

Supporting Information

Ferrocene - Diiron(I) Bis-Cyclopentadienyl Conjugates Exhibit Potent in Vitro Cytotoxicity and Selectivity Towards Cancer Cells Associated with ROS Scavenging

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IR and NMR spectra of diiron compounds

Figure S1. ^1H NMR spectrum (401 MHz, acetone- d_6) of **2** (*cis-E/cis-Z/trans-E/trans-Z* ratios = 4.3:2.3:1.2:1.0).

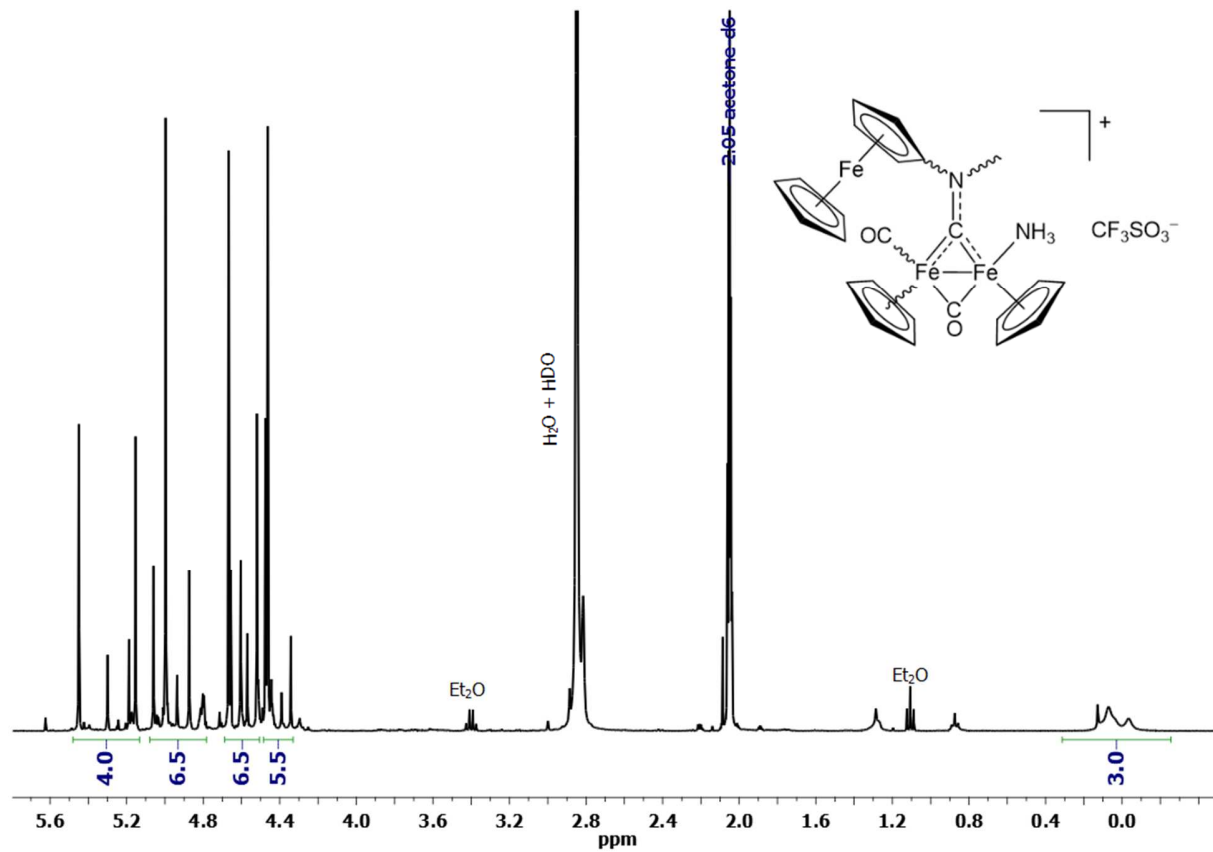


Figure S2. ^1H NMR spectrum (401 MHz, acetone- d_6) of **2**, highlighting the signals of the *cis-E* isomer.

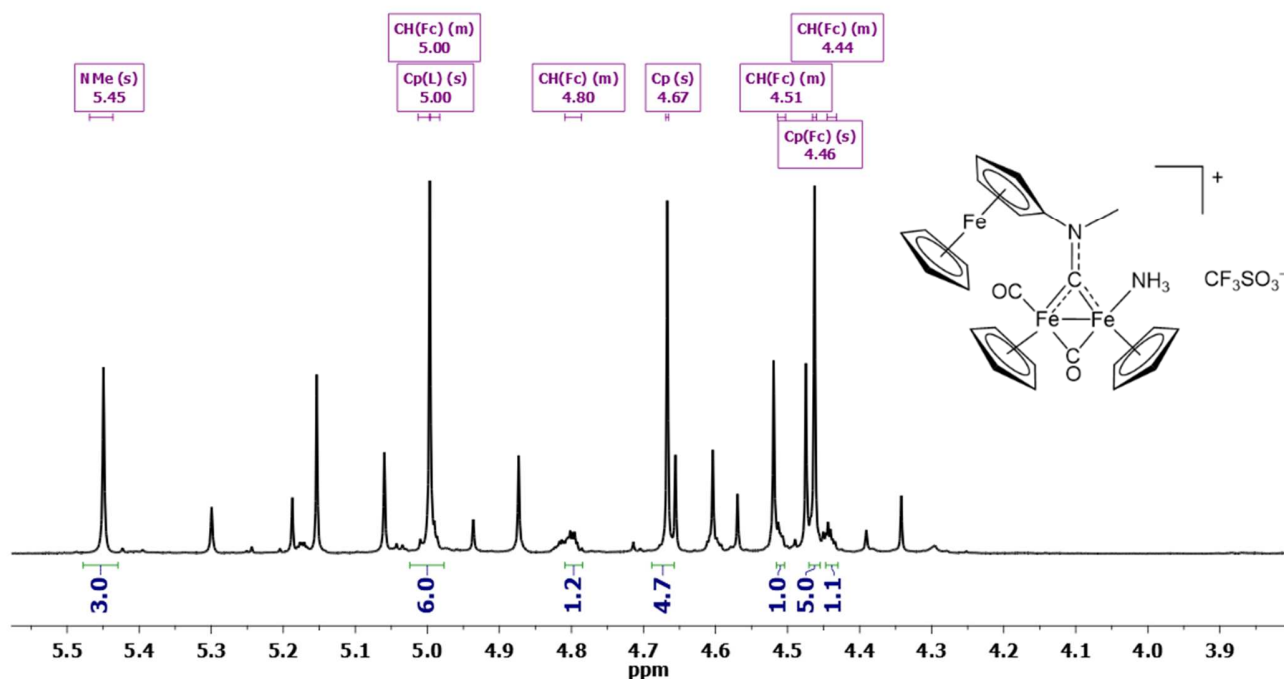


Figure S3. ^1H NMR spectrum (401 MHz, acetone- d_6) of **2**, highlighting the signals of the *cis-Z* isomer.

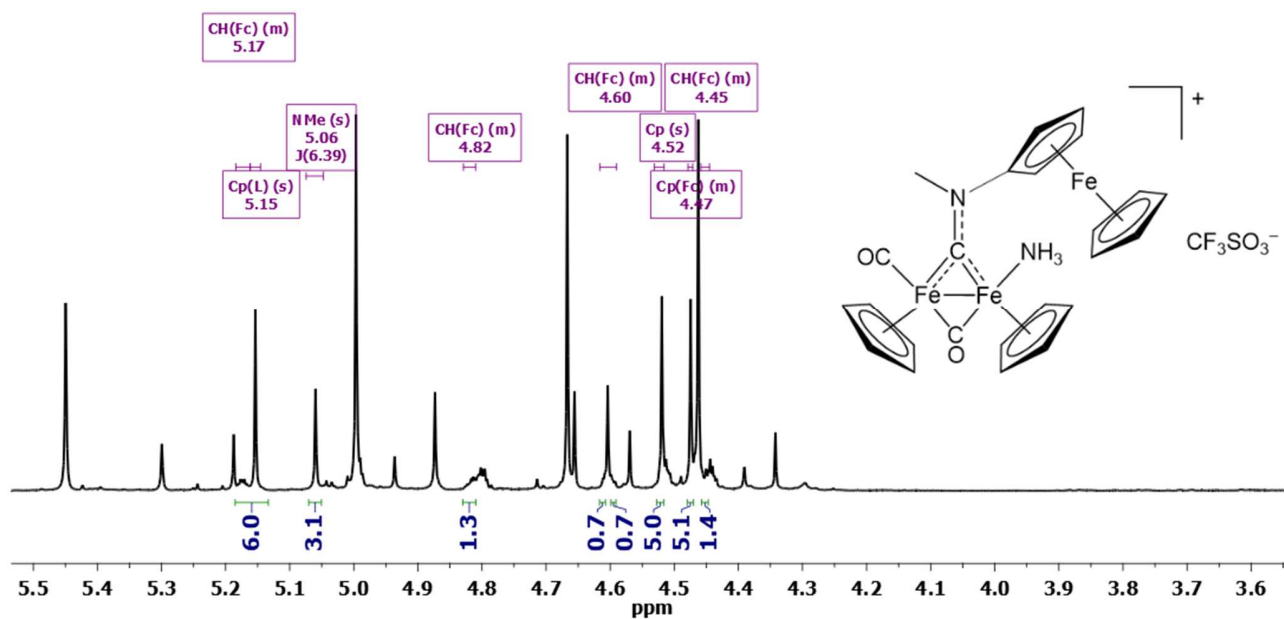


Figure S4. ^1H NMR spectrum (401 MHz, acetone- d_6) of **2**, highlighting the signals of the *trans-E* isomer.

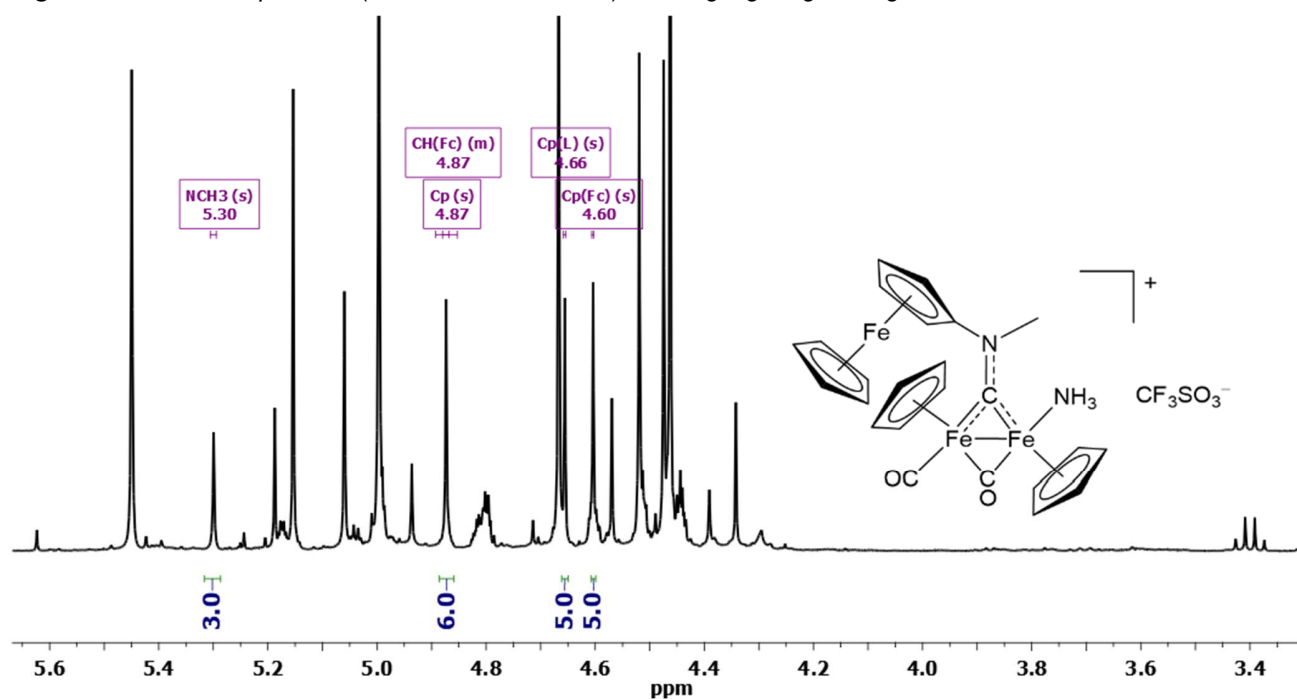


Figure S5. ^1H NMR spectrum (401 MHz, acetone- d_6) of **2**, highlighting the signals of the *trans-Z* isomer.

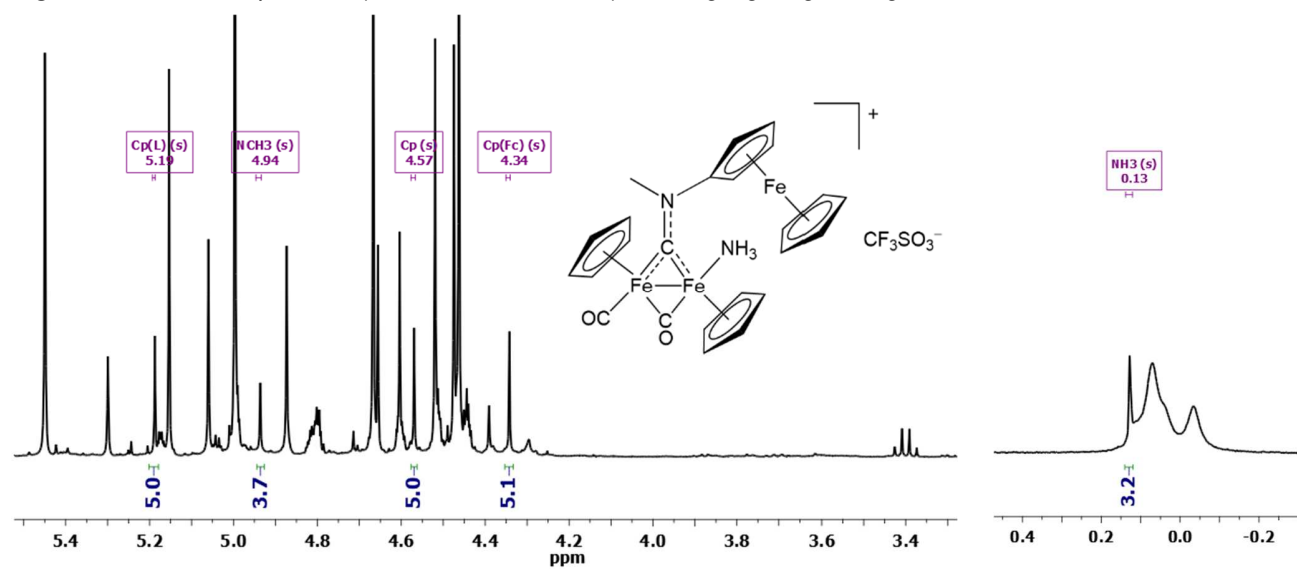


Figure S6. Black line: ^1H NMR spectrum (401 MHz, acetone- d_6) of **2**. Blue line: ^1H NOESY with irradiation at 4.67 ppm (Cp) of the *cis-E* isomer. Red line: ^1H NOESY with irradiation at 5.45 ppm (NCH₃) of the *cis-E* isomer. NOEs are indicated by the arrows.

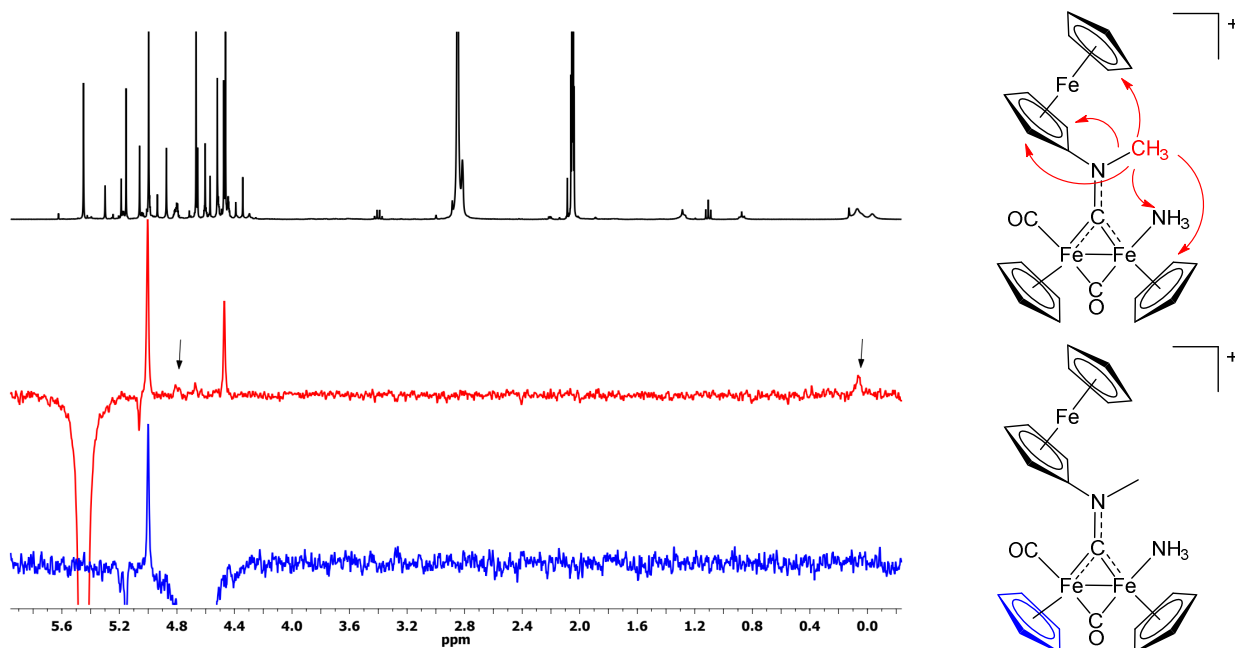


Figure S7. Black line: ^1H NMR spectrum (401 MHz, acetone- d_6) of **2**. Purple line: ^1H NOESY with irradiation at 5.15 ppm (Cp) of the *cis-Z* isomer. Green line: ^1H NOESY with irradiation at 5.06 ppm (NCH₃) of the *cis-Z* isomer. NOEs are indicated by the arrows.

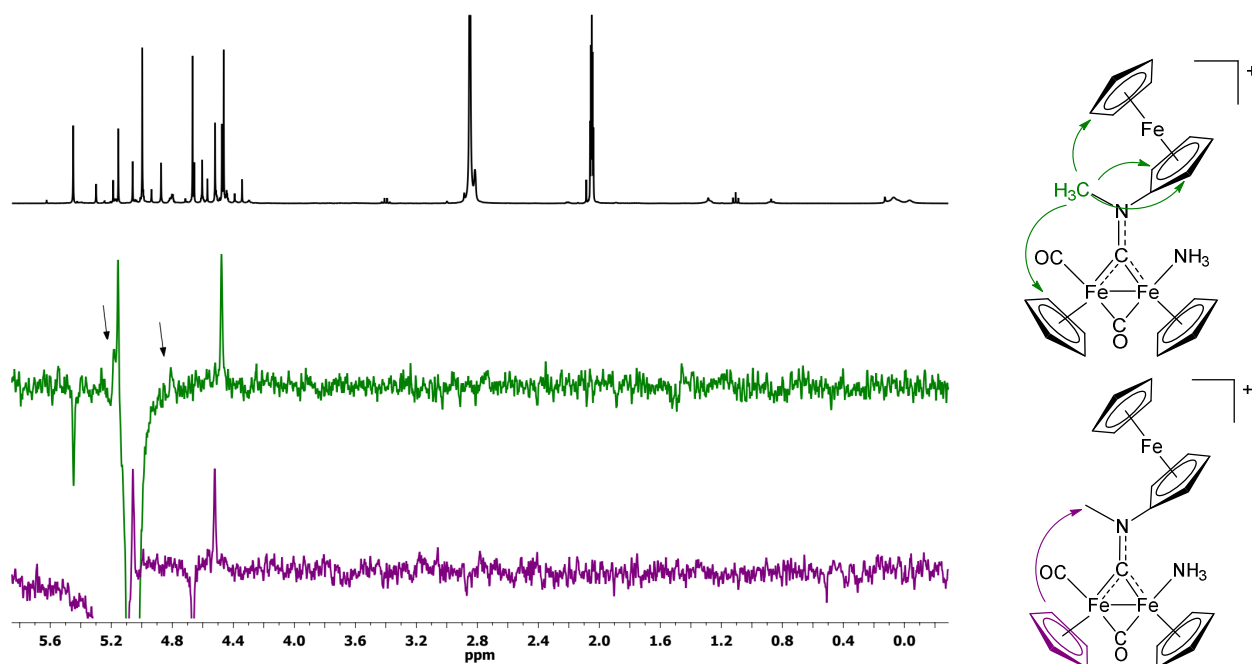


Figure S8. Black line: ^1H NMR spectrum (401 MHz, acetone- d_6) of **2**. Blue line: ^1H NOESY with irradiation at 4.67 ppm (Cp of the *cis-E* isomer). Purple line: ^1H NOESY with irradiation at 5.17 ppm (Cp of the *cis-Z* isomer). Red line: ^1H NOESY with irradiation at 5.45 ppm (NCH_3 of the *cis-E* isomer). Green line: ^1H NOESY with irradiation at 5.06 ppm (NCH_3 of the *cis-Z* isomer). Negative peaks (black arrows) indicate the chemical exchange underlying the partial magnetization transfer with the corresponding Cp ligand of the other diastereomer.

