. In this fashion, wide ranges of the visible and UV regions can be surveyed with high sensitivity in an automated fashion, typically by scanning overnight.

Once promising LIF signals were found in the 2D LIF spectra, these were investigated more thoroughly using a higher resolution Lambda-Physik Scanmate 2E dye laser, operated at medium resolution  $(0.1 - 0.2 \text{ cm}^{-1} \text{ linewidth})$  or optionally at high resolution  $(0.035 \text{ cm}^{-1} \text{ linewidth})$  by inserting an intracavity etalon. The medium resolution LIF spectra were calibrated (+/- 0.1 cm<sup>-1</sup> or better) with optogalvanic signals from a neon-filled hollow cathode lamp.

Emission spectra were recorded by fixing the Scanmate dye laser on a prominent Q-branch in the LIF spectrum and dispersing the fluorescence with a Spex 500M monochromator. The emission signals were recorded with an intensified, gated CCD camera (Andor iStar 320T, wavelength range of photocathode 280–760 nm) mounted in the exit focal plane of the instrument. With suitable gating, scattered laser light was completely suppressed and resonance fluorescence down to the initially pumped level could be measured, along with red-shifted emission to higher vibronic levels of the lower state. The spectra were calibrated to an accuracy of 1 - 2 cm<sup>-1</sup> with argon emission lines from a Li/Ar hollow cathode lamp.

Trimethylgallium (Sigma-Aldrich) was used as received. TMGa-d<sub>9</sub> was synthesized by the high temperature reaction of deuterated iodomethane with gallium in the presence of magnesium according to the equation  $6CD_3I + 3Mg + 2Ga \rightarrow 2(CD_3)_3Ga + 3MgI_2$  following the procedure of Zakharkin et al.<sup>8</sup> A 150 cm<sup>3</sup> stainless steel cylinder (Swagelok 304L-HDF4-150) was loaded with 1.44 g (21 mmol) of gallium metal chunks (Sigma-Aldrich), 0.93 g (38 mmol, slight excess) of magnesium powder (Sigma-Aldrich) and 8.97 g of CD<sub>3</sub>I (62 mmol, Sigma-Aldrich, >99.5% D). The cylinder was sealed with a high-temperature bellows valve (Swagelok SS-4H2, threaded surfaces sealed with several layers of 3.5 mil Teflon tape), connected to a vacuum line and thoroughly degassed. After demounting from the vacuum line, the exit of the valve was further sealed with a threaded stainless-steel cap, and the vessel was heated in an oven at 170°C for 70 hours. After cooling, the contents were expanded into vacuum, purified by trap-to-trap distillation and analyzed by gas phase FTIR spectroscopy. This procedure gave 2.0 g (78% yield) of trimethylgallium-d<sub>9</sub> with a purity of better than 95%, as judged by comparison with published IR spectra.<sup>9</sup>

## **III. THEORETICAL CALCULATIONS**

In previous theoretical work, we extensively explored the properties of the ground and several excited states of the isoelectronic AlCH<sub>2</sub> free radical<sup>2</sup>, which provided a conceptual framework for the present, less far-reaching studies of gallium methylene. *Ab initio* calculations of the energies, geometries and vibrational frequencies of the ground and lowest two doublet and the first quartet electronic excited states of GaCH<sub>2</sub> and GaCD<sub>2</sub> were performed using the Molpro<sup>10</sup> and CFOUR<sup>11</sup> quantum chemistry programs.

In Molpro, molecular orbitals were first obtained optimizing the ground and first two doublet excited electronic states in state averaged complete active space self-consistent field

AIP Publishing This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset

PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

(CASSCF)<sup>12,13</sup> computations (9 electrons in 10 orbitals). Energies were then calculated by the internally contracted multireference configuration interaction (ICMRCI)<sup>14,15</sup> method using the same active space as reference. The Davidson correction<sup>16,</sup> with relaxed reference<sup>17</sup> was applied to the ICMRCI energies. In CFOUR, we started from Unrestricted Hartree Fock (UHF) orbitals, checked to avoid SCF spatial instabilities,<sup>18</sup> followed by the coupled cluster singles and doubles with perturbative triple excitations<sup>19</sup> (CCSD(T)) method for the ground state, and equations of motion methods [EOM-CCSD] for the electronic excited states.<sup>20,21</sup> For both methods, the electron correlation treatment was either limited to valence electrons or extended to include the ten d-electrons of gallium.

For calculations with correlation treatment restricted to valence electrons, we used the augcc-pV(T+d)Z and aug-cc-pV(Q+d)Z<sup>22</sup> bases for gallium and the aug-cc-pVTZ and aug-cc-pVQZ<sup>23</sup> bases for carbon and hydrogen. For calculations including d-electron correlation, we used the augcc-pwCVTZ<sup>24,25</sup> bases for gallium and carbon and aug-cc-pVTZ for hydrogen

In the first stages of this work, we explored the relative energies and stabilities of other isomers of GaCH<sub>2</sub>. In Table I we report the geometries and frequencies of trans-HGaCH and CGaH<sub>2</sub>, calculated with both CASSCF/ICMRCI and CCSD(T) methods. For trans-HGaCH, the results (geometry and frequencies) from the two methods are very similar. The situation is different for CGaH<sub>2</sub> of  $C_{2\nu}$  symmetry which was found to be a transition state at the ICMRCI level, with an imaginary in-plane asymmetric CGaH bending frequency,  $\omega_6$ . Removing the symmetry constraint, the structural optimization reverted to the trans-HGaCH geometry. Since the trans-HGaCH isomer is predicted to be 42.8 – 45.1 kcal/mol higher in energy and the CCSD(T) planar CGaH<sub>2</sub> species 97.0 kcal/mol above the GaCH<sub>2</sub> global minimum, it is unlikely that either of these isomers would be observed in our experiments.

The CASSCF/aug-cc-pVTZ state-averaged ground state molecular orbitals calculated at the ICMRCI/ aug-cc-pVTZ equilibrium geometry are shown in the Supplementary Material. The orbital occupations of the first three electronic states of GaCH<sub>2</sub> are the same as those of AlCH<sub>2</sub><sup>2,3</sup>. The ground  $\tilde{X}^2B_1$  state is well-described by the configuration:

 $(\operatorname{core})(9a_1)^2(4b_2)^2(10a_1)^2(11a_1)^2(4b_1)^1$ 

in which the unpaired electron in the highest occupied molecular orbital (HOMO)  $4b_1$  is localized in an out-of-plane 2p orbital on the carbon atom. Promotion of an electron from the  $11a_1$  orbital ACCEPTED MANUSCRIPT

The Journal of Chemical Physics

AIP Publishing This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset.

PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

TABLE I. Ground state geometries, harmonic vibrational frequencies, and relative energies of the CGaH<sub>2</sub> and trans-HCGaH geometric isomers, calculated using Molpro at RHF/RCCSD(T) and CASSCF/ICMRCI levels of theory with the aug-cc-pV(T+d)Z basis.

CASSCF/ICMRCI <sup>a</sup> RHF/RCCSD(T)								
trans-HCGaH $\tilde{X}^2 A''(C_s)$								
r(GaC)/Å	1.8616	1.8668						
r(GaH)∕Å	1.5732	1.5747						
<i>r</i> (CH)/Å	1.0916	1.0918						
$\Theta$ (HGaC)/deg	144.3	144.8						
$\Theta(HCGa)/deg$	130.7	130.2						
$\omega_l(a')/cm^{-1}$	3132.3	3136.3						
$\omega_2(a')cm^{-1}$	1942.3	1956.6						
$\omega_3(a')/cm^{-1}$	750.3	757.6						
$\omega_4(a')/cm^{-1}$	600.7	601.5						
$\omega_5(a')/cm^{-1}$	393.7	386.9						
$\omega_6(a'')/cm^{-1}$	445.4	430.0						
$\Delta E (\text{kcal/mol})^{\text{b}}$	42.8	45.1						
	<b>CGaH</b> <sub>2</sub> $\tilde{X}^2 B_1$ (C	$(Z_{2\nu})$						
r(GaC)/Å	2.0377	2.0434						
<i>r</i> (CH)/Å	1.5855	1.5861						
$\Theta$ (HGaH)/deg	123.0	123.1						
$\omega_l (a_1)/cm^{-1}$	1926.2	1924.8						
$\omega_2(a_1)/cm^{-1}$	636.8	703.1						
$\omega_3(\mathbf{a}_1)/\mathbf{cm}^{-1}$	536.5	552.3						
$\omega_4 (b_1)/cm^{-1}$	444.5	508.9						
$\omega_5 (b_2)/cm^{-1}$	1922.4	1931.0						
$_{6}$ (b <sub>2</sub> )/cm <sup>-1</sup>	20 <i>i</i> <sup><i>c</i></sup>	279.1						
$\Delta E (\text{kcal/mol})^{\text{b}}$	95.4	97.0						

<sup>*a*</sup> CASSCF orbitals averaged for the three lowest doublet electronic states.

<sup>*b*</sup> Relative to GaCH<sub>2</sub>, calculated at the same level of theory.

<sup>*c*</sup> The imaginary frequency indicates that for CGaH<sub>2</sub> the  $C_{2\nu}$  geometry is a saddle point at the

CASSCF/ICMRCI level of theory. Lowering symmetry restrictions, geometry optimization leads

to the trans-HCGaH isomer.

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset

PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

AIP Publishing

(primarily a lone pair 4*s* orbital on the gallium atom) to the HOMO gives rise to the low-lying  $\tilde{A}^2 A_I$  excited state with configuration

$$(\operatorname{core})(9a_1)^2(4b_2)^2(10a_1)^2(11a_1)^1(4b_1)^2$$

Finally, promotion of an electron from the doubly occupied  $11a_1$  orbital to the empty lowest unoccupied molecular  $5b_2$  orbital (LUMO,  $\pi^*$  antibonding) produces the  $\tilde{B}^2A_2$  state with dominant configuration

$$(\operatorname{core})(9a_1)^2(4b_2)^2(10a_1)^2(11a_1)^1(4b_1)^1(5b_2)^1.$$

The structural and spectroscopic properties of the  ${}^{2}B_{1}$  ground state of GaCH<sub>2</sub>, calculated both at ICMRCI and CCSD(T) levels, are collected in Table II. For both methods, we took the results obtained with the aug-cc-pVTZ basis as reference and tested their stability on increasing the size of the basis (aug-cc-pVQZ basis) and the effects of the inclusion the gallium d-electrons in the correlation treatment (aug-cc-pwCVTZ basis). The geometrical parameters obtained from the six combinations of method/basis agree within 0.04 Å, 0.002 Å and 0.3° for the Ga-C and C-H bond lengths and the H-C-H angle, respectively. As expected, the inclusion of d-electron correlation has its largest effect on the Ga-C bond length. The vibrational frequencies have only modest fluctuations as well, with the out-of-plane bending mode  $\omega_4$  showing the largest deviation (about 50 cm<sup>-1</sup> for the H-isotopologue).

