

Supporting Information for

Testing Cuprophilicity with Tetrahedral Ruthenium Hydride Carbonyl Clusters

Cristiana Cesari,^{a*} Marco Bortoluzzi,^b Cristina Femoni,^a Maria Carmela Iapalucci,^a and Stefano Zacchini^{a*}

^a Dipartimento di Chimica Industriale "Toso Montanari", Università di Bologna, Via P. Gobetti 85 - 40129 Bologna, Italy. E-mail: cristiana.cesari2@unibo.it, stefano.zacchini@unibo.it

^b Dipartimento di Scienze Molecolari e Nanosistemi, Ca' Foscari University of Venice, Via Torino 155 – 30175 Mestre (Ve), Italy.

Page/s

IR spectra	S2-S8
¹ H NMR spectra	S9-S13
Crystallographic tables	S14-S22

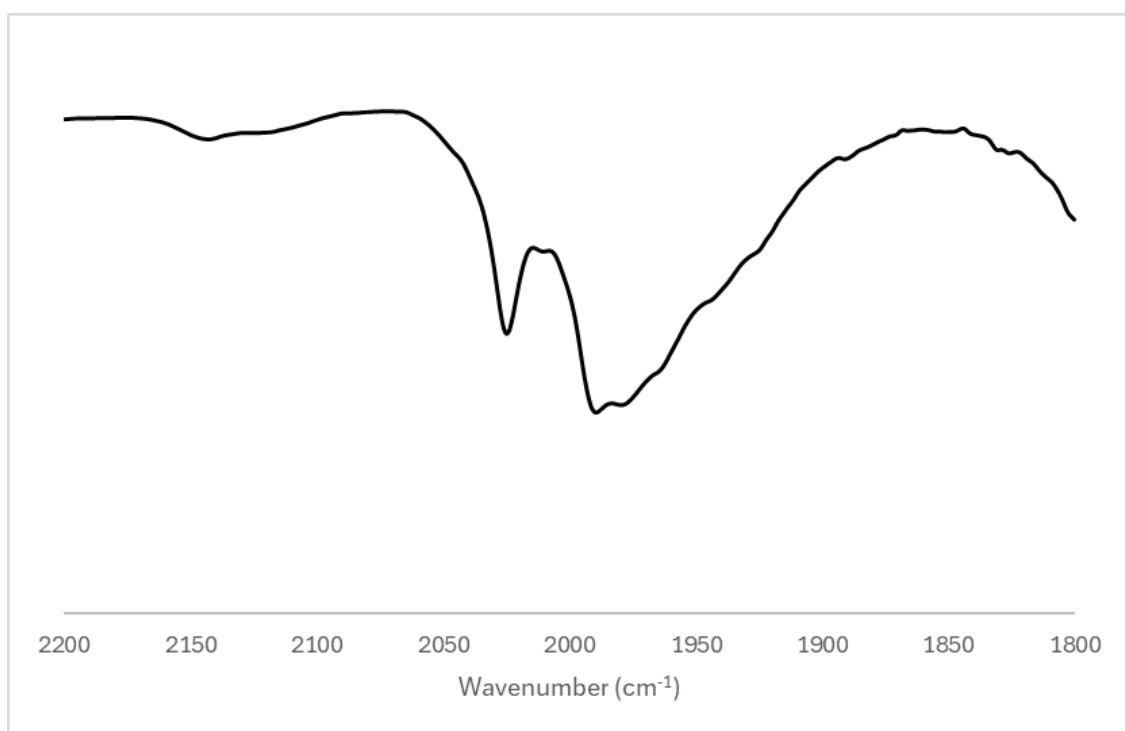


Figure S1. IR spectrum in the ν_{CO} region of $[\text{NEt}_4]_3[\text{HRu}_4(\text{CO})_{12}(\text{CuBr})_3]$ (**2**) in acetone.

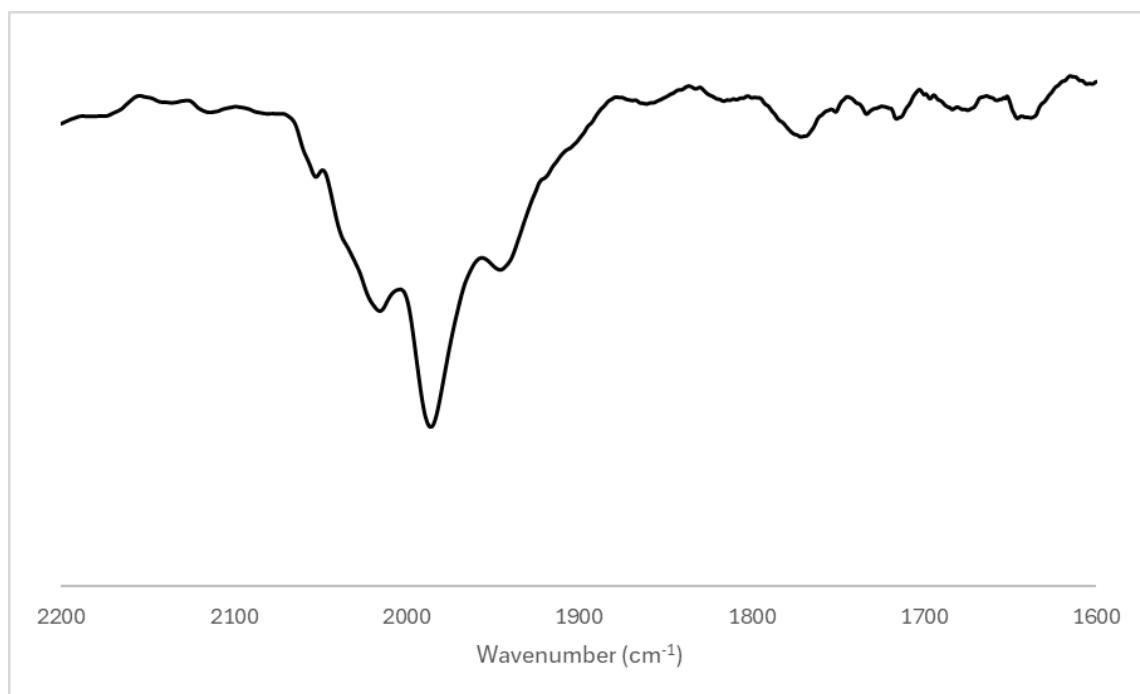


Figure S2. IR spectrum in the ν_{CO} region of $[\text{NEt}_4]_2[\text{H}_2\text{Ru}_8(\text{CO})_{24}\text{Cu}_7\text{Br}_3]$ (**3**) in CH_2Cl_2 .



Figure S3. IR spectrum in the ν_{CO} region of $[\text{NEt}_4]_3[\text{HRu}_8(\text{CO})_{24}\text{Cu}_6\text{Br}_2]$ (**4**) in CH_2Cl_2 .

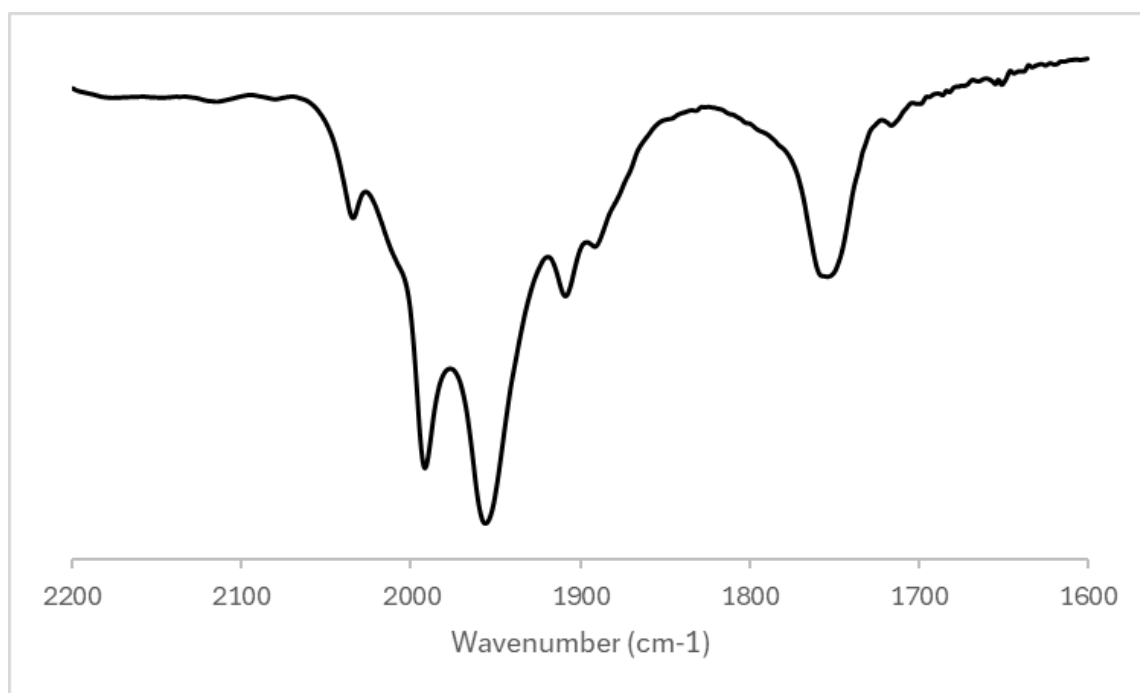


Figure S4. IR spectrum in the ν_{CO} region of $[\text{NEt}_4]_3[\text{HRu}_8(\text{CO})_{24}\text{Cu}_6\text{Br}_2]$ (**4**) in CH_3CN .