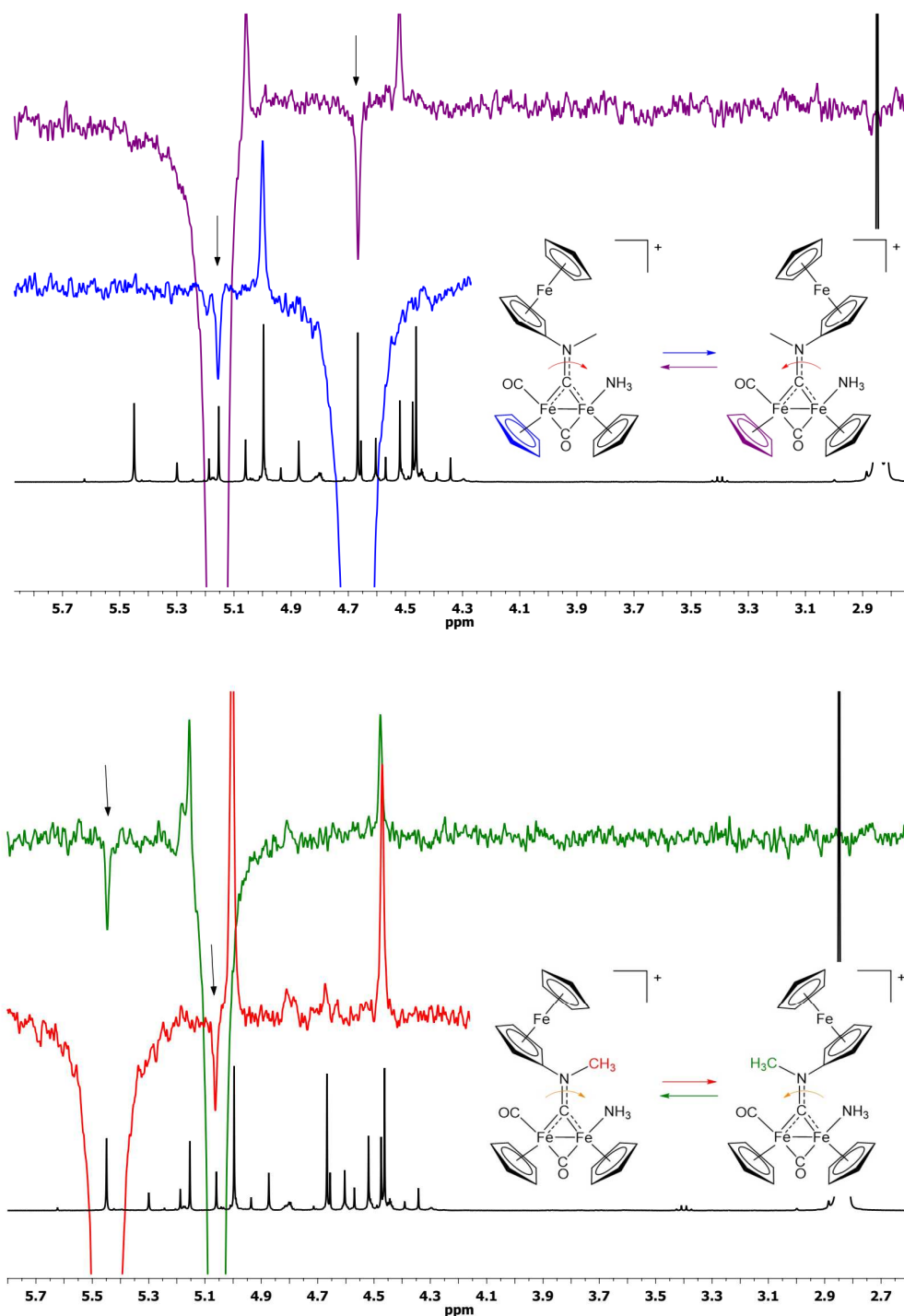


Figure S9. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum (101 MHz, acetone- d_6) of **2** (*cis-E/cis-Z/trans-E/trans-Z* ratios = 4.3:2.3:1.2:1.0). Only the resonances due to the major *cis-E/Z* isomers are highlighted.

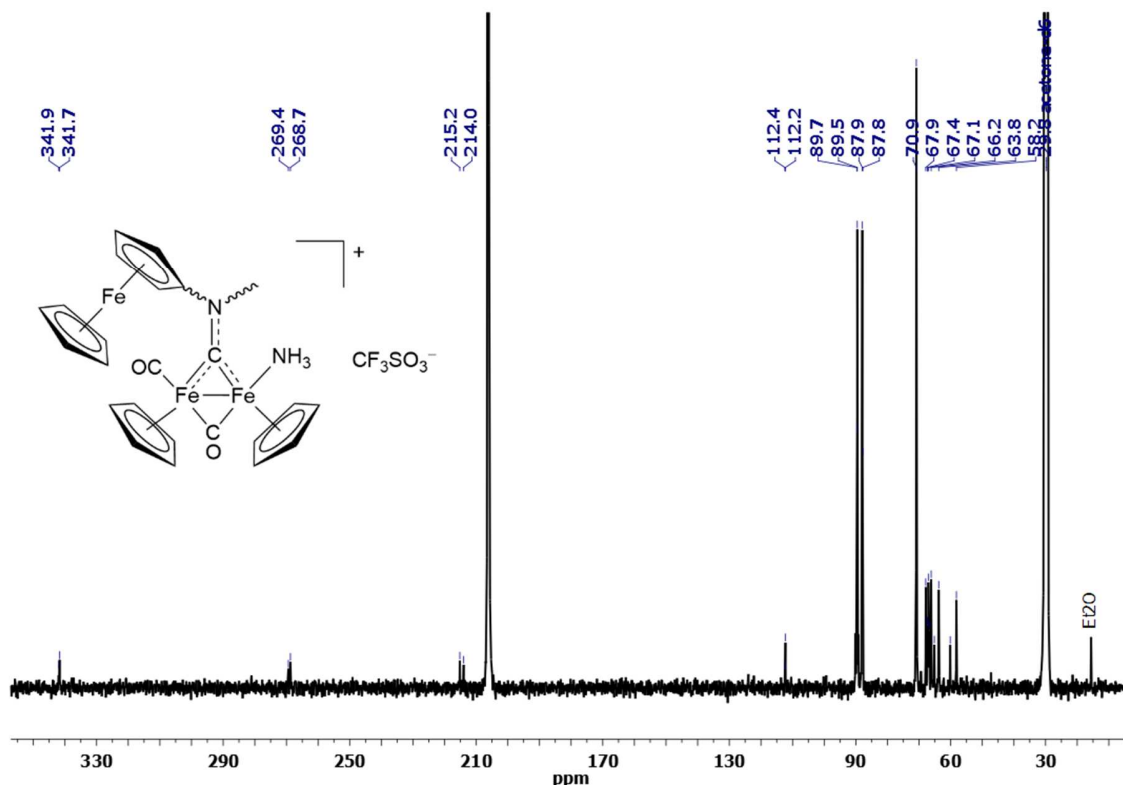


Figure S10. ^1H NMR spectrum (401 MHz, acetone- d_6) of $[\text{Fe}_2\text{Cp}_2(\text{CO})_2(\text{CNMe})\{\mu\text{-CNMe}(\text{Fc})\}]\text{CF}_3\text{SO}_3$, **3a** (*cis-E/Z* ratio ≈ 1 + minor *trans-E/Z* isomers). Only the resonances due to the *cis-E/Z* isomers are highlighted.

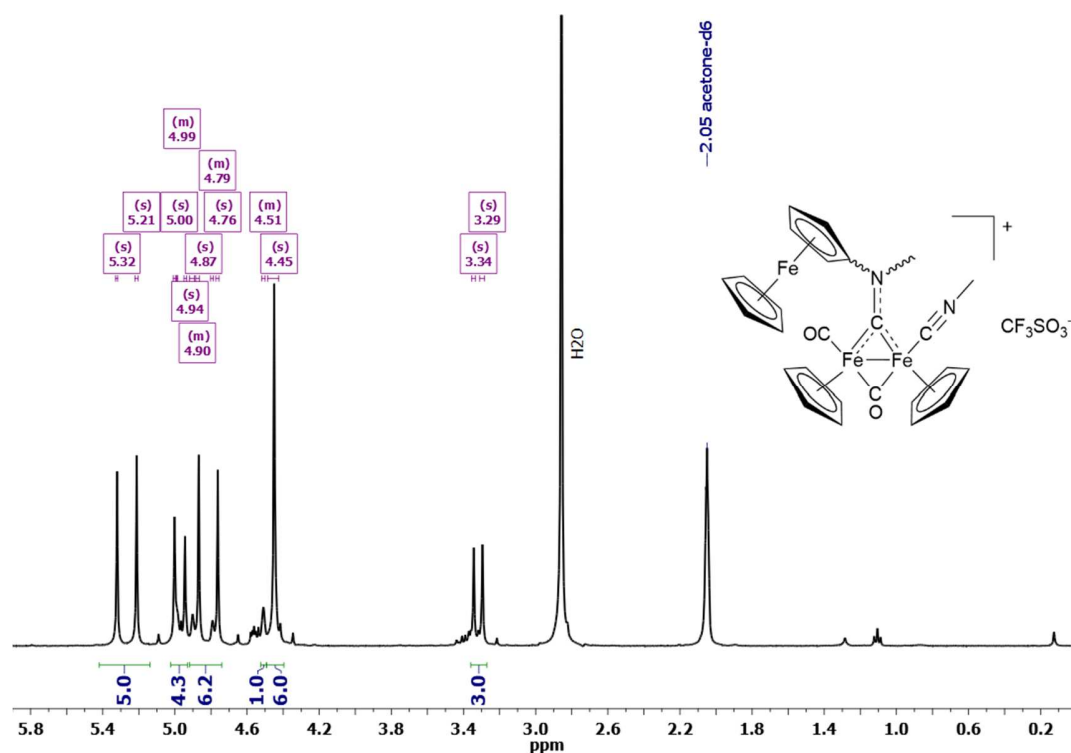


Figure S11. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum (101 MHz, acetone- d_6) of **3a** (*cis-E/Z* ratio ≈ 1 + minor *trans-E/Z* isomers). Only the resonances due to the *cis-E/Z* isomers are highlighted.

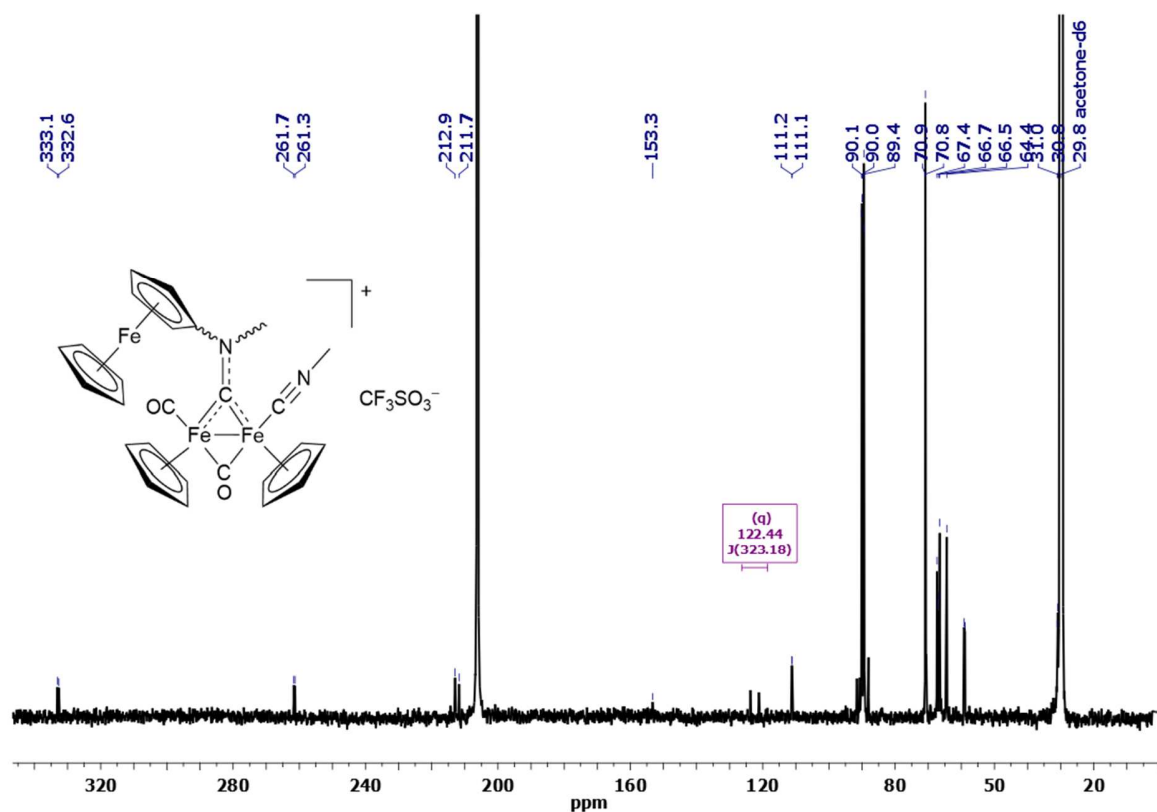


Figure S12. Solid-state IR spectrum (650-4000 cm^{-1}) of $[\text{Fe}_2\text{Cp}_2(\text{CO})_2(\text{CNCy})\{\mu\text{-CNMe}(\text{Fc})\}]\text{CF}_3\text{SO}_3$, **3b** (mixture of *cis-E/Z* isomers).

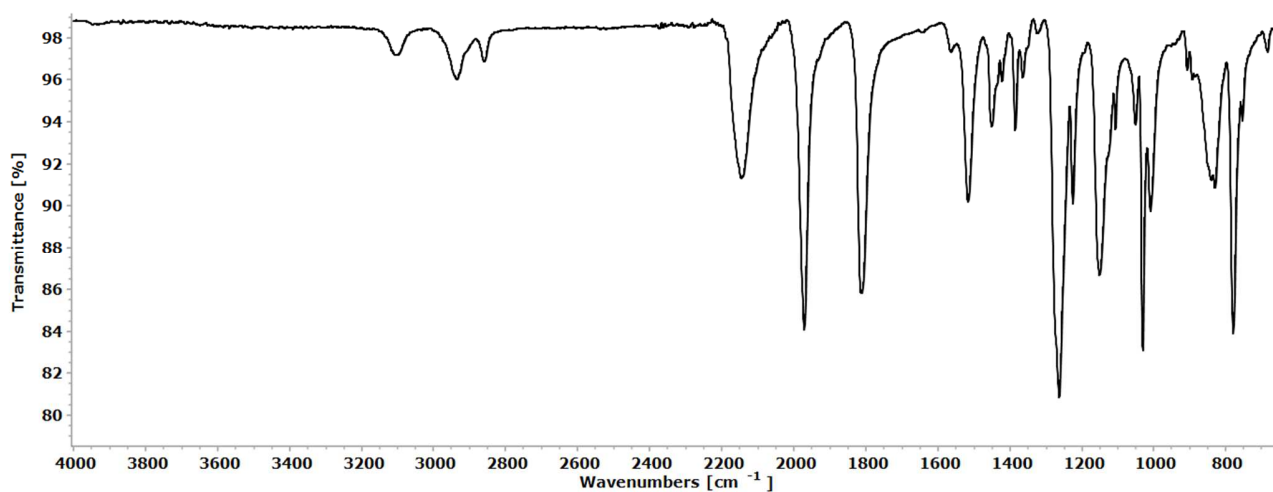


Figure S13. ^1H NMR spectrum (401 MHz, acetone- d_6) of **3b** (*cis-E/Z* ratio = 1.5).

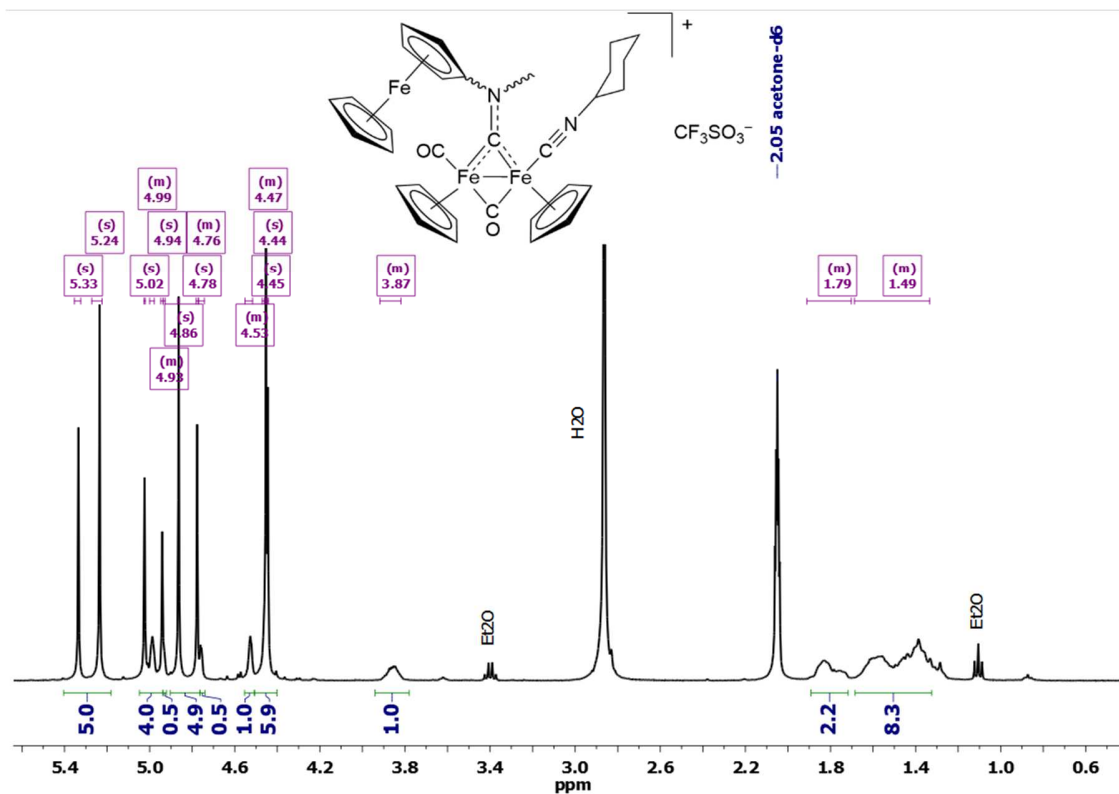


Figure S14. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum (101 MHz, acetone- d_6) of **3b** (*cis-E/Z* ratio = 1.5).