The  ${}^{2}A_{1}$  first excited electronic state has been studied using the same combinations of method/basis and the results are tabulated in Table II. The agreement between the various approaches for the geometrical parameters r(GaC), r(CH) and  $\theta$ (HCH) is a bit looser, compared to the ground state, being 0.04 Å, 0.004 Å and 2°, respectively. Excluding  $\omega_{4}$ , the vibrational frequencies show general agreement within 51 cm<sup>-1</sup>. In contrast, the out-of-plane mode,  $\omega_{4}$ , shows very large fluctuations, with frequencies around 1000 cm<sup>-1</sup> for ICMRCI and ranging from about 1300 cm<sup>-1</sup> to more than 2100 cm<sup>-1</sup> for EOM-CCSD, with a clearly unphysical huge jump when the d-electron correlation is included. The source of such behavior is not clear.

Similar studies of the second excited electronic state, spanning the same combinations of methods and bases, are summarized in Table II. Initially, we assumed a planar  $C_{2\nu}$  structure with

TABLE II. Optimized geometries, vibrational frequencies, dipole moments and electronic excitation energies of the ground and first two excited electronic states of <sup>69</sup>GaCH<sub>2</sub> and <sup>69</sup>GaCD<sub>2</sub> (values separated by a slash) calculated at the ICMRCI and CCSD(T) (EOM-CCSD for excited states) levels of theory, for three bases of different size. For aug-cc-pV(T+d)Z and aug-cc-pV(Q+d)Z bases only, valence electrons are correlated, while for aug-cc-pwCVTZ basis, both valence electron and gallium d-electrons are correlated. Quantities in square brackets are H-D isotope shifts for comparison with experiment.

		IC-MRCI (MOLPRO)			CCSD(T) or EOM-CCSD (CFOUR)					
		aug-cc- pV(T+d)Z valence	aug-cc-pV(Q+d)Z valence electrons	aug-cc-pwCVTZ d-electrons correlated	aug-cc- pV(T+d)Z valence	aug-cc-pV(Q+d)Z valence electrons	aug-cc-pwCVTZ all electrons correlated			
)		electrons			electrons					
4	$\widetilde{X}^2 B_1$									
250	r(GaC)/Å	2.0393	2.0335	2.0131	2.0454	2.0402	2.0097			
010	<i>r</i> (CH)/Å	1.0947	1.0932	1.0934	1.0942	1.0927	1.0934			
3/5.(	$\Theta(\text{HCH})/\text{deg}$	110.6	110.7	110.9	110.6	110.7	110.9			
063	µ/debye	-0.618	-0.672	-1.047	-0.424	-0.463	-0.504			
0.1	$\omega_l(a_1)/cm^{-1}$	3031.8/2198.2	3038.1/2202.8	3037.0/2202.1	3040.3/2204.4	3046.5/2208.8	3044.1/2207.1			
	$\omega_2(a_1)/cm^{-1}$	1378.3/1029.7	1378.6/1030.6	1384.1/1034.9	1375.5/1027.8	1375.1/1028.1	1375.0/1028.3			
		[348.6]	[348.4]	[340.0]	[347.7]	[347.0]	[346.7]			
AS	$\omega_3(a_1)/cm^{-1}$	523.4/488.6	528.5/493.8	532.7/497.6	527.1/492.7	532.9/497.8	530.7/495.6			
БЩ		[34.8]	[34.1]	[34.7]	[34.4]	[35.1]	[35.1]			
	$\omega_4 (b_1)/cm^{-1}$	413.9/323.8	410.8/321.4	399.6/312.8	447.5/350.1	449.6/351.8	439.1/343.8			
R		[90.1]	[89.7]	[89.4]	[97.4]	[97.8]	[95.3.3]			
IS P	$\omega_5 (b_2)/cm^{-1}$	3115.8/2309.7	3125.9/2317.5	3125.0/2317.5	3124.5/2316.3	3134.6/2324.0	3132.4/2322.4			
H	$\omega_6(b_2)/cm^{-1}$	419.5/316.3	426.6/321.7	419,3/316.4	423.1/319.0	430.7/324.7	420.2/317.1			
		[103.2]	[103.4]	[104.9]	[104.1]	[106.0]	[103.1]			
0	$\tilde{A}^2 A_1$									
SE	r(GaC)/Å	1.8636	1.8584	1.8363	1.8769	1.8730	1.8336			
	<i>r</i> (CH)/Å	1.0842	1.0830	1.0830	1.0817	1.0805	1.0806			
a	$\Theta(\text{HCH})/\text{deg}$	122.3	122.1	122.4	123.6	123.6	124.0			
2	µ/debye	2.113	2.126	2.102	1.949	1.942	1.781			
5	$\omega_l(a_1)/cm^{-1}$	3114.1/2245.0	3117.3/2247.6	3118.9/2248.6	3146.7/2266.2	3150.6/2269.1	3151.8/2270.3			
2	$\omega_2(a_1)/cm^{-1}$	1228.5/921.6	1235.2/928.3	1243.9/936.0	1220.3/912.3	1225.7/917.7	1226.4/920.7			
5	$\omega_3(a_1)/cm^{-1}$	673.5/629.7	683.0/638.3	692.9/646.8	671.0/631.1	679.5/638.2	687.5/643.8			
	$\omega_4(b_1)/cm^{-1}$	981.0/765.5	1005.4/784.7	1037.6/810.1	1264.9/986.2	1305.6/1017.9	2136.1/1666.2			
5	$\omega_5(b_2)/cm^{-1}$	3252.1/2423.7	3258.8/2428.9	3259.9/2430.0	3294.4/2457.3	3301.8/2462.8	3303.4/ 2464.3			
5	$\omega_6(b_2)/cm^{-1}$	519.1/389.6	527.5/396.0	530.7/398.6	530.5/397.5	538.4/403.4	534.5/400.9			
	$T_0 /{\rm cm}^{-1}$	7032/6944	6859/6768	6847/6749	10690/10565	10794/10665	13467/13244			
5				$\widetilde{B}^2 A_2$ (Planar)						
	r(GaC)/Å	1.9918	1.9856	1.9614	1.9995	1.9971	1.9662			
)	r(CH)/Å	1.0863	1.0851	1.0851	1.0839	1.0825	1.0828			
	$\Theta(\text{HCH})/\text{deg}$	116.9	116.9	117.1	117.1	117.1	117.7			
2	µ/debye	-0.588	-0.599	-0.846	-0.629	-0.633	-0.781			

The Journal of Chemical Physics

AIP Publishing

	10
	ö
J.	S
-	L L
na	<u> </u>
ur	g
ō	Ĕ
0	en
Å	S.
and the second s	<u> </u>

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset. **PLEASE CITE THIS ARTICLE AS DOI:** 10.1063/5.0182504

$\omega_l (a_1)/cm^{-1}$	3093.0/2233.4	3097.4/2236.8	3098.5/2237.5	3125.4/2256.6	3131.3/2260.9	3131.0/2260.1
$\omega_2(a_1)/cm^{-1}$	1347.8/997.9	1352.2/1002.2	1361.5/1009.5	1350.6/999.6	1352.7/1001.6	1350.2/998.8
	[349.9]	[350.0]	[323.0]	[351.0]	[351.1]	[351.4]
$\omega_3(a_1)/cm^{-1}$	557.2/527.1	566.5/536.0	576.2/545.0	557.5/528.3	561.7/532.0	556.6/527.9
	[30.1]	[30.5]	[30.5]	[29.2]	[29.7]	[28.7]
$\omega_4 (b_1)/cm^{-1}$	242.5 <i>i</i> /189.5 <i>i</i>	221.2 <i>i</i> /172.7 <i>i</i>	271.8 <i>i</i> /212.3 <i>i</i>	695.0/542.5	700.0/546.4	762.5/595.4
				[152.5]	[153.6]	[167.1]
$\omega_5 (b_2)/cm^{-1}$	3206.4/2387.9	3213.8/2393.6	3214.5/2394.6	3240.8/2414.2	3250.0/2421.1	3251.8/2424.4
$\omega_6(b_2)/cm^{-1}$	704.4/528.1	710.9/533.1	724.5/543.5	697.5/522.7	702.2/526.2	714.4/535.5
<b>T</b> ( 1	[176.4]	[177.8]	[181.0]	[17/4.8]	[176.0]	
$T_0$ /cm <sup>-1</sup>	а	а	а	21017/20929	21258/21171	24276/24177
			$\widetilde{\mathbf{p}}^2 \mathbf{A} / (\mathbf{O} \mathbf{I}_{\text{constrained}})$	[88]	[88]	[99]
$r(\mathbf{C} \circ \mathbf{C})/\mathbf{\hat{\lambda}}$	2 0015	1 0051	$B^{2}A^{\prime\prime}$ (Nonplanar)			
r(GaC)/A	2.0015	1.9951	1.9/10			
P(CH)/A	1.0000	1.08/4	1.0874			
0(HCH)/deg	24.0	24.0	24.1			
barriar/cm <sup>-1 c</sup>	150	24.0	24.1 75			
$\frac{\partial u n e^{-1}}{\partial u}$	3074 2/2221 0	3078 7/2224 4	3080 5/2225 6			
$\omega_1(a_1)/cm^{-1}$	1352 2/1002 7	1354 5/1005 0	1367.7/1015.1			
$\omega_2(a_1)/cm^2$	[349 5]	[349.5]	[352 6]			
$(a_1)/cm^{-1}$	543 9/509 6	551 6/516 8	561 7/526 3			
<i>w</i> <sub>3</sub> ( <i>a</i> <sub>1</sub> )/em	[34.3]	[34.8]	[35.4]			
$\omega_{4}(b_{1})/cm^{-1}$	334.6/262.8	341.4/268.0	337.3/264.7			
	[71.8]	[73.4]	[72.6]			
$\omega_5 (b_2)/cm^{-1}$	3181.2/2367.0	3188.1/2372.1	3188.6/2372.8			
$\omega_6(b_2)/cm^{-1}$	719.5/536.6	722.6/539.2	740.3/552.5			
	[182.9]	[183.4]	[187.8]			
$T_0 /{ m cm}^{-1}$	20668/20622	20791/20746	21551/21500			
	[46]	[45]	[51]			
	-	-	$\widetilde{a}^4 A_2$	•		
r(GaC)/Å	1.9971	1.9905				
<i>r</i> (CH)/Å	1.0861	1.0849				
$\Theta$ (HCH)/deg	121.5	121.5				
µ/debye	-0.775	-0.785				
$\omega_l (a_1)/cm^{-1}$	3098.7/2237.1	3104.8/2241.7				
$\omega_2(a_1)/cm^{-1}$	1341.3/992.2	1356.2/1004.3				
$\omega_3(a_1)/cm^{-1}$	540.9/ 512.1	558.6/528.3				
$\omega_{4}(b_{1})/cm^{-1}$	681.4/ 531.9	693.8/541.7				
$\omega_{\rm s}({\rm h_2})/{\rm cm^{-1}}$	3212 5/ 2392 8	3220.1/2398 7				
$\omega_{\rm s}(b_2)/cm^{-1}$	683 9/ 512 7	693 8/ 526 4				
$T_0/cm^{-1}$	183/1/18262	18878/18780				
10/CIII	10344/10202	100/0/10/09				

 $^{a}T_{0}$  not calculated.