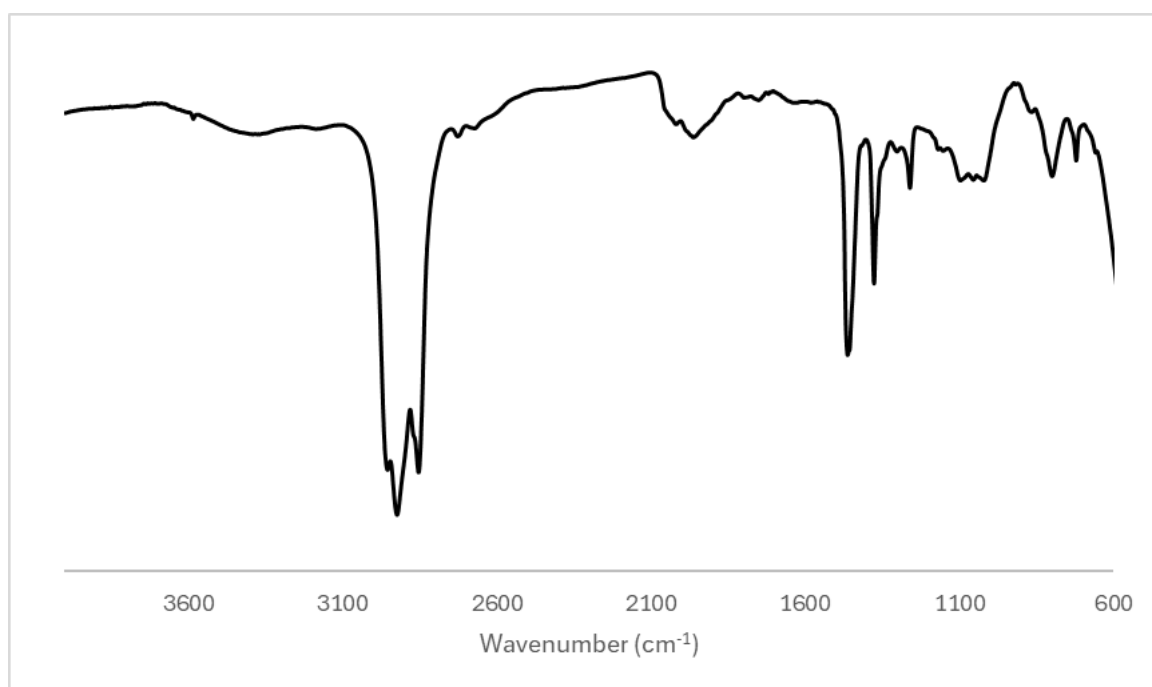


Figure S5. IR spectrum of crystals of $[\text{NEt}_4]_3[\text{HRu}_8(\text{CO})_{24}\text{Cu}_6\text{Br}_2]$ (**4**) in nujol mull.

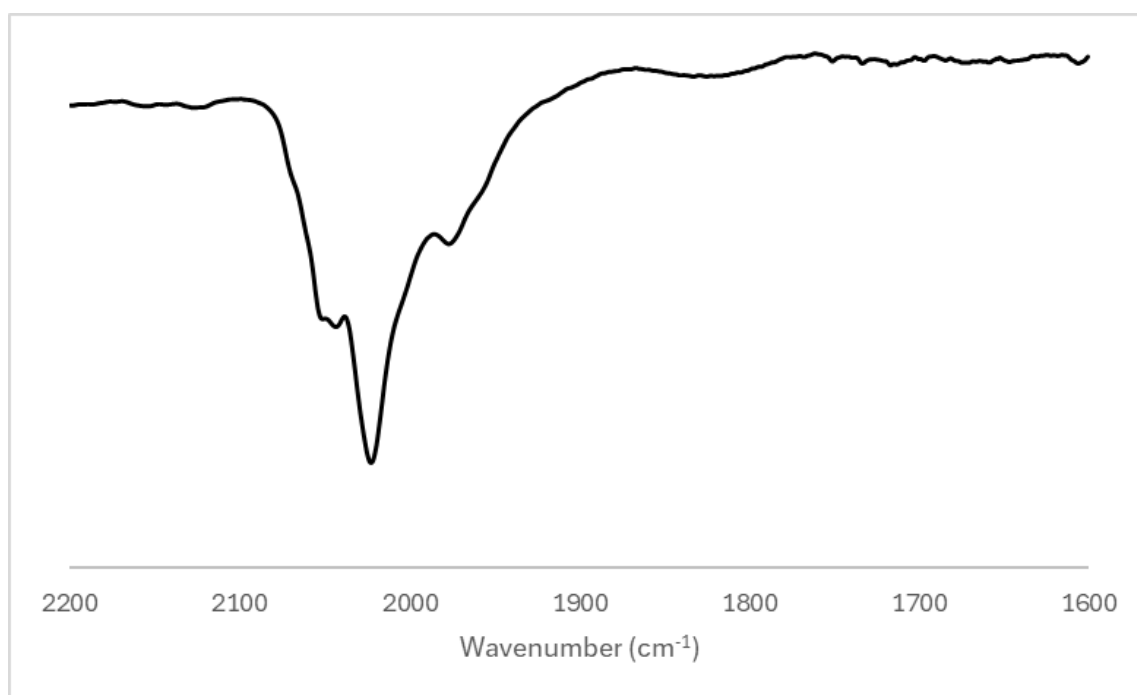


Figure S6. IR spectrum in the ν_{CO} region of $[\text{Ru}_5(\text{CO})_{15}(\text{CuMeCN})_2]$ (**5**) in CH_3CN .

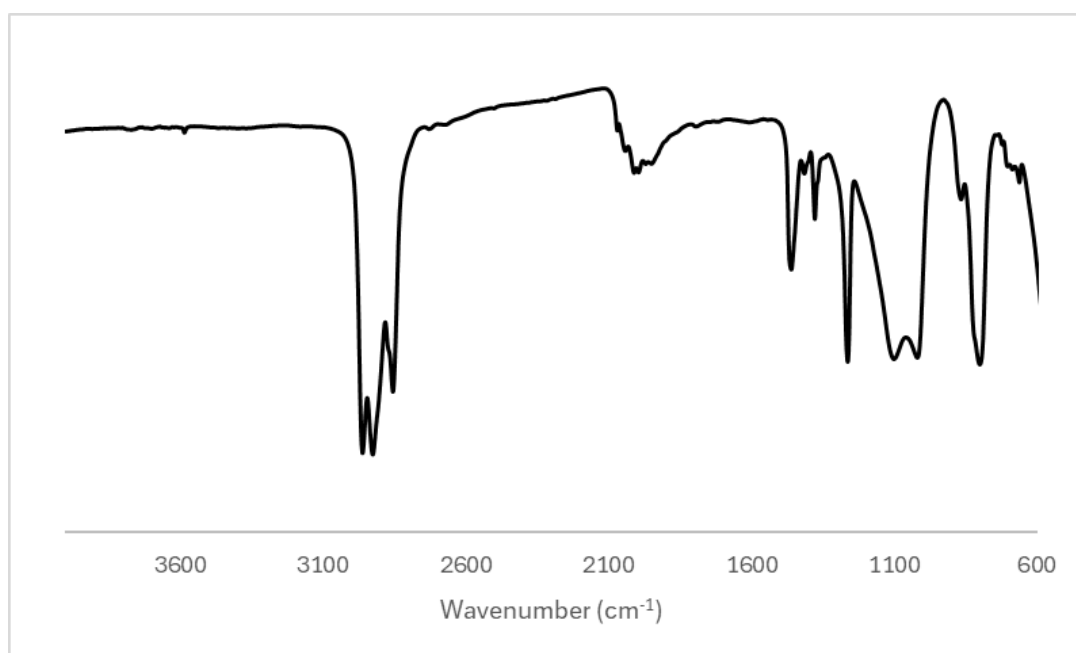


Figure S7. IR spectrum of crystals of $[\text{Ru}_5(\text{CO})_{15}(\text{CuMeCN})_2]$ (**5**) in nujol mull.

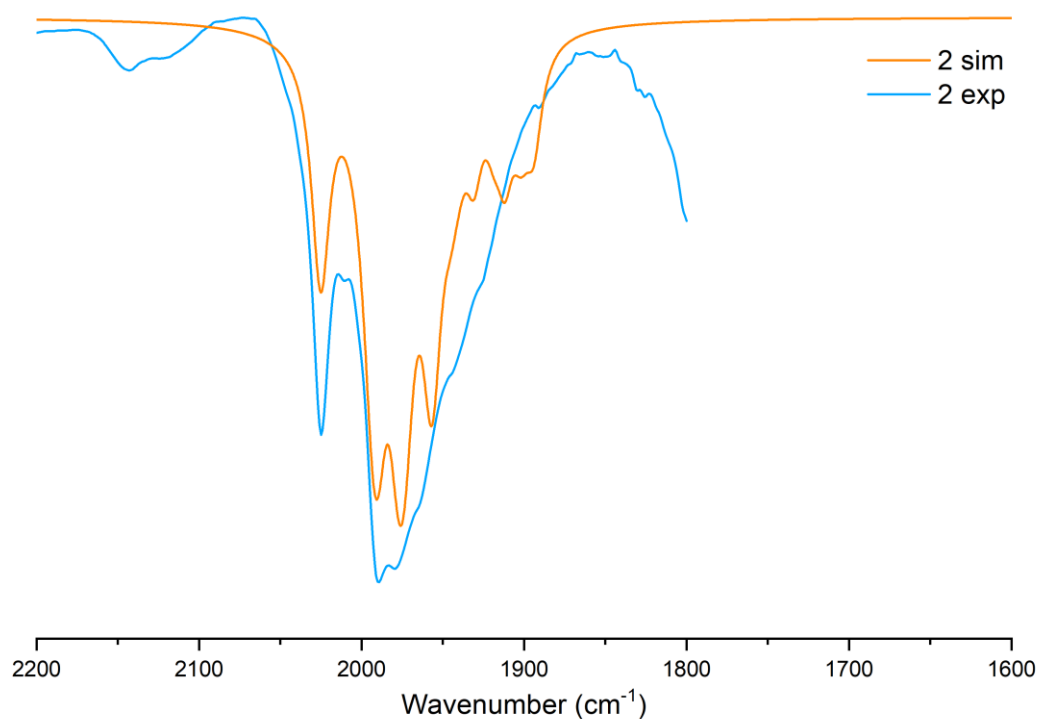


Figure S8. Superimposition of normalized experimental (light blue) and simulated (orange) IR spectra of $[\text{NEt}_4]_3[\text{HRu}_4(\text{CO})_{12}(\text{CuBr})_3]$ (**2**). Lorentzian broadening functions, $\text{FWHM} = 12 \text{ cm}^{-1}$.