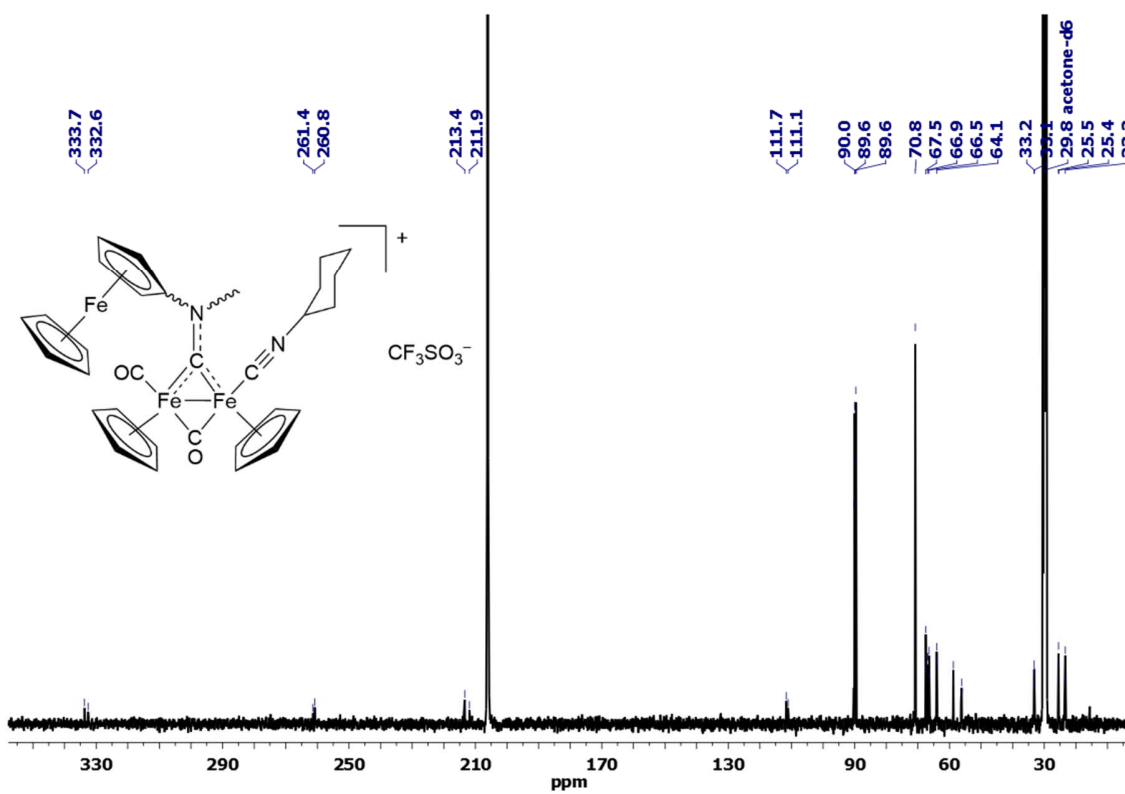


Figure S15. Black line: ^1H NMR spectrum (401 MHz, acetone- d_6) of **3b**. Blue line: ^1H NOESY with irradiation at 4.86 ppm (Cp of the *cis-E* isomer). Red line: ^1H NOESY with irradiation at 5.24 ppm (Cp⁺ of the *cis-E* isomer). Green line: ^1H NOESY with irradiation at 5.33 ppm (Cp of the *cis-Z* isomer). NOEs are indicated by the arrows.

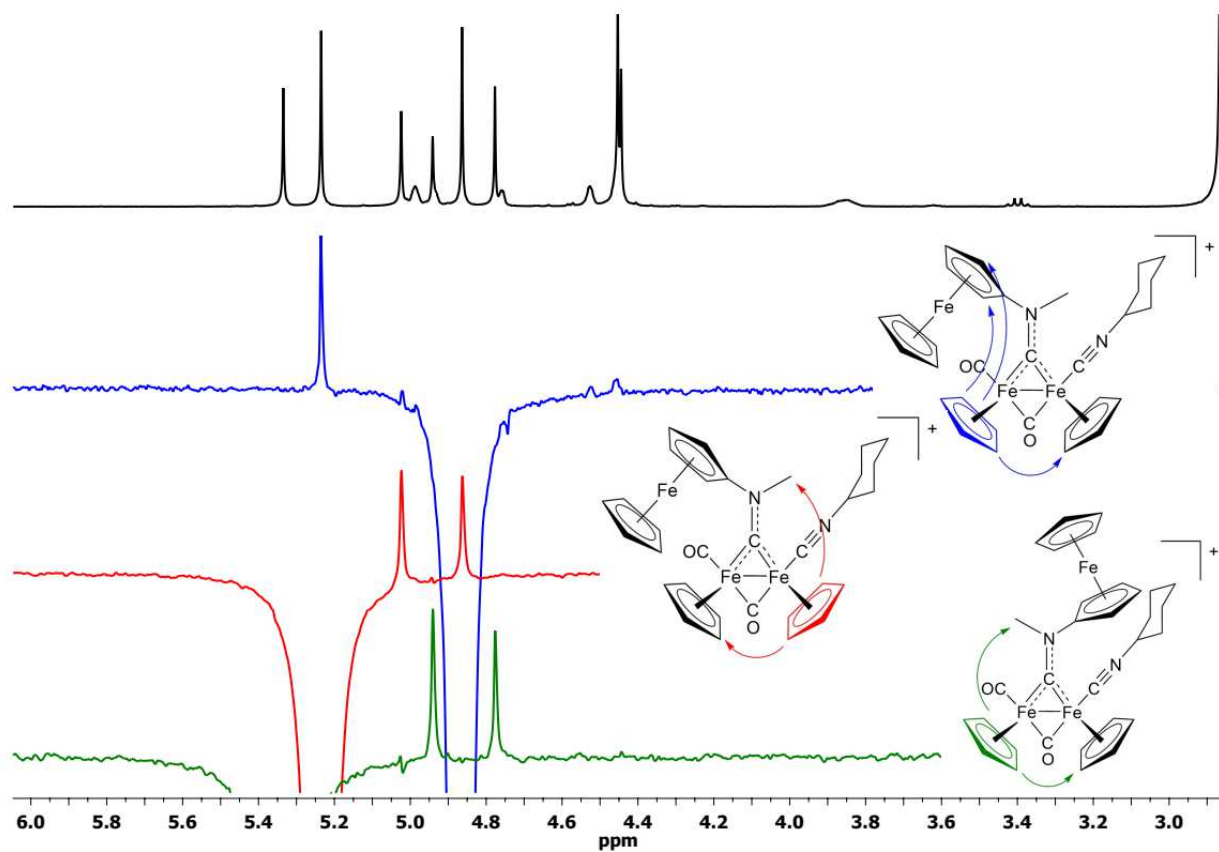


Figure S16. ^1H NMR spectrum (401 MHz, acetone- d_6) of $[\text{Fe}_2\text{Cp}_2(\text{CNMe})_2(\mu\text{-CO})\{\mu\text{-CNMe}(\text{Fc})\}]\text{CF}_3\text{SO}_3$, **4a** (*trans/cis* ratio = 1.2). Signals due to the $[\text{FeCp}(\text{CNMe})_3]^+$ by-product are marked with asterisk (*).

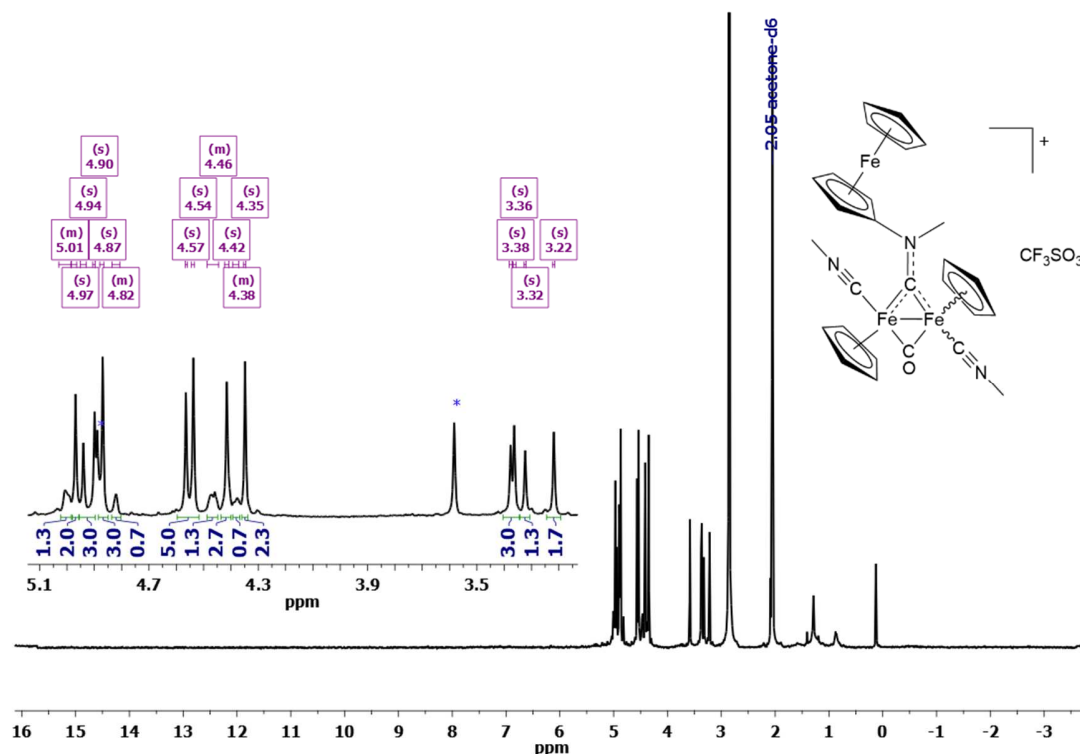


Figure S17. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum (101 MHz, acetone- d_6) of **4a** (*trans/cis* ratio = 1.2). Signals due to BHT (*) (2,6-di-*tert*-butyl-4-methylphenol, radical stabilizer present in commercial samples of THF) and the $[\text{FeCp}(\text{CNMe})_3]^+$ byproduct (*) are marked with asterisks.

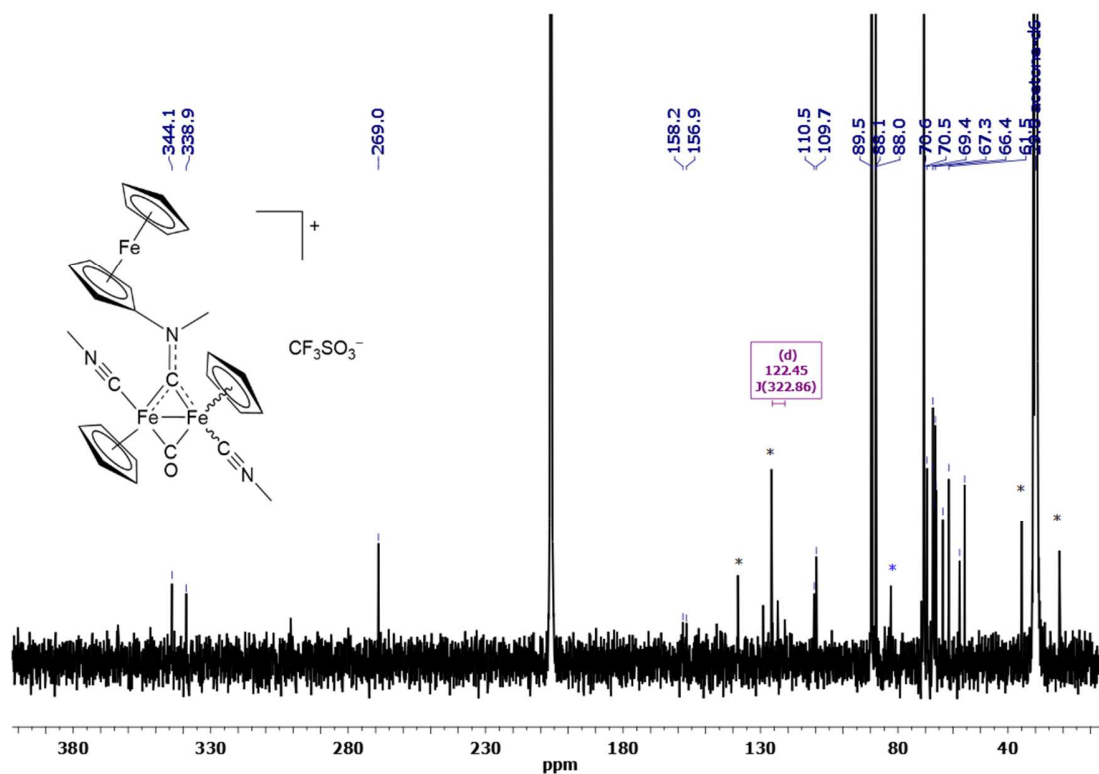


Figure S18. ^1H NMR spectrum (401 MHz, acetone- d_6) of $[\text{Fe}_2\text{Cp}_2(\text{CNCy})_2(\mu\text{-CO})\{\mu\text{-CNMe}(\text{Fc})\}]\text{CF}_3\text{SO}_3$, **4b** (*trans/cis* ratio = 3). Signals due to the $[\text{FeCp}(\text{CNCy})_3]^+$ by-product are marked with asterisk (*).

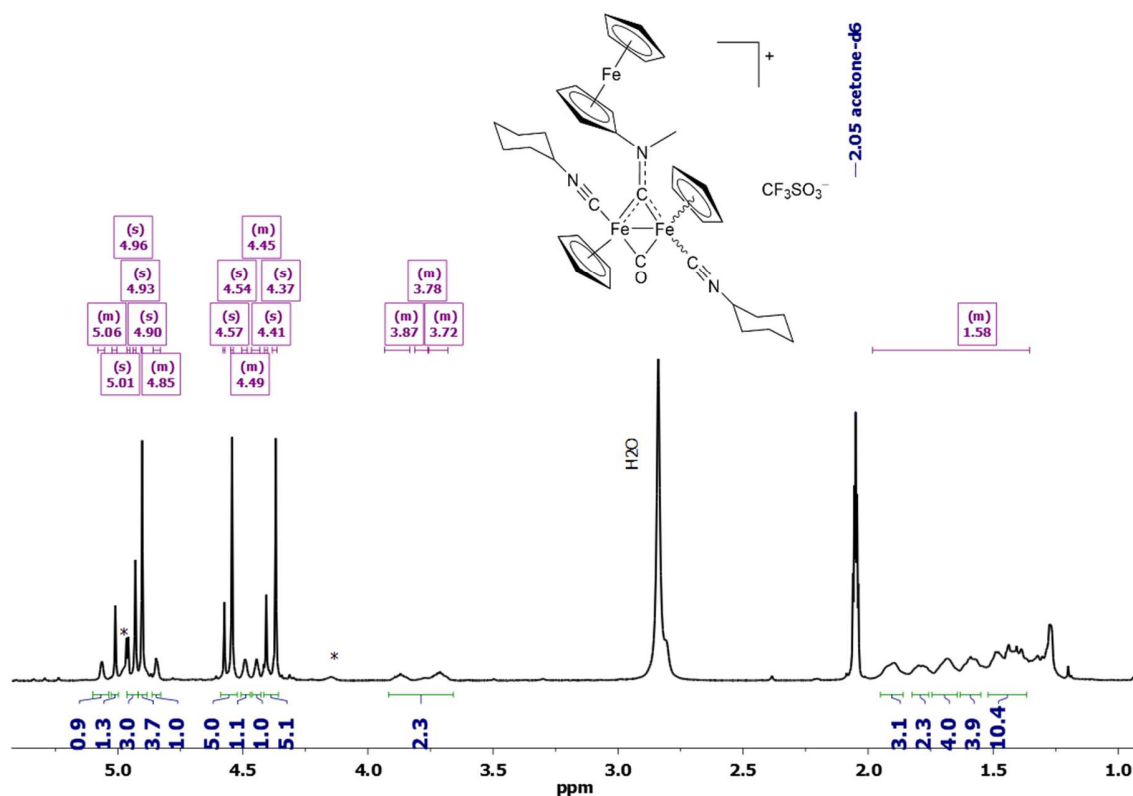


Figure S19. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum (101 MHz, acetone- d_6) of **4b** (*trans/cis* ratio = 3). Signals due to the $[\text{FeCp}(\text{CNCy})_3]^+$ by-product are marked with asterisk (*).

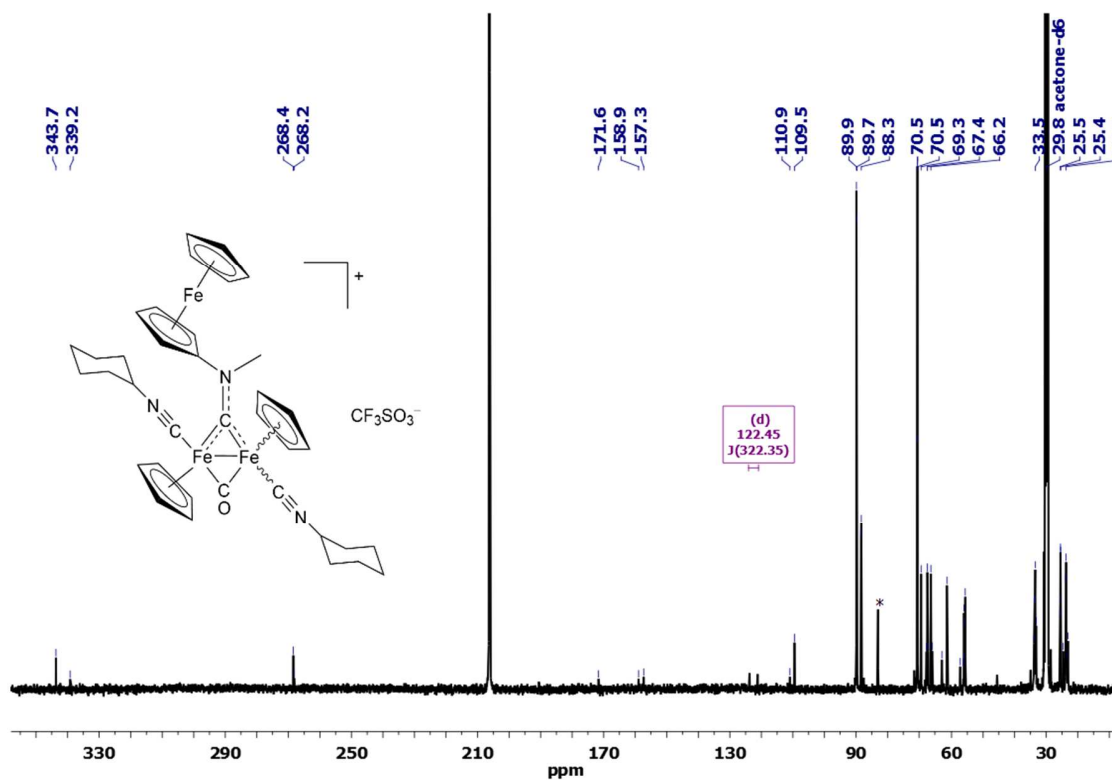


Figure S20. Black line: ^1H NMR spectrum (401 MHz, acetone- d_6) of **4b**. Red line: ^1H NOESY with irradiation at 5.01 ppm (Cp of the *cis* isomer). NOE is indicated by the arrow.

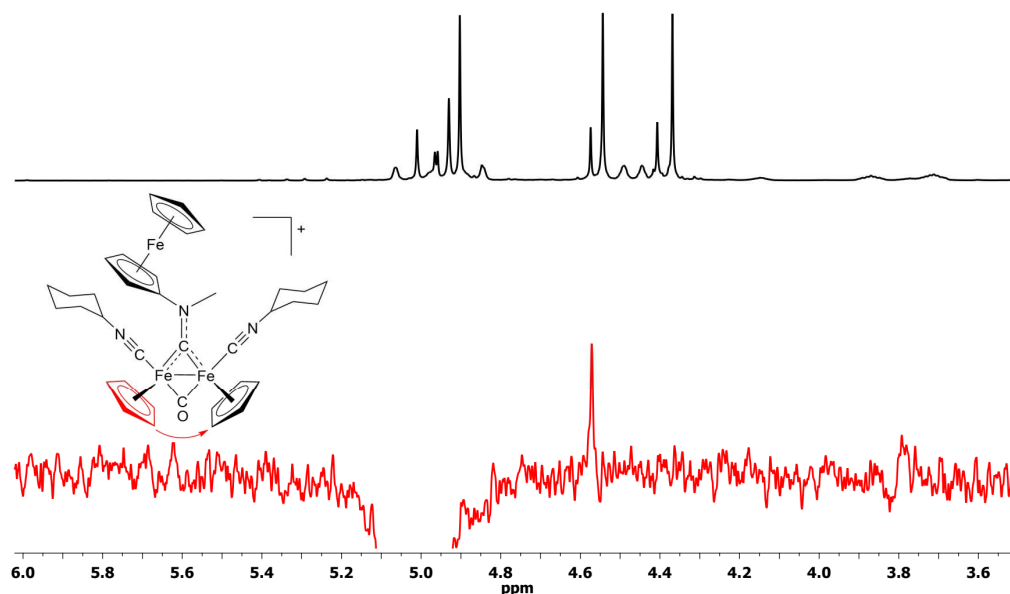


Figure S21. ^1H NMR spectrum (401 MHz, acetone- d_6) of $[\text{Fe}_2\text{Cp}_2\{\text{CN}(4\text{-C}_6\text{H}_4\text{OMe})\}_2(\mu\text{-CO})\{\mu\text{-CNMe}(\text{Fc})\}]\text{CF}_3\text{SO}_3$, **4c** (*cis/trans* ratio = 1.5). Signals due to the $[\text{FeCp}\{\text{CN}(4\text{-C}_6\text{H}_4\text{OMe})\}_3]^+$ by-product are marked with asterisk (*).