<sup>*b*</sup> Ga atom out of the HCH plane angle.

<sup>*c*</sup> Energy barrier to  $C_{2v}$  geometry.

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset

PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

term symbol  $\tilde{B}^2A_2$ . The geometric parameters obtained with this constraint are very similar, with r(GaC), r(CH) and  $\theta$ (HCH) differences within 0.04 Å, 0.004 Å and 0.8°, respectively. Moreover, all the vibrational frequencies, excluding  $\omega_4$ , agree within 46 cm<sup>-1</sup>. At the ICMRCI level, the  $\omega_4$  frequency is always imaginary, indicating that the imposed  $C_{2\nu}$  symmetry corresponds to a transition state. Only small differences are observed on changing the basis set. Conversely, at the EOM-CCSD level, the  $\omega_4$  frequency is definitely positive, with calculated frequencies around 700 cm<sup>-1</sup>. As we shall see, such high values are clearly unreliable, as they do not have any correspondence with experiment (*vide infra*), but nevertheless they are reported for completeness.

The question is: why are the ICMRCI and EOM-CCSD results so different for the  $\omega_4$  mode? An important clue comes from an examination of the spin expectation values in the CFOUR EOM-CCSD wavefunctions of the excited states. We restrict our discussion to the results from the aug-cc-pVTZ basis as the outcomes from other bases are very similar. For the  $\tilde{A}^2A_1$  state, the expectation value  $S^2$  calculated by CFOUR at the equilibrium geometry is 0.80, which is fairly close to the theoretical value for a pure doublet of 0.75. The situation is very different for the second excited state with CFOUR giving  $S^2 = 2.91$ , which is closer to the 3.75 expectation value of a pure quartet state. Clearly, the EOM-CCSD wavefunction of the second excited state is strongly spin contaminated, casting doubt on the calculated properties.

On the other hand, the results from Molpro do not suffer from spin contamination, since the CASSCF and ICMRCI method is based on Configuration State Functions (CSF) which are eigenfunctions of both  $S^2$  and  $S_z$  [see for example ref. 26]. On lowering the  $C_{2v}$  symmetry constraint in the calculation of the second excited state, the ICMRCI method finds a bent structure with small out-of-plane angle ( $\approx 24^\circ$ ) and very small barrier to planarity (159 cm<sup>-1</sup> with the aug-cc-pVTZ basis). It turns out that the ICMRCI calculated  $\omega_4$  frequency, its H/D isotopic shift and the computed  $T_0$  isotope shift are all in reasonable accord with experiment (*vide infra*), suggesting that they are the most reliable.

In summary, ICMRCI computations point to a slightly nonplanar structure for the second excited state. However, this finding is in conflict with experiment, since no single quanta transitions of  $\omega_4$  are observed. It can be argued that, in the presence of a barrier to planarity much lower than the vibrational zero-point energy, the molecule may behave as if it is effectively planar. Most probably, the recovered nonplanar structure is due to the limitations, both intrinsic and practical, of the ICMRCI method we used. In fact, we observed (see Table II) that the barrier tends

ACCEPTED MANUSCRIPT

The Journal of Chemical Physics

> AIP Publishing

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset

PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

to decrease either by increasing the basis size or by extending the correlation treatment to include gallium d-electrons, suggesting that a planar structure may be achieved by increasing both simultaneously.

Another pitfall of our treatment may be the neglect of spin-orbit interactions with neighboring quartet states. Indeed, the first quartet state of GaCH<sub>2</sub> ( $\tilde{a}^4A_2$ , see Table II) lies only ~2500 cm<sup>-1</sup> below the second excited doublet state. At planar  $C_{2\nu}$  geometries, no spin-orbit interaction between the  $\tilde{B}^2A_2$  and  $\tilde{a}^4A_2$  states can take place. However, for  $C_s$  non-planar geometries, a bending-dependent nonzero doublet-quartet spin-orbit interaction can occur through the  $LS_i$  component of the spin-orbit operator, <sup>27</sup> where *i* is the axis perpendicular to the symmetry plane of the molecule. The energy of the quartet state increases quickly upon bending (the  $\omega_4$ frequency of the  $\tilde{a}^4A_2$  state is ~700 cm<sup>-1</sup>, see Table II) as well as the extent of the interaction between the two states, resulting in a "push-up" effect on the out-of-plane bending potential of the  $\tilde{B}^2A_2$  state when spin-orbit interaction is included. In short, we speculate that the slight nonplanarity of the second excited state at the ICMRCI level is an artifact originating from a synergic effect of the many inherent approximations made in the calculation.

## **IV. RESULTS AND ANALYSIS**

## A. 2D LIF survey spectra

Fig. 1 shows a small portion of the 2D LIF spectrum obtained from the products of a discharge through TMGa vapor. The top panel shows the total LIF signal, and the bottom panel shows the color-coded emission spectrum obtained at each laser wavenumber. The emission spectra are displayed vertically as displacement (in cm<sup>-1</sup>) from the laser (laser wavenumber – emission wavenumber) which gives a direct measure of the lower state energy of each transition. The spectra show a series of three bands in the 22 200 – 23 700 cm<sup>-1</sup> region that all have similar rotational contours and similar emission spectra, very reminiscent of the 2D LIF spectrum of AlCH<sub>2</sub>.<sup>3</sup> The emission spectra show prominent intervals of 515 and 1340 cm<sup>-1</sup>, which can be

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset

PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

readily ascribed to the  $v_3''$  (Ga-C stretch) and  $v_2''$  (CH<sub>2</sub> scissors) vibrations (Table II) of GaCH<sub>2</sub>, respectively. If the first LIF feature is assigned as the 0-0 band, then the maximum at 22 230 cm<sup>-1</sup> compares favorably to the calculated  $T_0$  of 21 551 cm<sup>-1</sup> (nonplanar, aug-cc-pwCVTZ, d-electrons correlated Table II) for GaCH<sub>2</sub>. The 2D LIF spectrum is complicated by fluorescence from other impurity molecules formed in the discharge, particularly at energies above 23 000 cm<sup>-1</sup>. Nevertheless, all indications are that the three bands identified in the 2D LIF spectrum are due to the previously unidentified gallium methylene free radical.

## B. 0-0 band LIF spectra and confirmation of the gallium methylene assignment

Medium resolution LIF spectra of the 0-0 bands of  $GaCH_2$  and  $GaCD_2$  at two different rotational temperatures are shown in Fig. 2, along with calculated band contours based on the rotational constants obtained from the ground and excited state *ab initio* molecular geometries. The rotational structure was computed using the PGOPHER program,<sup>28</sup> assuming a linewidth of

0.4 cm<sup>-1</sup>, neglecting unpaired electron spin-rotation and gallium atom hyperfine effects, which are not resolved in our spectra, and including only the <sup>69</sup>Ga isotopologue in each case. For GaCH<sub>2</sub> with a <sup>2</sup>B<sub>1</sub> ground state, nuclear statistical weights favor even  $K_a$  values by a factor of 3, whereas in GaCD<sub>2</sub>, odd  $K_a$  values have twice as many nuclear spin states as even values. These expectations are robustly borne out in the GaCH<sub>2</sub> spectrum in Fig. 2, with a strong central <sup>*r*</sup>Q<sub>0</sub> branch flanked by weaker <sup>*p*</sup>Q<sub>1</sub> and <sup>*r*</sup>Q<sub>1</sub> branches. In GaCD<sub>2</sub>, the situation is reversed, with a weak central *Q* branch and stronger  $K_a''=1$  *Q* branches on either side, precisely as expected. Furthermore, the separation of the *Q* branches is consistent with the calculated molecular structures and the lack of resolved rotational structure is indicative of overlapping spectra from the <sup>69</sup>Ga (60.1 %) and <sup>71</sup>Ga (39.9 %) isotopologues in ACCEPTED MANUSCRIPT

The Journal of Chemical Physics A

AIP Publishing

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset. PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504



**FIG. 1.** A segment of the 2D LIF spectrum of the products of an electric discharge through a mixture of trimethylgallium vapor and high-pressure argon. The horizontal axis is the excitation laser wavenumber (cm<sup>-1</sup>). The top spectrum is the total LIF fluorescence, and the bottom panel shows the emission spectrum obtained at each excitation laser wavenumber. The emission spectra are arranged vertically and are plotted as displacement in cm<sup>-1</sup> from the laser wavenumber.

ACCEPTED MANUSCRIPT

The Journal of Chemical Physics

AIP Publishing

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset. PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504



**FIG. 2.** Low resolution LIF spectra of the 0-0 bands of GaCD<sub>2</sub> (top) and GaCH<sub>2</sub> (bottom). In each case, the top trace is the spectrum obtained at a rotational temperature of 8-10 K, the middle trace was obtained under milder expansion conditions (30-40 K) and the bottom trace is the spectrum calculated from the planar IC-MRCI (aug-cc-pwCVTZ d-electrons correlated) *ab initio* molecular structures in Table II. The letter H denotes GaCH<sub>2</sub> impurity features in the GaCD<sub>2</sub> spectra.

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset

PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

natural abundance. By fitting the observed *Q*-branch wavenumbers of the 30 K 0-0 bands (Fig. 2) to

$$\overline{\nu}(Q) = T_{00} + (A - \overline{B})'(K_a')^2 - (A - \overline{B})''(K_a'')^2$$
(1)

we obtained approximate values for the band origin and  $(A - \overline{B}) \approx A$  values of the upper and lower states, yielding GaCH<sub>2</sub>:  $(A - \overline{B})'' = 10.010 \pm 0.036$ ,  $(A - \overline{B})' = 9.332 \pm 0.018$  and GaCD<sub>2</sub>:  $(A - \overline{B})'' = 4.870 \pm 0.009$  and  $(A - \overline{B})' = 4.418 \pm 0.009$  cm<sup>-1</sup>. The calculated H/D 0-0 band isotope shift of 45-51 cm<sup>-1</sup> (Table II) is also in good accord with the experimental value of 32.4 cm<sup>-1</sup> derived from the fitted band origins.

