

Alma Mater Studiorum Università di Bologna
Archivio istituzionale della ricerca

Organocatalytic Strategies for the Synthesis of Axially Chiral Compounds

This is the final peer-reviewed author's accepted manuscript (postprint) of the following publication:

Published Version:

Bencivenni, G. (2015). Organocatalytic Strategies for the Synthesis of Axially Chiral Compounds. SYNLETT, 26(14), 1915-1922 [10.1055/s-0034-1378712].

Availability:

This version is available at: <https://hdl.handle.net/11585/515425> since: 2021-02-28

Published:

DOI: <http://doi.org/10.1055/s-0034-1378712>

Terms of use:

Some rights reserved. The terms and conditions for the reuse of this version of the manuscript are specified in the publishing policy. For all terms of use and more information see the publisher's website.

This item was downloaded from IRIS Università di Bologna (<https://cris.unibo.it/>).
When citing, please refer to the published version.

(Article begins on next page)

This is the final peer-reviewed accepted manuscript of:

Bencivenni, G. "Organocatalytic Strategies for the Synthesis of Axially Chiral Compounds." *Synlett*, vol. 26, no. 14, 2015, pp. 1915-1922.

The final published version is available online at : <http://dx.doi.org/10.1055/s-0034-1378712>

Rights / License:

The terms and conditions for the reuse of this version of the manuscript are specified in the publishing policy. For all terms of use and more information see the publisher's website.

This item was downloaded from IRIS Università di Bologna (<https://cris.unibo.it/>)

When citing, please refer to the published version.

Organocatalytic Strategies for the Synthesis of Axially Chiral Compounds

Giorgio Bencivenni*

Department of Industrial Chemistry "Toso Montanari", School of Science, University of Bologna, viale del Risorgimento 4, 40136, Bologna, Italy.

Tel: +390512096654; Fax: +390512093654.

E-mail: giorgio.bencivenni2@unibo.it.

Received: The date will be inserted once the manuscript is accepted.

Abstract: Atropisomers are compounds in which chirality arises from hindered rotation along carbon-carbon or carbon-heteroatom single bond. Recently organocatalysis appeared to be a rapid, valuable and efficient strategy for their preparation in an enantioselective manner. In this report a general overview of the most important organocatalyzed atroposelective transformations will be given pointing out in particular the specific role played by the catalyst.

1 Dynamic kinetic resolution

2 Desymmetrization of biaryls

3 Direct synthesis of biaryls

4 Asymmetric Friedel-Crafts amination of naphthols

5 *N*-alkylation via phase transfer catalysis

6 Desymmetrization of *N*-arylmalimides

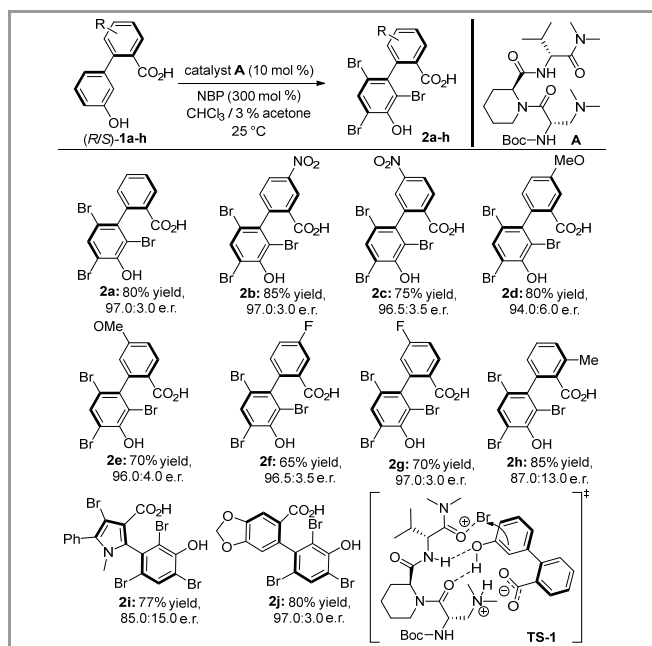
Key words: Atropisomers, Organocatalysis, Enantioselective synthesis, Desymmetrization, Dynamic kinetic resolution.

The enantioselective synthesis of axially chiral compounds is a stimulating research field mainly focused for the preparation of biaryl systems. Large number of methodologies have been successfully employed¹ and among all the Suzuki–Miyaura coupling is certainly the most diffuse.² The enantioselectivity can be generally induced using chiral auxiliary directly bonded to the substrates or using catalytic strategies in which chiral ligands surrounds the active metal core of the catalyst. Recently organocatalysis³ has emerged as one of the most versatile and powerful tool for the realization enantioenriched compounds. The recent successfully applications of organocatalysis to the synthesis of atropisomeric compounds⁴ confirm the potentiality expressed by small organic molecules to design novel and challenging chemical transformations. It is well known the importance that biaryl atropisomers have as ligands, catalysts or biological active compounds,⁵ so it is not strange that most of the organocatalytic synthesis of atropisomers focused on biaryl systems. However, during the last years, the attention of several research groups has been directed also to the enantioselective synthesis of non biaryls atropisomers such as amides, anilides and imides. In this report an overview of the most important organocatalyzed atroposelective transformations is given pointing out in particular the specific role played by the catalyst.