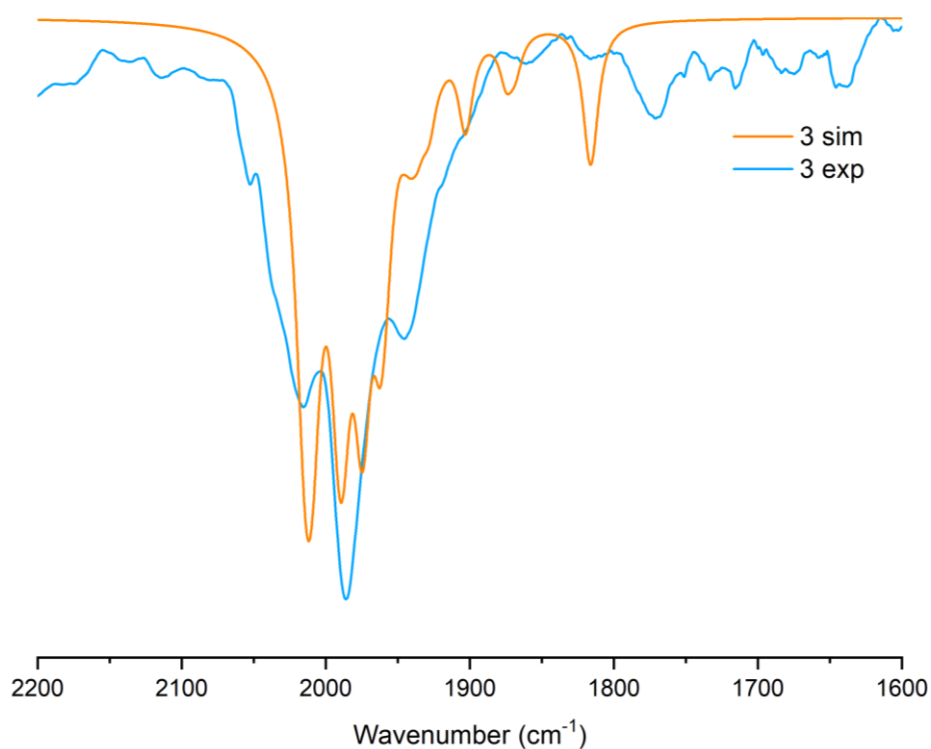


Figure S9. Superimposition of normalized experimental (light blue) and simulated (orange) IR spectra of $[\text{NEt}_4]_2[\text{H}_2\text{Ru}_8(\text{CO})_{24}\text{Cu}_7\text{Br}_3]$ (**3**). Lorentzian broadening functions, FWHM = 12 cm^{-1} .

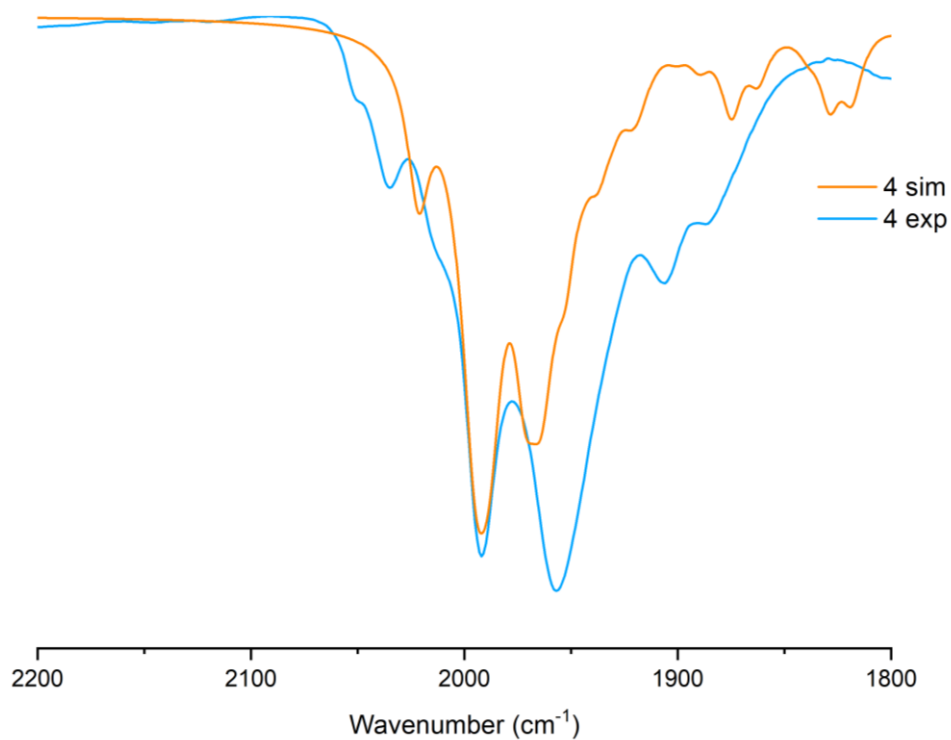


Figure S10. Superimposition of normalized experimental (light blue) and simulated (orange) IR spectra of $[\text{NEt}_4]_3[\text{HRu}_8(\text{CO})_{24}\text{Cu}_6\text{Br}_2]$ (**4**). Lorentzian broadening functions, FWHM = 12 cm^{-1} .

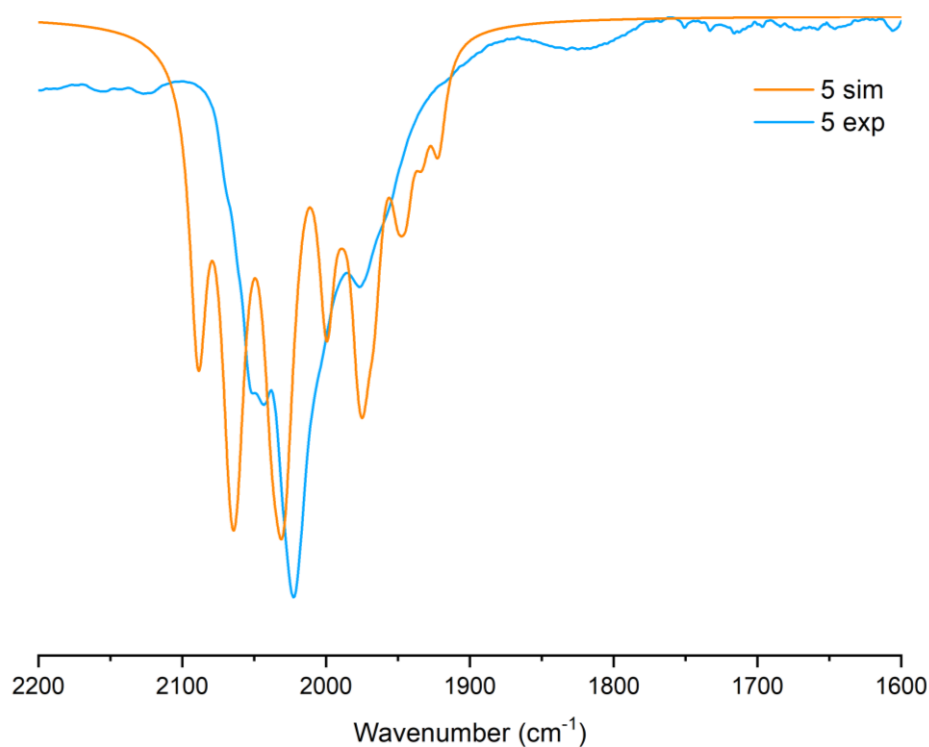


Figure S11. Superimposition of normalized experimental (light blue) and simulated (orange) IR spectra of $[\text{Ru}_5(\text{CO})_{15}(\text{CuMeCN})_2]$ (**5**). Lorentzian broadening functions, FWHM = 12 cm^{-1} .

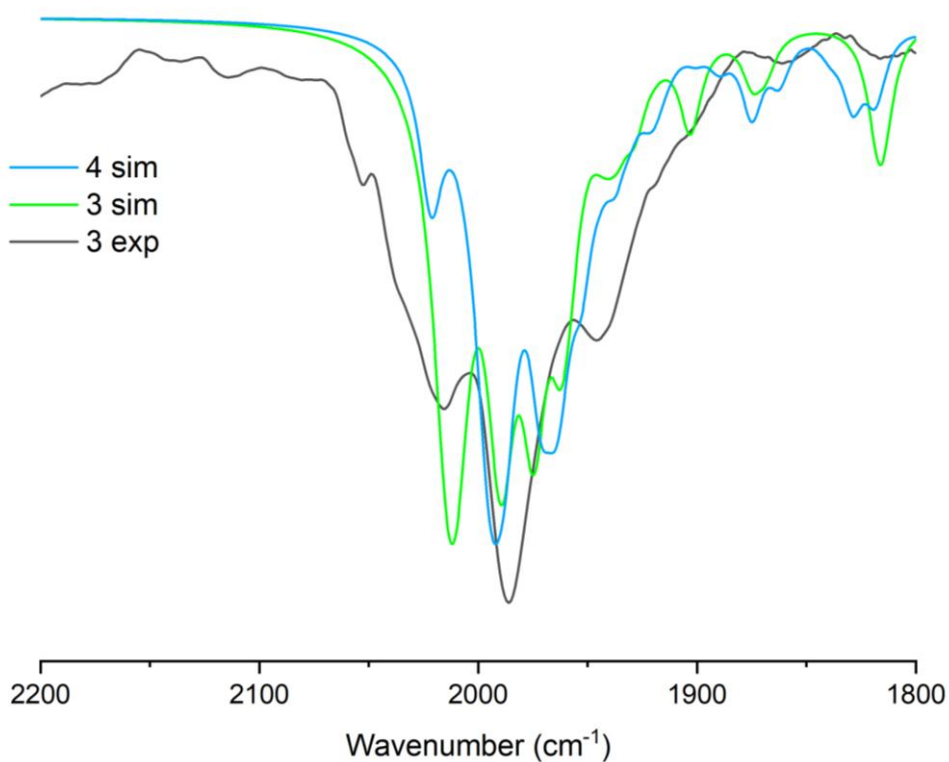


Figure S12. Overlay of normalized experimental IR spectrum (black) and simulated IR spectrum (green) of $[\text{NEt}_4]_2[\text{H}_2\text{Ru}_8(\text{CO})_{24}\text{Cu}_7\text{Br}_3]$ (**3**) along with the simulated IR spectrum of $[\text{NEt}_4]_3[\text{HRu}_8(\text{CO})_{24}\text{Cu}_6\text{Br}_2]$ (**4**). Lorentzian broadening functions, FWHM = 12 cm^{-1} .