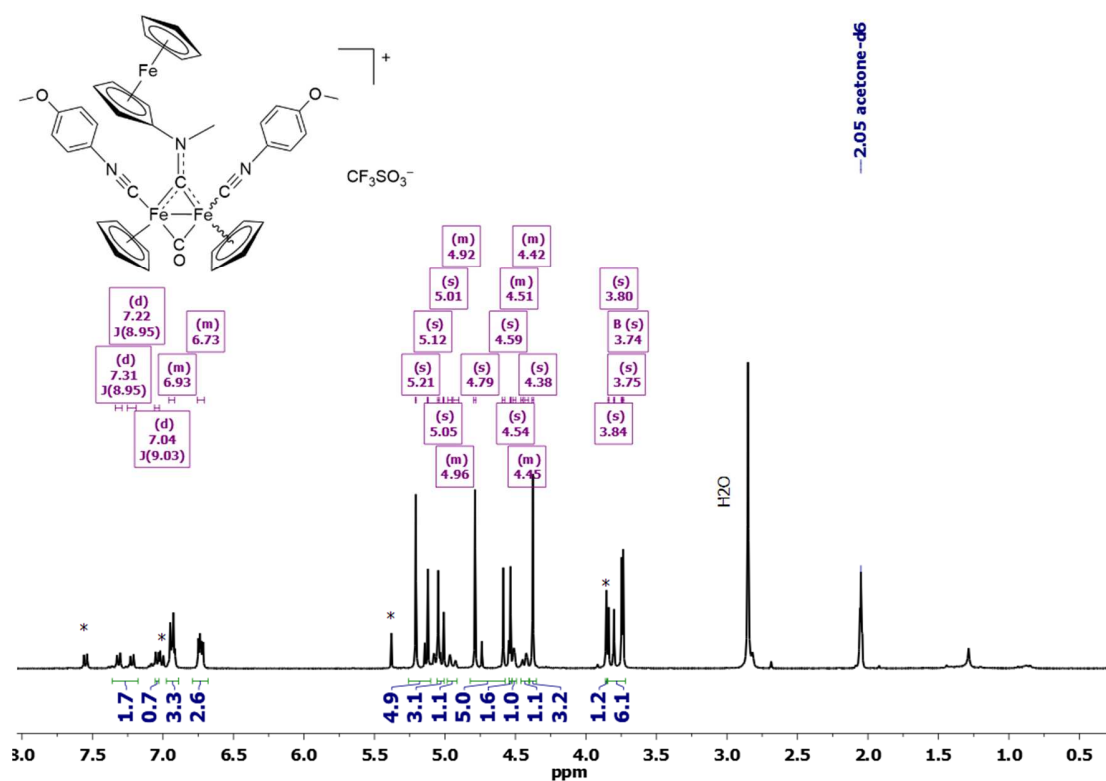


Figure S22. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum (101 MHz, acetone- d_6) of **4c** (*cis/trans* ratio = 1.5). Signals due to the $[\text{FeCp}\{\text{CN}(4\text{-C}_6\text{H}_4\text{OMe})\}_3]^+$ by-product are marked with asterisk (*).

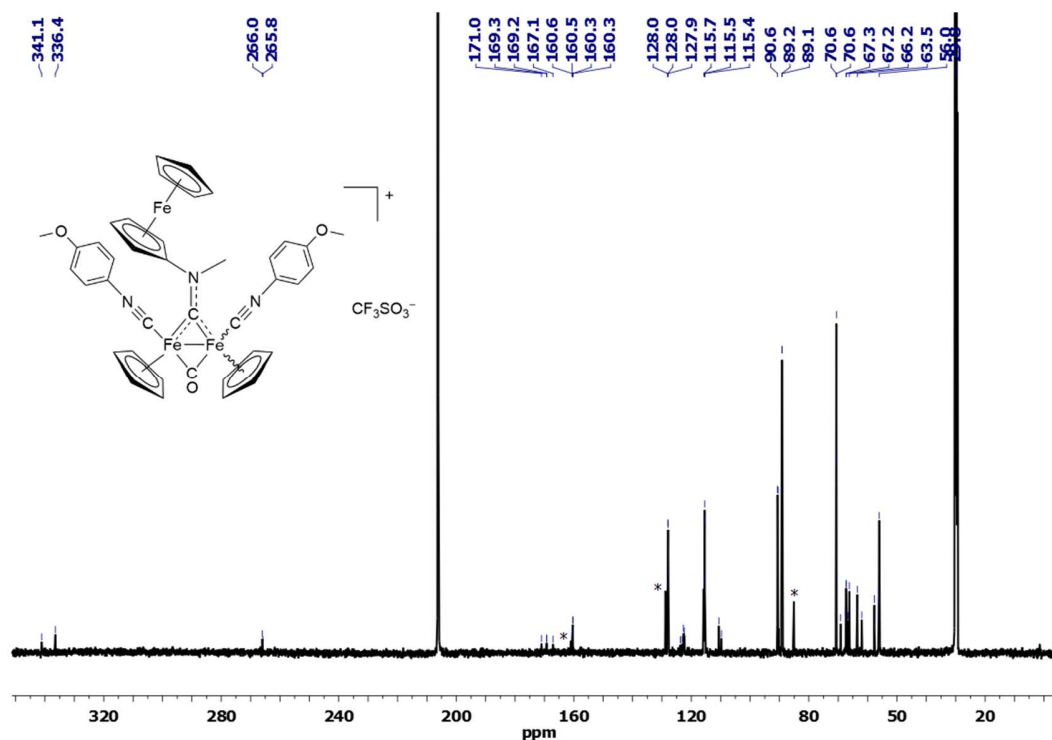


Figure S23. ^1H NOESY NMR spectra (401 MHz, acetone- d_6) of *trans*-**4c**. **Blue line:** irradiation at 5.12 ppm (Cp). **Red line:** irradiation at 4.59 ppm (Cp'). NOEs are indicated by the arrows; dotted lines indicate weak NOE effects.

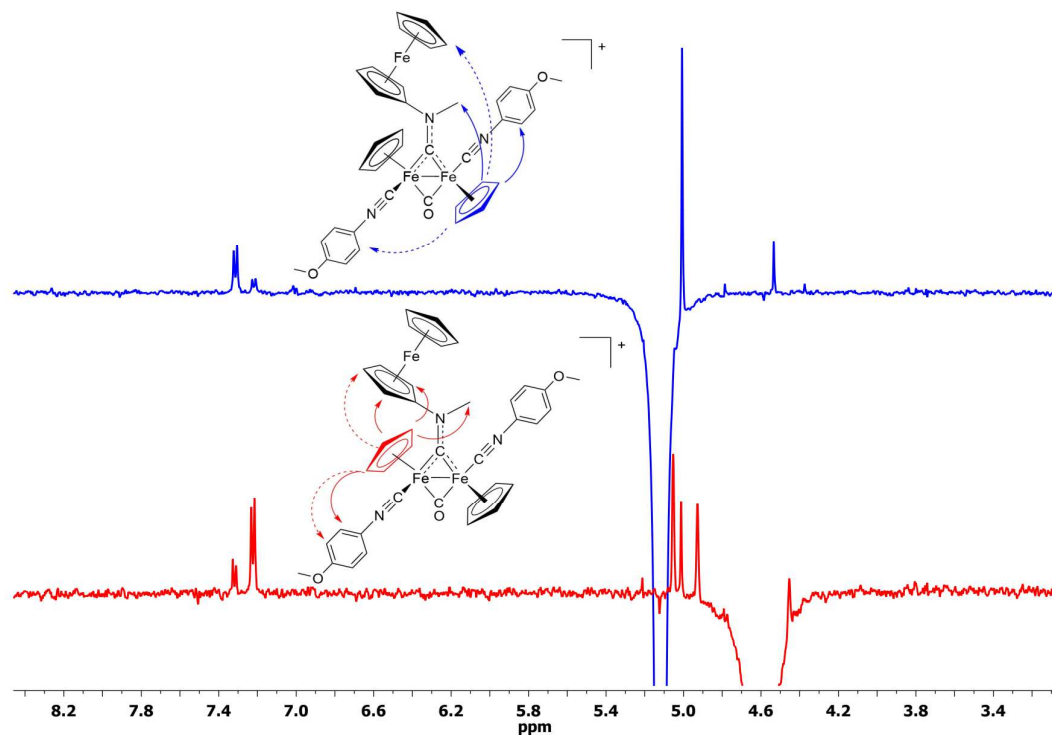


Figure S24. ^1H NOESY NMR spectra (401 MHz, acetone- d_6) of *cis*-**4c**. **Blue line:** irradiation at 5.21 ppm (Cp). **Red line:** irradiation at 4.79 ppm (Cp'). **Green line:** irradiation at 5.05 ppm (NCH₃). NOEs are indicated by the arrows; dotted lines indicate weak NOE effects.

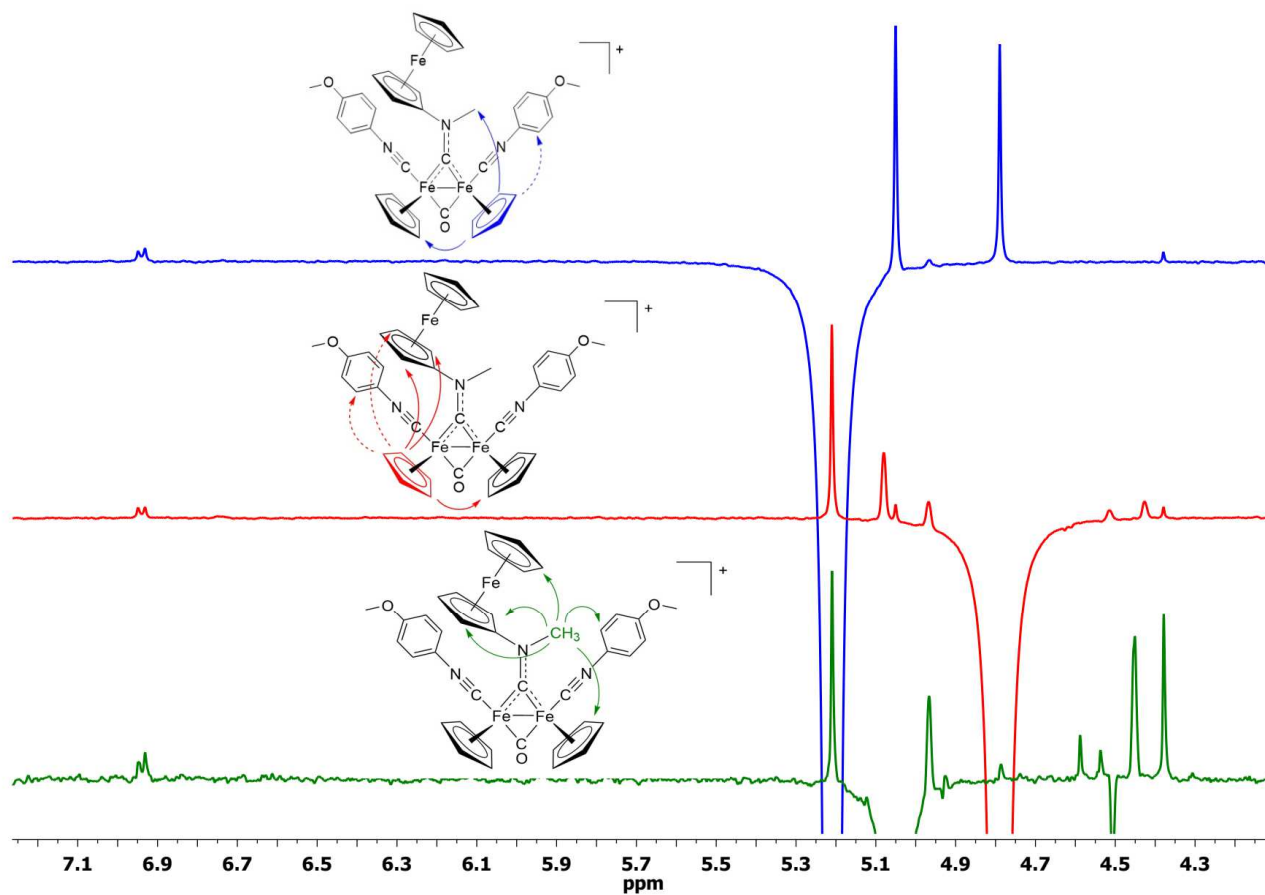


Figure S25. ^1H NMR spectrum (401 MHz, acetone- d_6) of $[\text{Fe}_2\text{Cp}_2(\text{CO})(\mu\text{-CO})\{\mu\text{-}\eta^1\text{:}\eta^3\text{-C}^3\text{MeC}^2\text{HC}^1\text{NMe}(\text{Fc})\}]\text{CF}_3\text{SO}_3$, **5a** (*cis-E/Z* ratio = 2.5 + traces of *trans* isomers – not highlighted).

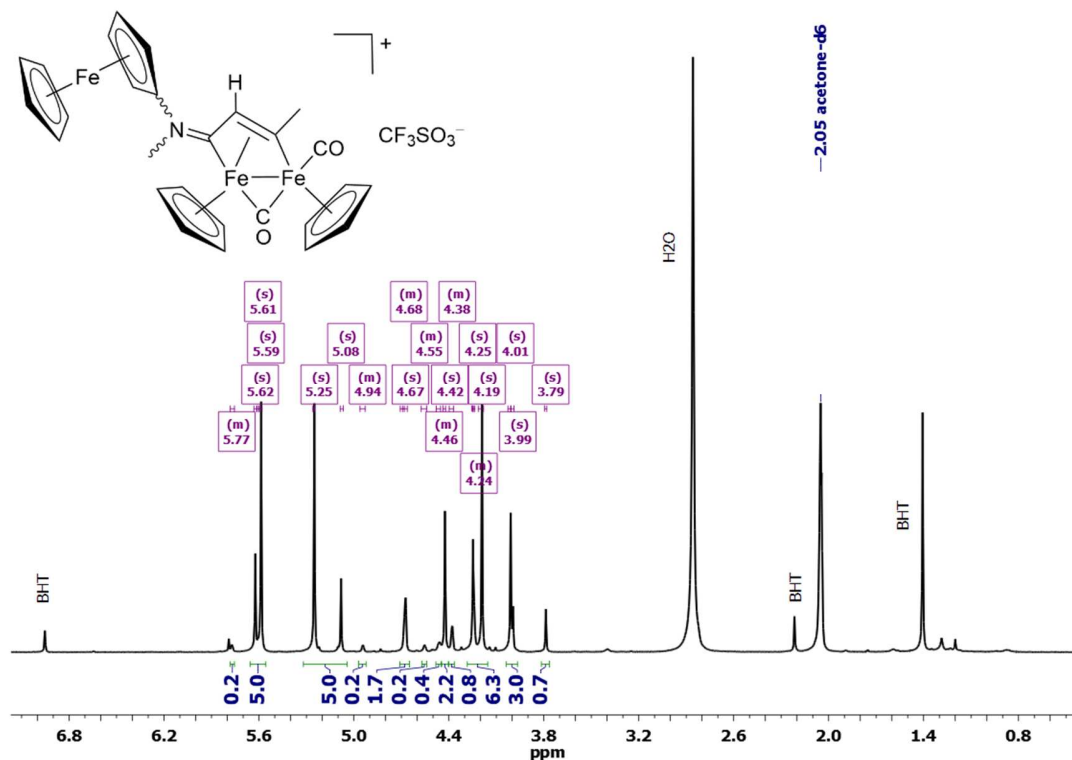


Figure S26. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum (101 MHz, acetone- d_6) of **5a** (*cis-E/Z* ratio = 2.5 + traces of *trans* isomers – not highlighted).

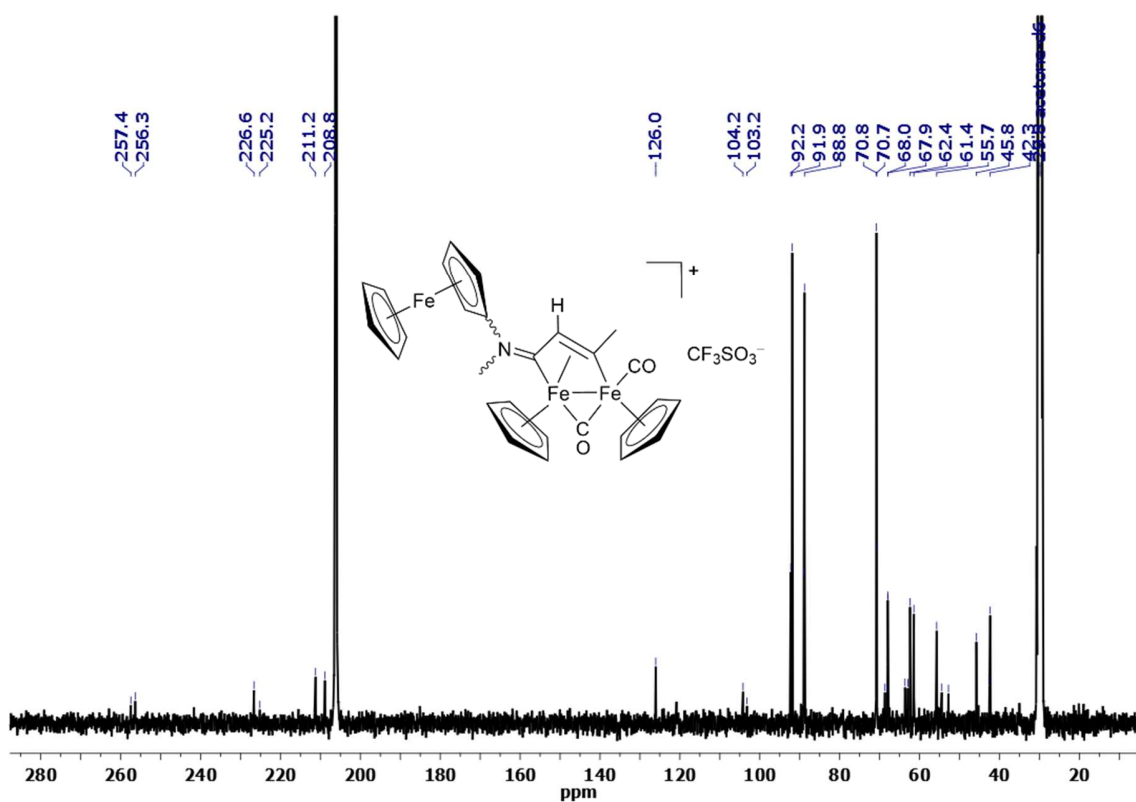


Figure S27. Black line: ^1H NMR spectrum (401 MHz, acetone- d_6) of **5a**. Blue line: ^1H NOESY with irradiation at 5.59 ppm (Cp) of the *cis-E* isomer. Red line: ^1H NOESY with irradiation at 5.25 ppm (Cp L) of the *cis-E* isomer. Green line: ^1H NOESY with irradiation at 5.08 ppm (Cp L) of the *cis-Z* isomer. NOEs are indicated by the arrows.