1 Dynamic Kinetic Resolution

The dynamic kinetic resolution (DKR) is one of the most important techniques to transform racemic axially chiral in biaryls into enantiopure atropisomers. For example in compound **1** the low energy barrier to rotation of the carbon-carbon single bond connecting

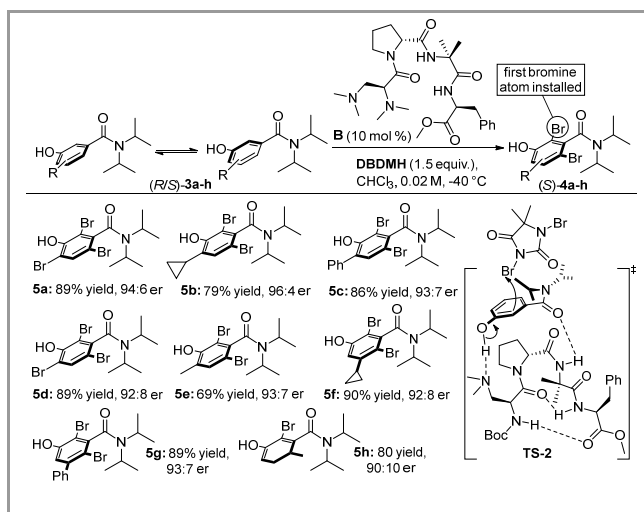
the two aryl units do not allow the isolation of stable atropisomers. By means of an appropriate regioselective reaction the barrier to racemization can be increased to a sufficiently high value so that the resulting compound can be isolated as stable atropisomer. This strategy was successfully applied by Miller and co-workers⁶ using the peptide catalyzed asymmetric bromination of carboxylic acids. Racemic biaryl benzoic acids **1a-h** were converted into non-racemic analogs using a simple bromination strategy employing 3 equivalents of *N*-bromophthalimide (NBP) as brominating agent. It was envisaged that peptidic based organocatalysts could be fundamental for the success of the atroposelective transformation. In particular their β -turn conformation could select the stereogenic axis orientation during the electrophilic aromatic substitution. Various tripeptide catalysts were screened and it was found that catalyst **A**, with an L-pipecolinic acid residue showed an excellent catalytic activity as well as a remarkable ability to control the axial chirality in the products. The reaction was performed using a large number of functional groups, was well tolerated and displayed good reactivity and enantioselectivity, however no mention on the effects of the substituents on the phenol ring was reported. The scope of the reaction is furthermore enriched by the presence of a pyrrole analog **2i** and catechol **2j** as precursors of natural bioactive substructures⁷ (scheme 1).



Scheme 1 Atroposelective bromination of carboxylic acids.

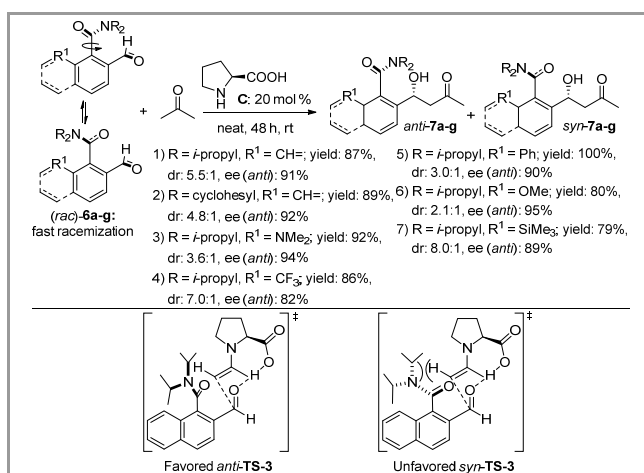
The proposed transition state **TS-1** is governed by several weak interactions involving **A** with the substrate. Thus **A** forms a quaternary ammonium salt with the carboxylic acid and also coordinates with the phenolic aryl unit by H-bonding interactions. The result is a twisted biaryl system which reacts with the bromonium ion activated by the *N,N*-dimethylamide. The increased steric hindrance resulting from the first bromination, probably occurring at the 2' position, enhanced the barrier to rotation avoiding a rapid racemization of the product. Further bromination at the 6' position definitely freezes the rotation leading to (*R*)-atropisomer.

Miller and coworkers⁸ confirmed the usefulness of a peptidic framework as catalyst for the dynamic kinetic resolution of racemic benzamides **3**. Various *N,N*-diisopropylbenzamides underwent atroposelective bromination in high yields and enantioselectivity with dibromodimethylhydantoin (DBDMH) as the brominating agent. In this transformation catalyst **B** based on a D-prolinic core was employed. Interestingly a different bromination pathway was observed in the presence and in the absence of catalyst that installed the first bromine atom in the most sterically demanding ortho position generating the stable chiral axis at this stage of the reaction. Based on the X-ray structure **TS-2** in which the deprotonation of the phenol by the NMe₂ group of **B** and an intermolecular hydrogen bond with the diisopropyl amide promote the stereoselective bromination was proposed (scheme 2).



Scheme 2 Atroposelective bromination of benzamides.

In 2-formyl-*N,N*-diisopropyl-1-naphthamide **6** the different orientations of the carbonyl group of the amide with respect to the plane entailed by the naphthyl moiety, reveals the existence of two possible atropisomers. However the orthogonal orientation of the freely rotating trigonal aldehyde substituent and the aryl allows the rapid interconversion of the two atropisomers by fast rotation along the single bond connecting the amidic carbonyl group and the aryl unit. The result is that **6a-g** exist as racemic compounds (scheme 3). A route to synthesize stable atropisomers is to increase the energy barrier to rotation by transforming the aldehyde from a trigonal-planar to a spherical-tetrahedral substituent.⁹ Walsh and co-workers¹⁰ exploited this concept and developed the first DKR of amides by means of an L-proline **C** catalyzed aldol reaction of acetone on the aldehydic substituent.