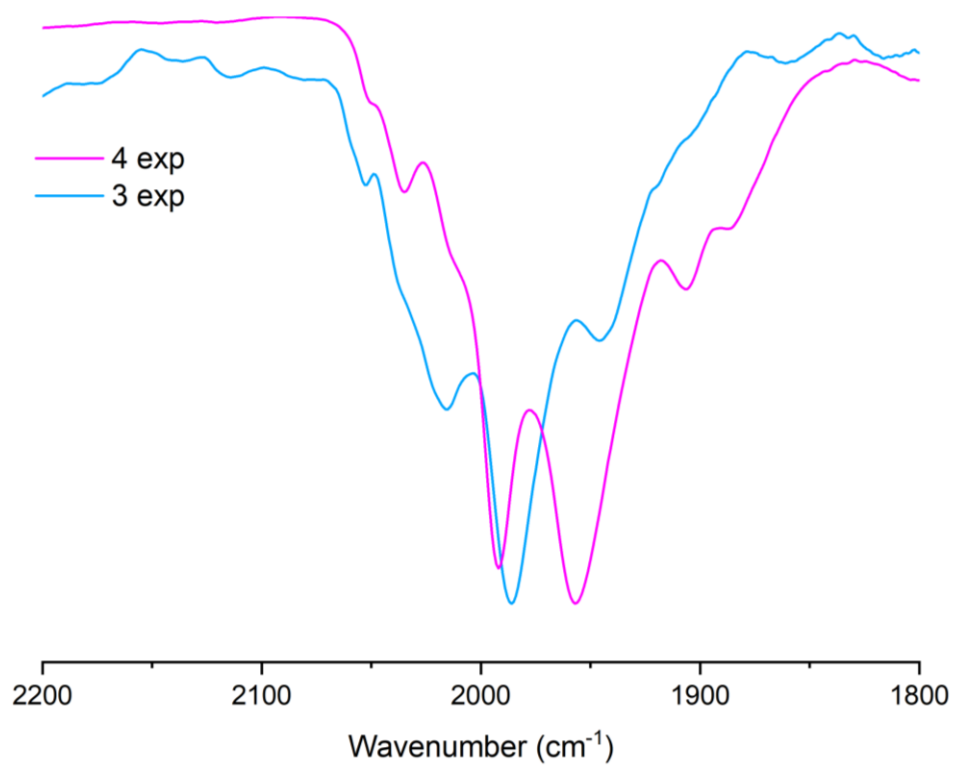


Figure S13. Comparison of normalized experimental IR spectra of $[\text{NEt}_4]_2[\text{H}_2\text{Ru}_8(\text{CO})_{24}\text{Cu}_7\text{Br}_3]$ (**3**) (light blue) and $[\text{NEt}_4]_3[\text{HRu}_8(\text{CO})_{24}\text{Cu}_6\text{Br}_2]$ (**4**) (purple).

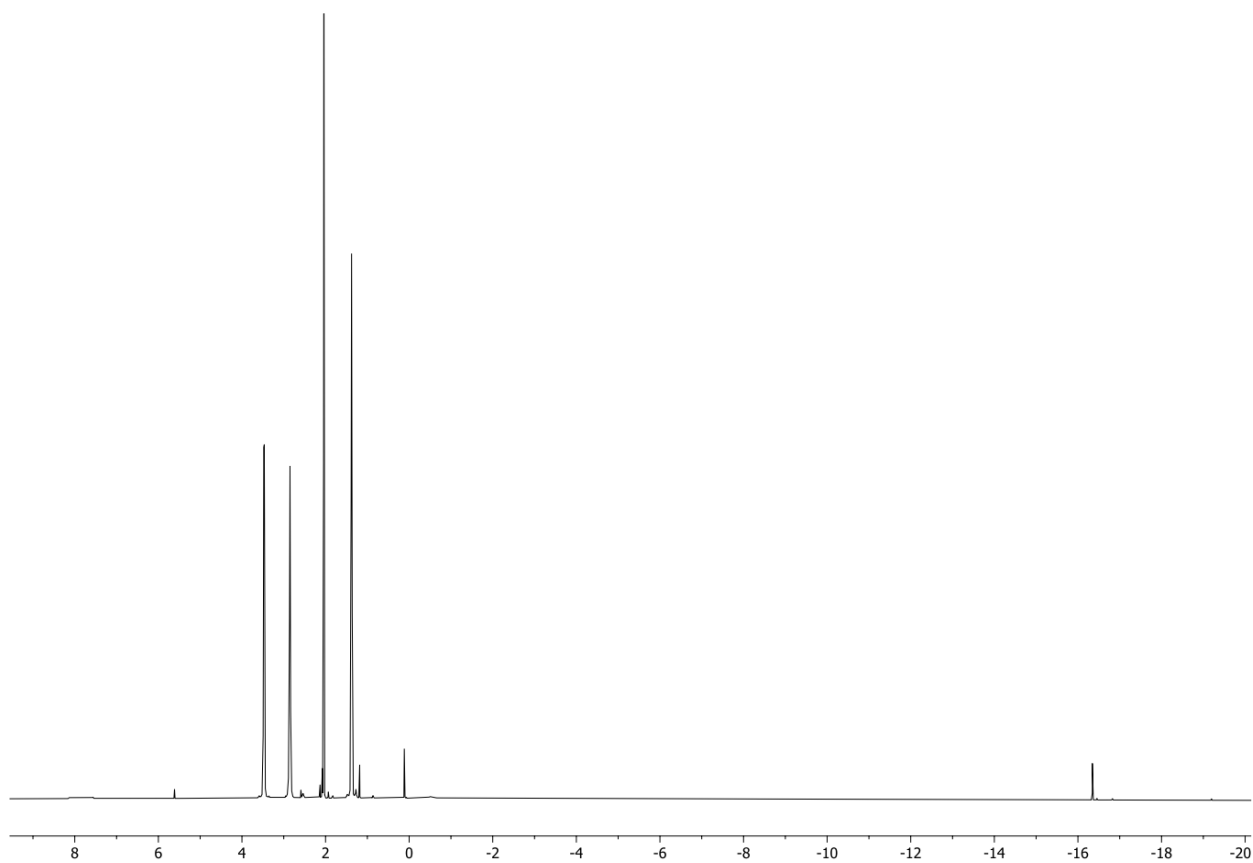


Figure S14. ^1H NMR spectrum of crystals of $[\text{NEt}_4]_3[\text{HRu}_4(\text{CO})_{12}(\text{CuBr})_3]$ (**2**) in acetone- d_6 at 298 K.

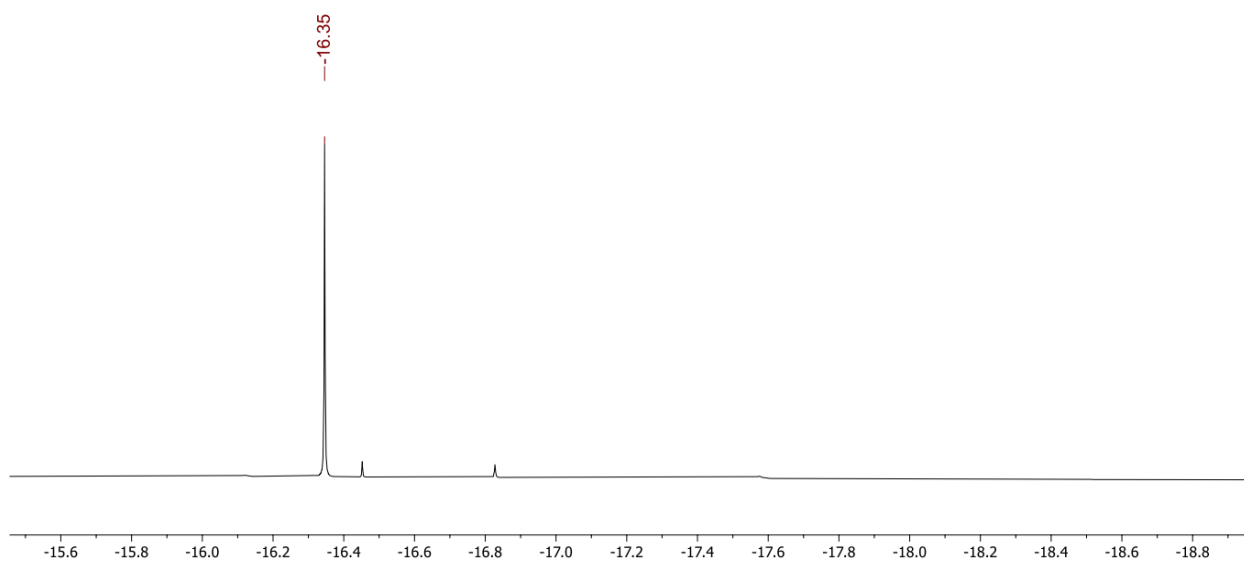


Figure S15. Hydride region of the ^1H NMR spectrum of crystals of $[\text{NEt}_4]_3[\text{HRu}_4(\text{CO})_{12}(\text{CuBr})_3]$ (**2**) in acetone- d_6 at 298 K.