We attempted to record high-resolution LIF spectra of the 0-0 band of GaCH<sub>2</sub>, as shown in Fig. 3. The  $K'_a = 1 - K''_a = 0$  *P*- and *R*-branches clearly show the splittings expected for a molecule in a doublet state, further confirming the spectrum as due to the gallium methylene free radical. Unfortunately, due to the presence of the two gallium isotopes in natural abundance, the crucial asymmetry-split branches ( ${}^{r}R_1$ ,  ${}^{r}P_1$ ,  ${}^{p}R_2$  and  ${}^{p}P_2$ ) were not resolvable at our highest resolution (0.035 cm<sup>-1</sup>) so a detailed rotational analysis and experimental determination of the molecular structure was not possible in the present investigation. We defer discussion of the vibronic assignments of the vibrationally resolved LIF spectra until after the analysis of the emission spectra.



**FIG. 3.** A high resolution LIF spectrum of the central portion of the 0-0 band of GaCH<sub>2</sub> showing resolved spin-splittings.

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset. PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

The Journal of Chemical Physics

AIP Publishing

AIP Publishing This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

## C. Emission spectra and the vibrational energy levels of the ground state

## 1. 0-0 band emission spectra and selection rules

The 0-0 band emission spectra of GaCH<sub>2</sub> and GaCD<sub>2</sub> are compared in Fig. 4, plotted as displacement from the laser excitation wavenumber, giving a direct measure of the ground state energy of each transition. We have also recorded extensive emission spectra from the various other LIF bands of both isotopologues and these are reported in the Supplementary Information. Since the LIF spectra show resolved rotational subbands, laser excitation of different branches gives somewhat different emission spectra. As shown in Fig. 4,  ${}^{p}Q_{I}$  excitation gives a single  $K'_{a} = 0 - K''_{a} = 1$  transition down to each lower state vibronic level. Due to the nuclear statistical weights and rotational cooling in the free jet expansion, these branches are intense in GaCD<sub>2</sub> but relatively weak in GaCH<sub>2</sub> (see Fig. 2). For comparison, we show in Fig. 4 the corresponding spectra obtained by GaCH<sub>2</sub>  ${}^{r}Q_{0}$  excitation, which gives two branches ( $K'_{a} = 1$  down to  $K''_{a} = 0$  and 2) with an approximate spacing of 4A" and GaCD<sub>2</sub>  ${}^{r}Q_{I}$  excitation (to  $K''_{a} = 1$  and 3 with splitting  $\approx 8 A''$ ).

The vibronic selection rules mandate unrestricted quantum number changes for the totally symmetric vibrational modes ( $v_1 - v_3$ ) of planar GaCH<sub>2</sub>. Although transitions to even quanta ( $4_2$ ,  $6_2$  etc.) of the non-totally symmetric modes are formally allowed, they are expected to be very weak unless the vibrational frequency changes substantially on electronic excitation, as is likely the case (Table II) for modes 4 and 6. In addition, vibronically induced bands may occur in the spectrum. The transition moment integral, with operator *M*, including vibronic interaction is

$$R_{e'v'e''v''} = \int \Psi_{ev}^{\prime*} \, M \, \Psi_{ev}^{\prime\prime} \, d\tau \tag{2}$$

AIP Publishing





FIG. 4. The 0-0 band emission spectra of GaCD<sub>2</sub> (top) and GaCH<sub>2</sub> (bottom). For GaCD<sub>2</sub>, the upper trace is emission from  $K'_a = 0$  (<sup>*p*</sup> $Q_I$  excitation) and the lower trace is emission from  $K'_a = 2$  (<sup>*r*</sup> $Q_I$ excitation). For GaCH<sub>2</sub>, the upper trace is emission from  $K'_a = 0$  (<sup>*p*</sup>Q<sub>1</sub> excitation) and the lower trace is emission from  $K'_a = 1$  (<sup>*r*</sup> $Q_0$  excitation). Features with the same lower state vibrational assignments are linked by dashed lines.

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset

PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

which, for a  $b_2$  GaCH<sub>2</sub> vibration ( $v_5$  and  $v_6$ ), can only be nonzero for a transition moment along the *a*-axis (C<sub>2</sub> axis) such that  $A_2 \times B_1 \times b_2 = A_1$ . Thus, the  $6_1^0$  parallel band can be vibronically induced by borrowing intensity from an allowed  $B_1 \leftarrow B_1$  electronic transition. In direct contrast, for the  $v_4$  vibration of  $b_1$  symmetry, the product of the vibronic symmetries is  $A_2 \times B_1 \times b_1 = A_2$ and there is no transition moment of  $A_2$  symmetry in  $C_{2v}$  so the  $4_1^0$  band cannot be vibronically induced. Finally, for a combination band such as  $4_1^0 6_1^0$ , the direct product is  $A_2 \times B_1 \times b_1 \times b_2 =$  $B_1$  which is allowed for a transition involving *c*-type rotational selection rules. In summary, vibronically induced transitions involving odd quantum number changes in the  $b_2$  modes can occur as parallel bands and those involving odd combinations of  $b_1 + b_2$  modes will appear as perpendicular bands indistinguishable at modest resolution from the allowed *b*-type bands in the spectrum.

If GaCH<sub>2</sub> is slightly nonplanar ( $C_s$  symmetry) in the excited state, as predicted by our IC-MRCI calculations, then v<sub>1</sub> - v<sub>4</sub> are all of a' symmetry and unrestricted quantum number changes in these modes are allowed, giving rise to perpendicular bands. The greater the nonplanar distortion, the greater the expected Franck-Condon activity in the v<sub>4</sub> out-of-plane bending mode. Although rigorously forbidden in  $C_{2v}$  symmetry, the 4<sup>0</sup><sub>1</sub> band would be expected to gain intensity in proportion to the extent of nonplanarity.

Before assigning the emission spectra, it is also necessary to consider the rovibronic structure of the observed bands in more detail, as illustrated in Fig. 5. The spectra are complicated by the observation of resolved subbands and the near degeneracy of  $v_4''$  and  $v_6''$ . What we measure is the displacement of each band from the excitation laser wavenumber which gives a direct measure of the ground state vibrational energy. As shown on the LHS of Fig. 5,  ${}^{p}Q_{1}$  excitation

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504 gives a displacement  $\mathbf{D}_1$  (laser wavenumber – transition wavenumber) which measures the interval between  $K_a = 1$  of the ground state vibrational level of interest (3<sub>1</sub> in the Fig.) and  $K_a = 1$  of the zero-point level whereas  ${}^{r}Q_0$  excitation gives a more direct measure  $\mathbf{D}_2$  of the  $K_a = 0$  separation. As long as the  $(A - \overline{B})$  values are comparable in the two ground state vibrational levels, as they usually are,  $\mathbf{D}_1 = \mathbf{D}_2$  within the  $\pm 1$  cm<sup>-1</sup> accuracy of the emission measurement. For example, we measure 3<sub>1</sub> to have a vibrational energy of 516.6 cm<sup>-1</sup> from both 0-0 band  ${}^{p}Q_1$  and  ${}^{r}Q_0$  emission spectra. Since  ${}^{p}Q_1$  excitation results in single  $K'_a = 0 - K''_a = 1$  transitions, with many fewer overlapping lines, we usually choose such spectra for displacement measurements.

In the case of vibronically induced bands, such as  $6_1^0$ , the emission bands obey *a*-type ( $\Delta K_a = 0$ ) selection rules and, in the absence of perturbations, one would expect all the *Q*-branch transitions to be coincident. However,  $6_1$  is very likely perturbed by the  $4_1$  level just below it as shown in the middle of Fig. 5. The product of the  $6_1$  and  $4_1$  vibrational symmetries is  $b_2 \times b_1 = A_2$ , which mandates an *a*-axis Coriolis interaction with matrix element  $\xi_{46}^a K_a$ . This would not affect the  $6_1 K_a = 0$  levels, but  $K_a = 1$ , 2 would be progressively displaced upward, as illustrated. In this case, the  ${}^pQ_1$  displacement **D**\_3 needs to be corrected by adding the ground state  $(A - \overline{B})$  value, whereas the  ${}^rQ_0$  displacement **D**\_4 is larger than necessary by the  $6_1 K_a = 1 - K_a = 0$  interval.

> AIP Publishing

ACCEPTED MANUSCRIPT

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504



**FIG. 5.** Schematic energy level diagram showing the intervals that define the measured emission displacements (**D**<sub>1</sub>, **D**<sub>2</sub> etc.) subsequent to laser excitation of the  ${}^{p}Q_{1}$  and  ${}^{r}Q_{0}$  rotational branches of the 0-0 band of gallium methylene. The dotted lines denote Coriolis interactions between adjacent vibrational levels. The dashed lines indicate  ${}^{p}Q_{1}$  and  ${}^{r}Q_{0}$  laser excitation transitions.

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset

PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

Finally, we consider emission transitions down to the 4<sub>2</sub>, 4<sub>1</sub>6<sub>1</sub> and 6<sub>2</sub> triad of levels, which appear in many of the spectra. In this case, all the transitions involve  $\Delta K_a = \pm 1$ , but there are Coriolis complications, with  $\Delta K_a = 0$  *a*-axis interactions possible between 4<sub>2</sub> and 4<sub>1</sub>6<sub>1</sub> and between 4<sub>1</sub>6<sub>1</sub> and 6<sub>2</sub>. These would have the effect of compressing the *K* stacks in 4<sub>2</sub>, inflating them in 6<sub>2</sub> and conceivably leaving them largely unchanged in 4<sub>1</sub>6<sub>1</sub>, as illustrated in Fig. 5. For emission down to 4<sub>2</sub>, the <sup>*p*</sup>Q<sub>1</sub> displacement **D**<sub>5</sub> is then less than the true vibrational interval **D**<sub>6</sub>. The energy level diagram indicates that the <sup>*p*</sup>Q<sub>1</sub> displacement  $\approx {}^{$ *r* $}Q_0$  displacement for 4<sub>1</sub>6<sub>1</sub>, and the 6<sub>2</sub> emission is the opposite of 4<sub>2</sub> (<sup>*p*</sup>Q<sub>1</sub> displacement >  ${}^{$ *r* $}Q_0$  displacement). In all cases, the <sup>*r*</sup>Q<sub>0</sub> displacement gives the correct vibrational interval, whereas the 4<sub>2</sub> and 6<sub>2</sub>  ${}^{$ *p* $}Q_1$  displacements have to be corrected for Coriolis effects.