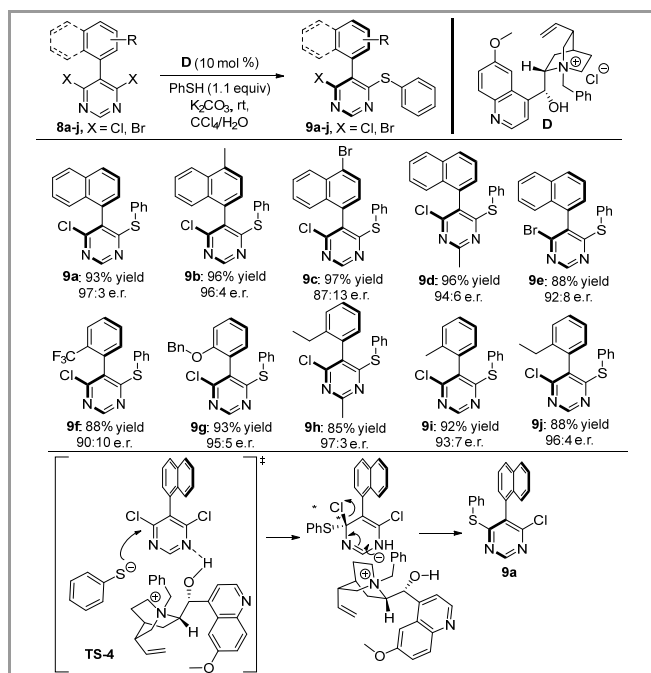
Scheme 3 Atroposelective DKR catalyzed by L-proline.

The reaction proceeded with high enantiocontrol but with moderate *anti*-diastereoselectivity and can be easily applied also to substituted benzamides with similar results. Several experiments revealed that **C** gave high enantiocontrol during the aldol reaction but

it was not so efficient to control of the stereogenic axis. The favored *anti* product is probably the result of a **TS-3** where the interactions of the amide substituents with the axial hydrogen in the Zimmerman-Traxler transition state model¹¹ are **absent**.

2 Desymmetrization of biaryls

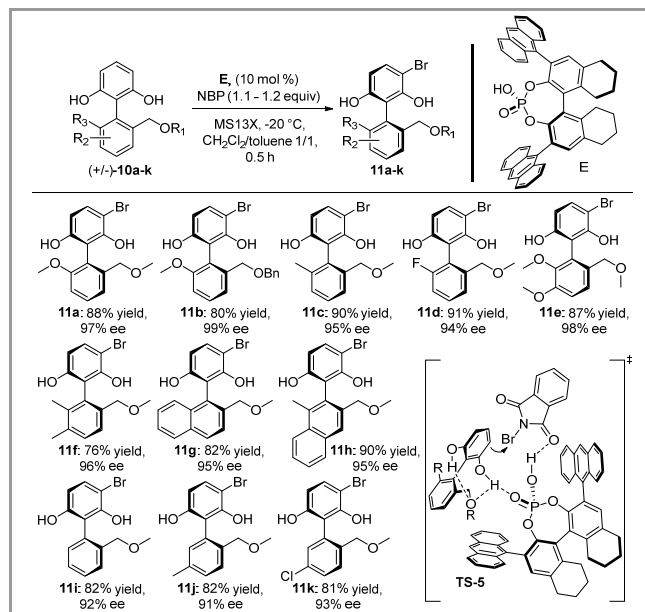
The desymmetrization is a largely investigated strategy for the preparation of chiral biaryls. An interesting example was reported by the group of Smith¹² using phase transfer catalysis (PTC).¹³



Scheme 4 Atroposelective desymmetrization via NAS.

Under the control of commercially available *N*-benzylquininium chloride **D** one enantiotopic chlorine atom of electron poor dichloropyrimidines **8a-j** was selectively displaced by thiophenol. The nucleophilic aromatic substitution (NAS)¹⁴ was highly efficient and gave high yields and atroposelectivities with a large series of substituents. Using an excess of thiol the atroposelectivity was increased because of a desymmetrization/kinetic resolution that transformed the minor enantiomer to an achiral species after displacement of the second chlorine atom. The chiral ion pair represented by the catalyst cation and the thiophenol anion (**TS-4**) promoted the formation of a Meisenheimer complex¹⁴ in which a chiral axis and a stereogenic center are present.

Akiyama¹⁵ developed an interesting example of desymmetrization/dynamic kinetic resolution (DKR) performing an electrophilic aromatic bromination reaction of *ad-hoc* designed racemic biaryls catalyzed by chiral phosphoric acid **E**.

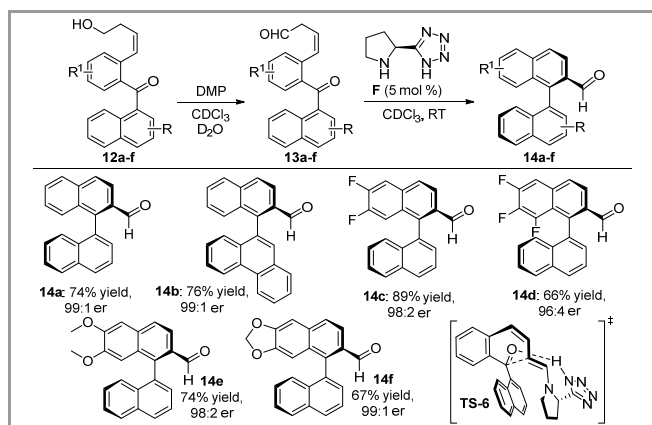


Scheme 5 Desymmetrization catalyzed by chiral phosphoric acid.

The two hydroxy groups could establish a hydrogen bond network made of two concomitant interactions with *N*-bromophthalimide (NBP) and the catalyst and with the etheral substituent of the other aromatic ring. While the first interaction allows the reagents to react under the stereochemical control of **E**, the second gives the required rigidity to the complex during the generation of the chiral axis (**TS-5**). Using an excess of NBP the reaction **proceeded** in a desymmetrization/kinetic resolution sequence leading to chiral biaryls in excellent enantioselectivity. The postulated hydrogen bond network was supported by computational calculations that highlighted the role of both hydroxy and CH₂OMe groups (scheme 5).