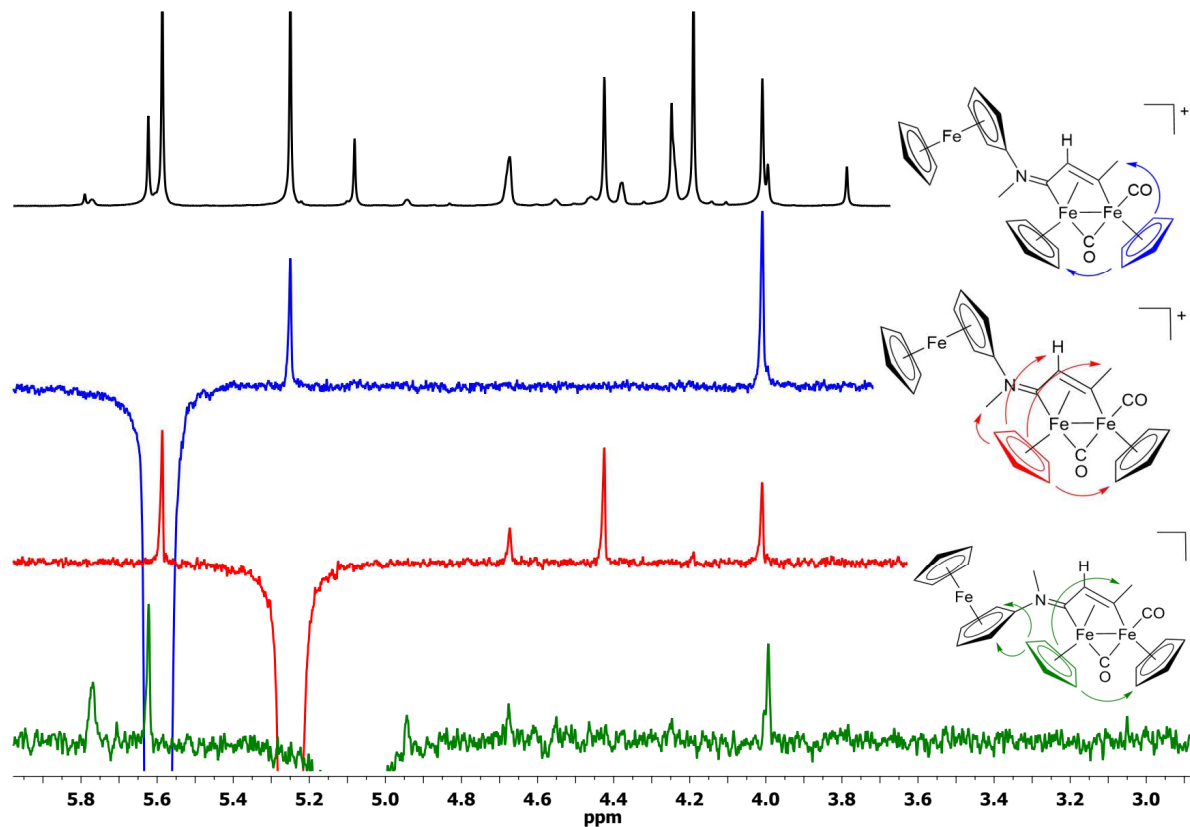


Figure S28. Solid-state IR spectrum (650-4000 cm^{-1}) of $[\text{Fe}_2\text{Cp}_2(\text{CO})(\mu\text{-CO})\{\mu\text{-}\eta^1\text{:}\eta^3\text{-C}^3\text{PhC}^2\text{HC}^1\text{NMe}(\text{Fc})\}]\text{CF}_3\text{SO}_3$, *cis-E-5b*.

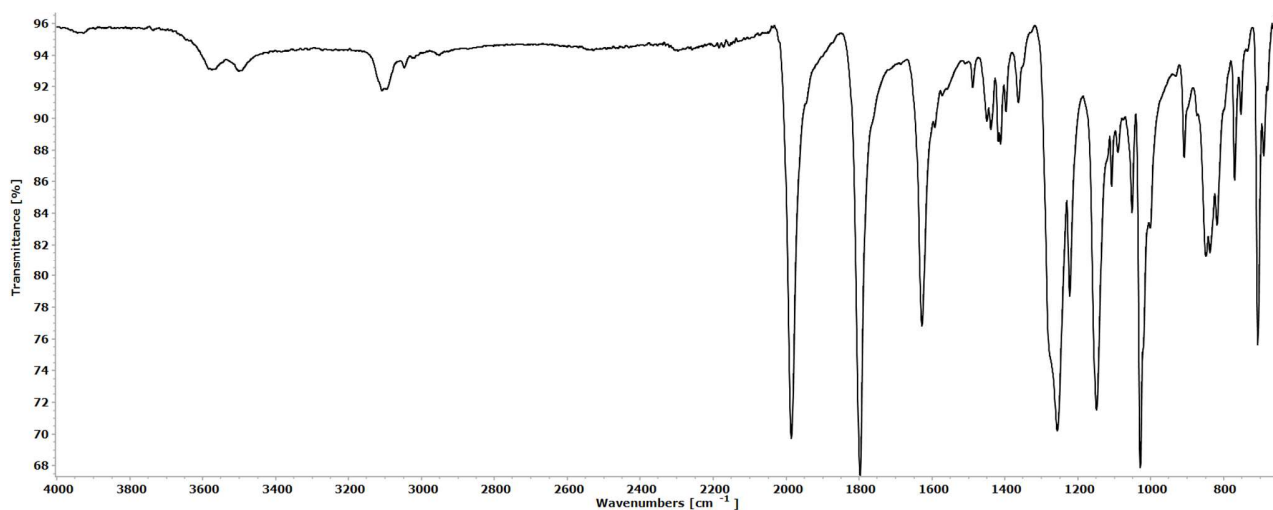


Figure S29. ^1H NMR spectrum (401 MHz, acetone- d_6) of *cis-E-5b*. Traces of *trans-E-5b* are indicated with the asterisk (*).

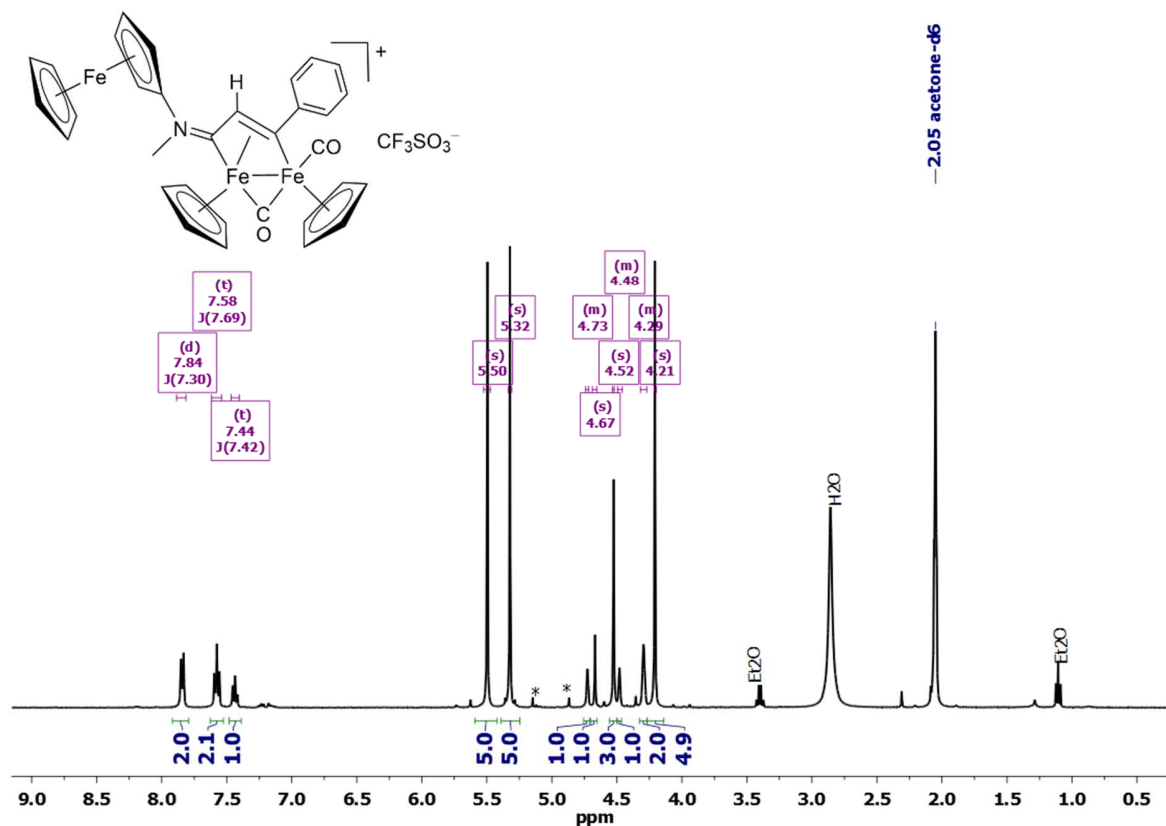


Figure S30. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum (101 MHz, acetone- d_6) of *cis-E-5b*.

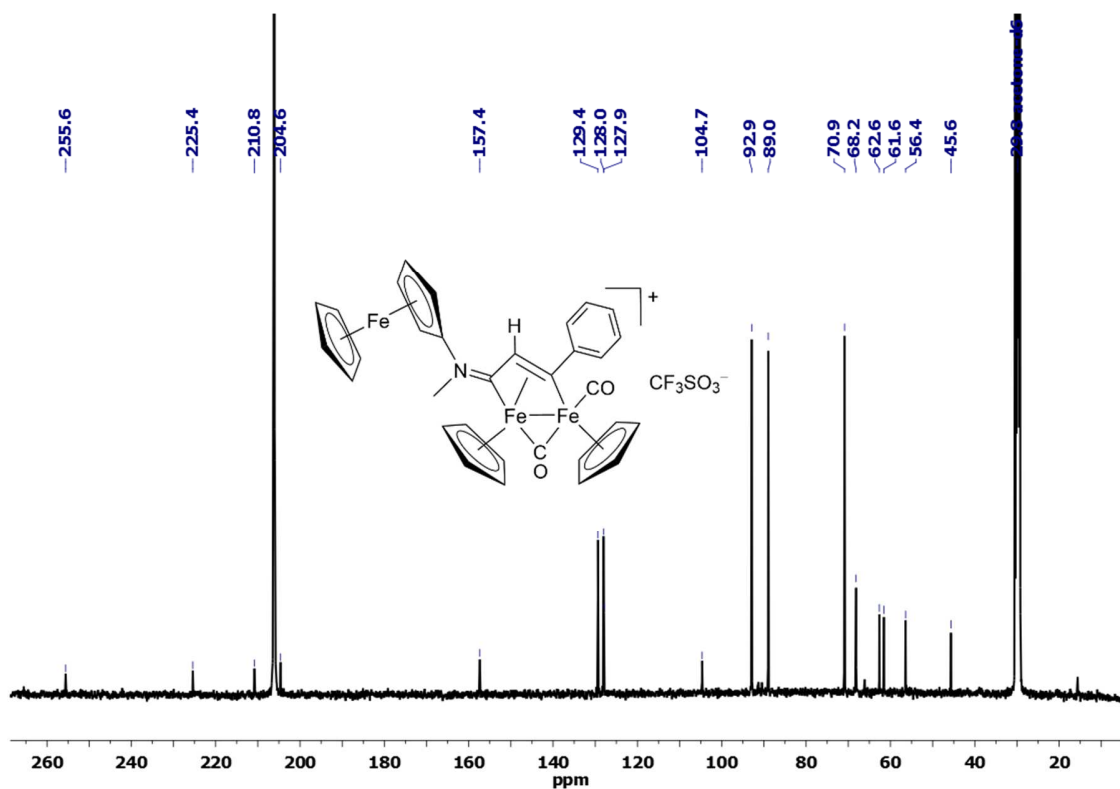


Figure S31. Black line: ^1H NMR spectrum (401 MHz, acetone- d_6) of *cis-E-5b*. Blue line: ^1H NOESY with irradiation at 5.50 ppm (Cp^L). Red line: ^1H NOESY with irradiation at 5.32 ppm (Cp). **A:** NOEs are indicated by the arrows. **B:** Negative peaks (black arrows) indicate the chemical exchange underlying the partial magnetization transfer with the corresponding Cp ligand of the *trans-E* diastereomer.

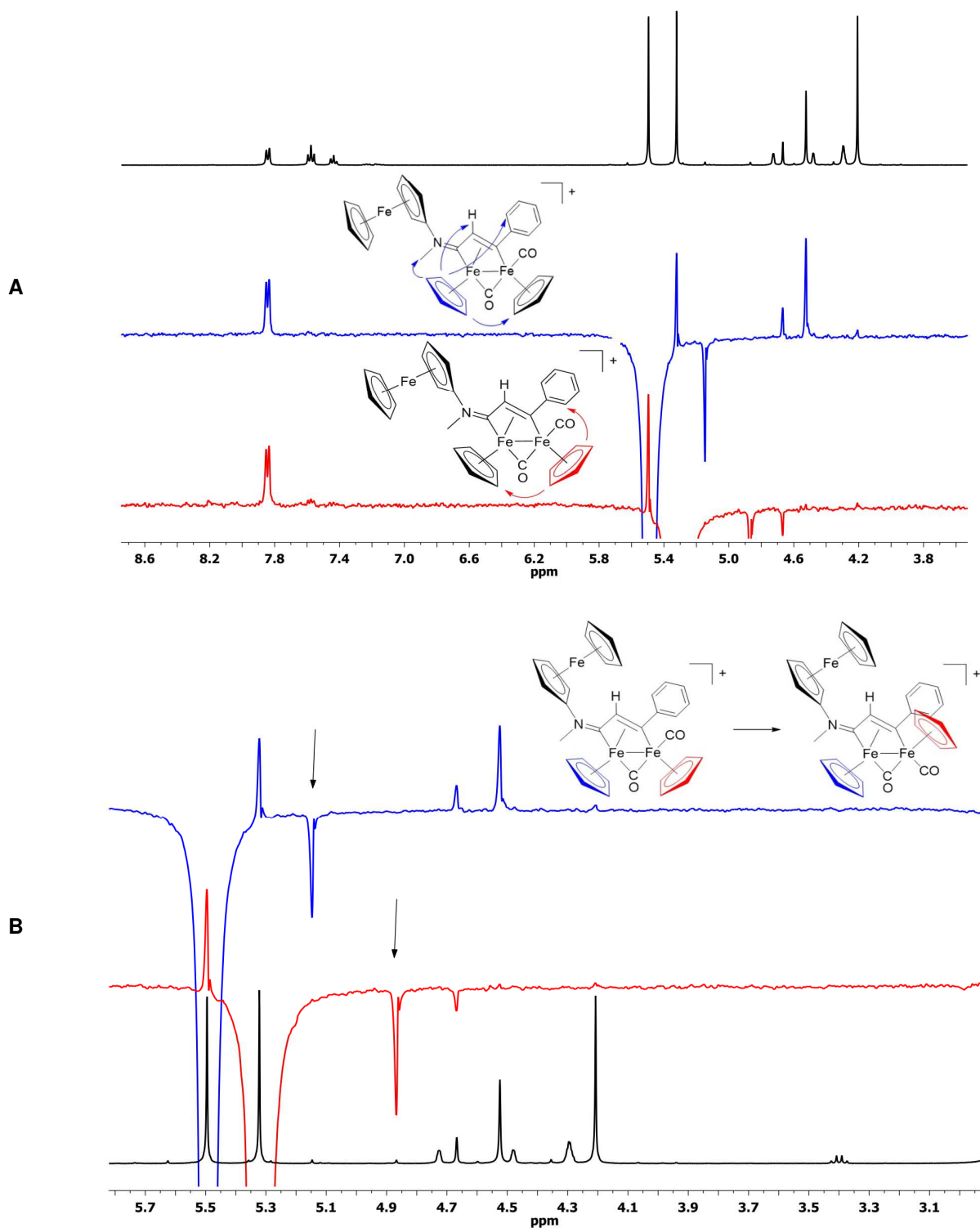


Figure S32. Solid-state IR spectrum (650-4000 cm^{-1}) of $[\text{Fe}_2\text{Cp}_2(\text{CO})(\mu\text{-CO})\{\mu\text{-}\eta^1\text{:}\eta^3\text{-C}^3\text{MeC}^2\text{MeC}^1\text{NMe}(\text{Fc})\}\text{CF}_3\text{SO}_3, \mathbf{5c}$ (mixture of *cis*-*E/Z* isomers).

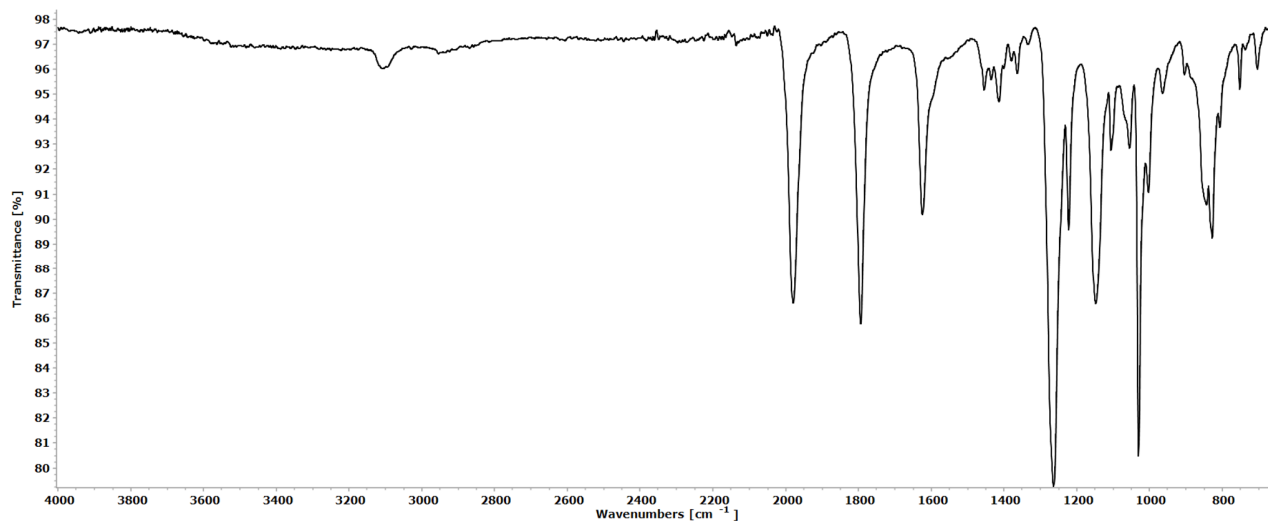


Figure S33. ^1H NMR spectrum (401 MHz, acetone- d_6) of $\mathbf{5c}$ (*cis*-*E/Z* ratio = 7.6).

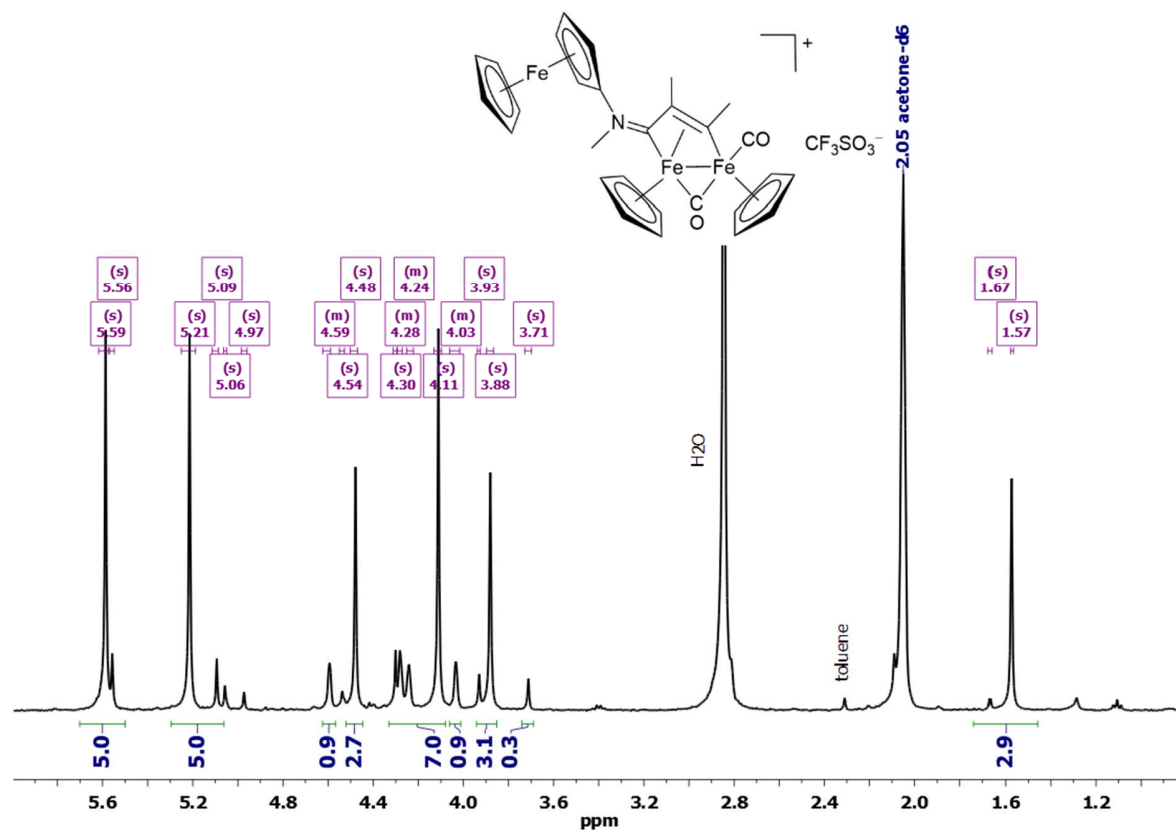


Figure S34. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum (101 MHz, acetone- d_6) of **5c** (*cis-E/Z* ratio = 7.6).