## 2. Vibrational assignments

For brevity in our discussion of the analysis of the emission spectra, comparisons of the vibrational frequencies will subsequently be denoted as GaCH<sub>2</sub>/GaCD<sub>2</sub> and, except where noted, all quoted *ab initio* values refer to the IC-MRCI aug-cc-pwCVTZ, d-electrons correlated results from Table II. The resonance fluorescence down to the zero-point level dominates the 0-0 band spectra (Fig. 4), consistent with *ab initio* predictions that the molecular structure changes little on electronic excitation. The second most intense feature is readily assigned as the transition to  $v_2'' = 1347/1011 \text{ cm}^{-1}$  (*ab initio* = 1384/1035 cm<sup>-1</sup>). The 2<sub>n</sub> progression can be readily followed out to 3 quanta in GaCH<sub>2</sub> and 5 quanta in the various emission spectra of GaCD<sub>2</sub>. The next prominent feature occurs at 517/478 cm<sup>-1</sup> (*ab initio* = 533/498 cm<sup>-1</sup>) and the lack of a substantial deuterium isotope effect identifies it as  $v_3''$  (Ga-C stretch) which can be followed in various spectra out to 3<sub>5</sub> in both isotopologues. Other prominent features in the emission spectra are readily identified as

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset.

PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

transitions down to combinations of  $v_2$  and  $v_3$ . The  $v_1''$  frequency is predicted to be very high (*ab initio* = 3037/2202 cm<sup>-1</sup>) and has not been positively identified in any of our spectra.

The 0-0 band emission spectra exhibit very weak  ${}^{r}Q_{0}$  excitation features at 404/316 cm<sup>-1</sup> and overlapping clusters of bands in the 800-852/622-648 cm<sup>-1</sup> regions, all of which must involve the low-frequency, non-totally symmetric modes v<sub>4</sub> and v<sub>6</sub>. The 404/316 cm<sup>-1</sup> intervals can only be assigned as single quanta, which must be either the 4<sup>0</sup><sub>1</sub> (*ab initio* = 400/313 cm<sup>-1</sup>) perpendicular transitions or the vibronically induced 6<sup>0</sup><sub>1</sub> (*ab initio* = 419/316) parallel transitions. Since experimentally  ${}^{p}Q_{I}$  excitation gives single features and  ${}^{r}Q_{0}$  excitation gives doublets, we assign the 404/316 cm<sup>-1</sup> features as the 4<sup>0</sup><sub>1</sub> bands which gain intensity due to the slight nonplanarity of the excited state. The 4<sup>0</sup><sub>1</sub>  ${}^{r}Q_{0}$  splittings are anomalously small (19/8 cm<sup>-1</sup> compared to 42.2/20.5 cm<sup>-1</sup> for the 3<sup>0</sup><sub>1</sub> emission bands), which we hypothesize must be due to v<sub>4</sub>:v<sub>6</sub> Coriolis coupling, much as occurs in formaldehyde.<sup>29</sup> Thus, if 6<sub>1</sub> is slightly above 4<sub>1</sub>, the *a*-axis Coriolis interaction does not perturb the  $K_{a} = 0$  levels but the  $K_{a} = 1$  levels repel each other, accounting for the small 4<sub>1</sub> splitting (see Fig. 5).

Building on the v<sub>6</sub> identification, GaCH<sub>2</sub>  ${}^{r}Q_{0}$  features at 808 (2 × 404), 855 (2 × 427.5) and 831 (404 + 427) cm<sup>-1</sup> can then be readily assigned to the levels 4<sub>2</sub>, 6<sub>2</sub> and 4<sub>1</sub>6<sub>1</sub>. Similar GaCD<sub>2</sub>  ${}^{r}Q_{0}$ features were found at 629 (2 × 314.5) [4<sub>2</sub>], 655 (2 × 327.5) [6<sub>2</sub>] and 642 (314.5 + 327.5) [4<sub>1</sub>6<sub>1</sub>] cm<sup>-1</sup>. These energy levels are shown schematically on the RHS of Fig. 5.

The  $6_2^0$  bands derive their intensity from the substantial difference in the  $v_6$  frequencies in the combining electronic states. Using the harmonic oscillator approximation, an expression for the  $6_2^0$  to  $0_0^0$  band intensity ratio can be derived as<sup>30</sup>

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset

$$\frac{I(6_2^0)}{I(0_0^0)} = \frac{1}{2} \frac{\overline{v}^3(6_2^0)}{\overline{v}^3(0_0^0)} \left[ \frac{\omega_6' - \omega_6''}{\omega_6' + \omega_6''} \right]^2 \tag{3}$$

where the  $\overline{v}$  symbols are the vibronic transition frequencies (cm<sup>-1</sup>) and the  $\omega$ 's are the vibrational frequencies in the two states. Using GaCH<sub>2</sub> *ab initio* frequencies of  $\omega'_6 = 740.3$  and  $\omega''_6 = 419.3$  cm<sup>-1</sup> (Table II) gives a ratio of 0.034, about a factor of 5 larger than the experimental value of 0.006. Further combinations and overtones of these intervals can be found throughout the spectra.

Fig. 6 shows a comparison of the emission spectra from two of the weaker, higher wavenumber LIF bands, one of GaCD<sub>2</sub> ( ${}^{p}Q_{1} = 23570.8 \text{ cm}^{-1}$ ) and one of GaCH<sub>2</sub> ( ${}^{p}Q_{1} = 23363.2 \text{ cm}^{-1}$ ) These spectra differ dramatically from those of the 0-0 bands (Fig. 4), with very weak fluorescence down to 0<sub>0</sub>, 3<sub>1</sub> and 2<sub>1</sub> but very prominent transitions to the 4<sub>2</sub>, 4<sub>1</sub>6<sub>1</sub> and 6<sub>2</sub> triplet of levels. In addition, the GaCD<sub>2</sub> overtones 4<sub>4</sub>, 4<sub>3</sub>6<sub>1</sub> and 6<sub>4</sub> are also readily identified. In addition, some of the emission spectra of GaCD<sub>2</sub> (see Supplementary Information) exhibit strong transitions down to levels at 1313 and 1333 cm<sup>-1</sup> in the ground state . The only viable assignments on energetic grounds are 2<sub>1</sub>4<sub>1</sub> and 2<sub>1</sub>6<sub>1</sub>, respectively, although such bands would normally be expected to be very weak.

Although the emission spectra extend to displacements of  $3\ 000 - 4\ 000\ \text{cm}^{-1}$ , in many cases assignments of the less prominent bands above 1 500 cm<sup>-1</sup> proved quite difficult, due to Coriolis coupling, Fermi resonances, overlap and multiple possible quantum number



This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset. PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

Chemical Physics

The Journal of

AIP Publishing



**FIG. 6.** Portions of the emission spectra of GaCD<sub>2</sub> (top,  ${}^{p}Q_{1}$  branch at 23570.8 cm<sup>-1</sup>) and GaCH<sub>2</sub> (bottom,  ${}^{p}Q_{1}$  branch at 23363.2 cm<sup>-1</sup>) that show strong transitions down to levels involving  $v_{4}^{\prime\prime}$  and  $v_{6}^{\prime\prime}$ .

> AIP Publishing

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504 combinations. It also proved impossible to differentiate between the  $4_{3}6_{1}$  and  $4_{1}6_{3}$  levels, which are nearly degenerate, so we have simply assigned them as the former. We present in Table III the measured ground state vibrational energy levels of both isotopologues along with our best attempts at assignments.

Although the  $\tilde{A}^2 A_1 - \tilde{X}^2 B_1$  electronic transition is orbitally forbidden, we speculated that the emission from the upper state might be observable as a series of weak vibronically induced bands, which would pinpoint the location of the  $\tilde{A}$  state. For this purpose, we fixed the laser on the intense  ${}^{r}Q_0$  branch of GaCH<sub>2</sub> and stepped our monochromator in overlapping segments from 550 – 850 nm, recording any emission with our gated CCD detector. No GaCH<sub>2</sub> fluorescence was observed.

## D. Analysis of the LIF spectra

Fig. 7 shows the complete LIF spectra of GaCH<sub>2</sub> and GaCD<sub>2</sub>, recorded at a rotational temperature of 8-10 K. Above 22 800 cm<sup>-1</sup> the spectra were overlapped by impurity bands (GaH, GaD, CH, CD etc.) with fluorescence lifetimes much shorter than the  $\sim$ 2 - 4 µs decays of gallium methylene. We have used temporal gating to eliminate these strong impurity bands from the spectra in Fig. 7. In addition, we recorded emission spectra from all the bands that had the appropriate gallium methylene fluorescence lifetime and rotational contour as an aid to making upper state assignments. We note that the emission spectra were useful when there were overlapping bands (cf. particularly 23 400 -23 600 cm<sup>-1</sup> of GaCD<sub>2</sub> LIF) as the number of peaks

AIP Publishing

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset. PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

TABLE III. Summary of the ground state vibrational levels of GaCH<sub>2</sub> and GaCD<sub>2</sub> measured from single vibronic level emission spectra.