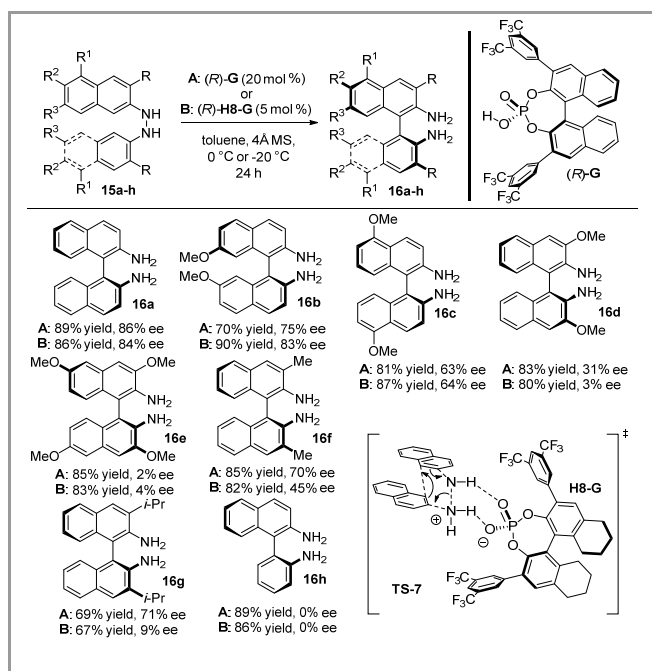
3 Direct synthesis of biaryls

Sparr¹⁶ developed a new atroposelective synthesis of 1,1'-binaphthalene-2-carbaldehydes **14a-f** using an intramolecular aldol condensation of ketoaldehyde of type **13** catalyzed by pyrrolidine-tetrazole **F**. Despite the instability of the starting materials, prepared *in situ* from alcohols **12a-f**, the reaction gave binaphthyles in excellent yields and atroposelectivity with 5 mol% of **F**. The geometry of the substrate places the enamine in a suitable position to react with the ketone. The plausible coordination of the ketone by the NH group of **F** twists the carbonyl and the naphthyl groups out of the plane (**TS-6**). In this way during the dehydration/aromatization path the stereochemical information is transferred from the catalyst to the chiral axis (scheme 6).



Scheme 6 Atroposelective intramolecular aldol condensation.

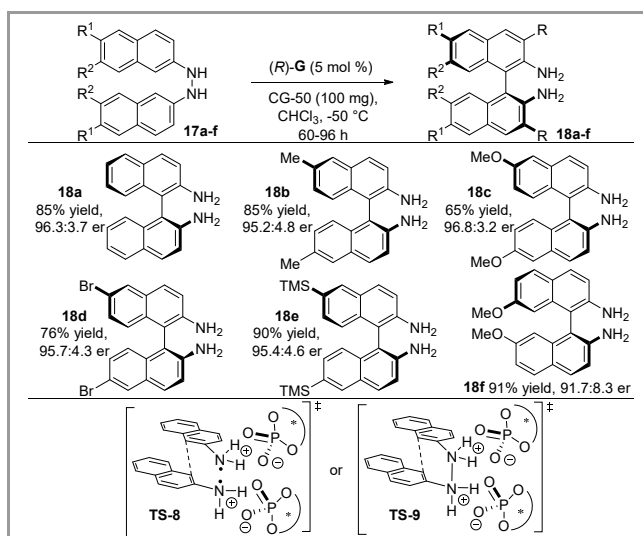
In 2013, Kürti¹⁷ presented the first organocatalytic atroposelective coupling reaction of *N,N'*-biaryl hydrazines 15a-h for the synthesis of BINAM¹⁸ derivatives 16.

Scheme 7 [3,3]-Sigmatropic rearrangement of *N,N'*-biaryl hydrazines catalyzed by chiral phosphoric acid.

The strategy of this synthesis exploited the ability of phosphoric acid **G** to promote the [3,3]-sigmatropic rearrangement giving enantioenriched 1,1'-binaphthyls as single regioisomers. The reaction could be catalyzed also by **H8-G** that furnished similar conversions but different enantiocontrol when 3,3'-disubstituted hydrazines were used (scheme 7). A DFT calculated model for the TS-7 is presented. The stereoselectivity is the result of steric and electronic effects. It was observed a complete proton transfer from the catalyst to the nitrogen of the hydrazine with the phosphate counterion that locks the protonated intermediate via H-bonding and π - π aryl interactions

into a sole conformation until deprotonation, assisted by the same phosphate occurs.

List *et al*¹⁹ presented the same reaction using the same catalyst albeit with some differences that lead to an increment of yields and atroposelectivity (scheme 8). Loading of (*R*)-**G** was reduced to 5 mol%, chloroform was used as the solvent and the temperature was reduced to -50 °C. Moreover to ensure complete conversion acidic CG-50 resin was employed. However the scope is limited only to 6,6'- and 7,7'-dinaphthylhydrazines. Based on the study on the nonlinear effects,²⁰ the authors presented experimental evidences supporting a dicationic or radical cation rather than a monocationic pathway (TS-8 or TS-9).



Scheme 8 Synthesis of BINAM derivatives via dication/ diradical intermediate.

4 Asymmetric Friedel-Crafts amination of naphthols

After the seminal report by Curran,²¹ atropisomerism in which axial chirality originates from restricted rotation along carbon-nitrogen single bond became an important topic of asymmetric synthesis. Jørgensen and Bella²² developed the first organocatalytic atroposelective synthesis of anilides via Friedel-Crafts (FC) amination of 2-naphthols using atropisomeric 5'-aminated dihydrocupreidine-DHQD **H** and dihydrocupreine-DHQ **I** as catalyst (Figure 1).