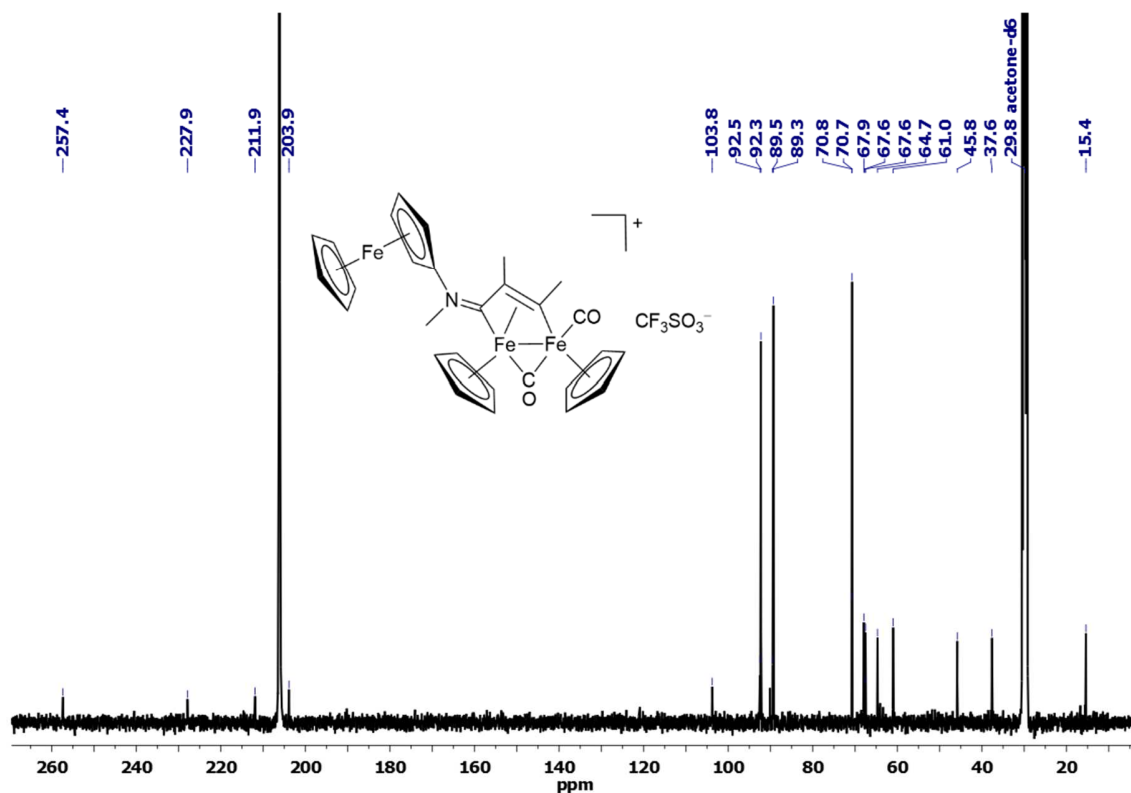


Figure S35. Black line: ^1H NMR spectrum (401 MHz, acetone- d_6) of **5c**. Blue line: ^1H NOESY with irradiation at 5.59 ppm (Cp of *cis-E-5c*). Red line: ^1H NOESY with irradiation at 5.21 ppm (Cp^L of *cis-E-5c*). Green line: ^1H NOESY with irradiation at 5.09 ppm (Cp^L of *cis-Z-5c*). NOEs are indicated by the arrows.

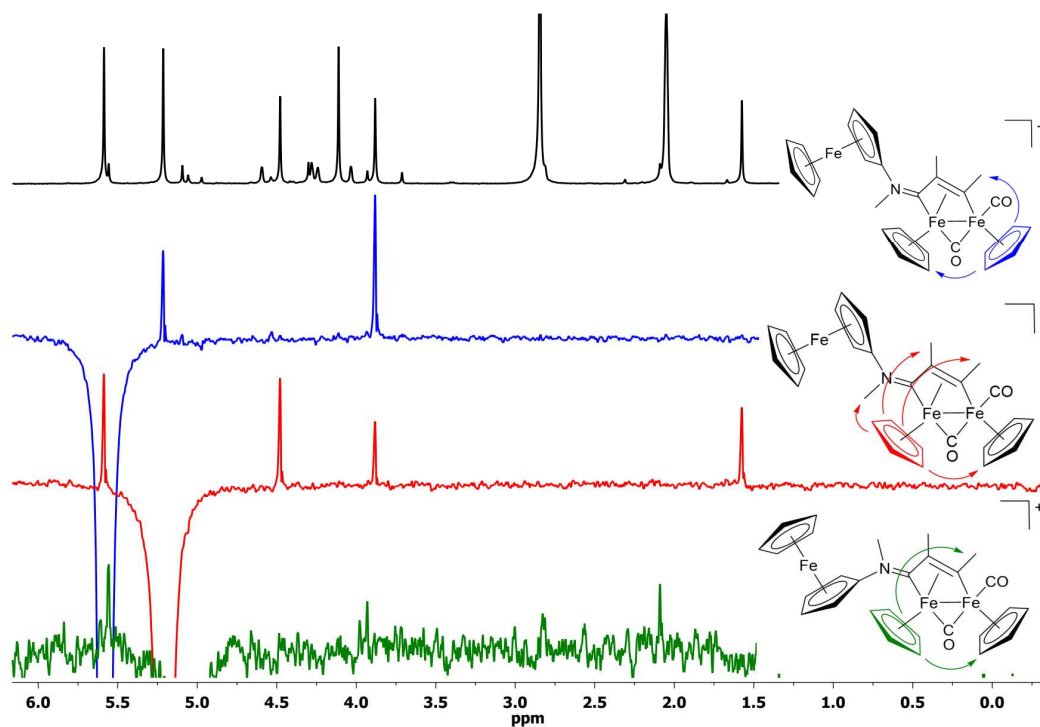
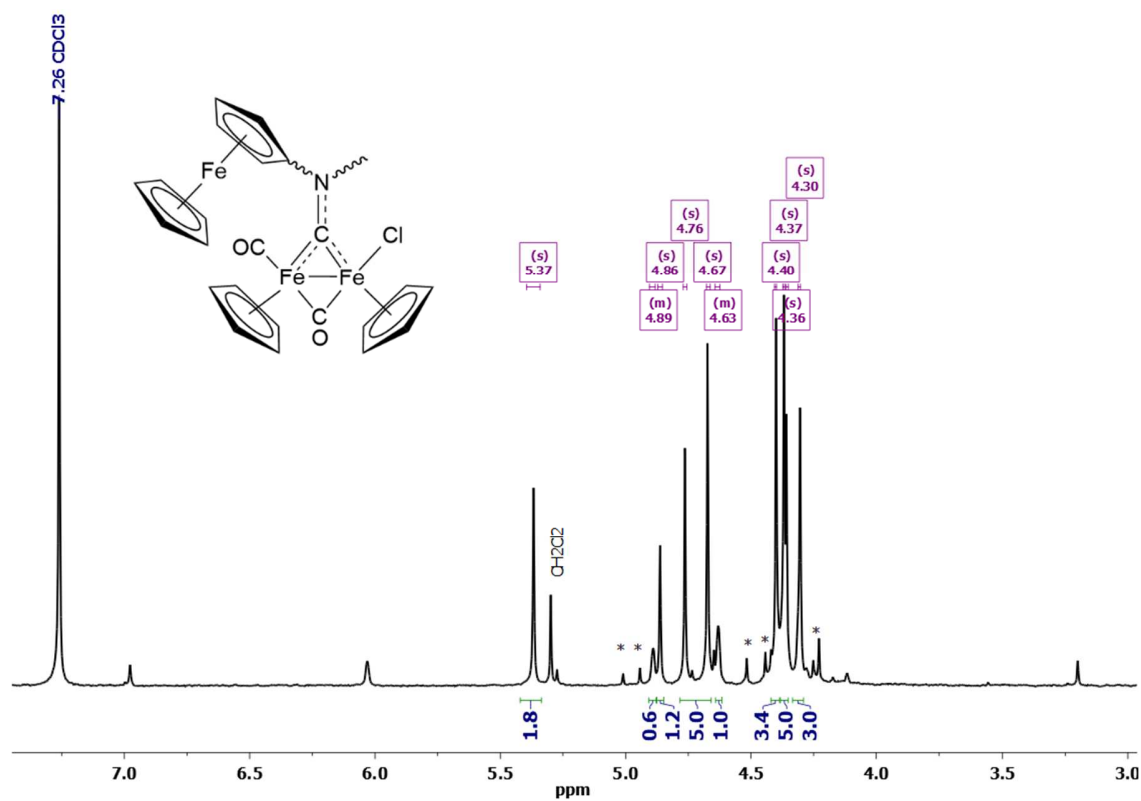


Figure S36. ^1H NMR spectrum (401 MHz, CDCl_3) of $[\text{Fe}_2\text{Cp}_2\text{Cl}(\text{CO})_2\{\mu\text{-CNMe}(\text{Fc})\}]$, **6** (*cis-E/Z* ratio = 1.5). Signals due to *trans* isomers are marked with asterisk (*).



qNMR purity assessment

Figure S37. ^1H NMR spectrum (401 MHz, acetone- d_6) of **2** (*cis*-*E/Z* ratio = 1.8 + ca. 6 % *trans* isomers) + 1,3,5-trimethoxybenzene (standard for qNMR).

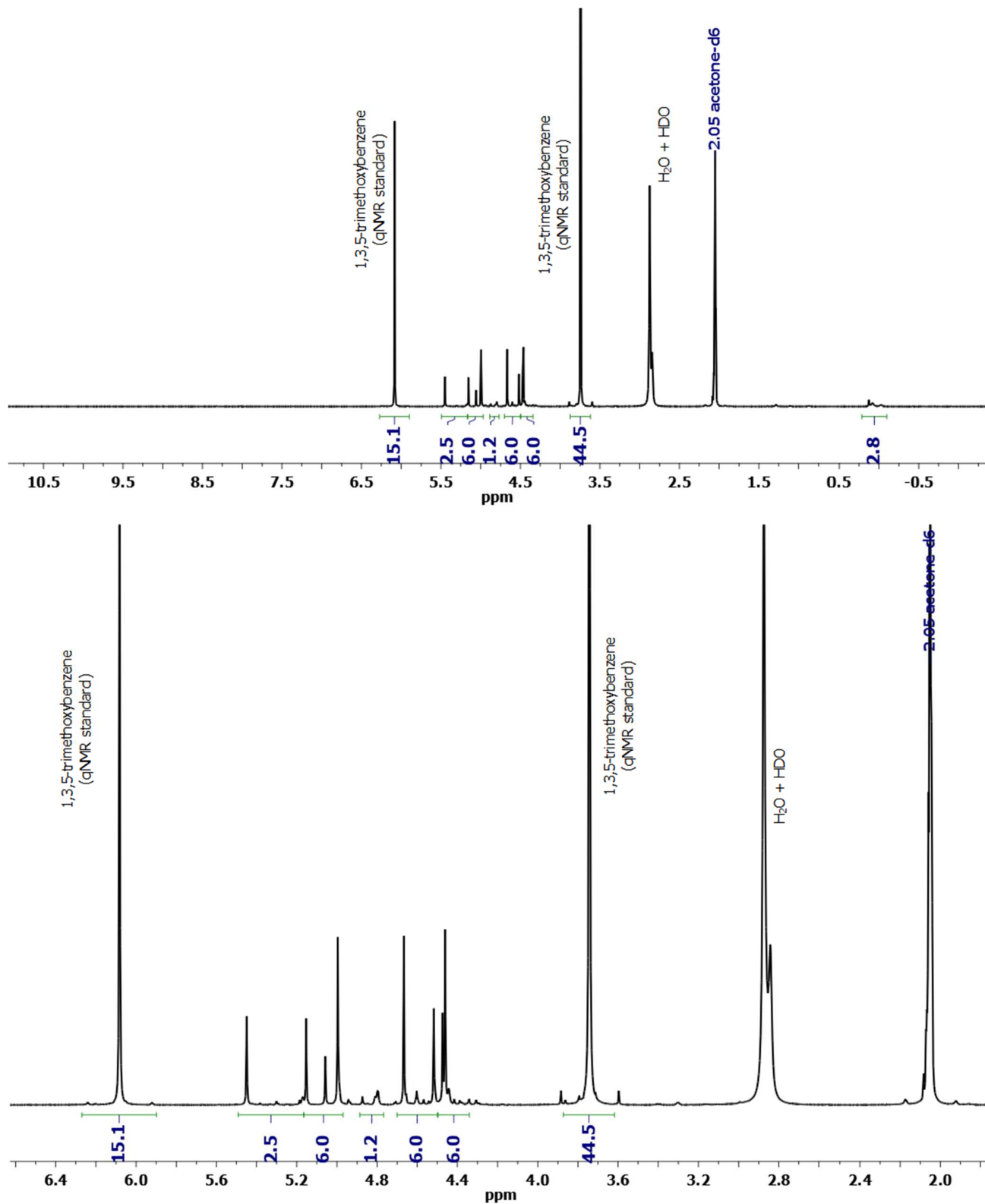


Figure S38. ^1H NMR spectrum (401 MHz, acetone- d_6) of **3a** (*cis*-*E/Z* ratio ≈ 1 + traces of *trans* isomers) + 1,3,5-trimethoxybenzene (standard for qNMR).

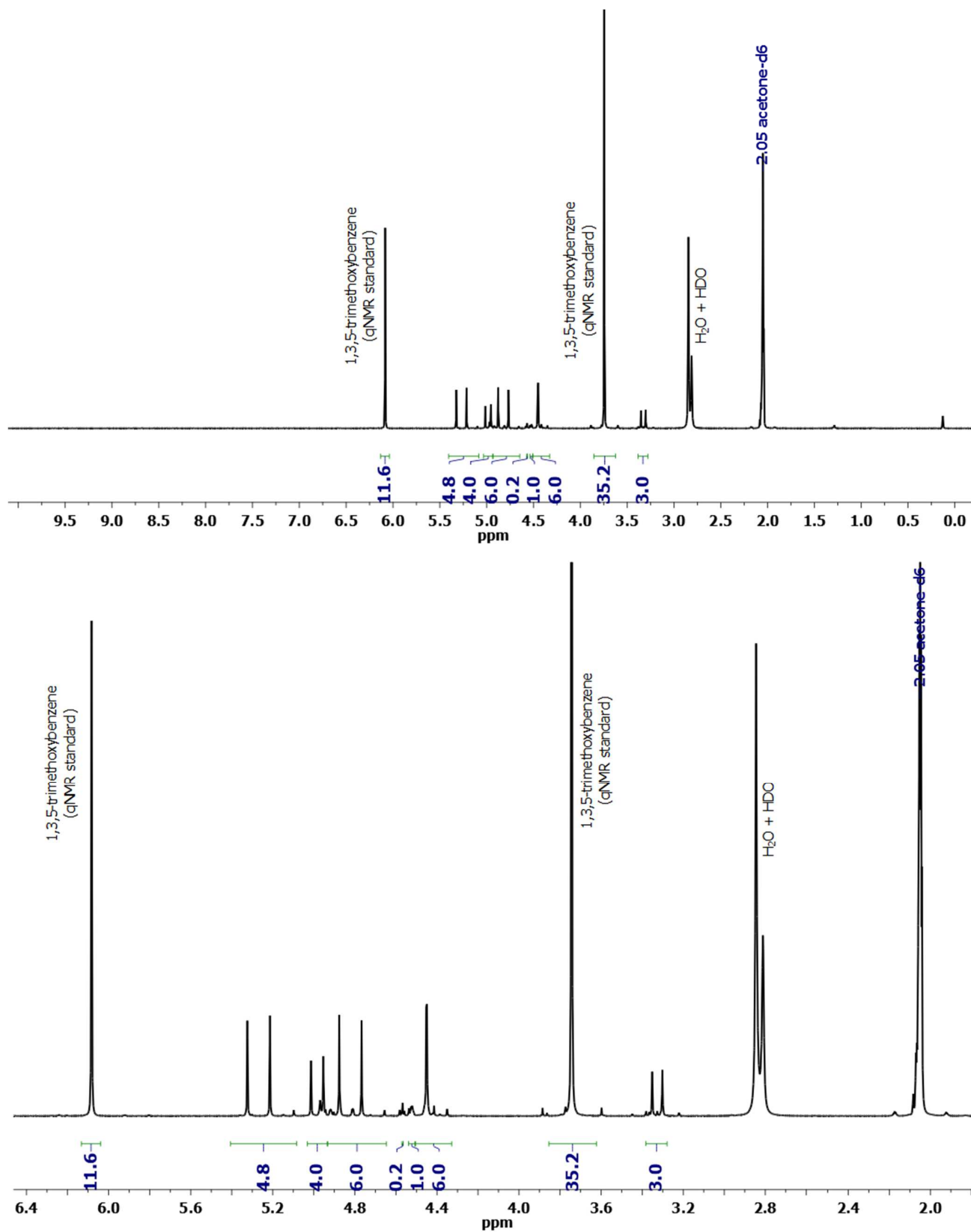


Figure S39. ^1H NMR spectrum (401 MHz, acetone- d_6) of **3b** (*cis*-*E/Z* ratio = 1.5 + *trans* isomer; *cis/trans* ratio = 4.1; signals marked with asterisk *) + 1,3,5-trimethoxybenzene (standard for qNMR).

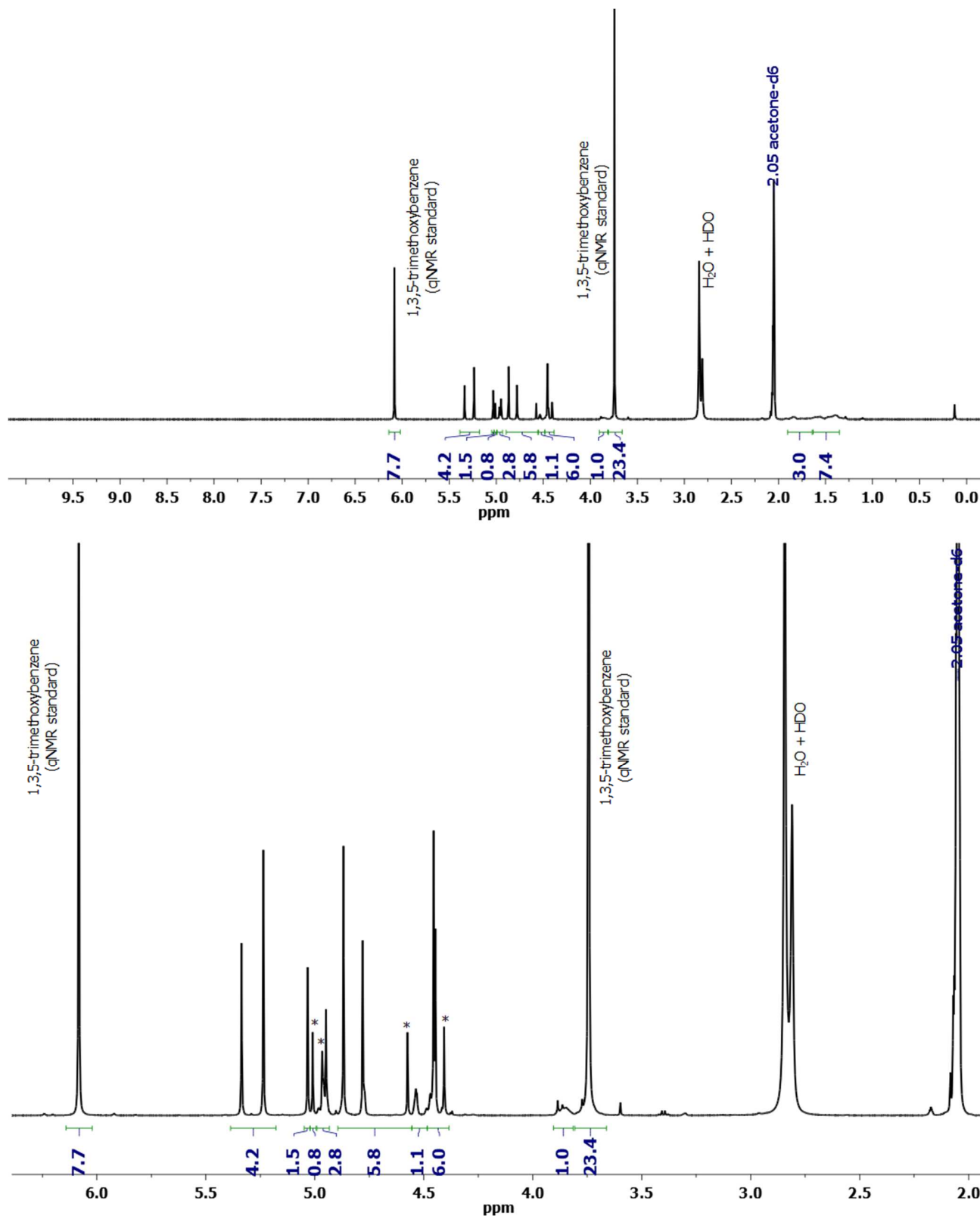


Figure S40. ^1H NMR spectrum (401 MHz, CD_3CN) of **5a** (*cis-E/Z* ratio = 2.2) + 1,3,5-trimethoxybenzene (standard for qNMR).