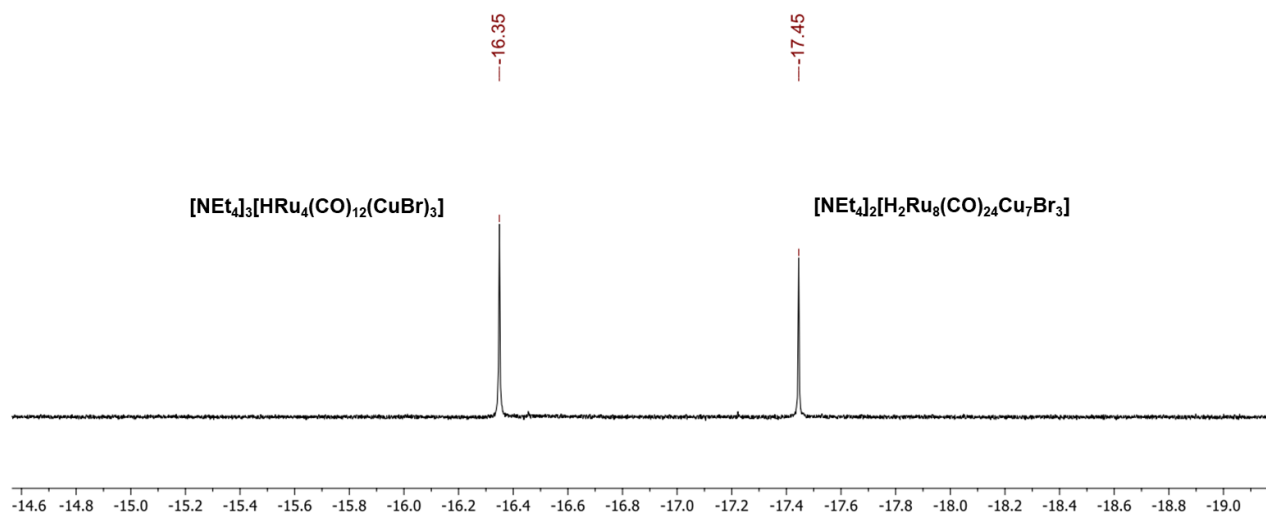


Figure S16. Hydride region of the ^1H NMR spectrum of a mixture of $[\text{NEt}_4]_3[\text{HRu}_4(\text{CO})_{12}(\text{CuBr})_3]$ (**2**) and $[\text{NEt}_4]_2[\text{H}_2\text{Ru}_8(\text{CO})_{24}\text{Cu}_7\text{Br}_3]$ (**3**) in acetone- d_6 at 298 K.

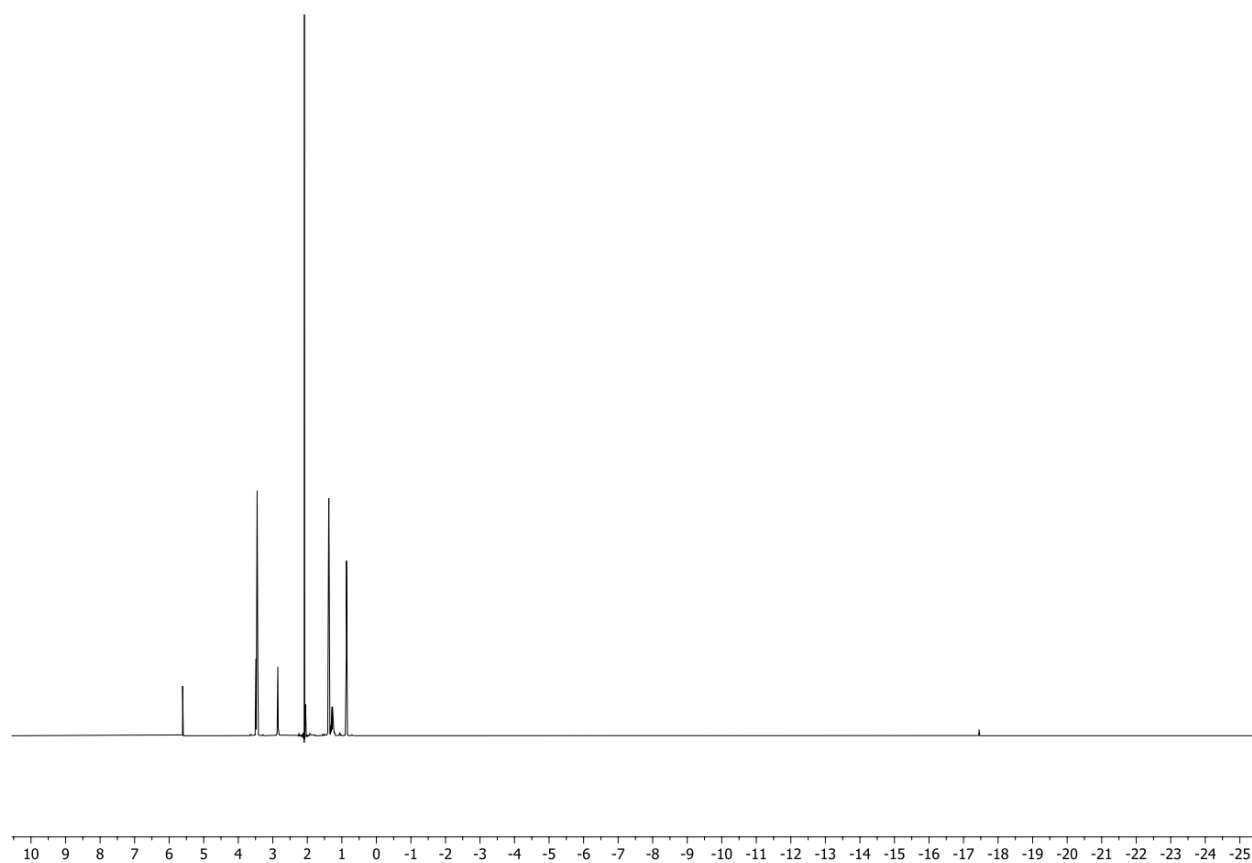


Figure S17. ^1H NMR spectrum of crystals of $[\text{NEt}_4]_2[\text{H}_2\text{Ru}_8(\text{CO})_{24}\text{Cu}_7\text{Br}_3]$ (**3**) in acetone- d_6 at 298 K.

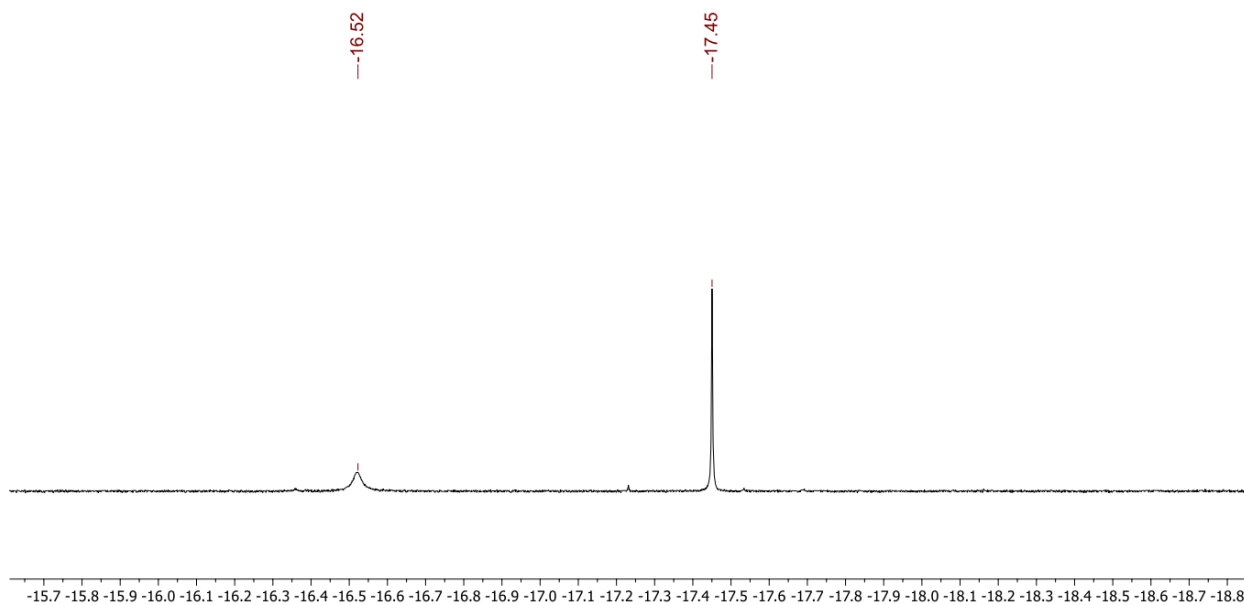


Figure S18. Hydride region of the ^1H NMR spectrum of crystals of $[\text{NEt}_4]_2[\text{H}_2\text{Ru}_8(\text{CO})_{24}\text{Cu}_7\text{Br}_3]$ (**3**) in acetone- d_6 at 298 K.