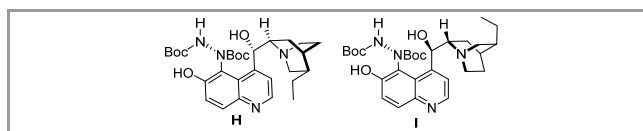
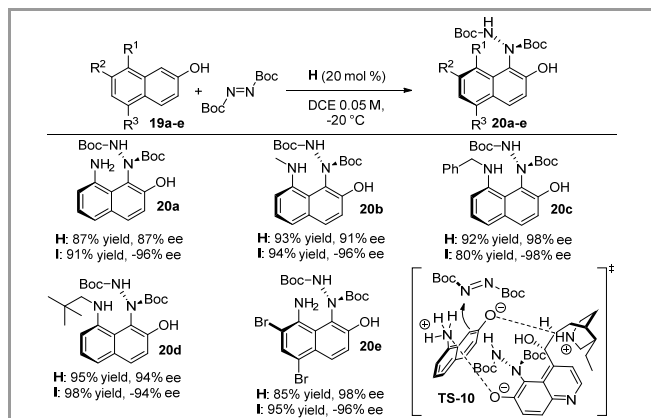


Figure 1 Catalysts for the Friedel-Crafts atroposelective amination.

The reaction was highly enantioselective and various naphthylanilides **20a-e** were prepared. The presence of large substituents in the 8-position of naphthols and at both carbonyl groups of the azodicarboxylate derivative, avoided racemization of the chiral axis

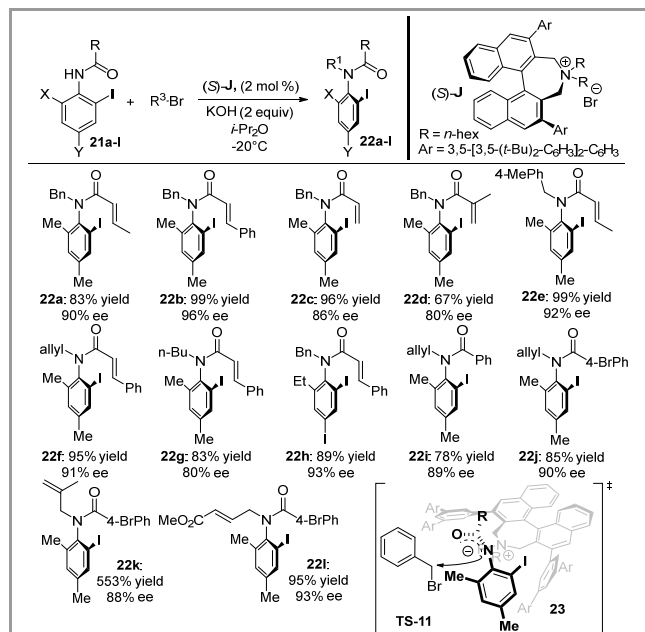
(scheme 9). The reaction mechanism is expected to work through deprotonation of the naphtholic hydroxy group by the quinuclidine ring of **H** followed by the FC amination. However because various substituents, different from amino group at the 8-position of naphthols gave racemic mixtures, a double zwitterionic species closely associated in a two-point contact ion pair, is believed to be the intermediate that undergoes the FC amination (**TS-10**).^{22b}



Scheme 9 Atroposelective Friedel-Crafts amination of naphthols.

5 N-alkylation via phase transfer catalysis

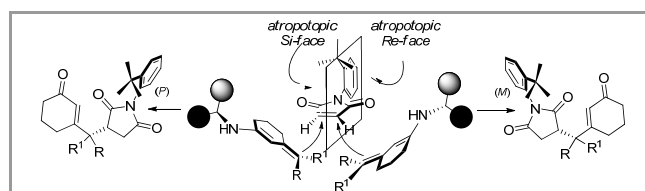
Maruoka²³ realized the first synthesis of axially chiral *o*-iodoanilides using PTC. A chiral binaphthyl-modified ammonium salt (**S**)-**J** promoted the *N*-alkylation reaction of substituted *o*-iodoanilides **21a-l** in excellent enantioselectivity and yields. X-ray analysis on the chiral ammonium anilides **23** elucidates that the alkylation takes place from the opposite side shielded by the catalyst cation (**TS-11**) which recognizes the different steric hindrance between iodide and methyl group in salt **23** giving rise to the observed axial chirality (scheme 10).



Scheme 10 Selected examples for the alkylation of *o*-iodoanilides.

6 Desymmetrization of *N*-arylmaleimides

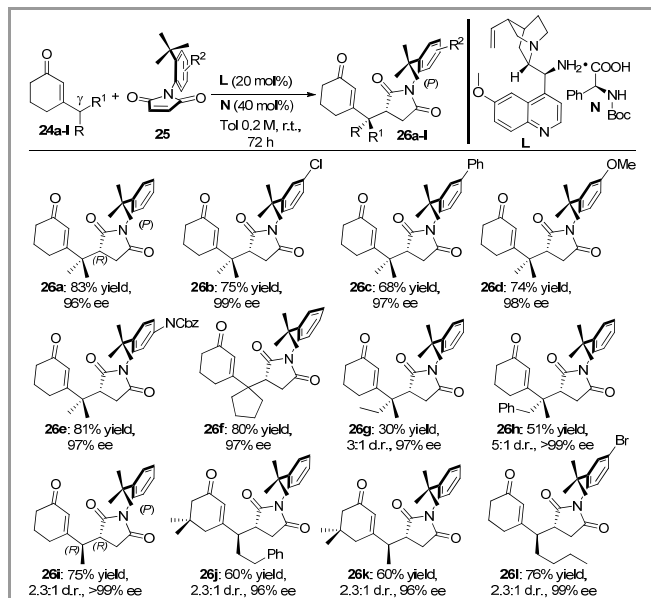
Inspired by the report of Curran on the addition of *tert*-butyl radicals to *N*-arylmaleimides,²¹ we realized the atroposelective desymmetrization of *N*-(2-*tert*-butylphenyl)maleimides via vinylogous Michael addition of 3-substituted cyclohexenones.²⁴ This class of compound has a hindered rotation of the C_{Ar}-N single bond. This implies the existence of a plane of symmetry which can be desymmetrized through nucleophilic addition at one of the two carbon atoms of the double bond, with the consequent generation of a stereogenic axis in the resulting succinimide. The ability of a catalyst to distinguish between the two atropotopic faces of the maleimide and direct the reaction to only one of the two carbon atoms that are at the same time prochiral for central and axial chirality is fundamental (scheme 11).