	GaCH <sub>2</sub>			GaCD <sub>2</sub>	
Energy (cm <sup>-1</sup> )	Assignment	Comment	Energy (cm <sup>-1</sup> )	Assignment	Comment
404	41	$v_{6} = 404$	314	41	$v_4 = 314$
517	31	$v_3 = 517$	478	31	$v_3 = 478$
808	42	404+404	622	42	$4_1 + 308$
831	4161	$4_1 + 427$	634	4161	$4_1 + 320$
855	62	$v_6 = 428$	648	62	$v_6 = 324$
1032	32	3 <sub>1</sub> + 515	951	32	$3_1 + 473$
1298	3142	31 + 781	1011	21	$v_2 = 1011$
1318	314161	$3_1 + 801$	1092	3142	$3_1 + 614$
1347	21	$v_2 = 1347$	1107	314161	$3_1 + 629$
1542	33	$3_2 + 510$	1118	3162	$3_1 + 640$
1632	44	$4_2 + 824$	1255	44	$4_2 + 633$
1647	4361	$4_{1}6_{1} + 816$	1270	4361	$4_{1}6_{1} + 636$
1677	64	$6_2 + 822$	1292	64	$4_2 + 644$
1791	3242	$3_2 + 759$	1313	2141	$2_1 + 302$
1816	324161	$3_2 + 784$	1333	2161	$2_1 + 322$
1864	2131	$2_1 + 517$	1417	33	$3_2 + 466$
2048	34	$3_3 + 506$	1486	2131	$2_1 + 475$
2105	3144	$4_4 + 473$	1544	3242	$3_14_2 + 452$
2120	314361	$4_{3}6_{1} + 473$	1577	324161	314161 +470
2156	3164	$6_4 + 479$	1583	3262	$3_{1}6_{2} + 465$
2171	2142	$2_1 + 824$	1637	2142	$2_1 + 626$
2183	214161	$2_1 + 836$	1659	214161	$2_1 + 648$
2214	2162	$2_1 + 867$	1667	2162	$2_1 + 656$
2286?			1737	3144	$4_4 + 482$
2376	2132	$3_2 + 1344$	1745	314361	$4_{3}6_{1} + 475$
2464	46	$4_4 + 832$	1759	3164	$6_4 + 467$
2499	66	$6_4 + 822$	1765?		
2550	35	34 + 502	1878	34	$3_3 + 461$
2581?			1952	2132	$2_13_1 + 466$
2658	$2_13_14_2$	$2_13_1 + 794$	1997?		
2673	$2_2$	$2_1 + 1326$	2007	$2_2$	$2_1 + 996$
2685?			2020	3342	$3_3 + 602$
2708	$2_13_14_2$	$2_13_1 + 844$	2097?		
2885	2133	$2_13_2 + 509$	2108	213142	$2_13_1 + 622$
2971	$2_{1}4_{4}$	$4_4 + 1339$	2123	21314161	$2_13_1 + 637$
2994	214361	$4_{3}6_{1} + 1347$	2145	213162	$2_13_1 + 659$
3008	2164	$6_4 + 1331$	2258?		
3021?			2331?		

# The Journal of Chemical Physics

# ACCEPTED MANUSCRIPT

ypeset.	
ypese	
ypes	
ype	
$\leq$	
÷,	
$\overline{a}$	
ő	
$\neq$	
0	
U	
$\geq$	
0	
8	
$\cup$	
U	
Ð	
$\sim$	
$\overline{a}$	
Ð	
2	
5	
$\cup$	
$\circ$	
5	
$\Theta$	
$\geq$	
$\sim$	
$\rightarrow$	$ \rightarrow $
$\subseteq$	Õ
	10
0	Ñ
4	$\infty$
	<u> </u>
	$\bigcirc$
2	10
$\overline{\mathbb{O}}$	
1	$\sim$
	2
<u> </u>	$\circ$
Q	<u> </u>
$\Box$	$\bigcirc$
	<u></u>
$\sim$	
>	_
0	$\bigcirc$
5	$\overline{\frown}$
8	_
	S
<u> </u>	$\triangleleft$
L	
$\circ$	щ
	_
$\overline{\circ}$	$\odot$
	_
<u> </u>	
Φ	Lr.
$\geq$	$\triangleleft$
U	10
	~
	-
onli	F
e onli	E
he onli	E E
the onli	NTE T
r, the onli	CITE TI
er, the onli	E CITE TI
ver, the onli	SE CITE TI
lever, the onli	ASE CITE TI
wever, the onli	ASE CITE TI
owever, the onli	EASE CITE TI
However, the onli	LEASE CITE TI
. However, the onli	PLEASE CITE TI
ot. However, the onli	PLEASE CITE TI
ript. However, the onli	PLEASE CITE TI
cript. However, the onli	PLEASE CITE TI
script. However, the onli	PLEASE CITE TI
uscript. However, the onli	PLEASE CITE TI
nuscript. However, the onli	PLEASE CITE TI
anuscript. However, the onli	PLEASE CITE TI
manuscript. However, the onli	PLEASE CITE TI
I manuscript. However, the onli	PLEASE CITE TI
ed manuscript. However, the onli	PLEASE CITE TI
ted manuscript. However, the onli	PLEASE CITE TI
pted manuscript. However, the onli	PLEASE CITE TI
epted manuscript. However, the onli	PLEASE CITE TI
cepted manuscript. However, the onli	PLEASE CITE TI
iccepted manuscript. However, the onli	PLEASE CITE TI
accepted manuscript. However, the onli	PLEASE CITE TI
I, accepted manuscript. However, the onli	PLEASE CITE TI
ed, accepted manuscript. However, the onli	PLEASE CITE TI
ved, accepted manuscript. However, the onli	PLEASE CITE TI
wed, accepted manuscript. However, the onli	PLEASE CITE TI
iewed, accepted manuscript. However, the onli	PLEASE CITE TI
viewed, accepted manuscript. However, the onli	PLEASE CITE TI
eviewed, accepted manuscript. However, the onli	PLEASE CITE TI
reviewed, accepted manuscript. However, the onli	PLEASE CITE TI
er reviewed, accepted manuscript. However, the onli	PLEASE CITE TI
ser reviewed, accepted manuscript. However, the onli	PLEASE CITE TI
seer reviewed, accepted manuscript. However, the onli	PLEASE CITE TI
s peer reviewed, accepted manuscript. However, the onli	PLEASE CITE TI
's peer reviewed, accepted manuscript. However, the onli	PLEASE CITE TI
or's peer reviewed, accepted manuscript. However, the onli	PLEASE CITE TI
hor's peer reviewed, accepted manuscript. However, the onli	PLEASE CITE TI
thor's peer reviewed, accepted manuscript. However, the onli	PLEASE CITE TI
uthor's peer reviewed, accepted manuscript. However, the onli	PLEASE CITE TI
author's peer reviewed, accepted manuscript. However, the onli	PLEASE CITE TI
e author's peer reviewed, accepted manuscript. However, the onli	PLEASE CITE TI
he author's peer reviewed, accepted manuscript. However, the onli	PLEASE CITE TI
the author's peer reviewed, accepted manuscript. However, the onli	PLEASE CITE TI
is the author's peer reviewed, accepted manuscript. However, the onli	PLEASE CITE TI
s is the author's peer reviewed, accepted manuscript. However, the onli	PLEASE CITE TI
is is the author's peer reviewed, accepted manuscript. However, the onli	PLEASE CITE TI

3152?			23512		
3192.	2231	$2_2 + 522$	2409	2133	$2_{1}3_{2} + 457$
3393	$2_{1}3_{4}$	$3_4 + 1345$	2483	$2_23_1$	$2_2 + 476$
3484	2242	$2_2 + 811$	2620?		
3516	224161	$2_2 + 843$	2631	2242	$2_2 + 624$
3538	2262	$2_2 + 865$	2664	2262	$2_2 + 657$
3956?			2670?		
3973	23	$2_2 + 1300$	2697?		
			2723?		
			2763?		
			2947	2232	$2_23_1 + 464$
			2988	23	$2_2 + 981$
			3379?		
			3398?		
			3449	2331	$2_3 + 461$
			3470?		
			3612?		
			3662?		
			3962	24	$2_3 + 974$
			4430	2431	24 + 468
			4468?		
			4925	25	$2_4 + 963$

ACCEPTED MANUSCRIPT

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset. PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

Chemical Physics

The Journal of

AIP Publishing



**FIG. 7.** The medium resolution LIF spectra of  $GaCD_2$  (top) and  $GaCH_2$  (bottom). These spectra were recorded over three laser dyes and are uncorrected for laser power or detector efficiency, so the relative intensities are not very meaningful. An asterisk denotes a LIF feature not attributable to gallium methylene.

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset

PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

and their separation in each emission band were diagnostic of the *Q*-branch being excited by the laser ( ${}^{p}Q_{1} = 1$  peak,  ${}^{r}Q_{0} = 2$  features split by 4A'',  ${}^{r}Q_{1} = 2$  peaks separated by 8A'' etc.). Careful study of the LIF rotational contours shows that they are all perpendicular bands, indicative of upper state vibrational levels involving v<sub>1</sub> - v<sub>3</sub> and even quantum number combinations of v<sub>4</sub> and v<sub>6</sub> (such as  $4_{0}^{2}$ ,  $6_{0}^{2}$  and  $4_{0}^{1}6_{0}^{1}$ ) as discussed in section IV.C. Vibronically induced parallel bands, such as  $6_{0}^{1}$ , are not evident in our spectra.

Schooled by our analysis of the emission spectra, we then tackled the LIF spectra which are actually quite complicated. Fig. 7 shows that there is a strong band at +1278/+949 cm<sup>-1</sup> (GaCH<sub>2</sub>/GaCD<sub>2</sub> displacement above the 0-0 bands at 22 221.5/22 189.1 cm<sup>-1</sup>), intervals comparable to those calculated for  $v'_2$  (1368/1015 cm<sup>-1</sup>). The emission spectra give further credence to this assignment as Franck-Condon considerations suggest that "like should emit to like" and both of these bands show strong emission down to 2<sub>1</sub> and 2<sub>2</sub>. The first two bands in the GaCD<sub>2</sub> LIF spectrum occur at +309 and +462 cm<sup>-1</sup>, whereas in GaCH<sub>2</sub> they appear as a pair of overlapping features at +445 and +484 cm<sup>-1</sup>. The +484/+462 cm<sup>-1</sup> bands have very similar excited state energies, with only a small H/D isotope shift (22 cm<sup>-1</sup>) and both emit weakly to 0<sub>0</sub> and strongly to 3<sub>1</sub>, suggesting an assignment of 3<sup>1</sup><sub>0</sub>. *Ab initio* frequencies of 562/526 cm<sup>-1</sup> and an isotope shift of 35 cm<sup>-1</sup> are in general accord with experiment. Using the emission spectra as a guide, we were able to follow the anharmonic  $v'_3$  progression out to three members with 3<sup>2</sup><sub>0</sub> emitting strongly to 3<sub>2</sub> and 3<sup>3</sup><sub>0</sub> emitting primarily to 3<sub>3</sub>.