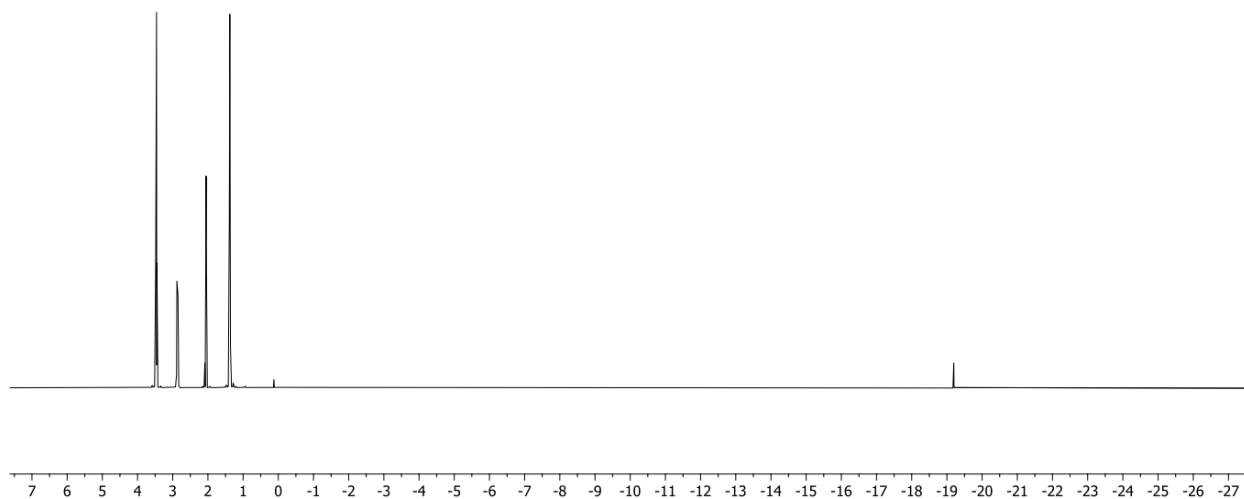


Figure S19. ^1H NMR spectrum of crystals of $[\text{NEt}_4]_3[\text{HRu}_8(\text{CO})_{24}\text{Cu}_6\text{Br}_2]$ (**4**) in acetone- d_6 at 298 K.

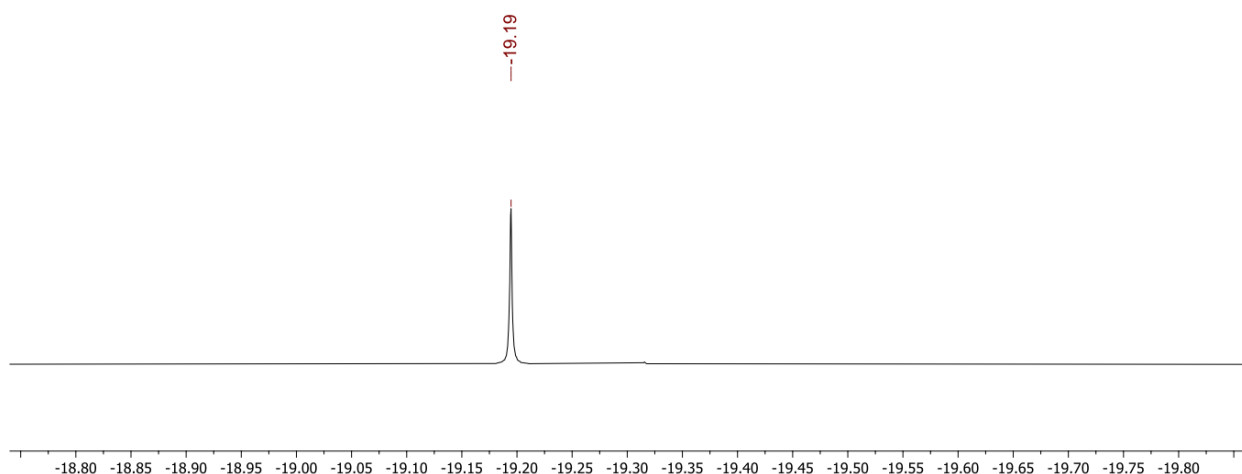


Figure S20. Hydride region of the ^1H NMR spectrum of crystals of $[\text{NEt}_4]_3[\text{HRu}_8(\text{CO})_{24}\text{Cu}_6\text{Br}_2]$ (**4**) in acetone- d_6 at 298 K.

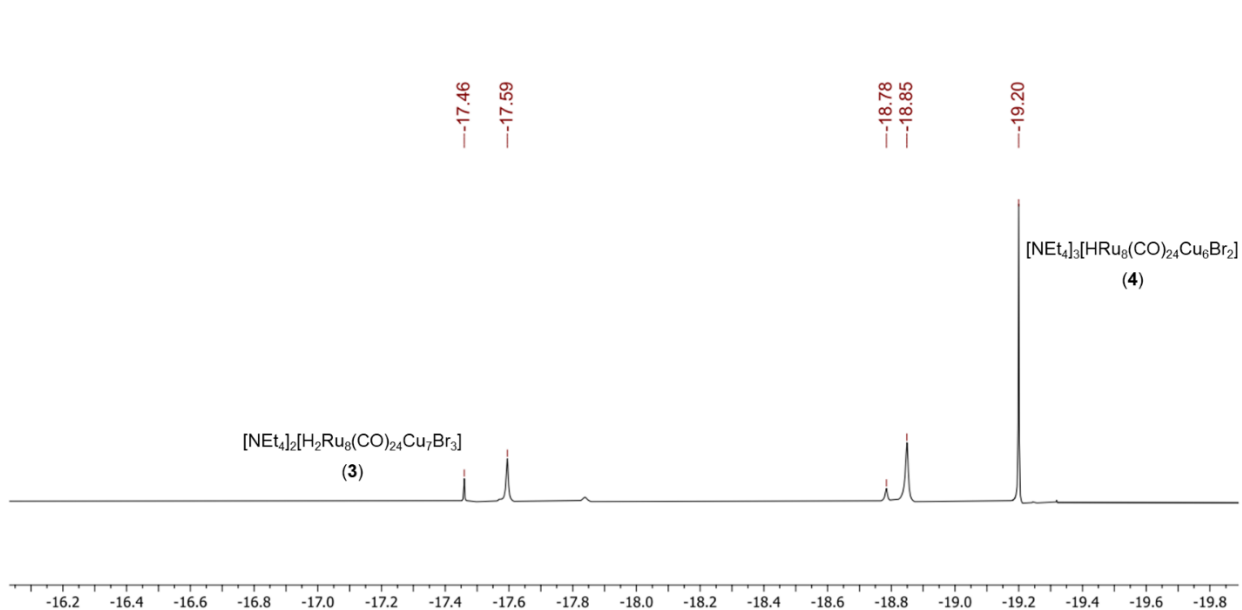


Figure S21. ^1H NMR spectrum, in acetone- d_6 at 298 K, of the crude product of the reaction after the addition of 1 equivalent of $[\text{Cu}(\text{CH}_3\text{CN})_4][\text{BF}_4]$ to $[\text{NEt}_4]_3[\text{HRu}_4(\text{CO})_{12}]$ (**1**).

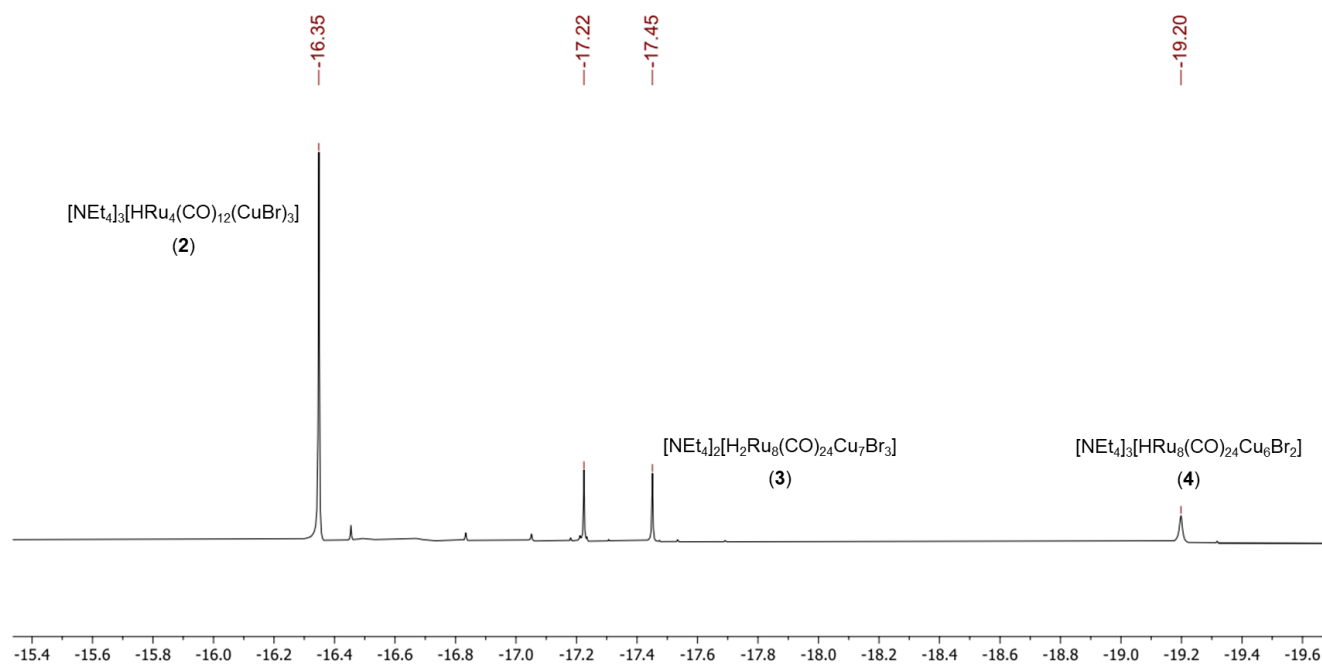


Figure S22. ^1H NMR spectrum in acetone- d_6 at 298 K of the mixture extracted in acetone at the end of the reaction of Figure S21.

Table S1. Main bond distances (Å) and angles (°) of [HRu₄(CO)₁₂(CuBr)₃]³⁻ (**2**).