Scheme 11 Desymmetrization of maleimides via Michael addition.

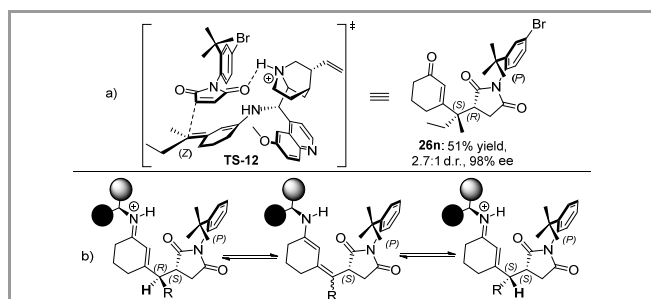
9-Amino(9-deoxy)*epi*-quinine **L** in combination with *N*-Boc-L-Phenylglycine **N** promoted the vinylogous reaction of various substituted cyclohexenones and maleimides with complete remote control over all the elements of chirality (scheme 12). After several experiments and with the absolute configuration in hand, a mechanism based on a hydrogen bonding activation of the carbonyl group lying in the *Re* face of the plane of symmetry of the maleimide was proposed (**TS-12**). In this way catalyst **L** directed the

approach of enone to the bottom face of the double bond with the control of both the chiral axis and the chiral centers (scheme 13a).



Scheme 12 Selected examples for the desymmetrization of maleimides. Only major diastereoisomers are reported.

We demonstrated that an equilibrium of epimerization at the exocyclic chiral center accounted for the moderate diastereoselection when γ -monosubstituted enones were employed (scheme 13b).



Scheme 13. a) Proposed TS; b) Epimerization parasitic reaction.

Using a similar strategy we recently studied the formal Diels-Alder reaction of α,β -unsaturated ketones 27a-j with *N*-(2-*t*-butylphenyl)maleimides.²⁵

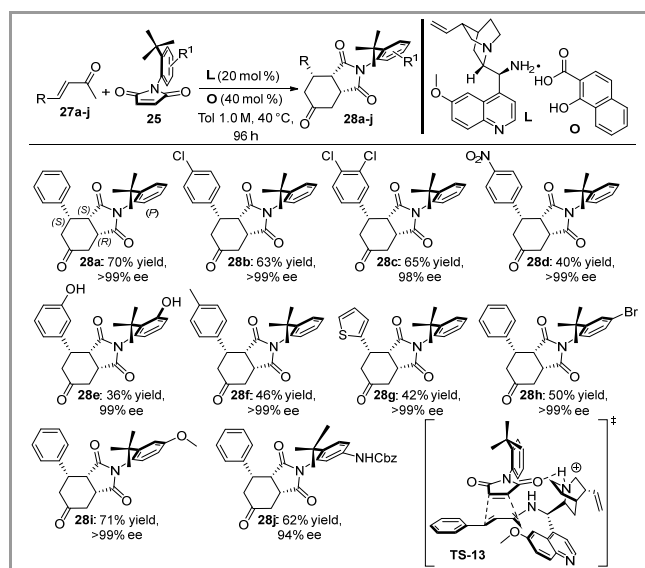
Acknowledgment

University of Bologna is gratefully acknowledge for the financial support. I would like to express my gratitude to Prof. P. Melchiorre and Prof. P. Righi for precious teachings and support during these years.

References

- (1) For detailed reviews see: (a) Baudoin, O.; *Eur. J. Org. Chem.* **2005**, 4223. (b) Bringmann, G.; Mortimer, A. J. P.; Keller, P. A.; Gresser, M. J.; Garner, J.; Breuning, M.; *Angew. Chem. Int. Ed.* **2005**, 44, 5384. (c) Bringmann,

Catalyst **L** in combination with 1-hydroxy-2-naphthoic acid **O** catalyzed the cycloaddition process of a large series of α,β -unsaturated enamines giving rise only to *endo*-diastereoisomers with complete control of axial and central chirality (scheme 14). The approach of the enamines to the double bond of the maleimide gives rise to an opposite stereochemical outcome at the chiral centers from the non-atroposelective examples reported by Melchiorre.²⁶ Probably the reaction is strongly influenced by the presence of the *tert*-butyl group which forces the chiral enamine to add to a different side of the maleimide leading to a site specific reaction pathway (TS-13).



Scheme 14 Selected examples for the atroposelective formal Diels-Alder reaction of α,β -unsaturated ketones with *N*-aryl maleimides.

In summary several methods for the preparation of atropisomers have been developed using diverse organocatalytic strategies. In addition to the well established methodologies mainly focused on the use of metal or chiral auxiliaries, organocatalysis offers different but powerful tools for the enantioselective approach to various classes of atropisomers.