The first LIF bands at +445/+309 cm<sup>-1</sup> must involve  $v'_4$  or  $v'_6$ , although the calculated frequency of the latter (740/553 cm<sup>-1</sup>) is too high to merit serious consideration. The only possible

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset

PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

1

assignment on energetic grounds is then  $4_0^2$ , which owes its intensity to the large change in the v<sub>4</sub> frequency on electronic excitation (Table II). Unfortunately, the emission spectra of these bands are somewhat contradictory. In GaCD<sub>2</sub> 4<sup>2</sup> emits strongly to 0<sub>0</sub>, 2<sub>1</sub> and combinations of v<sub>2</sub>, v<sub>4</sub> and v<sub>6</sub>. In GaCH<sub>2</sub>, strong emission occurs to 0<sub>0</sub>, 3<sub>1</sub> and combinations of v<sub>4</sub> and v<sub>6</sub> with little activity in v<sub>2</sub>. We speculate that in GaCH<sub>2</sub> 3<sup>1</sup> and 4<sup>2</sup> (perturbed separation = 38 cm<sup>-1</sup>) are coupled by Fermi resonance which is why 3<sub>1</sub> shows up prominently in the 4<sup>2</sup> emission. In GaCD<sub>2</sub>, these bands are 143 cm<sup>-1</sup> apart, so Fermi resonance complications are minimized. In any event, the prominence of  $v''_4$  and  $v''_6$  combinations in both emission spectra indicates they have the same upper state assignment and that it must involve a non-totally symmetric mode, *viz*  $4_0^2$ . This assignment gives  $v'_4 = 223/155$  cm<sup>-1</sup>, rather smaller than the *ab initio* values of 337/265 cm<sup>-1</sup> although the calculated isotope shift (73 cm<sup>-1</sup>) is in very good agreement with the experimental value of 68 cm<sup>-1</sup>. We speculate that the neglect of spin-orbit interactions and the nonplanar  $\tilde{B}$  state *ab initio* structure both contribute to the mismatch in the observed and calculated out-of-plane bending frequencies.

In the GaCH<sub>2</sub> LIF spectrum, the pattern of bands with low-frequency intervals is repeated above 23 700 cm<sup>-1</sup>, with weak bands +433 (emitting down to  $2_1$  and  $2_13_1$ ) and +479 (emitting down to  $0_0$  and  $2_1$ ) cm<sup>-1</sup> above  $2_0^1$ , leading to assignments of  $2_0^1 4_0^2$  and  $2_0^1 3_0^1$ , respectively. The analogous LIF bands are found in GaCD<sub>2</sub> at +318 (emitting down to  $2_1$  and  $2_13_1$ ) and +451 (emitting down to  $2_2$  and  $2_23_1$ ) cm<sup>-1</sup> above  $2_0^1$ . A further band at +1585 cm<sup>-1</sup> above  $0_0^0$  has prominent emission down to  $3_2$  and  $3_3$  and the correct interval to be assigned as  $3_0^3 4_0^2$  [1287 ( $3^3$ ) + 299 ( $4^2$ )]. Finally, there is a weak GaCH<sub>2</sub> band at +1075 cm<sup>-1</sup>, which has prominent emission down to  $3_1$  and  $2_13_1$ , which we assign as  $3_0^1 4_0^2$ , although the interval is rather higher than 443 ( $4^2$ ) + 483 ( $3^1$ ) = 926 cm<sup>-1</sup>

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset. PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504 The only remaining unassigned bands in the LIF spectra (GaCH<sub>2</sub> = +1152 and +1427 cm<sup>-1</sup> and GaCD<sub>2</sub> = +812 and 1072 cm<sup>-1</sup>) are those that show strong emission down to levels involving v<sub>4</sub> and v<sub>6</sub>, signaling that the upper states must also involve these modes. The higher energy bands can be assigned with confidence as  $6_0^2$ , giving GaCH<sub>2</sub>:  $v'_6 = 714$  cm<sup>-1</sup> (*ab initio* = 740 cm<sup>-1</sup>) and GaCD<sub>2</sub>:  $v'_6 = 536$  cm<sup>-1</sup> (*ab initio* = 553 cm<sup>-1</sup>) and a theoretical H/D isotope shift of 188 cm<sup>-1</sup>, very close to the experimental value of 178 cm<sup>-1</sup>. There is a further GaCD<sub>2</sub> band at +1386 cm<sup>-1</sup>, with a similar emission spectrum, that can be identified as  $4_0^2 6_0^2$  [1072 ( $6^2$ ) + 314 ( $4^2$ ) cm<sup>-1</sup>].

At this point, the only fly in the ointment is the identity of the medium intensity bands at +1152 (GaCH<sub>2</sub>) and +812 (GaCD<sub>2</sub>) cm<sup>-1</sup>. Both show strong emission down to the 4<sub>2</sub>, 4<sub>1</sub>6<sub>1</sub> and 6<sub>2</sub> cluster of levels and it is likely that they have the same upper state. The only excited state levels not involving  $v'_4$  or  $v'_6$  in this locale, assuming negligible anharmonicity, are 4<sup>4</sup> (890/618 cm<sup>-1</sup>), 4<sup>1</sup>6<sup>1</sup> (936/690 cm<sup>-1</sup>) and 4<sup>2</sup>6<sup>1</sup> (1158/845 cm<sup>-1</sup>). Although the latter is energetically realistic, our previous discussion of the emission spectra showed that vibronically induced transitions involving 6<sup>1</sup> should follow *a*-type selection rules whereas both LIF features are unequivocally perpendicular bands. A  $4_0^1 6_0^1$  assignment, involving *c*-type bands, would solve the difficulty but the energy mismatch is unreasonably large. Either the energy level structure in the upper state is substantially perturbed in some fashion or  $4_0^1 6_0^2$  gains intensity through some more involved mechanism. We were unable to come to a resolution on this point so have simply included both possibilities in our list of assignments. The assignments, band origins, major emission features, separations from the 0-0 band and summary comments on each LIF band are presented in Table IV.

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset. PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

	GaCH2			GaCD <sub>2</sub>			
Assignment	$T_0{}^a$	Emits strongly to	Comments	$T_0^{\ b}$	Emits strongly to	Comments	
000	22 221.5	00,21		22 189.1	00, 21	H/D shift = 32.4	
$4_0^2$	22 666.6	00, 31, 44	+445.1 [v <sub>4</sub> =222.6]	22 498.1	$0_0, 2_1, 2_14_1$	+309.0 [v <sub>4</sub> '= 154.5]	
$3_0^1$	22 705.0	$3_1, 2_1, 2_{13_1}$	+483.5 [v <sub>3</sub> =483.5]	22 650.8	$3_1, 2_1, 2_{1}3_1$	+461.7 [v'_3=461.7]	
3 <sub>0</sub> <sup>2</sup>	23 139.6	32	+918.1 [3 <sup>1</sup> <sub>0</sub> +434.6]	23 074.1	32	+885.0 [3 <sup>1</sup> <sub>0</sub> +423.3]	
$3_0^1 4_0^2?$	23 296.6	31, 2131	+1075.1 [3 <sup>1</sup> <sub>0</sub> +591.6]				
$4_0^1 6_0^1 / 4_0^1 6_0^2$	23 373.2	4 <sub>2</sub> , 4 <sub>6</sub>	+1151.7 [61+437]	23 001.0	6 <sub>2</sub> , 4 <sub>1</sub> 6 <sub>1</sub>	+811.9 [6 <sup>1</sup> +276.1]	
$2_0^1$	23 499.2	$2_1, 2_2$	+1277.7 [ $v_2'$ =1277.7]	23 138.0	$2_1, 2_2$	+948.9 $[v_2'=948.9]$	
3 <sup>3</sup> <sub>0</sub>	23 531.8	33	$+1310.3 [3_0^2 + 392.2]$	23 475.9	33	+1286.8 [3 <sup>2</sup> <sub>0</sub> +401.8]	
$6_0^2$	23 648.0	4161, 44	+1426.5 [ $v_6'$ =713.3]	23 260.7	42, 4161	+1071.6 [v <sub>6</sub> =535.8]	
$4_0^2 6_0^2$ ?				23 574.9	$4_2, 4_16_1, 6_2$	+1385.8 [6 <sup>2</sup> <sub>0</sub> +314.2]	
$2_0^1 4_0^2$	23 932.5	2131	+1711 [2 <sup>1</sup> <sub>0</sub> +433.3]	23 455.5	$2_1, 2_13_1, 2_2$	+1266.4 [2 <sup>1</sup> <sub>0</sub> +317.5]	
$2^1_0 3^1_0$	23 978.1	$0_0, 2_1$	+1756.6 [2 <sup>1</sup> <sub>0</sub> +478.9]	23 589.4	$2_2, 2_23_1$	$+1400.3 [2_0^1 + 451.4]$	
3342				23 774.5	3 <sub>2</sub> , 3 <sub>3</sub>	+1585.4 [3 <sup>3</sup> <sub>0</sub> +298.6]	

TABLE IV: Assignments and approximate band origins (cm<sup>-1</sup>) of the LIF bands of GaCH<sub>2</sub> and GaCD<sub>2</sub>.

<sup>a</sup>For GaCH<sub>2</sub>,  $T_0$  calculated as  ${}^pQ_1$  maximum +  $(A - \overline{B})'' = 10.00$  cm<sup>-1</sup> or  ${}^rQ_0$  maximum –  $(A - \overline{B})'$ 

 $= 9.33 \text{ cm}^{-1}.$ 

<sup>c</sup>For GaCD<sub>2</sub>,  $T_0$  calculated as  ${}^pQ_1$  maximum +  $(A - \overline{B})'' = 4.87$  cm<sup>-1</sup>.

AIP Publishing

The Journal of Chemical Physics

## **V. DISCUSSION**

## A. Ground state molecular structure and vibrational frequencies

We can compare our *ab initio* values of the GaCH<sub>2</sub> ground state structure to similar values from the literature. For trimethylgallium, the experimental bond lengths from a gas-phase electron diffraction study<sup>31</sup> are Ga-C =  $1.967 \pm 0.002$  Å and C-H =  $1.0821 \pm 0.003$  Å, only slightly shorter than out best IC-MRCI [aug-cc-pwCVTZ, d-electrons correlated] calculated values (2.013 and 1.093 Å). The ground-state bond length of GaC (<sup>4</sup> $\Sigma$ ) is presently unknown but the best calculated<sup>32</sup> equilibrium value [CASSCF/MRCI, aug-cc-pV5Z basis set] is 2.025 Å, consistent with our predictions (2.010 - 2.040 Å, Table II) for GaCH<sub>2</sub>.

Our experimentally determined ground state Ga-C stretching frequencies  $(v''_3)$  are 517/478 cm<sup>-1</sup> (GaCH<sub>2</sub>/GaCD<sub>2</sub>), in general accord with the Ga-C frequencies of Ga(CH<sub>3</sub>)<sub>3</sub>/Ga(CD<sub>3</sub>)<sub>3</sub> obtained from IR and Raman studies<sup>33</sup> as 527/478 cm<sup>-1</sup>. For GaC, the calculated ground state vibrational fundamental<sup>31</sup> is 549 cm<sup>-1</sup>, only slightly larger than the GaCH<sub>2</sub> value. The experimentally determined CH<sub>2</sub> symmetric bending frequencies of aluminum methylene<sup>3</sup> (1353/1016) and gallium methylene (1347/1011) are virtually identical, suggesting that the methylene group is only mildly perturbed by the metal atom.