Ru1-Ru2	2.8008(5) (1.13)	Ru1-Ru3	2.7873(5) (1.12)
Ru1-Ru4	2.7901(5) (1.12)	Ru2-Ru3	2.9067(5) (1.17)
Ru2-Ru4	2.9537(5) (1.19)	Ru3-Ru4	2.9430(5) (1.19)
Cu1-Ru2	2.6380(7) (1.09)	Cu1-Ru3	2.7808(7) (1.15)
Cu1-Ru4	2.6854(7) (1.11)	Cu2-Ru2	2.6041(7) (1.08)
Cu2-Ru4	2.6211(7) (1.09)	Cu3-Ru2	2.6181(7) (1.08)
Cu3-Ru3	2.6198(7) (1.09)	Cu1-Cu2	2.6380(8) (1.12)
Cu1-Cu3	2.5809(8) (1.10)	Cu1-Br1	2.3092(7) (1.00)
Cu2-Br2	2.3018(9) (0.99)	Cu3-Br3	2.2938(7) (0.99)
H1-Ru3	1.83(3) (1.26)	H1-Ru4	1.83(3) (1.26)
H1-Cu1	1.85(3) (1.32)	C6-Ru2	1.913(5) (0.95)
C6-Cu1	2.558(5) (1.32)	C6-Cu2	2.589(5) (1.33)
C6-Cu3	2.603(5) (1.34)	C9-Ru3	1.918(5) (0.95)
C9-Cu3	2.593(5) (1.33)	C12-Ru4	1.915(5) (0.95)
C12-Cu2	2.657(5) (1.37)	Ru2-C6-O6	170.1(4)
Ru3-C9-O9	172.0(4)	Ru4-C12-O12	171.7(5)
$\alpha_{\text{CO}6}$	0.337(6) [0.373(6)]	$\alpha_{\text{CO}9}$	0.352(14) [0.387(14)]
$\alpha_{\text{CO}12}$	0.388(6) [0.423(6)]		

Cotton's FSRs are reported in parentheses. Pauling's atomic radii are employed: Cu 1.173 Å, Ru 1.241 Å, Br 1.142 Å, C 0.771 Å, H 0.230 Å.

α corrected for the different atom radii of Cu (1.173) and Ru (1.241) in parentheses (covalent radii from Pauling).

Table S2. Main bond distances (Å) and angles (°) of [H₂Ru₈(CO)₂₄Cu₇Br₃]²⁻ (**3**).

Ru1-Ru2	2.847(2) (1.15)	Ru5-Ru8	2.844(2) (1.15)
Ru1-Ru3	2.861(2) (1.15)	Ru5-Ru7	2.858(2) (1.15)
Ru1-Ru4	2.808(2) (1.13)	Ru5-Ru6	2.804(2) (1.13)
Ru2-Ru3	2.889(2) (1.16)	Ru7-Ru8	2.885(2) (1.16)
Ru2-Ru4	2.976(2) (1.20)	Ru6-Ru8	2.960(2) (1.19)
Ru3-Ru4	2.897(2) (1.16)	Ru6-Ru7	2.922(2) (1.18)
Cu2-Ru2	2.674(3) (1.11)	Cu6-Ru8	2.700(3) (1.12)
Cu2-Ru3	2.781(3) (1.15)	Cu6-Ru7	2.796(3) (1.16)
Cu2-Ru4	2.706(3) (1.12)	Cu6-Ru6	2.711(3) (1.12)
Cu1-Ru2	2.638(3) (1.09)	Cu5-Ru8	2.640(3) (1.09)
Cu1-Ru3	2.668(3) (1.11)	Cu5-Ru7	2.650(3) (1.10)
Cu3-Ru3	2.653(3) (1.10)	Cu7-Ru7	2.652(3) (1.10)
Cu3-Ru4	2.620(3) (1.09)	Cu7-Ru6	2.627(3) (1.09)
Cu4-Ru3	2.584(3) (1.07)	Cu4-Ru7	2.591(3) (1.07)
Cu1-Cu2	2.562(4) (1.09)	Cu5-Cu6	2.560(3) (1.09)
Cu2-Cu3	2.569(3) (1.10)	Cu6-Cu7	2.588(3) (1.10)
Cu4-Cu1	2.580(4) (1.10)	Cu4-Cu5	2.568(4) (1.09)
Cu4-Cu2	2.572(3) (1.10)	Cu4-Cu6	2.551(3) (1.09)
Cu4-Cu3	2.629(4) (1.12)	Cu4-Cu7	2.646(4) (1.13)
Cu1-Cu7	2.722(4)	Cu2-Cu6	2.786(3)

	(1.16)		(1.19)
Cu3-Cu5	2.721(4) (1.16)		
Cu1-Br1	2.353(3) (1.02)	Cu7-Br1	2.357(3) (1.02)
Cu2-Br2	2.370(3) (1.02)	Cu6-Br2	2.381(3) (1.03)
Cu3-Br3	2.364(3) (1.02)	Cu5-Br3	2.358(3) (1.02)
H2-Ru2	1.70(2) (1.16)	H1-Ru8	1.70(2) (1.16)
H2-Ru4	1.71(2) (1.16)	H1-Ru6	1.70(2) (1.16)
H2-Cu2	1.60(2) (1.14)	H1-Cu6	1.60(2) (1.16)
C6-Ru2	1.99(2) (0.99)	C24-Ru8	1.96(2) (0.97)
C6-Cu1	2.49(2) (1.28)	C24-Cu5	2.47(2) (1.27)
C10-Ru4	1.91(2) (0.95)	C18-Ru6	1.94(2) (0.96)
C10-Cu3	2.47(2) (1.27)	C18-Cu7	2.54(2) (1.31)
C7-Ru3	1.90(2) (0.94)	C20-Ru7	1.91(3) (0.95)
C7-Cu1	2.53(2) (1.30)	C20-Cu5	2.50(3) (1.29)
C8-Ru3	1.85(2) (0.92)	C21-Ru7	1.91(2) (0.95)
C8-Cu3	2.35(2) (1.21)	C21-Cu7	2.44(2) (1.26)
C8-Cu4	2.35(2) (1.21)	C21-Cu4	2.33(2) (1.20)
C9-Ru1	2.28(2) (1.14)	C19-Ru5	2.25(2) (1.12)
C9-Ru3	1.97(2) (0.98)	C19-Ru7	1.97(2) (0.98)
Ru2-C6-O6	165.2(19)	Ru8-C24-O24	169(2)
Ru4-C10-O10	169(2)	Ru6-C18-O18	171(2)
Ru3-C7-O7	172(2)	Ru7-C20-O20	171(2)

Ru3-C8-O8	167(2)	Ru7-C21-O21	166.1(19)
Ru3-C9-O9	145(2)	Ru5-C19-O19	130.5(18)
α_{CO6}	0.25(2) [0.28(2)]	α_{CO24}	0.26(2) [0.29(2)]
α_{CO10}	0.29(2) [0.33(2)]	α_{CO18}	0.31(2) [0.34(2)]
α_{CO7}	0.33(2) [0.37(2)]	α_{CO20}	0.31(2) [0.34(2)]
α_{CO8}	0.27(2) [0.31(2)]	α_{CO21}	0.28(2) [0.31(2)]
α_{CO9}	0.16(2)	α_{CO19}	0.14(2)

Cotton's FSRs are reported in parentheses. Pauling's atomic radii are employed: Cu 1.173 Å, Ru 1.241 Å, Br 1.142 Å, C 0.771 Å, H 0.230 Å.

α corrected for the different atom radii of Cu (1.173) and Ru (1.241) in parentheses (covalent radii from Pauling).

Table S3. Main bond distances (Å) and angles (°) of [HRu₈(CO)₂₄Cu₆Br₂]³⁻ (**4**).

Ru1-Ru2	2.9130(10) (1.17)	Ru1-Ru3	2.8804(11) (1.16)
Ru1-Ru4	2.8341(11) (1.14)	Ru2-Ru3	2.9998(11) (1.21)
Ru2-Ru4	2.8075(11) (1.13)	Ru3-Ru4	2.8519(11) (1.15)
Ru5-Ru6	3.0584(11) (1.23)	Ru5-Ru7	2.8441(11) (1.15)
Ru5-Ru8	2.7983(11) (1.13)	Ru6-Ru7	2.8228(11) (1.14)
Ru6-Ru8	2.9352(11) (1.18)	Ru7-Ru8	2.8146(11) (1.13)
Cu1-Ru1	2.6896(14) (1.11)	Cu1-Ru3	2.6652(14) (1.10)
Cu2-Ru1	2.8642(14) (1.19)	Cu2-Ru2	2.7221(14) (1.13)
Cu2-Ru3	2.7383(14) (1.13)	Cu2-Ru5	2.6337(14) (1.09)
Cu3-Ru1	2.6627(14) (1.10)	Cu3-Ru2	2.5947(14) (1.07)
Cu4-Ru1	2.6461(14) (1.10)	Cu4-Ru5	2.7710(14) (1.15)
Cu4-Ru6	2.6058(14)	Cu5-Ru5	2.6506(14)

	(1.08)		(1.10)
Cu5-Ru6	2.6403(14) (1.09)	Cu5-Ru8	2.6889(14) (1.11)
Cu6-Ru5	2.6077(13) (1.08)	Cu6-Ru6	2.7099(14) (1.12)
Cu6-Ru7	2.6457(14) (1.10)	Cu1-Cu2	2.5432(16) (1.10)
Cu1-Cu4	2.4669(16) (1.07)	Cu1-Cu6	2.6596(17) (1.13)
Cu2-Cu3	2.5712(17) (1.10)	Cu2-Cu4	2.4669(16) (1.05)
Cu3-Cu4	2.5590(16) (1.09)	Cu3-Cu5	2.7095(17) (1.15)
Cu4-Cu5	2.5762(16) (1.10)	Cu4-Cu6	2.6492(17) (1.13)
Cu1-Br1	2.3915(15) (1.03)	Cu6-Br1	2.3864(15) (1.03)
Cu3-Br2	2.3459(15) (1.01)	Cu5-Br2	2.3816(15) (1.03)
H1-Ru2	1.83(10) (1.24)	H1-Ru3	1.95(10) (1.33)
H1-Cu2	1.79(10) (1.28)	C1-Ru1	1.894(11) (0.94)
C1-Cu3	2.461(10) (1.27)	C1-Cu4	2.241(10) (1.15)
C2-Ru1	1.892(9) (0.94)	C2-Cu1	2.546(10) (1.31)
C4-Ru2	1.911(11) (0.95)	C4-Cu3	2.501(10) (1.29)
C7-Ru3	1.906(11) (0.95)	C7-Cu1	2.301(10) (1.18)
C13-Ru5	1.882(10) (0.94)	C13-Cu2	2.145(10) (1.10)
C16-Ru6	1.879(11) (0.93)	C16-Cu4	2.359(10) (1.21)
C16-Cu6	2.595(10) (1.33)	C17-Ru6	1.907(10) (0.95)
C17-Cu5	2.619(10) (1.35)	C19-Ru7	1.899(11) (0.94)
C19-Cu6	2.379(10)	C22-Ru8	1.906(10)