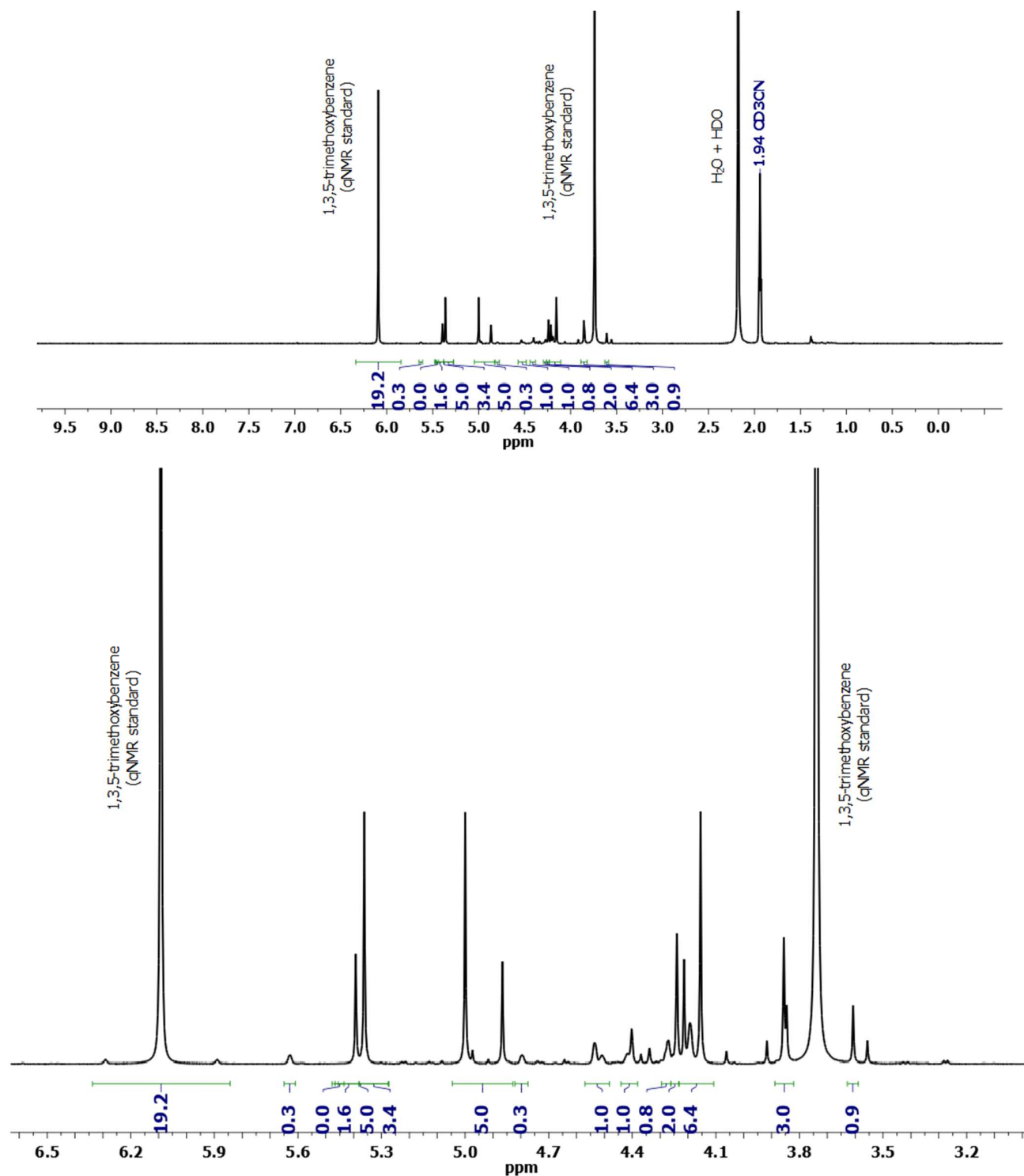


Figure S41. ^1H NMR spectrum (401 MHz, acetone- d_6) of **5b** (*cis*-*E/Z* ratio = 9.1; signals of *cis*-**Z-5b** are marked with asterisk *) + 1,3,5-trimethoxybenzene (standard for qNMR).

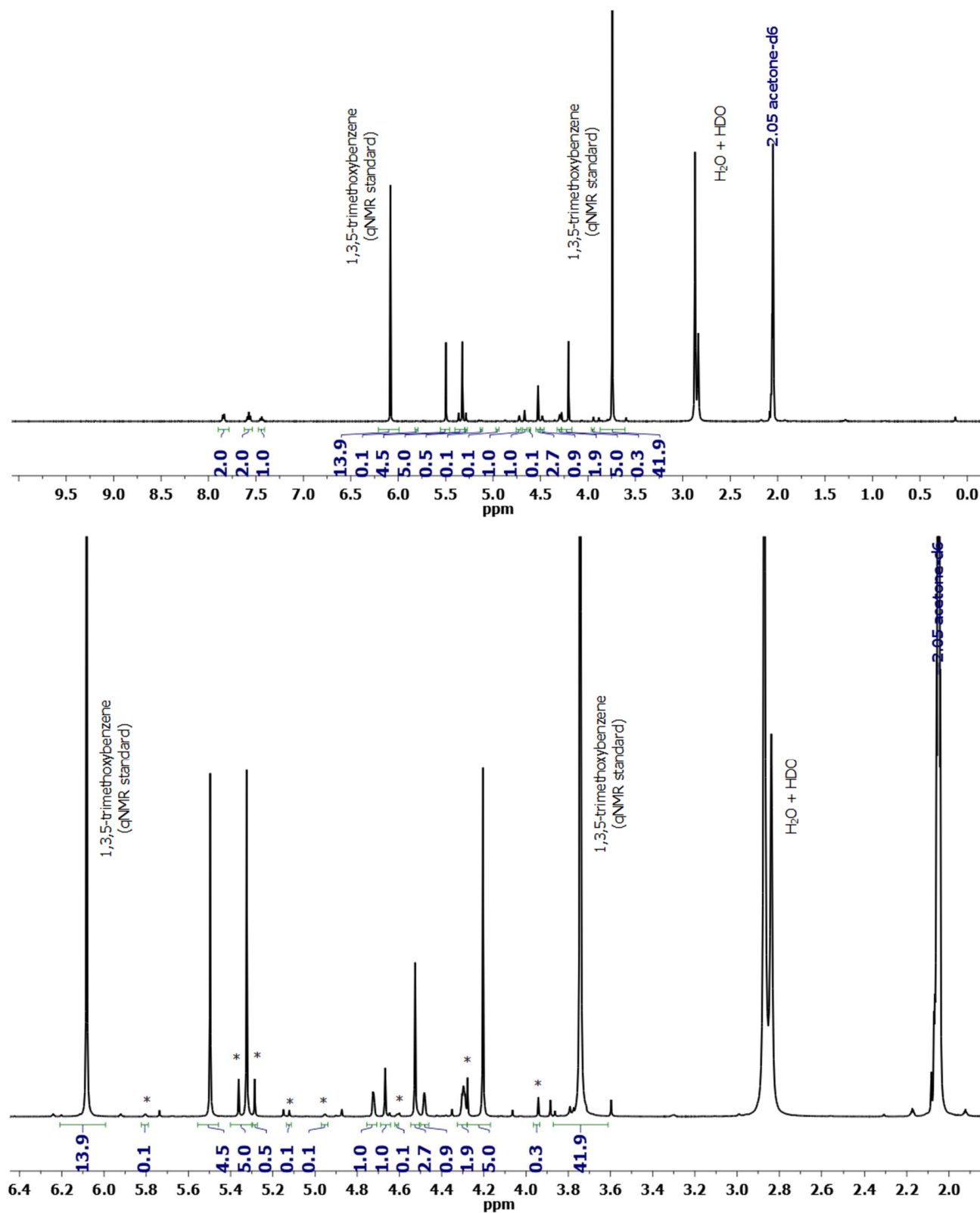


Figure S42. ^1H NMR spectrum (401 MHz, acetone- d_6) of **5c** (*cis*-*E/Z* ratio = 7.6) + 1,3,5-trimethoxybenzene (standard for qNMR). The N-Me resonance of the *cis*-*Z* isomer (marked with asterisk *) is partially superimposed to the OMe resonance of 1,3,5-trimethoxybenzene.

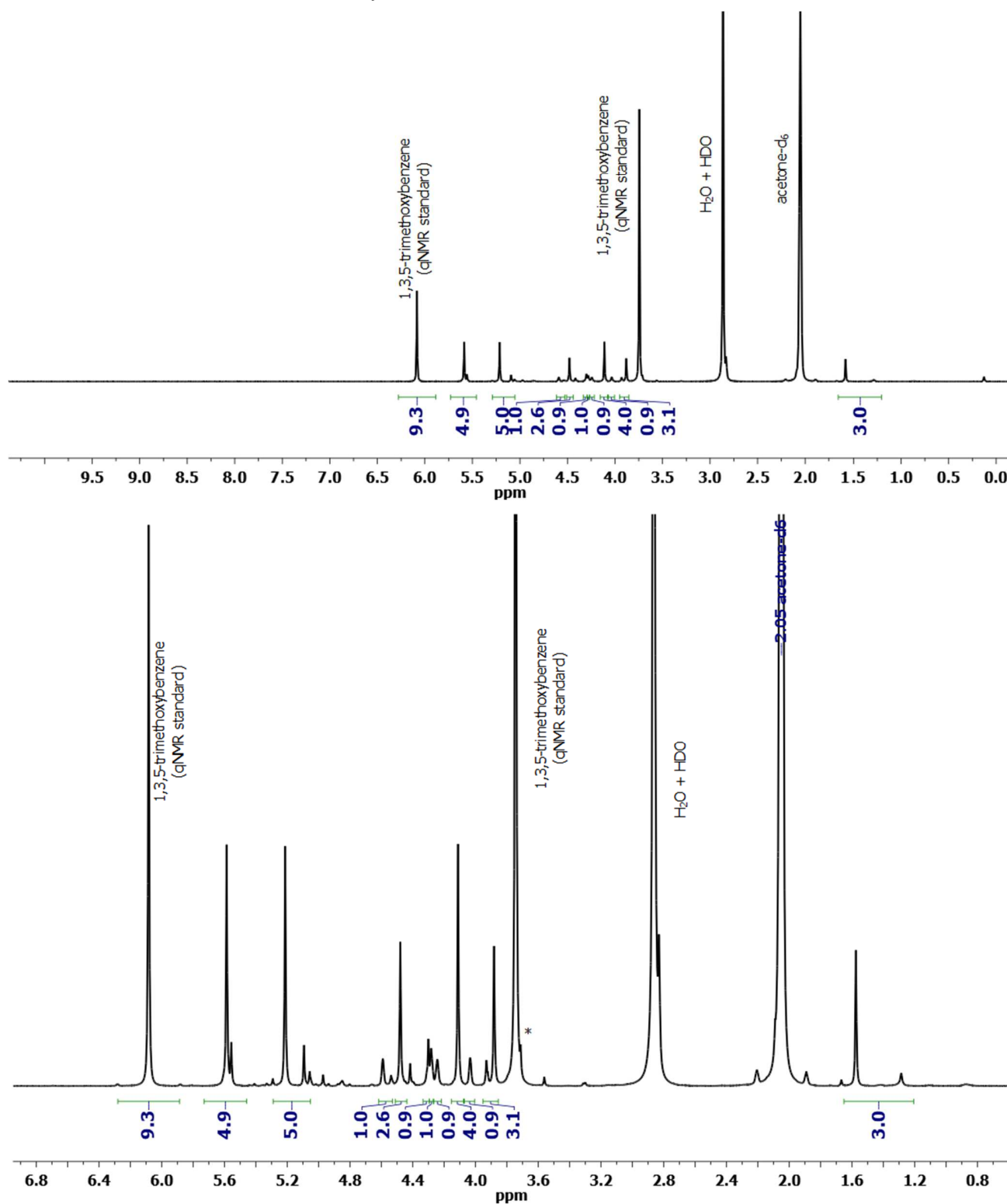


Table S1. Purity determination of **2**, **3a,b**, **5a-c** via qNMR analyses in acetone-d₆ or CD₃CN solution with 1,3,5-trimethoxybenzene as standard.

$$\%P_{Fe} = \frac{(I_{Fe}/N_{Fe})}{(I_{std}/N_{std})} \cdot \frac{(m_{std}/M_{std})}{(m_{Fe}/M_{Fe})} \cdot \%P_{std}$$

Fe compound (analyte)			1,3,5-trimethoxybenzene (standard)			NMR data				Fe compound purity (%)
Sample	molar mass (g/mol)	mass (g)	molar mass (g/mol)	mass (g)	Purity (%)	Relative integrals (I) and corresponding number of protons (N)				
	m _{Fe}	M _{Fe}	m _{std}	M _{std}	%P _{std}	I _{std}	N _{std}	I _{Fe}	N _{Fe}	
2	690.073	2.046	168.19	2.347	99.8	44.52	9	6	6	95.0
3a	714.094	1.716	168.19	1.508	99.8	35.20	9	6	6	95.2
3b	781.203	2.705	168.19	1.442	99.8	23.40	9	6	6	95.0
5a	713.106	2.086	168.19	2.992	99.8	19.19	3	5	5	94.9
5b	775.175	2.047	168.19	1.97	99.8	41.91	9	5	5	95.1
5c	727.133	4.001	168.19	2.727	99.8	9.27	3	3	3	95.2

Table S2. Hydrogen bonds for *cis-E-5b*·H₂O [Å and °].

D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
O(101)-H(111)...O(13)#1	0.88(2)	2.06(2)	2.923(4)	164(5)
O(101)-H(112)...O(12)#2	0.88(2)	2.11(2)	2.965(5)	165(5)

Symmetry transformations used to generate equivalent atoms:

#1 x+1, y, z; #2 -x+1, -y, -z+1.

Table S3. Crystal data and measurement details for *cis-E-3a*, *cis-4b* and *cis-E-5b*·H₂O.

	<i>cis-E-3a</i>	<i>cis-4b</i>	<i>cis-E-5b</i> ·H ₂ O
Formula	C ₂₇ H ₂₅ F ₃ Fe ₃ N ₂ O ₅ S	C ₃₈ H ₄₄ F ₃ Fe ₃ N ₃ O ₄ S	C ₃₃ H ₃₁ F ₃ Fe ₃ NO ₆ S
FW	714.10	863.37	794.20
T, K	100(2)	100(2)	100(2)
λ , Å	0.71073	0.71073	0.71073
Crystal system	Monoclinic	Orthorhombic	Monoclinic
Space group	<i>P</i> 2 ₁ / <i>n</i>	<i>P</i> 2 ₁ 2 ₁ 2 ₁	<i>P</i> 2 ₁ / <i>c</i>
<i>a</i> , Å	8.9391(3)	11.2389(7)	11.7307(3)
<i>b</i> , Å	18.5516(7)	16.8628(11)	11.0006(3)
<i>c</i> , Å	16.6455(6)	19.0064(12)	24.6106(6)
β , °	104.1780(10)	90	101.3380(10)
Cell Volume, Å ³	2676.32(17)	3602.1(4)	3113.89(14)
Z	4	4	4
<i>D</i> _c , g·cm ⁻³	1.772	1.592	1.694
μ , mm ⁻¹	1.746	1.311	1.512
F(000)	1448	1784	1620
Crystal size, mm	0.18×0.15×0.12	0.16×0.14×0.11	0.15×0.12×0.09
θ limits, °	1.672-25.498	1.614-25.999	1.688-25.999
Reflections collected	36161	49721	43404
Independent reflections	4981 [<i>R</i> _{int} = 0.0278]	6997 [<i>R</i> _{int} = 0.0399]	6124 [<i>R</i> _{int} = 0.0564]
Data / restraints / parameters	4981 / 0 / 372	6997 / 0 / 471	6124 / 2 / 431
Goodness on fit on F ² ^a	1.109	1.119	1.050
<i>R</i> ₁ (<i>I</i> > 2 σ (<i>I</i>)) ^b	0.0449	0.0282	0.0426
<i>wR</i> ₂ (all data) ^c	0.1052	0.0718	0.1159
Largest diff. peak and hole, e Å ⁻³	1.447 / -0.740	0.656 / -0.264	1.715 / -1.269

^a Goodness on fit on F² = $[\sum w(F_o^2 - F_c^2)^2 / (N_{ref} - N_{param})]^{1/2}$, where $w = 1/[\sigma^2(F_o^2) + (aP)^2 + bP]$, where $P = (F_o^2 + 2F_c^2)/3$; N_{ref} = number of reflections used in the refinement; N_{param} = number of refined parameters. ^b $R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$. ^c $wR_2 = [\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)^2]^{1/2}$, where $w = 1/[\sigma^2(F_o^2) + (aP)^2 + bP]$, where $P = (F_o^2 + 2F_c^2)/3$.

Simulated IR and NMR spectra

Figure S43. Simulated IR spectra of *cis* and *trans* isomers of **4a** (C-PCM/r²SCAN-3c, acetone as continuous medium). Lorentzian broadening functions, FWHM = 8 cm⁻¹.

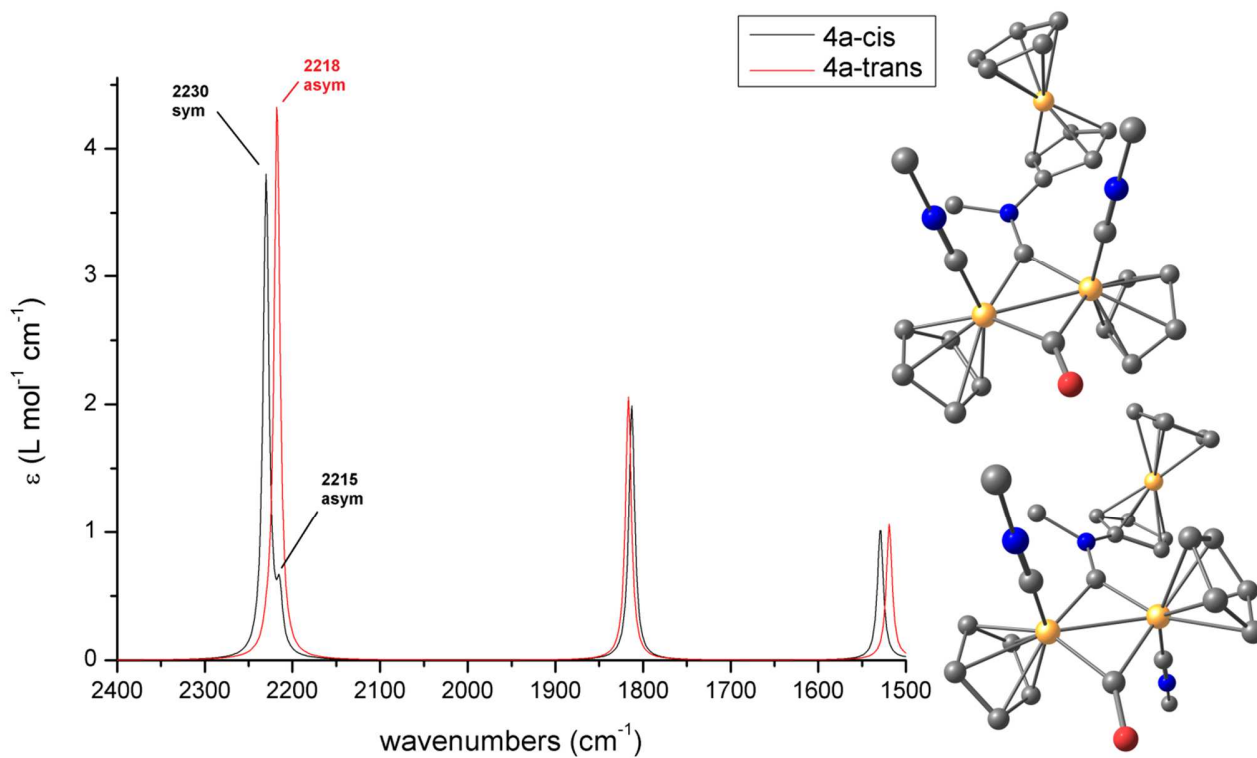


Figure S44. Simulated IR spectra of the isomers of **5c** (C-PCM/r²SCAN-3c, acetone as continuous medium). Lorentzian broadening functions, FWHM = 8 cm⁻¹.