## B. Comparison of the excited state properties of AlCH2 and GaCH2

Table V shows a comparison of the changes in geometric structure and vibrational frequencies for the various AlCH<sub>2</sub> and GaCH<sub>2</sub> excited states derived from *ab initio* theory and the available experimental data. The low-lying  $\tilde{A}^2A_1$  states are a result of promoting an electron from the metal-based lone pair *s* orbital to an out-of-plane 2*p* orbital on carbon, yielding a •M=CH<sub>2</sub> structure. In both molecules, this results in a substantial decrease in the metal-carbon bond length

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset.

PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset. PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504 and a concomitant increase in the HCH angle. The  $\tilde{A} - \tilde{X}$  transitions of both radicals occur in the near infrared in a region which will be difficult to interrogate with conventional LIF or emission techniques. The changes in vibrational frequencies are as expected with a decrease in v<sub>2</sub> and an increase in v<sub>3</sub> on electronic excitation.

The  $\tilde{B}$  state is formed by promotion of an electron from the lone pair *s* orbital on the metal to the empty lowest unoccupied molecular orbital, with only minor changes in the molecular structure, including an approximately 5° increase in the HCH angle. The GaCH<sub>2</sub> excited state is almost 3 000 cm<sup>-1</sup> higher in energy than AlCH<sub>2</sub> and this may account for the slightly different behavior of the vibrational frequencies on electronic excitation in the two radicals.

An interesting aspect of the present work concerns the *ab initio* finding that the  $\tilde{B}$  state has a nonplanar structure. The experimental data suggest that, if there is some nonplanarity, it is slight, as evidenced by the fact that the  $4_0^1$  bands are so weak in emission. However, a flat or distinctly anharmonic excited state out-of-plane bending potential would probably explain the difficulties we had in assigning the LIF spectrum based on our expectation of only slightly anharmonic progressions. This aspect of the GaCH<sub>2</sub> excited state awaits the day when rotationally resolved spectra can be obtained and the excited state structure can be derived from such data.

## **VI. CONCLUSIONS**

The GaCH<sub>2</sub> and GaCD<sub>2</sub> free radicals have been detected spectroscopically for the first time as products of an electric discharge through trimethylgallium vapor in high pressure argon. The electronic transition at 450 nm detected by LIF spectroscopy has been assigned as  $\tilde{B}^2A_2 - \tilde{X}^2B_1$ based on the high-level *ab initio* calculations reported in this work, the observed H/D isotope and nuclear statistical weight effects in the spectra, and the correspondence with the electronic spectra



This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset. PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504 of AlCH<sub>2</sub>. Four of the six vibrational frequencies ( $v_2$ ,  $v_3$ ,  $v_4$  and  $v_6$ ) have been established in the  $\tilde{X}$  and  $\tilde{B}$  electronic states of GaCH<sub>2</sub> and GaCD<sub>2</sub>. The observed LIF and emission spectra provide sensitive methods of detecting and monitoring the gallium methylene free radicals in the gas phase.

	$\widetilde{X}^2 B_1$ Ground State								
	r(M-C) Å	r(C-H) Å	θ(HCH) °	<i>T</i> <sub>θ</sub> (cm <sup>-1</sup> )	v <sub>2</sub> (cm <sup>-1</sup> ) CH <sub>2</sub> bend	v <sub>3</sub> (cm <sup>-1</sup> ) M-C stretch	Source		
AlCH <sub>2</sub>	1.959	1.106	110.4	0	1353	612	Expt. Ref. 3		
GaCH <sub>2</sub>	2.013 <sup>a</sup>	1.093 <sup>a</sup>	110.9 <sup>a</sup>	0	1347 <sup>b</sup>	517 <sup>b</sup>	This work		
	$\widetilde{A}^2 A_I$ Excited State								
	Δr(M-C) Å	Δr(C-H) Å	Δθ(HCH) °	$T_{\theta}$	$\Delta v_2$	Δν3	Source		
AlCH <sub>2</sub>	-0.128	-0.019	+8.0	6533	-109	+148	Theory. Ref. 2		
GaCH <sub>2</sub>	-0.177 <sup>a</sup>	-0.010 <sup>a</sup>	+11.5 <sup>a</sup>	6847 <sup>a</sup>	-103	+176	This work.		
			$\widetilde{B}^2 A_2 \operatorname{Ex}$	cited Sta	te				
	<b>Δr(M-C)</b> Å	Δr(C-H) Å	Δθ(HCH) °	Tθ	$\Delta v_2$	Δν3	Source		
AlCH <sub>2</sub>	-0.016	-0.015	+5.0	19 572	+24	+51	Expt. Ref. 3		
GaCH <sub>2</sub>	-0.042 <sup>a</sup>	-0.006 <sup>a</sup>	+4.3 <sup>a</sup>	22222 <sup>b</sup>	-70 <sup>b</sup>	-34 <sup>b</sup>	This work.		

TABLE V. Comparison of the ground state molecular structures, band origins and vibrational frequencies and their changes on electronic excitation for  $MCH_2$  (X = Al, Ga) molecules.

<sup>a</sup> Theory. Table II, slightly nonplanar geometry in the  $\tilde{B}$  state.

<sup>b</sup>Experiment.

The Journal of Chemical Physics This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset. PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

AIP Publishing

## SUPPLEMENTARY MATERIAL

See the supplementary material for a plot of the  $GaCH_2$  molecular orbitals and pdf copies of the extensive emission spectra recorded during the course of this work.

## ACKNOWLEDGEMENTS

The authors thank Crystal Stover for proofreading the manuscript. R.T. acknowledges financial support from the University of Bologna. This research was funded by Ideal Vacuum Products.

## AUTHOR DECLARATIONS

## **Conflict of Interest**

The authors have no conflicts to disclose.

## **Author Contributions**

Tony. C. Smith: Conceptualization, data acquisition and analysis, funding acquisition. Riccardo Tarroni: *Ab initio* calculations, data analysis, manuscript writing and editing. Dennis Clouthier: Conceptualization, data acquisition and analysis, manuscript writing and editing.

## DATA AVAILABILITY

The data that support the findings of this study are available within the article and in the supplementary material.

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset

PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset

Chemical Physics

The Journal of

AIP Publishing PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

## References

1. T. J. Mountziaris and K. F. Jensen, J. Electrochem. Soc. 138, 2426 (1991).

2. R. Tarroni and D. J. Clouthier, J. Chem. Phys. 153, 014301 (2020).

3. F. X. Sunahori, T. C. Smith and D. J. Clouthier, J. Chem. Phys. 157, 044301 (2022)

4. H. Harjanto, W.W. Harper, and D. J. Clouthier, J. Chem. Phys. 105, 10189 (1996).

5. W. W. Harper and D. J. Clouthier, J. Chem. Phys. 106, 9461 (1997).

6. D. L. Michalopoulos, M. E. Geusic, P. R. R. Langridge-Smith, and R. E. Smalley, J. Chem. Phys. **80**, 3556 (1984).

7. T. C. Smith, M. Gharaibeh and D. J. Clouthier, J. Chem. Phys. 157, 204306 (2022).

8. L. I. Zakharkin, V. V. Gavrilenko and N. P. Fatyushina, Russ. Chem. Bull. 46, 379 (1997).

9. G. A. Atiya, A. S. Grady, D. K. Russell and T. A. Claxton, Spectrochim. Acta., **47A**, 467 (1991).

10. Molpro, version 2010.1, a package of ab initio programs, H. -J. Werner, P. J. Knowles, and others, see https://www.molpro.net.

11. CFOUR, version 1.0, a quantum-chemical program package, J. F. Stanton, J. Gauss, L.

Cheng, M.E. Harding, D.A. Matthews, P.G. Szalay, and others, See http://www.cfour.de.

12. H.-J. Werner and P. J. Knowles, J. Chem. Phys. 82, 5053 (1985)

13. P. J. Knowles and H.-J. Werner, Chem. Phys. Lett. 115, 259 (1985).

14. H.-J. Werner and P. J. Knowles, J. Chem. Phys. 89, 5803 (1988).

15. P. J. Knowles and H.-J. Werner, Chem. Phys. Lett. 145, 514 (1988).

16. E. R. Davidson and D. W. Silver, Chem. Phys. Lett. 52, 403 (1977).

17. H.-J. Werner, M. Kállay, and J. Gauss, J. Chem. Phys. 128, 034305 (2008).

18. J. F. Stanton and J. Gauss, Adv. Chem. Phys. 125, 101 (2003).

> AIP Publishing

This is the author's peer reviewed, accepted manuscript. However, the online version of record will be different from this version once it has been copyedited and typeset

PLEASE CITE THIS ARTICLE AS DOI: 10.1063/5.0182504

19. K. Raghavachari, G.W. Trucks, J.A. Pople and M. Head-Gordon, Chem. Phys. Lett. **157**, 479 (1989).

20. J. F. Stanton and R.J. Bartlett, J. Chem. Phys. 98, 7029 (1993).

21. J. F. Stanton and J. Gauss, Theor. Chim. Acta **91**, 267 (1995).

22. A. K. Wilson, D. E. Woon, K. A. Peterson, T. H. Dunning, J. Chem. Phys. 110, 7667 (1999).

23. T. H. Dunning, J. Chem. Phys. 90, 1007 (1989).

24. K. A. Peterson and T.H. Dunning, Jr., J. Chem. Phys. 117, 10548 (2002).

25 N. J. DeYonker, K. A. Peterson and A. K. Wilson, J. Phys. Chem. A 111, 11383 (2007).

26. F. L. Pilar, *Elementary Quantum Chemistry*, 2nd edition, Dover, 2003.

27. D. G. Fedorov, S. Koseki, M. W. Schmidt and M. S. Gordon, Int. Rev. Phys. Chem. 22, 551 (2003).

28. PGOPHER, A Program for Simulating Rotational, Vibrational and Electronic Spectra, C. M. Western, *Journal of Quantitative Spectroscopy and Radiative Transfer*, **186** 221-242 (2016).

29. D. J. Clouthier and D. A. Ramsay, Annu. Rev. Phys. Chem. 34, 31 (1983).

30. J. Coon, R. E. DeWames, and C. M. Lloyd, J. Mol. Spectrosc. 8, 285 (1962).

31. B. Beagley, D. G. Schmidling and I. A. Steer, J. Mol. Struct. 21, 437 (1974).

32. G. F. S. Fernandes, M. A.P. Pontes, M. H. de Oliveira, L. F.A. Ferrão and F. B.C. Machado, Chem, Phys. Lett. **687**, 171 (2017).

33. D. C. McKean, G. P. McQuillan, J. L. Duncan, N. Shephard, B. Munro, V. Fawcett and H.G. M. Edwards, Spectrochim. Acta, 43A, 1405 (1987).