- G.; Menche, D.; *Acc. Chem. Res.*, **2001**, 34, 615. (d) Lipshutz, B. H.; Kayser, F.; Liu, Z.-P.; *Angew. Chem. Int. Ed.* **1994**, 33, 1842. (e) Spring, D. R.; Krishnan, S.; Schreiber, S. L.; *J. Am. Chem. Soc.* **2000**, 122, 5656. (f) Surry, D. S.; Su, X.; Fox, D. J.; Franckevicius, V.; Macdonald, S. J. F.; Spring, D. R.; *Angew. Chem. Int. Ed.* **2005**, 117, 1904. (g) Meyers, A. I.; Lutowski, K. A.; *J. Am. Chem. Soc.* **1982**, 104, 879. (h) Meyers, A. I.; Himmelsbach, R. J.; *J. Am. Chem. Soc.* **1985**, 107, 682.
- (2) (a) Miyaura, N.; Suzuki, A.; *Chem. Rev.* **1995**, 95, 2457. (b) Kamikawa, K.; Uemura, M.; *Synlett* **2000**, 938. (c) Lipshutz, B. H.; Keith, J. M.; *Angew. Chem. Int. Ed.* **1999**, 38, 3530. (d) Baudoin, O.; Décor, A.; Cesario, M.; Guéritte, F.; *Synlett* **2003**, 2009. (e) Broutin, P.-E.;

- Colobert, F.; *Org. Lett.* **2003**, *5*, 3281. (f) Nicolaou, K. C.; Li, H.; Boddy, C. N. C.; Ramanjulu, J. M.; Yue, T.-Y.; Natarajan, S.; Chu, X.-J.; Bräse, S.; Rübsam, F.; *Chem. Eur. J.* **1999**, *5*, 2584. (g) Cammidge, A. N.; Crépy, K. V. L.; *Chem. Commun.* **2000**, 1723. (h) Yin, J.; Buchwald, S. L.; *J. Am. Chem. Soc.* **2000**, *122*, 12051.
- (3) *Comprehensive Enantioselective Organocatalysis*; Dalko, P. I.; Ed. Wiley-VCH: Weinheim, **2013**.
- (4) (a) Christie, G. H.; Kenner, J. H.; *J. Chem. Soc.* **1922**, *121*, 614. (c) Clayden, J.; Moran, W. J.; Edwards, P. J.; LaPlante, S. R.; *Angew. Chem. Int. Ed.* **2009**, *48*, 6398. (d) Kozłowski, M. C.; Morgan, B. J.; Linton, E. C.; *Chem. Soc. Rev.* **2009**, *38*, 3193. (e) Clayden, J.; Fletcher, S. P.; McDouall, J. J. W.; Rowbottom, S. J. M.; *J. Am. Chem. Soc.* **2009**, *131*, 5331. (f) Tetrahedron Symposium: Atropisomerism, Jonathan Clayden, Ed. *Tetrahedron* **2004**, *60*, 4325. (g) Cozzi, P. G.; Emer, E.; Gualandi, A.; *Angew. Chem. Int. Ed.* **2011**, *50*, 3847.
- (5) (a) Miyashita, A.; Yasuda, A.; Takaya, H.; Toriumi, K.; Ito, T.; Souchi, T.; Noyori, R.; *J. Am. Chem. Soc.* **1980**, *102*, 7932. (b) Rosini, C.; Franzini, L.; Raffaelli, A.; Salvadori, P.; *Synthesis*, **1992**, *6*, 503. (c) Clayden, J.; *Angew. Chem. Int. Ed.* **1997**, *36*, 949. (d) Clayden, J.; Johnson, P.; Pink, J. H.; Helliwell, M.; *J. Org. Chem.* **2000**, *65*, 7033. (e) LaPlante, S. R.; Edwards, P. J.; Fader, L. D.; Jakalian, A.; Hucke, O.; *ChemMedChem* **2011**, *6*, 505. (f) Bringmann, G.; Gulder, T.; Gulder, T. A. M.; Breuning, M.; *Chem. Rev.* **2011**, *111*, 563. (g) Bhat, V.; Wang, S.; Stoltz, B. M.; Virgil, S. C.; *J. Am. Chem. Soc.* **2013**, *135*, 16829. (h) Perron, Q.; Alexakis, A.; *Adv. Synth. Catal.* **2010**, *352*, 2611. (i) Usanov, D. L.; Yamamoto, H.; *Angew. Chem. Int. Ed.* **2010**, *49*, 8169. (j) Aikawa, K.; Mikami, K.; *Angew. Chem. Int. Ed.* **2003**, *42*, 5458. (k) Huang, H.; Okuno, T.; Tsuda, K.; Yoshimura, M.; Kitamura, M.; *J. Am. Chem. Soc.* **2006**, *128*, 8716. (l) Kumarasamy, E.; Raghunathan, R.; Jockusch, S.; Ugrinov, A.; Sivaguru, J.; *J. Am. Chem. Soc.* **2014**, *136*, 8729. (m) Rios, R.; Jimeno, C.; Carroll, P. J.; Walsh, P. J.; *J. Am. Chem. Soc.* **2002**, *124*, 10272.
- (6) Gustafson, J. L.; Lim, D.; Miller, S. J.; *Science* **2010**, *328*, 1251.
- (7) (a) Martin, R.; Jäger, A.; Böhl, M.; Richter, S.; Fedorov, R.; Manstein, D. J.; Gutzeit, H. O.; Knölker, H.-J.; *Angew. Chem. Int. Ed.* **2009**, *48*, 8042. (b) Kende, A. S.; Liebeskind, L. S.; *J. Am. Chem. Soc.* **1976**, *98*, 267.
- (8) Barrett, K. T.; Miller, S. J.; *J. Am. Chem. Soc.* **2013**, *135*, 2963.
- (9) Ahmed, A.; Bragg, R. A.; Clayden, J.; Lai, L. W.; McCarthy, C.; Pink, J. H.; Westlund, N.; Yasin, S. A. *Tetrahedron* **1998**, *54*, 13277.
- (10) Chan, V.; Kim, J. G.; Jimeno, C.; Carroll, P. J.; Walsh, P. J.; *Org. Lett.* **2004**, *6*, 2051.
- (11) Zimmerman, H. E.; Traxler, M. D. *J. Am. Chem. Soc.* **1957**, *79*, 1920.
- (12) Armstrong, R. J.; Smith, M. D.; *Angew. Chem. Int. Ed.* **2014**, *53*, 12822.
- (13) (a) Ooi, T.; Maruoka, K.; *Angew. Chem. Int. Ed.* **2007**, *46*, 4222. (b) Shirakawa S.; Maruoka K.; *Angew. Chem. Int. Ed.* **2013**, *52*, 4312.
- (14) Burnett, F.; Zahler, R. E.; *Chem. Rev.*, **1951**, *49*, 273.
- (15) Mori, K.; Ichikawa, Y.; Kobayashi, M.; Shibata, Y.; Yamanaka, M.; Akiyama, T.; *J. Am. Chem. Soc.* **2013**, *135*, 3964.
- (16) Link, A.; Sparr, C.; *Angew. Chem. Int. Ed.* **2014**, *53*, 5458.
- (17) Li, G.-Q.; Gao, H.; Keene, C.; Devonas, M.; Ess, D. H.; Kürti, L.; *J. Am. Chem. Soc.* **2013**, *135*, 7414.
- (18) Quideau, S.; Lyvinec, G.; Marguerit, M.; Bathany, K.; Ozanne-Beaudenon, A.; Buffeteau, T.; Cavagnat, D.; Chénéde, A.; *Angew. Chem. Int. Ed.*, **2009**, *48*, 4605.
- (19) De, C. K.; Pesciaoli, F.; List, B.; *Angew. Chem. Int. Ed.* **2013**, *52*, 9293.
- (20) Satyanarayana, T.; Abraham, S.; Kagan, H. B. *Angew. Chem. Int. Ed.* **2009**, *48*, 456.
- (21) Curran, D. P.; Qi, H.; Geib, S. J.; DeMello, N. C.; *J. Am. Chem. Soc.* **1994**, *116*, 3131.
- (22) (a) Brandes, S.; Bella, M.; Kjærsgaard, A.; Jørgensen, K. A.; *Angew. Chem. Int. Ed.* **2006**, *45*, 1147. (b) Brandes, S.; Niess, B.; Bella, M.; Prieto, A.; Overgaard, J.; Jørgensen, K. A.; *Chem. Eur. J.* **2006**, *12*, 6039.
- (23) Shirakawa, S.; Liu, K.; Maruoka, K.; *J. Am. Chem. Soc.* **2012**, *134*, 916.
- (24) Di Iorio, N.; Righi, P.; Mazzanti, A.; Mancinelli, M.; Ciogli, A.; Bencivenni, G.; *J. Am. Chem. Soc.* **2014**, *136*, 10250.
- (25) Eudier, F.; Righi, P.; Mazzanti, A.; Ciogli, A.; Bencivenni, G.; *Org. Lett.* **2015**, *17*, 1728.
- (26) Wu, L.; Bencivenni, G.; Mancinelli, M.; Mazzanti, A.; Bartoli, G.; Melchiorre, P.; *Angew. Chem., Int. Ed.* **2009**, *48*, 7196.