	(1.22)		(0.95)
C22-Cu5	2.535(10) (1.30)	C3-Ru1	1.935(10) (0.96)
C3-Ru4	2.340(10) (1.16)	C15-Ru5	1.959(9) (0.97)
C15-Ru8	2.180(10) (1.08)	Ru1-C1-O1	163.0(9)
Ru1-C2-O2	173.8(9)	Ru2-C4-O4	169.6(9)
Ru3-C7-O7	164.0(9)	Ru5-C13-O13	157.7(8)
Ru6-C16-O16	168.6(9)	Ru6-C17-O17	171.5(9)
Ru7-C19-O19	167.1(10)	Ru8-C22-O22	173.1(9)
Ru1-C3-O3	149.5(8)	Ru5-C15-O15	141.1(8)
α_{CO1}	0.183(12) [0.219(12)]	α_{CO2}	0.346(12) [0.382(12)]
α_{CO4}	0.309(13) [0.344(13)]	α_{CO7}	0.207(12) [0.243(12)]
α_{CO13}	0.141(11) [0.176(11)]	α_{CO16}	0.255(13) [0.292(13)]
α_{CO17}	0.373(12) [0.409(12)]	α_{CO19}	0.253(13) [0.289(13)]
α_{CO22}	0.330(12) [0.366(12)]		
α_{CO3}	0.209(11)	α_{CO15}	0.113(10)

Cotton's FSRs are reported in parentheses. Pauling's atomic radii are employed: Cu 1.173 Å, Ru 1.241 Å, Br 1.142 Å, C 0.771 Å, H 0.230 Å.

α corrected for the different atom radii of Cu (1.173) and Ru (1.241) in parentheses (covalent radii from Pauling).

Table S4. Main bond distances (Å) and angles (°) of $[\text{Ru}_5(\text{CO})_{15}(\text{CuMeCN})_2]$ (**5**).

Ru1-Ru2	2.8526(12) (1.15)	Ru1-Ru3	2.9839(12) (1.20)
Ru1-Ru4	2.6894(12) (1.08)	Ru2-Ru3	2.7792(12) (1.12)
Ru2-Ru4	2.9112(13) (1.17)	Ru2-Ru5	2.8226(13) (1.14)
Ru3-Ru4	2.9056(12) (1.17)	Ru3-Ru5	2.9325(12) (1.18)
Ru4-Ru5	2.6944(12) (1.09)	Cu1-Ru1	2.7336(16) (1.13)
Cu1-Ru3	2.5655(16)	Cu1-Ru4	2.7180(15)

	(1.06)		(1.13)
Cu2-Ru3	2.5860(17) (1.07)	Cu2-Ru4	2.6729(16) (1.11)
Cu2-Ru5	2.7358(16) (1.13)	Cu1-Cu2	2.6726(18) (1.11)
Cu1-N1	1.920(9) (0.99)	Cu2-N2	1.925(10) (0.99)
C7-Ru3	1.886(12) (0.94)	C7-Cu1	2.497(11) (1.28)
C9-Ru3	1.894(13) (0.94)	C9-Cu2	2.404(13) (1.24)
C11-Ru4	1.901(12) (0.94)	C11-Cu2	2.610(13) (1.34)
C8-Ru2	2.512(12) (1.29)	C8-Ru3	1.903(13) (0.95)
C10-Ru1	2.393(11) (1.19)	C10-Ru4	1.929(12) (0.96)
Ru3-C7-O7	171.9(10)	Ru3-C9-O9	170.5(11)
Ru4-C11-O11	174.2(10)	Ru3-C8-O8	159.4(10)
Ru4-C10-O10	156.2(10)	α_{CO7}	0.324(14) [0.360(14)]
α_{CO9}	0.269(16) [0.305(16)]	α_{CO11}	0.373(16) [0.409(16)]
α_{CO8}	0.320(15)	α_{CO10}	0.241(13)

Cotton's FSRs are reported in parentheses. Pauling's atomic radii are employed: Cu 1.173 Å, Ru 1.241 Å, Br 1.142 Å, C 0.771 Å, H 0.230 Å.

α corrected for the different atom radii of Cu (1.173) and Ru (1.241) in parentheses (covalent radii from Pauling).

Table S5. Crystal data and experimental details for [NEt₄]₃[HRu₄(CO)₁₂(CuBr)₃] (**2**), [NEt₄]₂[H₂Ru₈(CO)₂₄Cu₇Br₃]·3CH₃COCH₃ (**3**·3CH₃COCH₃), [NEt₄]₃[HRu₈(CO)₂₄Cu₆Br₂]·CH₂Cl₂ (**4**·CH₂Cl₂), and [Ru₅(CO)₁₅(CuMeCN)₂]·CH₂Cl₂ (**5**·CH₂Cl₂).

	[NEt ₄] ₃ [HRu ₄ (CO) ₁₂ (CuBr) ₃]	[NEt ₄] ₂ [H ₂ Ru ₈ (CO) ₂₄ Cu ₇ Br ₃]·3CH ₃ COCH ₃
Formula	C ₃₆ H ₆₁ Br ₃ Cu ₃ N ₃ O ₁₂ Ru ₄	C ₄₉ H ₆₀ Br ₃ Cu ₇ N ₂ O ₂₇ Ru ₈
<i>F</i> _w	1562.50	2602.06
T, K	100(2)	100(2)
λ, Å	0.71073	0.71073
Crystal system	Monoclinic	Monoclinic
Space Group	<i>P</i> 2 ₁ / <i>n</i>	<i>P</i> 2 ₁ / <i>c</i>
<i>a</i> , Å	12.6129(5)	14.292(3)
<i>b</i> , Å	27.7074(10)	23.458(5)
<i>c</i> , Å	14.6211(6)	22.443(5)
α, °	90	90
β, °	96.0210(10)	93.240(9)
γ, °	90	90
Cell Volume, Å ³	5081.5(3)	7512(3)
<i>Z</i>	4	4
<i>D</i> _c , g cm ⁻³	2.042	2.301
μ, mm ⁻¹	4.802	5.159
<i>F</i> (000)	3048	4976
Crystal size, mm	0.20×0.16×0.14	0.22×0.16×0.13
θ limits, °	1.582-26.000	1.648-25.200
Index ranges	-15 ≤ <i>h</i> ≤ 15 -34 ≤ <i>k</i> ≤ 34 -18 ≤ <i>l</i> ≤ 18	-17 ≤ <i>h</i> ≤ 17 -28 ≤ <i>k</i> ≤ 28 -26 ≤ <i>l</i> ≤ 26
Reflections collected	72054	80943
Independent reflections	9987 [R _{int} = 0.0445]	13546 [R _{int} = 0.0868]
Completeness to θ max	100.0%	100.0%
Data / restraints / parameters	9987 / 4 / 569	13546 / 605 / 852
Goodness on fit on F ²	1.213	1.078
R ₁ (<i>I</i> > 2σ(<i>I</i>))	0.0369	0.0929
wR ₂ (all data)	0.0752	0.2460
Largest diff. peak and hole, e Å ⁻³	1.278 / -0.620	3.188 / -2.388

	[NEt₄]₃[HRu₈(CO)₂₄Cu₆Br₂]·CH₂Cl₂	[Ru₅(CO)₁₅(CuMeCN)₂]·CH₂Cl₂
Formula	C ₄₉ H ₆₃ Br ₂ Cl ₂ Cu ₆ N ₃ O ₂₄ Ru ₈	C ₂₀ H ₈ Cl ₂ Cu ₂ N ₂ O ₁₅ Ru ₅
<i>F</i> _w	2498.54	1219.61
T, K	100(2)	100(2)
λ, Å	0.71073	0.71073
Crystal system	Triclinic	Triclinic
Space Group	<i>P</i> $\bar{1}$	<i>P</i> $\bar{1}$
a, Å	12.5448(6)	9.0648(10)
b, Å	13.5388(6)	11.3026(13)
c, Å	22.7631(10)	17.3140(19)
α, °	87.1610(10)	107.816(4)
β, °	85.5450(10)	93.116(4)
γ, °	67.5230(10)	109.764(4)
Cell Volume, Å ³	3560.8(3)	1564.7(3)
Z	2	2
D _c , g cm ⁻³	2.330	2.589
μ, mm ⁻¹	4.662	3.921
F(000)	2400	1148
Crystal size, mm	0.12×0.09×0.07	0.13×0.10×0.06
θ limits, °	1.628-26.407	1.992-25.099
Index ranges	-15 ≤ h ≤ 15 -16 ≤ k ≤ 16 -28 ≤ l ≤ 28	-10 ≤ h ≤ 10 -13 ≤ k ≤ 13 -20 ≤ l ≤ 20
Reflections collected	34169	11164
Independent reflections	14580 [R _{int} = 0.0674]	5534 [R _{int} = 0.0510]
Completeness to θ max	100.0%	99.3%
Data / restraints / parameters	14580 / 48 / 862	5534 / 0 / 417
Goodness on fit on F ²	1.081	1.061
R ₁ (I > 2σ(I))	0.0640	0.0588
wR ₂ (all data)	0.1265	0.1534
Largest diff. peak and hole, e Å ⁻³	1.888 / -1.585	1.946 / -1.326