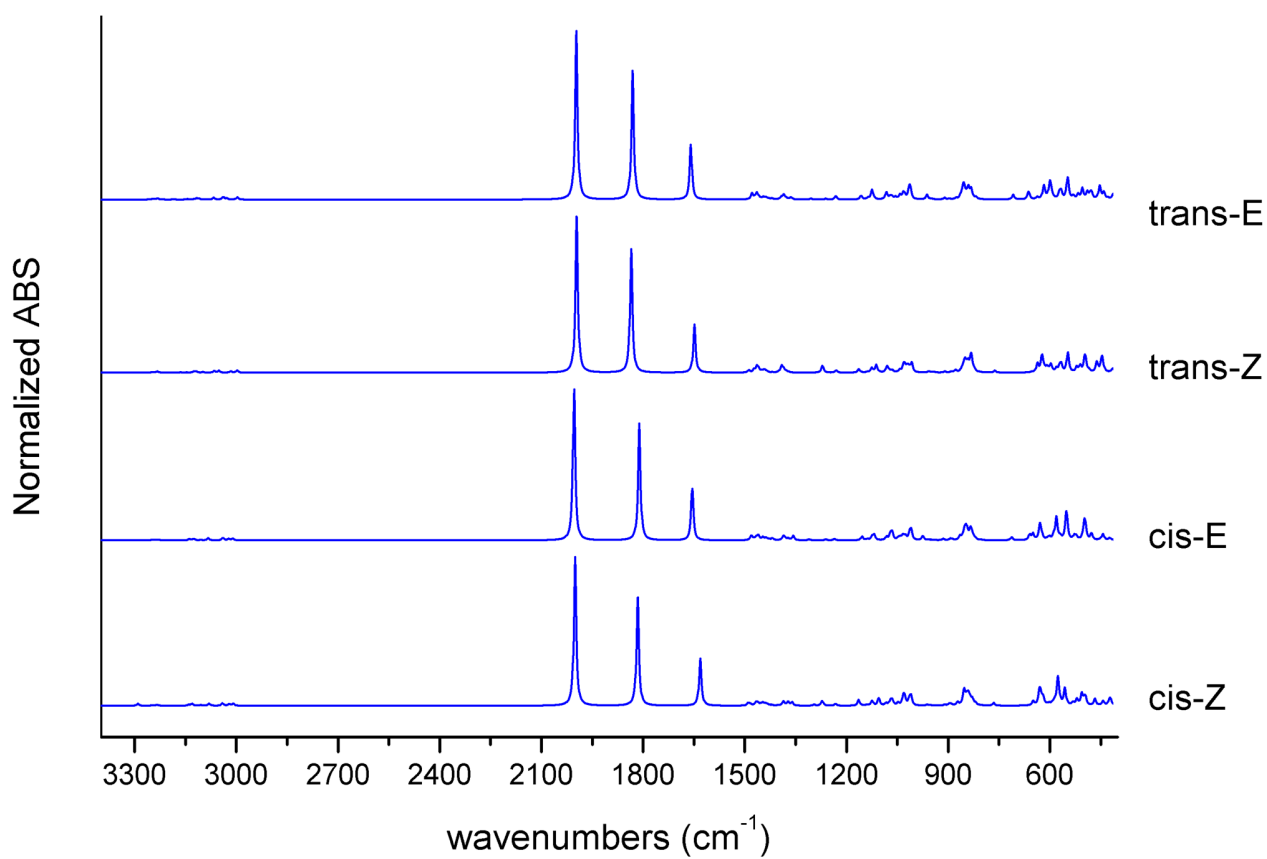


Figure S45. Simulated ^1H NMR spectra of *cis-E-5c* (top) and *cis-Z-5c* (bottom); experimental spectrum of **5c** (*cis-E/Z* ratio = 7.6) in acetone- d_6 .

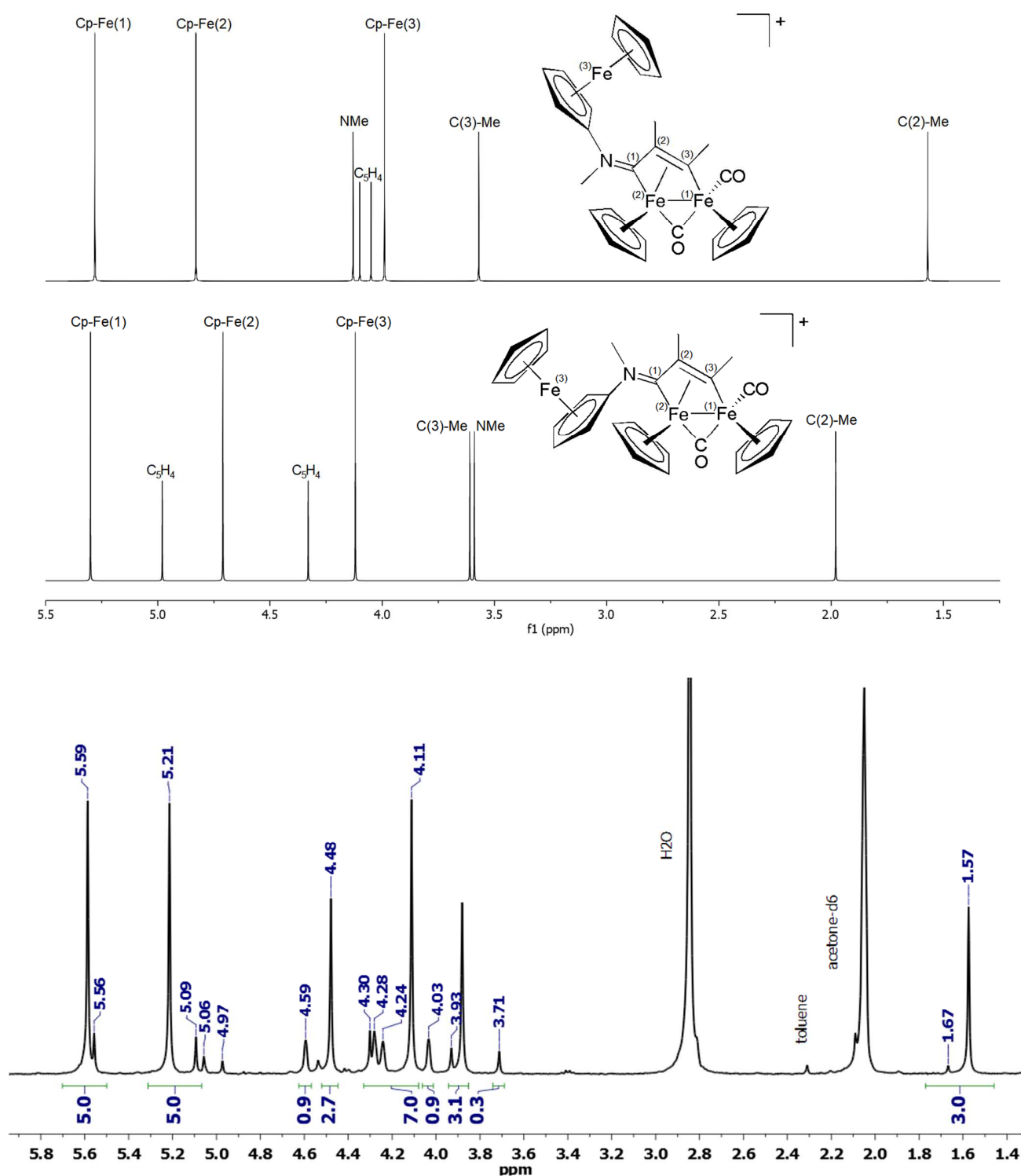
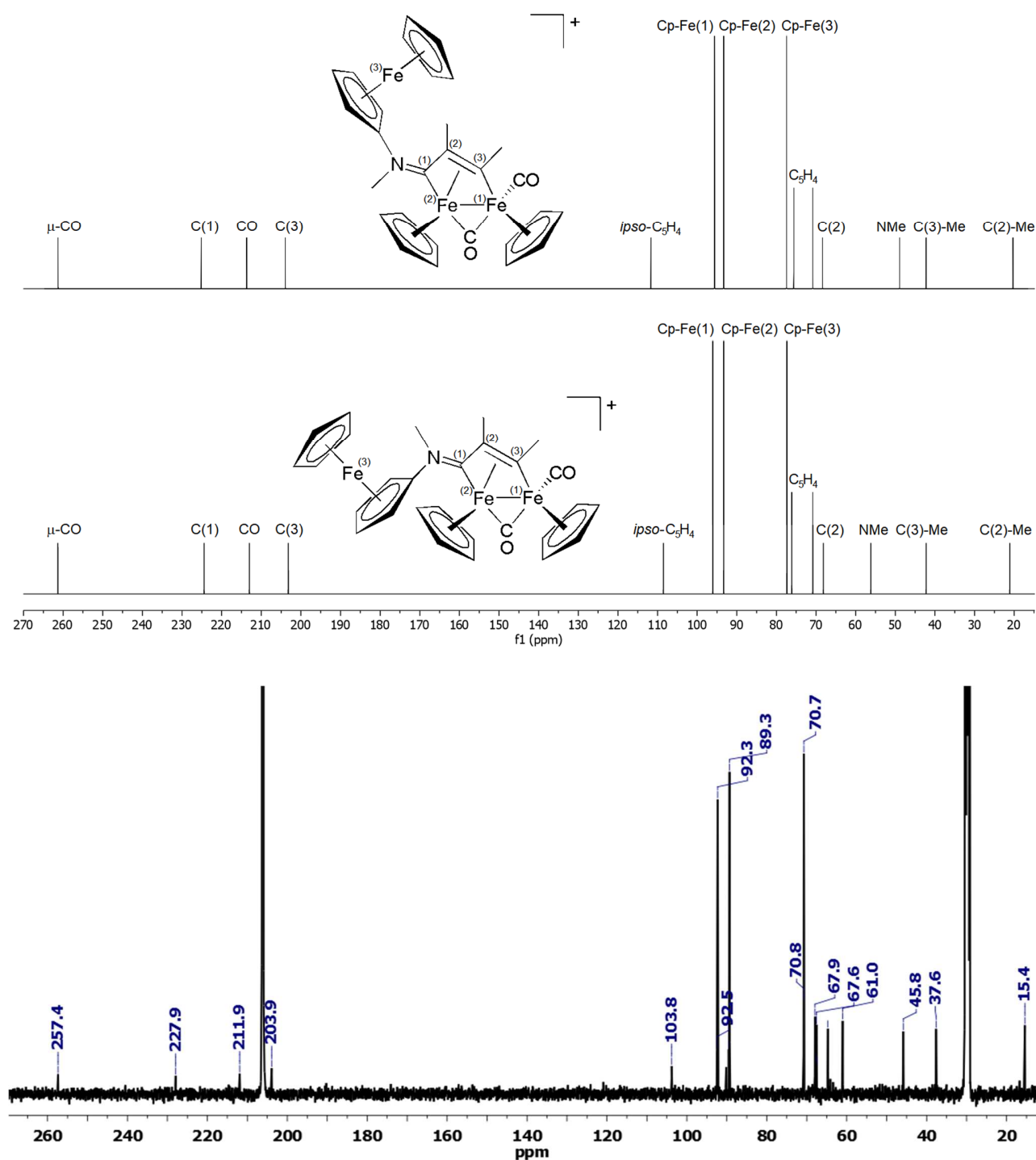


Figure S46. Simulated $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of *cis-E-5c* (top) and *cis-Z-5c* (middle); experimental spectrum of **4c** (*cis-E/Z* ratio = 7.6) in acetone- d_6 .



Cyclic voltammetry

Figure S47. Cyclic voltammetry of **5b** recorded at a Pt electrode in 0.2 M $[\text{N}^n\text{Bu}_4]\text{PF}_6/\text{CH}_2\text{Cl}_2$ solution between -1.75 and $+1.15$ V (vs FeCp_2). Scan rate: 0.1 V/s.

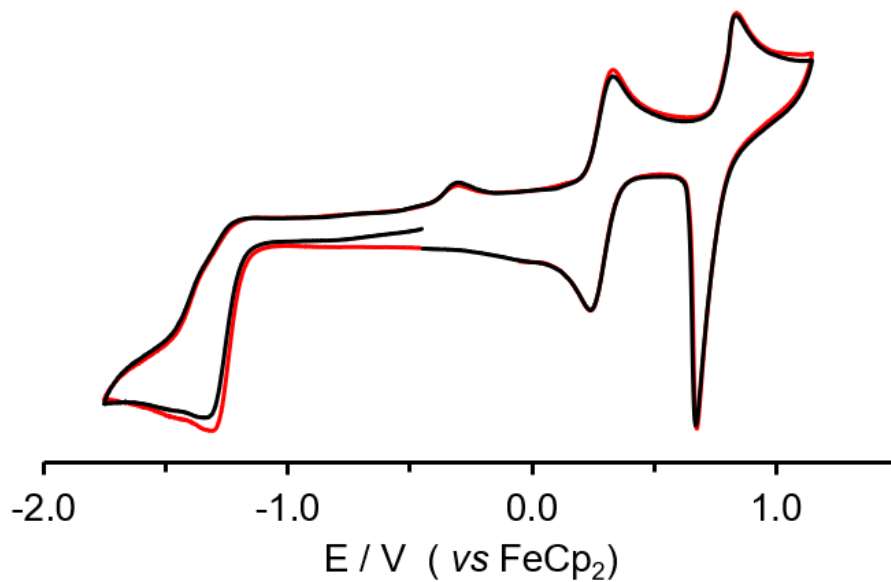
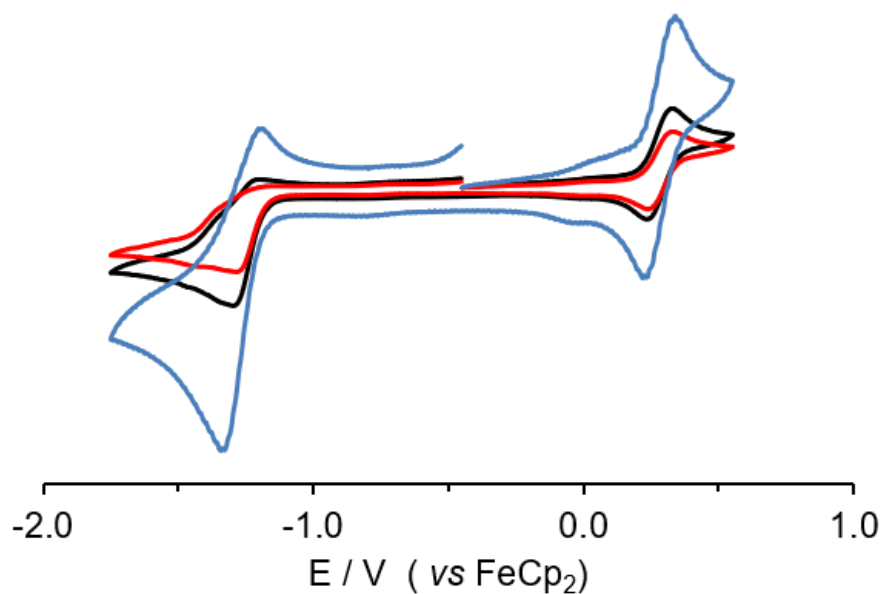


Figure S48. Cyclic voltammograms of **5b** recorded at a Pt electrode in 0.2 M $[\text{N}^n\text{Bu}_4]\text{PF}_6/\text{CH}_2\text{Cl}_2$ solution (vs Ag/AgCl) between -1.75 and $+0.55$ V. Scan rate: 0.1 V/s (red line), 0.2 V/s (black line), 1 V/s (blue line).



Cytotoxicity studies

Figure S49. Dose-response curves of distinct cell lines treated with the investigated triiron compounds.

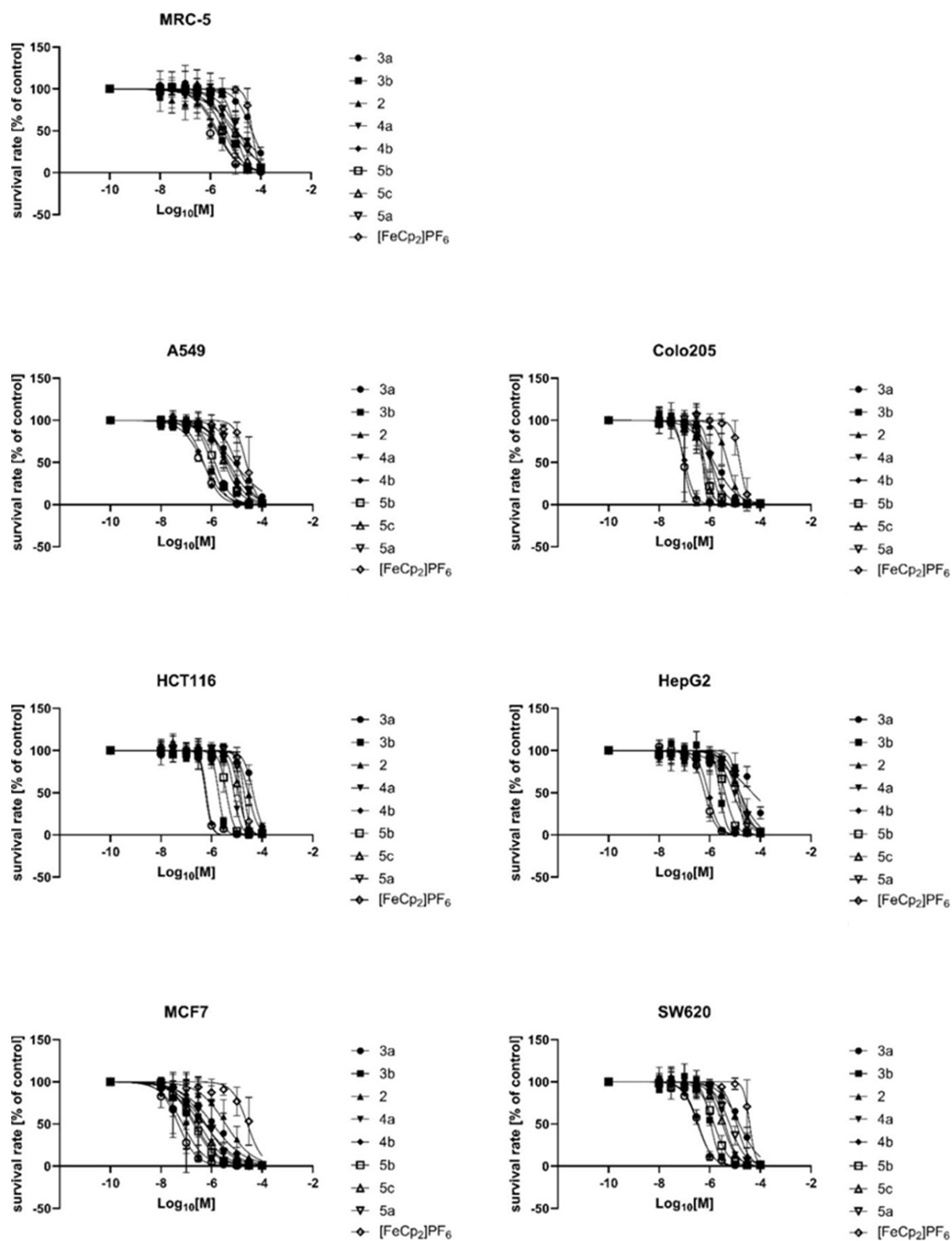


Figure S50. Correlation between $\text{Log } P_{ow}$ values (vertical axis) and median IC_{50} values (μM , horizontal axis) of investigated triiron complexes.