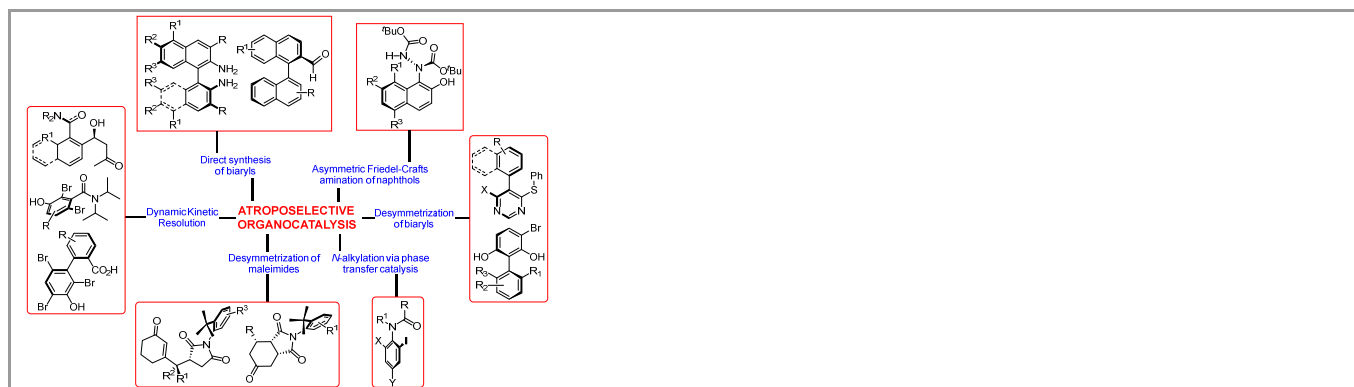
Giorgio Bencivenni was born in 1978. He graduated in 2003 in Industrial Chemistry at the University of Bologna where obtained his PhD degree under the supervision of Prof. D. Nanni in 2008. During his doctoral studies he was a visiting student in the group of Prof. John C. Walton at the University of St. Andrews. He joined the group of Prof. G. Bartoli at the University of Bologna working as a postdoctoral associate from 2008 to 2011. From 2012 to present he works as postdoctoral associate in the group of Prof. Paolo Righi at the Industrial Chemistry department of the University of Bologna.



Please place the biographical sketch here, this will be placed on the first (Synfacts) or second page (Accounts) of the article at the galley proof stage. You may also place the photograph(s) of the authors here, alternatively you can send them by regular mail to the editorial office.

Please place the graphical abstract and short title of the article here. The short title will be used as a running header.

Short title: Organocatalytic Synthesis of Atropisomers



Manuscript submission checklist

- Statement of significance of work.
- Full mailing address, telephone, and fax numbers, and email address of the corresponding author.
- Graphical abstract.
- 5 key words.
- Original Word file.
- Original graphics files.
- Biographical sketches of the authors.
- Photographs of the authors.

Proceed to submit your article via our online submission system at <http://mc.manuscriptcentral.com/synlett>. When prompted to "Add an Editor", please select the Editor who invited you to submit this Account or Synfacts article.